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# Supporting Information for Facile Synthesis of a Light Switchable Polymers with Diazo- cine Units in the Main Chain

- Experimental Data -

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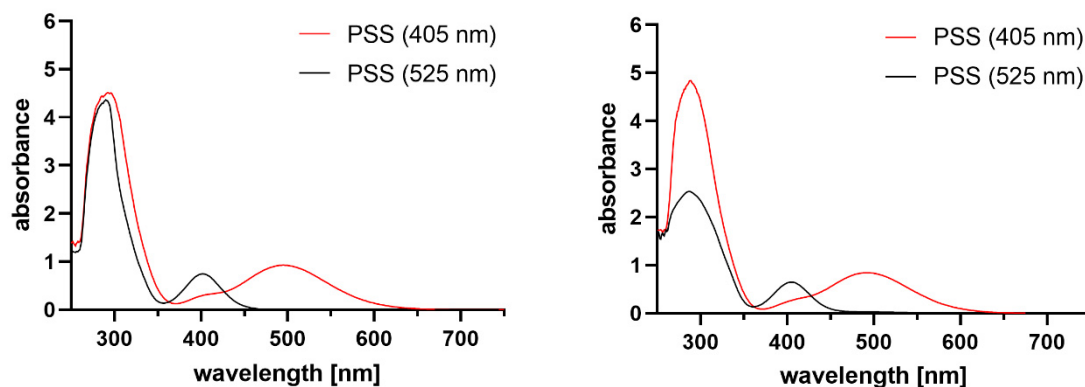
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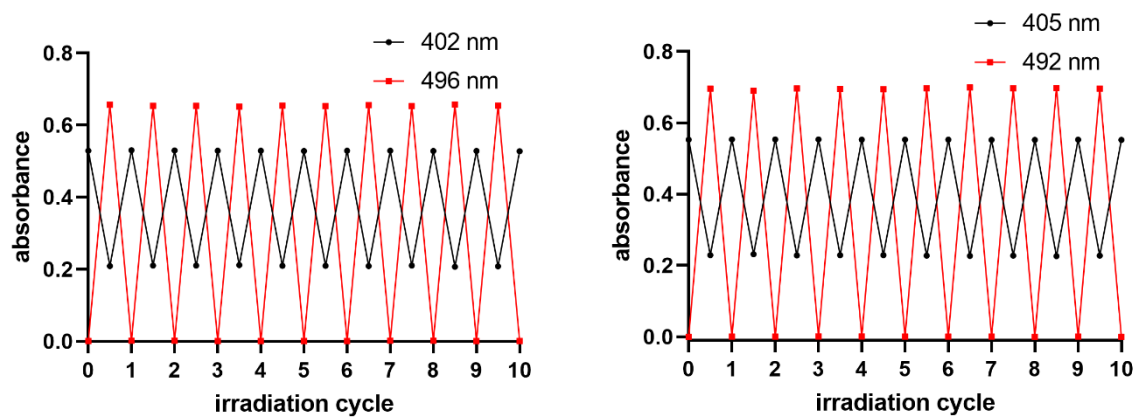
The use of abbreviations follows the conventions from the ACS Style guide [1]:

1. *The ACS Style Guide: Effective Communication of Scientific Information*; Coghill, A.M., Garson, L.R., American Chemical Society, Eds.; 3rd ed.; American Chemical Society ; Oxford University Press: Washington, DC : Oxford ; New York, 2006; ISBN 9780841239999 9780841274006.

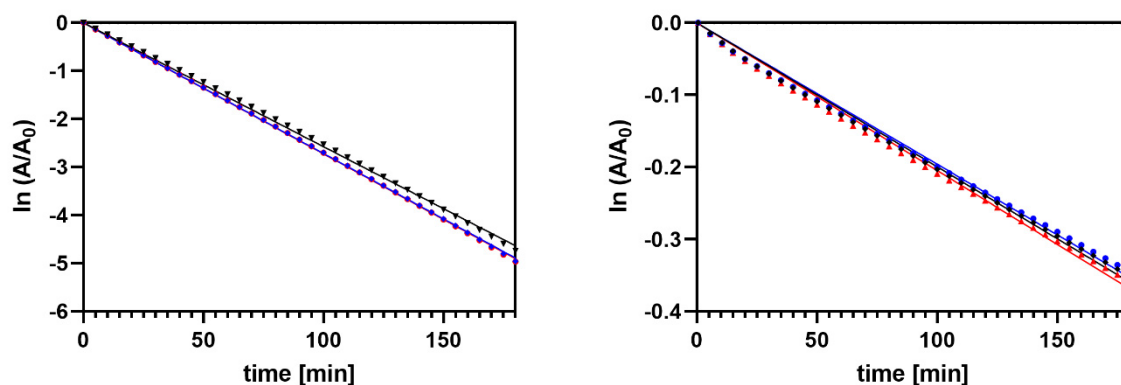
### UV-Vis Spectra of the Products



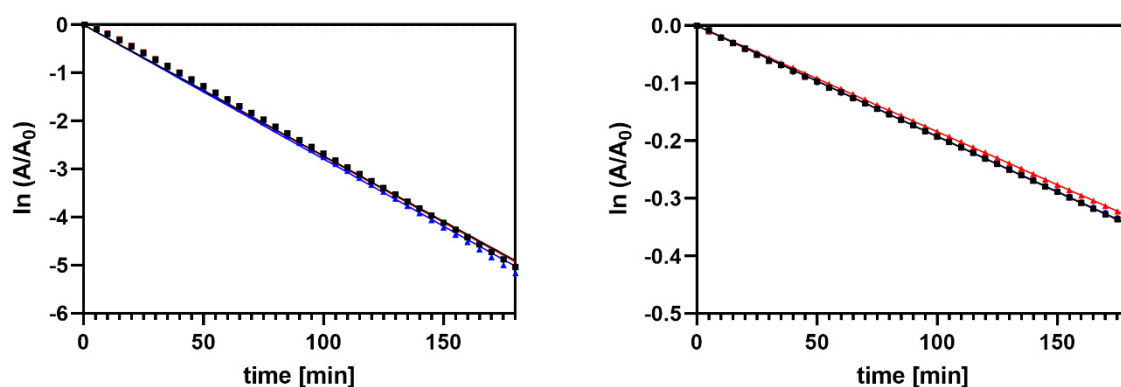
**Figure S1.** UV-vis spectra of compounds **M1** (left) and **M2** (right) after light irradiation at 405 nm (red) and 525 nm wavelength (black) at a concentration of 1 mM in THF.



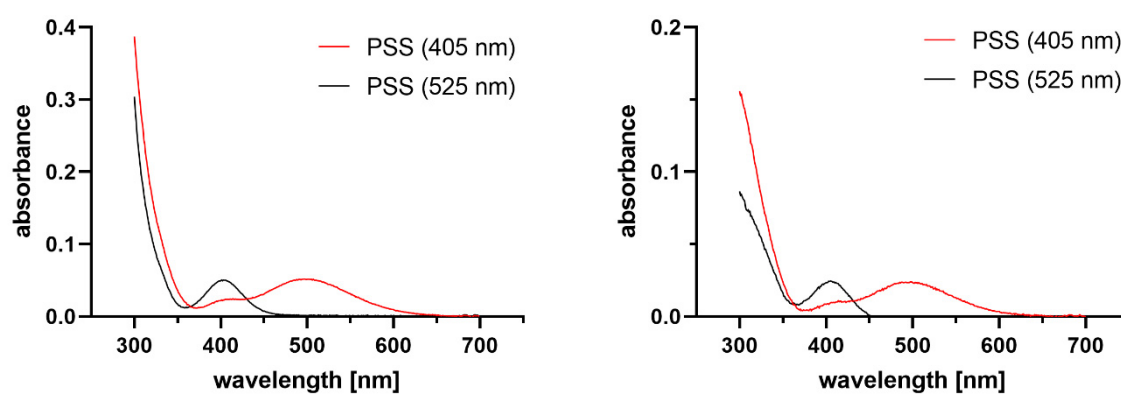
**Figure S2.** Cyclic UV-vis measurements of polymers **P1** (left) and **P2** (right) after light irradiation at 405 nm and 525 nm wavelength monitoring the absorption at  $\lambda_{\max}(Z)$  (black) and  $\lambda_{\max}(E)$  (red) at a concentration of 0.5 mg/mL in THF.



**Figure S3.** First-order thermal relaxation kinetics of **M1** (left) and **M2** (right) from PSS (405 nm) wavelength at  $\lambda_{\max}(E)$  at a concentration of 3 mg/mL in THF.

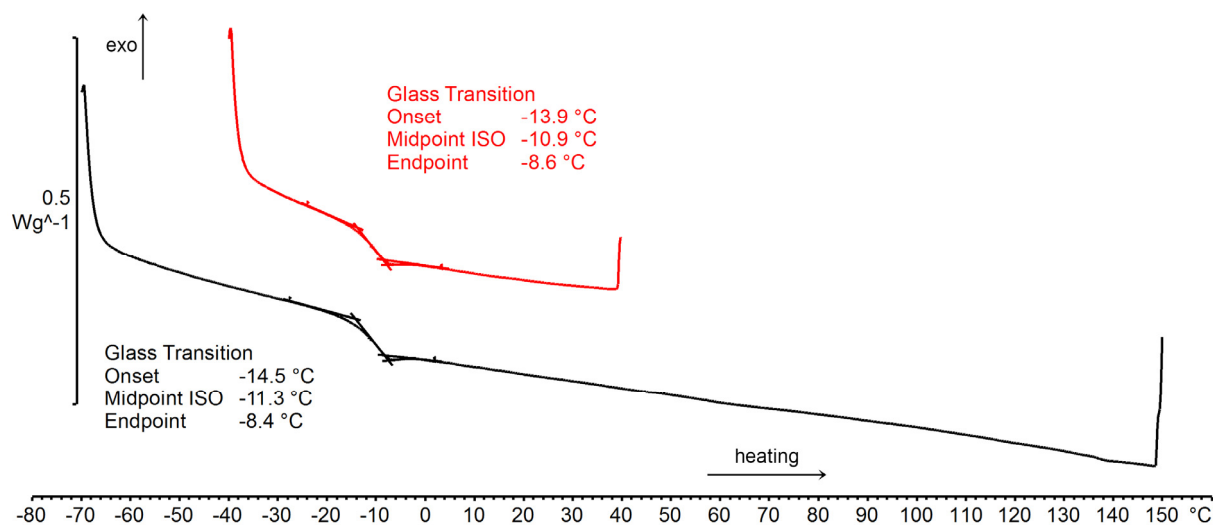


**Figure S4.** First-order thermal relaxation kinetics of **P1** (left) and **P2** (right) from PSS (405 nm) wavelength at  $\lambda_{\max}(E)$  at a concentration of 3 mg/mL in THF.

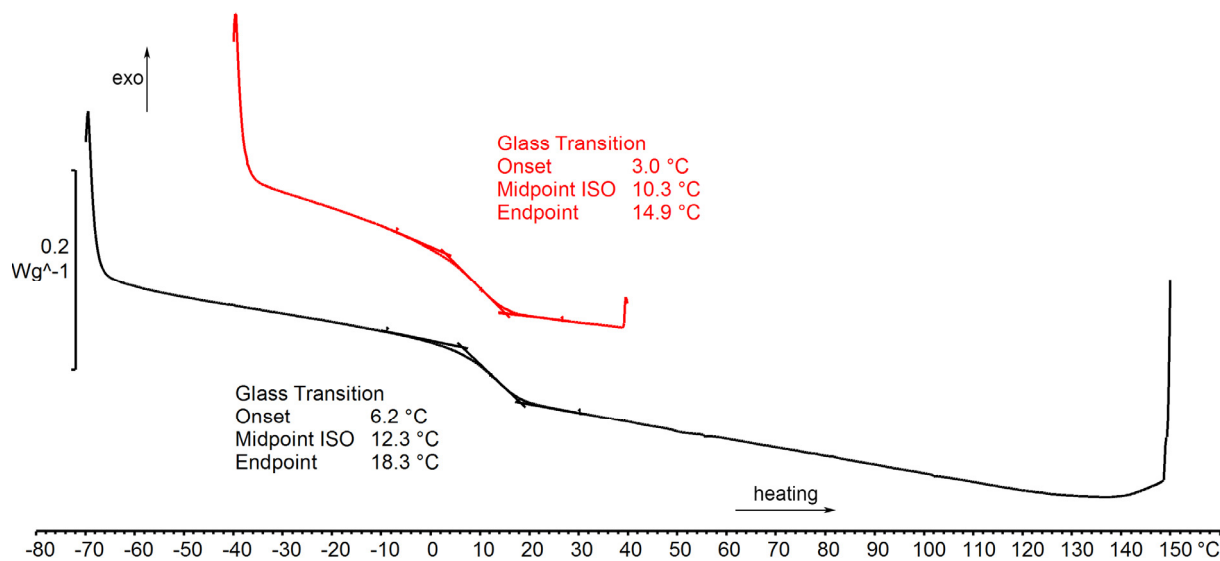


**Figure S5.** UV-vis spectra of polymers **P1** (left) and **P2** (right) after light irradiation at 405 (red) and 525 nm wavelength (black) as spin-coated thin films.

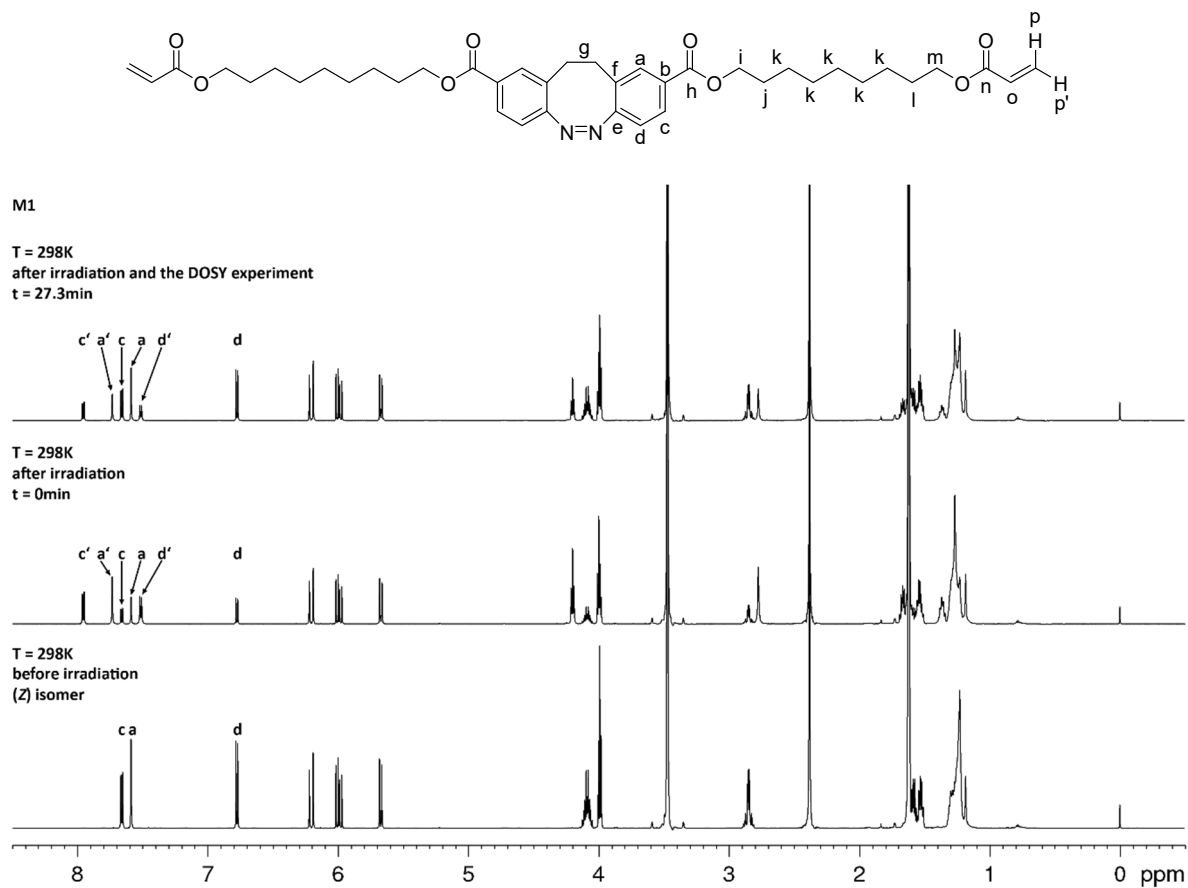
## DSC Plots of Polymers P1 and P2



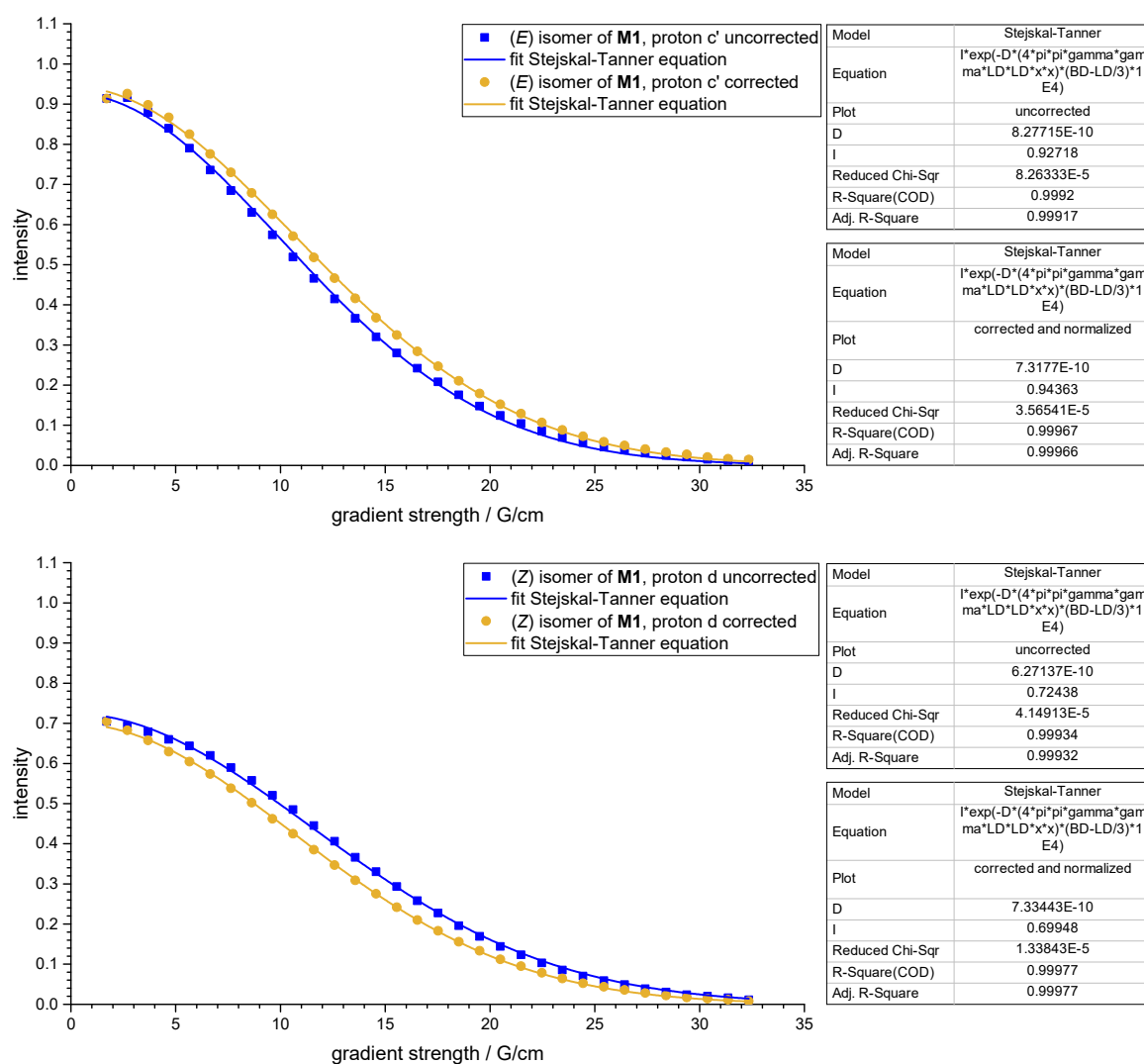
**Figure S6.** DSC plots of polymer **P1** at PSS (525 nm) (black) and PSS (405 nm) (red) indicating the glass transition temperature  $T_g$ . The DSC measurements were cycled between -70 to 150 ° and -40 to 40 °C, respectively.



**Figure S7.** DSC plots of polymer **P2** at PSS (525 nm) (black) and PSS (405 nm) (red) indicating the glass transition temperature  $T_g$ . The DSC measurements were cycled between -70 to 150 °C and -40 to 40 °C, respectively.

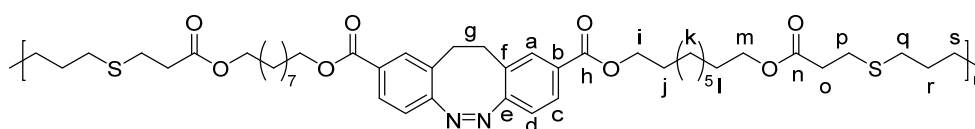
$^1\text{H}$  DOSY NMR Correction Factors and Fitting GraphsBis(9-(acryloyloxy)nonyl) (Z)-11,12-dihydrodibenzo[*c,g*][1,2]diazocine-2,9-dicarboxylate (**M1**)

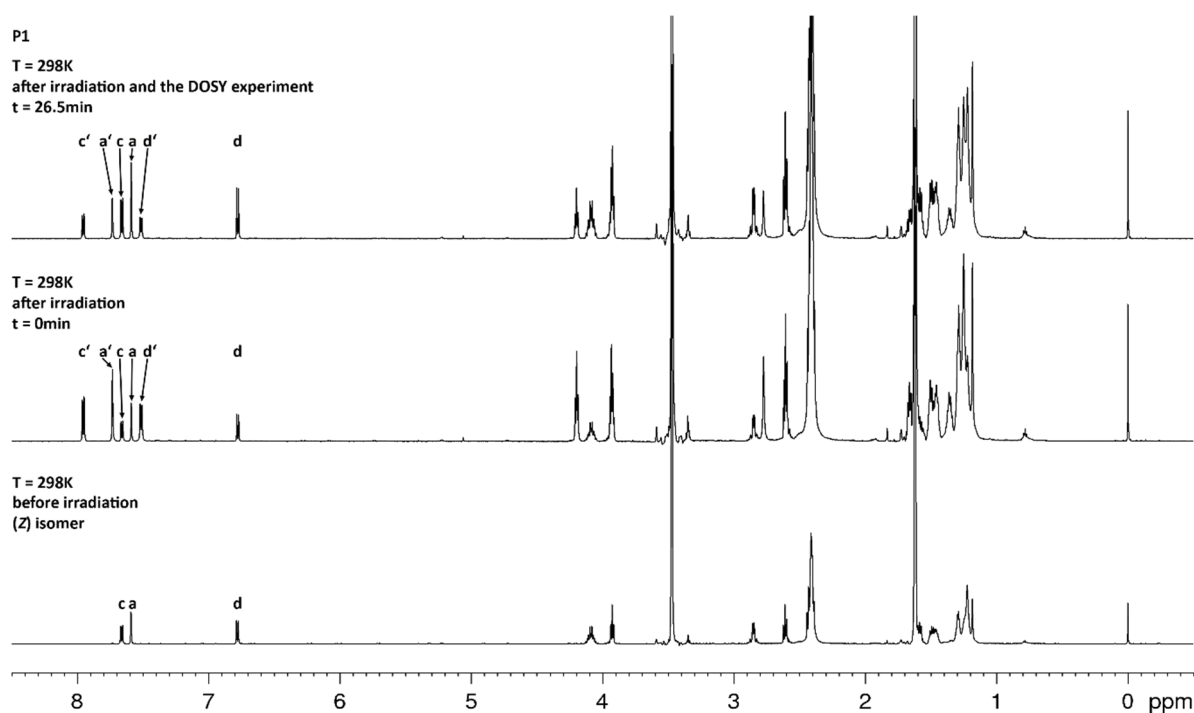
**Figure S8.** Stacked plot of the 1D  $^1\text{H}$  NMR spectra of **M1** from 8.5 to -0.5 ppm at T = 298 K. Bottom: before irradiation, centre: after irradiation at t = 0 min and top: after irradiation and the DOSY experiment at t = 27.3 min.



**Figure S9.** Example graphs are presented to highlight the effect of data intensity correction of one peak of the (E) (top) and (Z) isomer (bottom) after light irradiation at 405 nm wavelength of M1. Fit statistics are shown for the individual fits of these resonances.

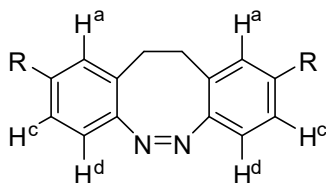
Poly[3,3'-hexane-1,6-diylbis(sulfanediyl) bis(propionyloxynonyl) (Z)-(11,12-dihydrodibenzo[c,g][1,2]diazocine-2,9-dicarboxylate) (P1)





**Figure S10.** Stacked plot of the 1D  $^1\text{H}$  NMR spectra of polymer **P1** from 8.5 to -0.5 ppm at  $T = 298\text{ K}$ . Bottom: before irradiation, centre: after irradiation at  $t = 0\text{ min}$  and top: after irradiation and the DOSY experiment at  $t = 26.5\text{ min}$ .

Aromatic proton assignments of **M1**, **P1**, **M2**, **P2** for  $^1\text{H}$  DOSY NMR:



**Table S1.** Corrected diffusion coefficients  $D$  [ $10^{-6}\text{ cm}^2\text{ s}^{-1}$ ] of the (Z) and (E) isomers of **M1**, **P1**, **M2**, **P2** after light irradiation at 405 and 525 nm wavelength from  $^1\text{H}$  DOSY NMR measurements.

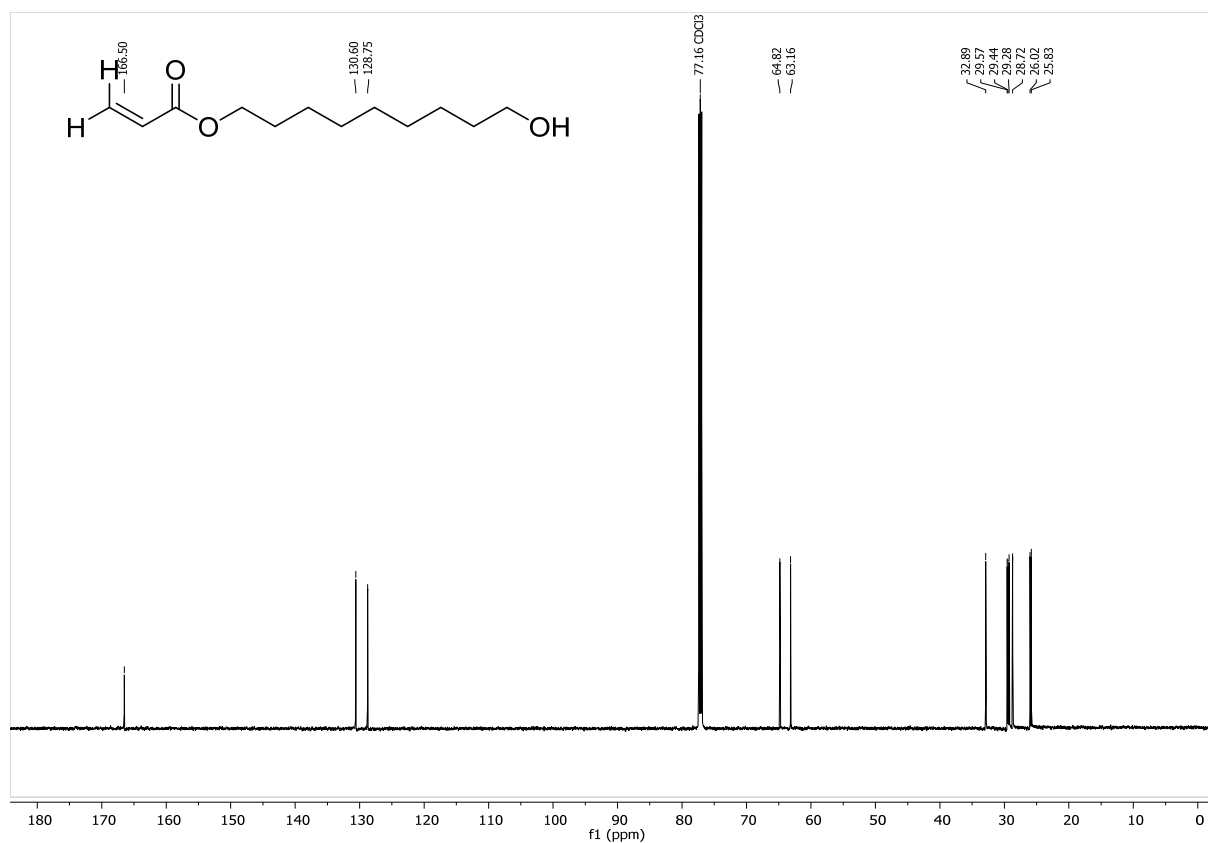
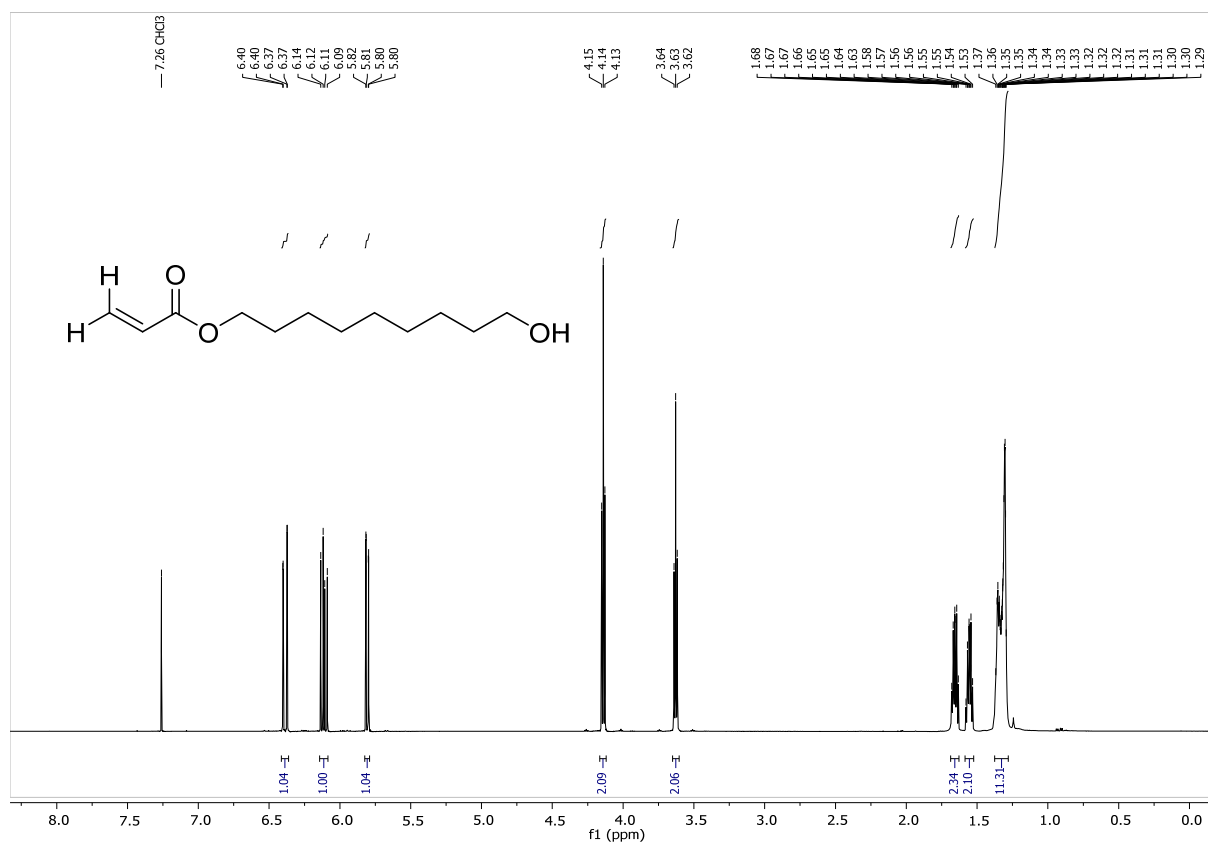
		PSS (405 nm)				PSS (525 nm)	
		Z (average)	Z (c; a; d)	E (average)	E (c'; a'; d')	Z (average)	Z (c; a; d)
<b>M1</b> <sup>1</sup>		7.45±0.14	7.60±0.04	7.39±0.07	7.32±0.04	7.46±0.11	7.58±0.03
			7.43±0.05		7.41±0.06		7.42±0.02
			7.33±0.04		7.45±0.03		7.37±0.03
<b>P1</b> <sup>2</sup>		1.90±0.03	1.90±0.03	1.91±0.02	1.89±0.03	1.96±0.05	2.00±0.03
			1.87±0.03		1.93±0.03		1.91±0.03
			1.92±0.03		1.91±0.03		1.98±0.03
<b>M2</b>		10.4±0.16	10.3±0.26	11.0±0.26	11.3±0.04	10.6±0.09	10.7±0.05
			10.3±0.10		10.8±0.04		10.5±0.04
			10.6±0.06		10.8±0.11		10.7±0.06
<b>P2</b>		2.72±0.21	2.91±0.03	2.75±0.07	2.79±0.03	2.84±0.08	2.84±0.02
			2.74±0.03		2.79±0.03		2.93±0.03
			2.49±0.02		2.67±0.03		2.76±0.03

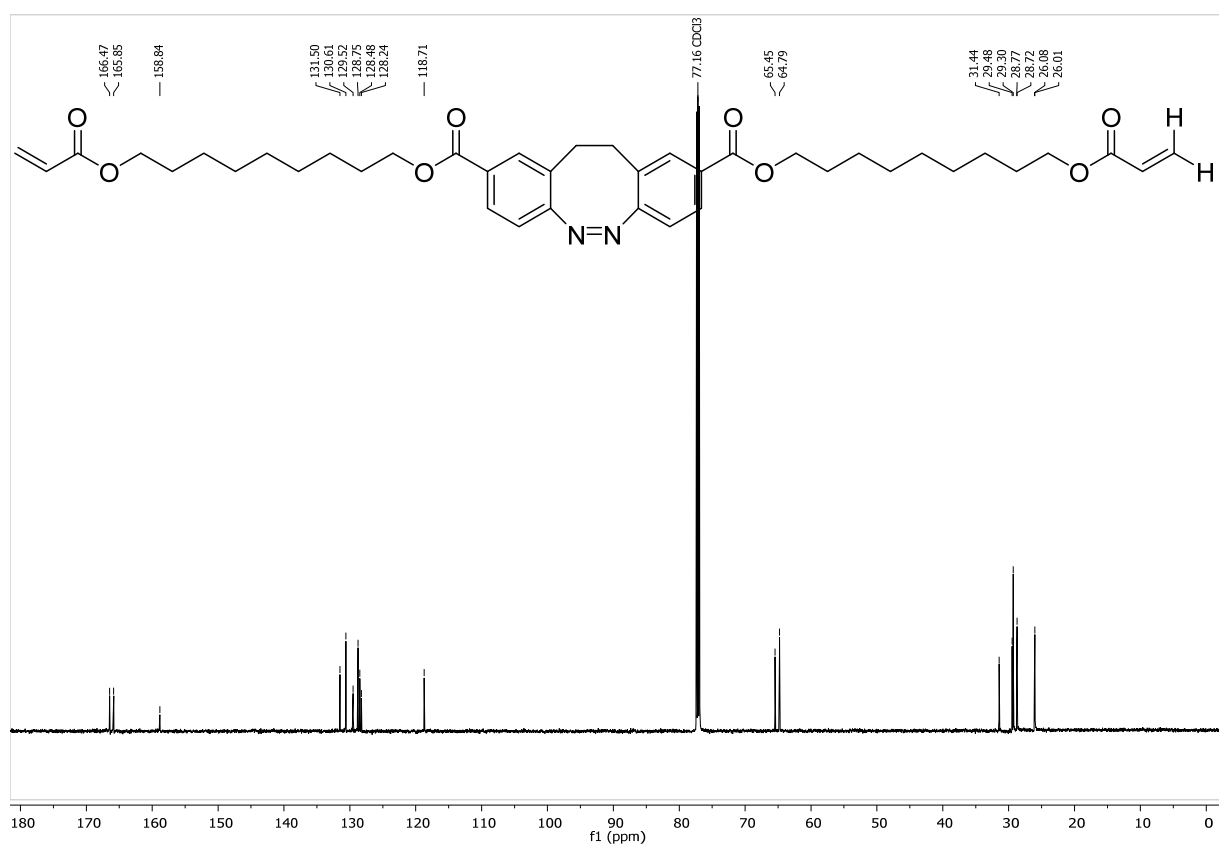
<sup>1</sup>Amount of (E) at  $t = 0$ : 68.5%;  $f = 22.93\text{ min}$ . <sup>2</sup>Amount of (E) at  $t = 0$ : 67.2%;  $f = 17.20\text{ min}$ .

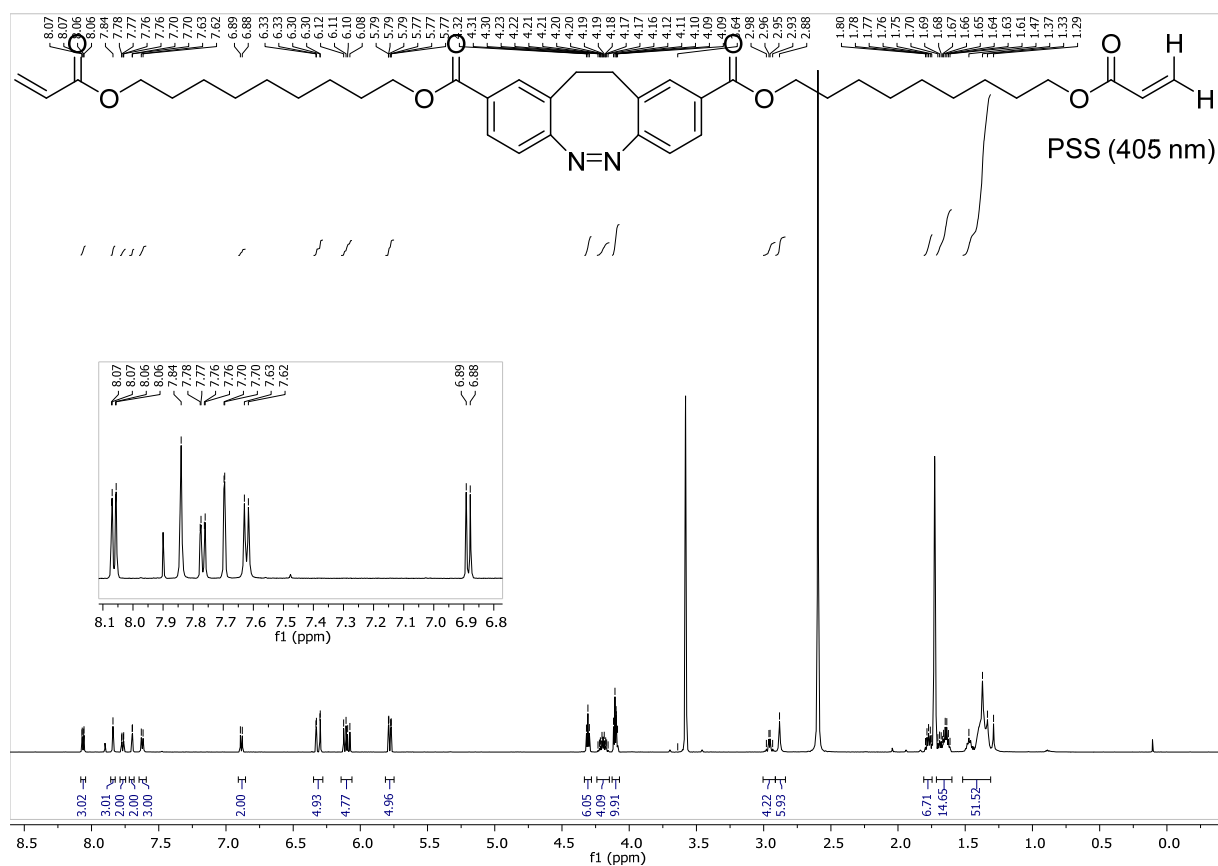
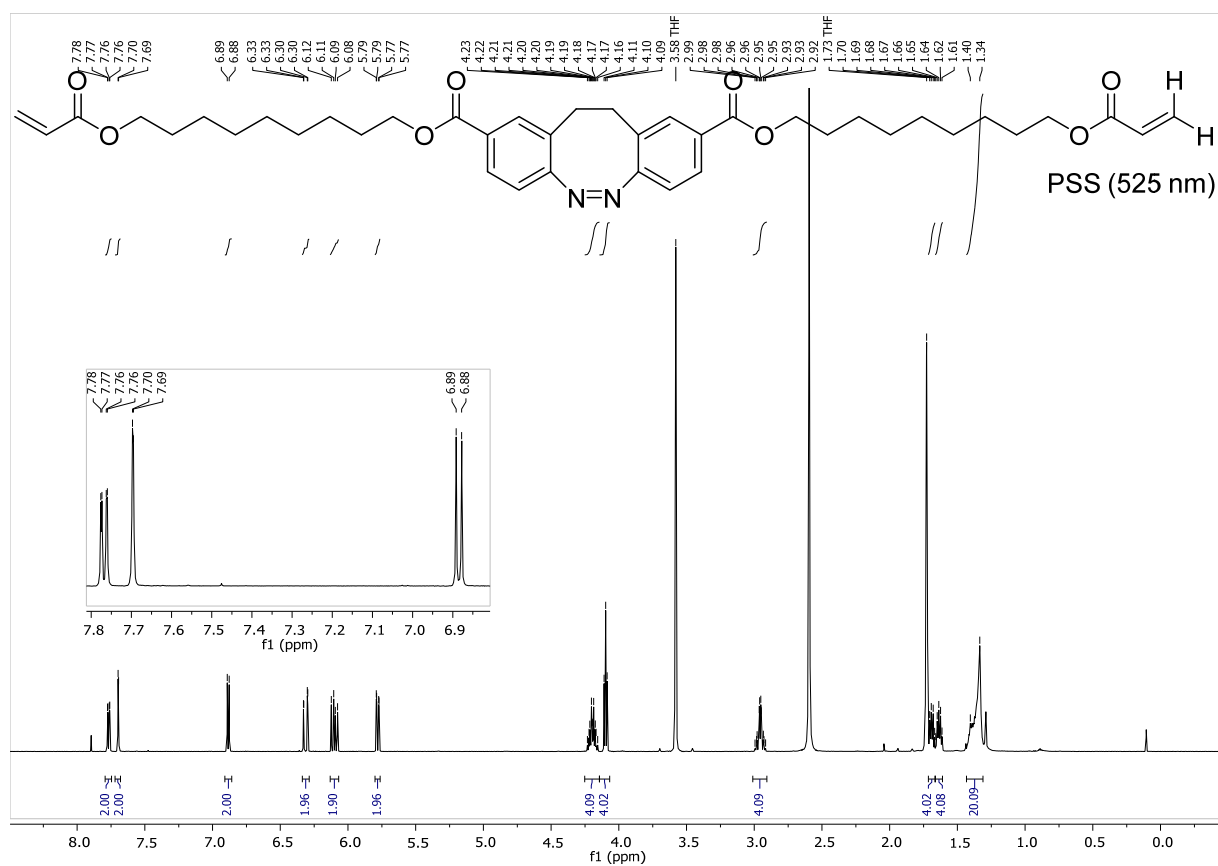


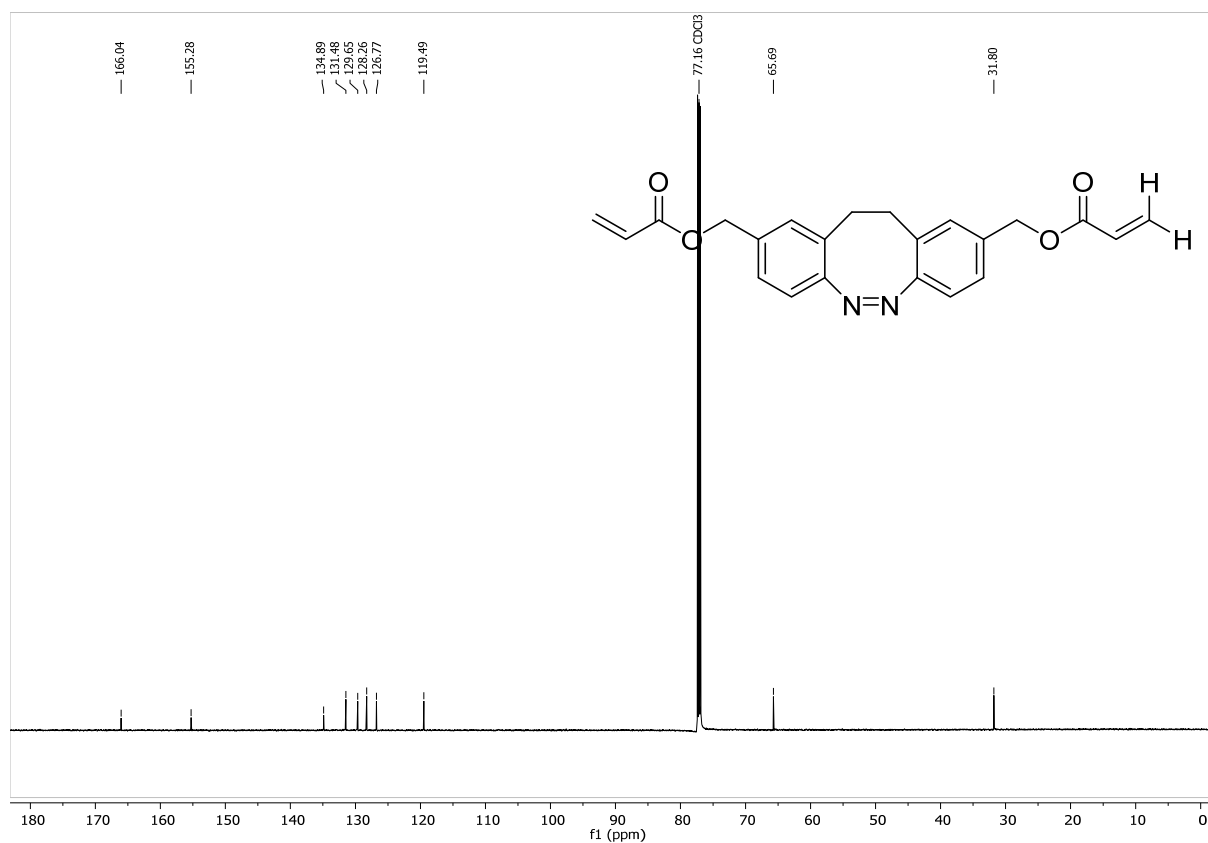
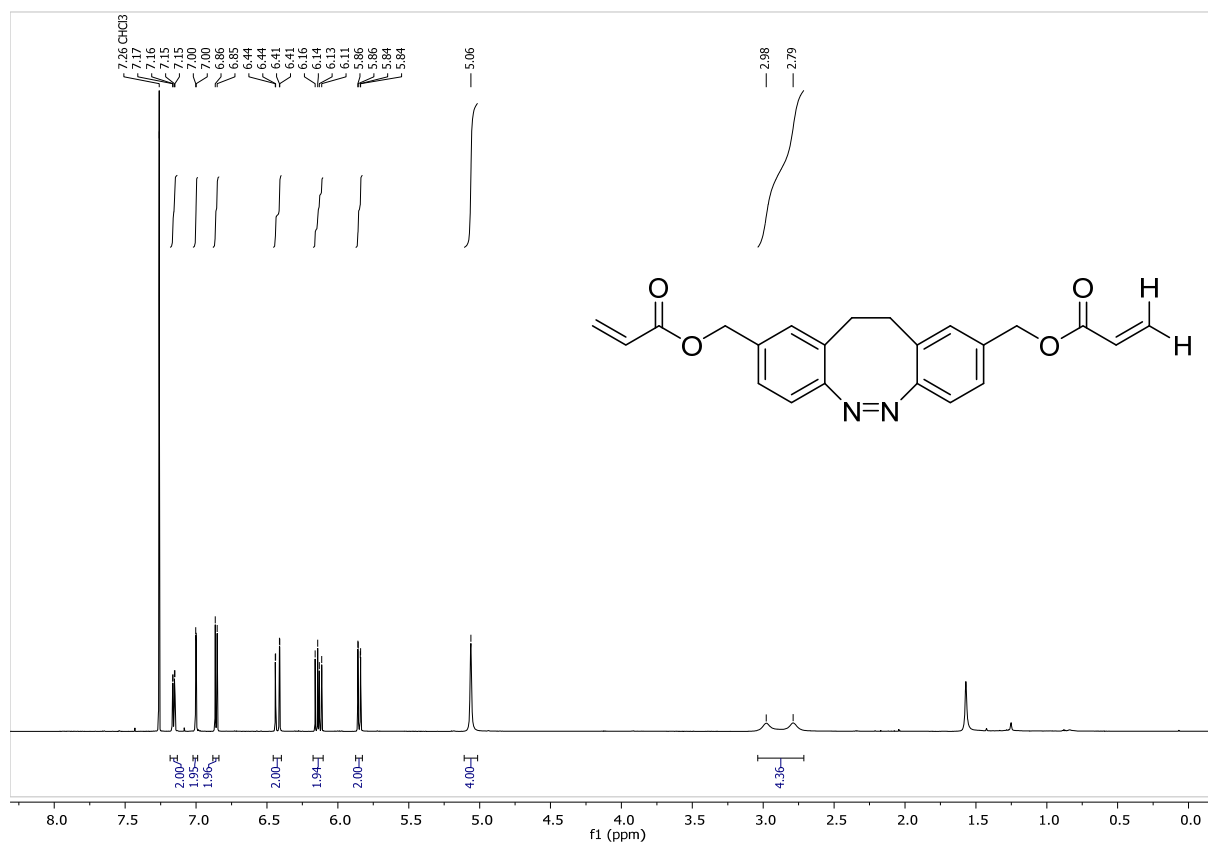
$^1\text{H}$  and  $^{13}\text{C}\{^1\text{H}\}$  NMR Spectra of the Products

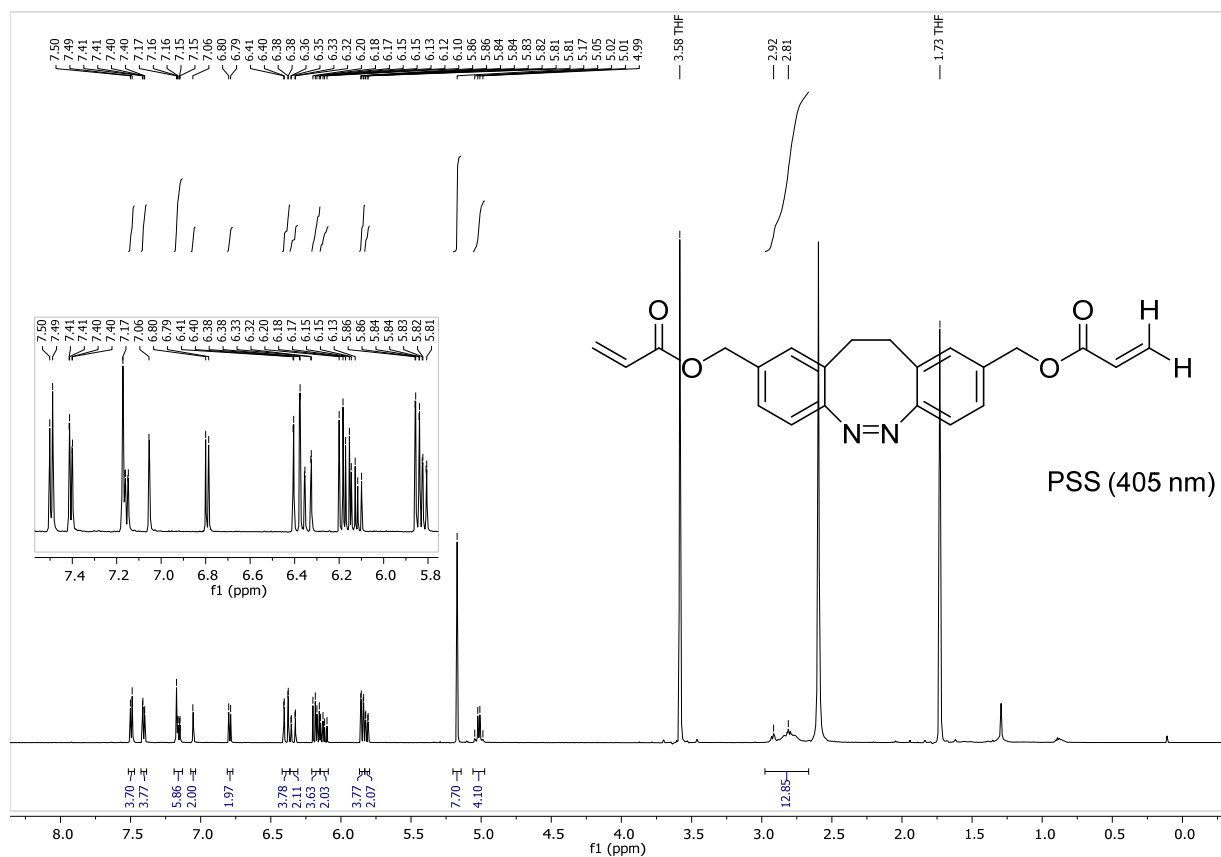
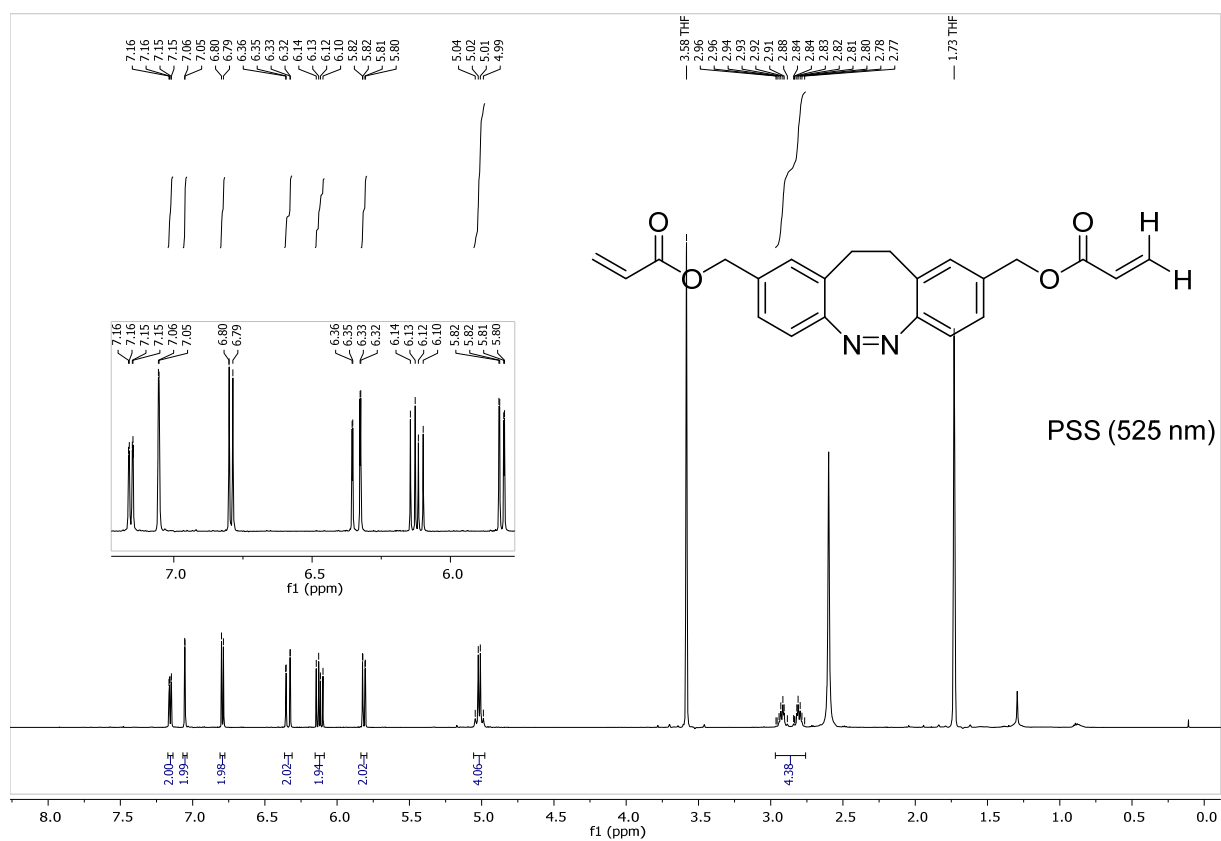
## 9-Hydroxynonyl acrylate



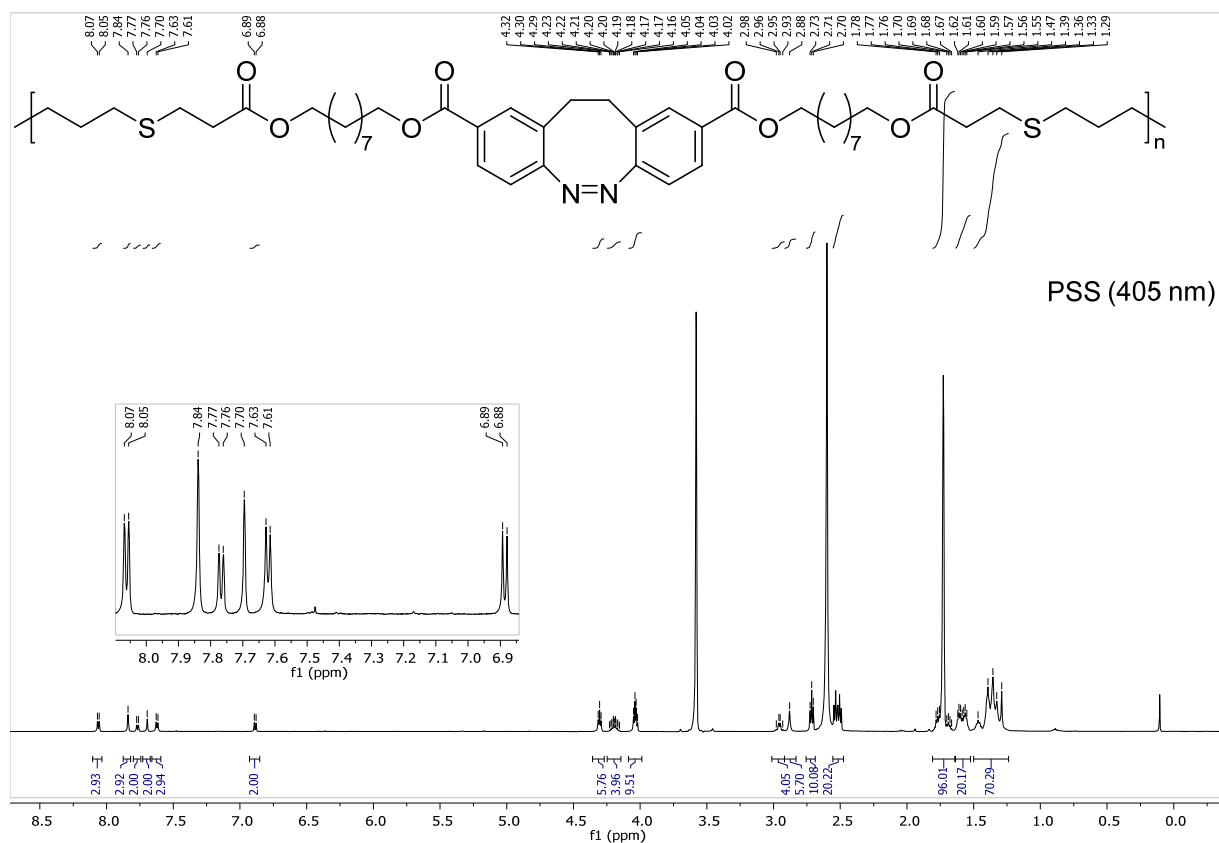
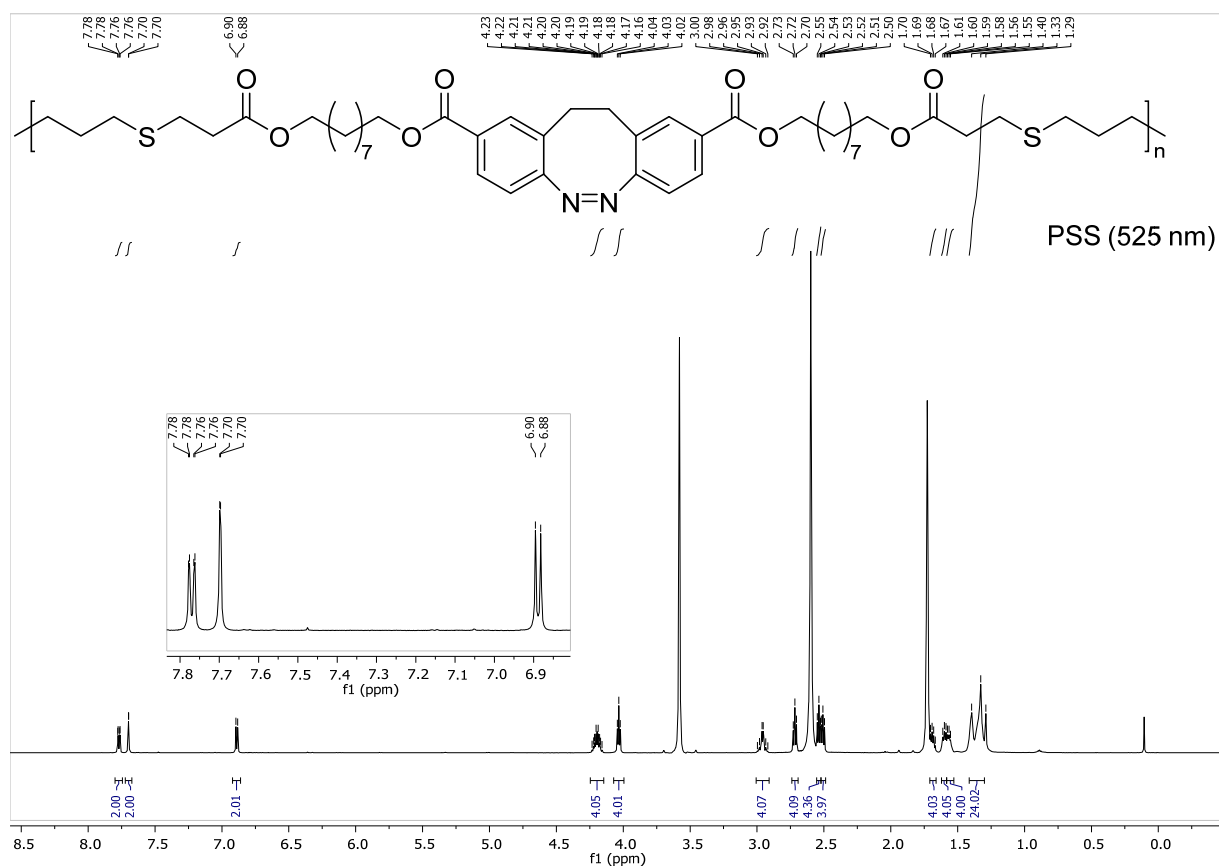
**Bis(9-(acryloyloxy)nonyl) (Z)-11,12-dihydrodibenzo[c,g][1,2]diazocine-2,9-dicarboxylate (M1)**

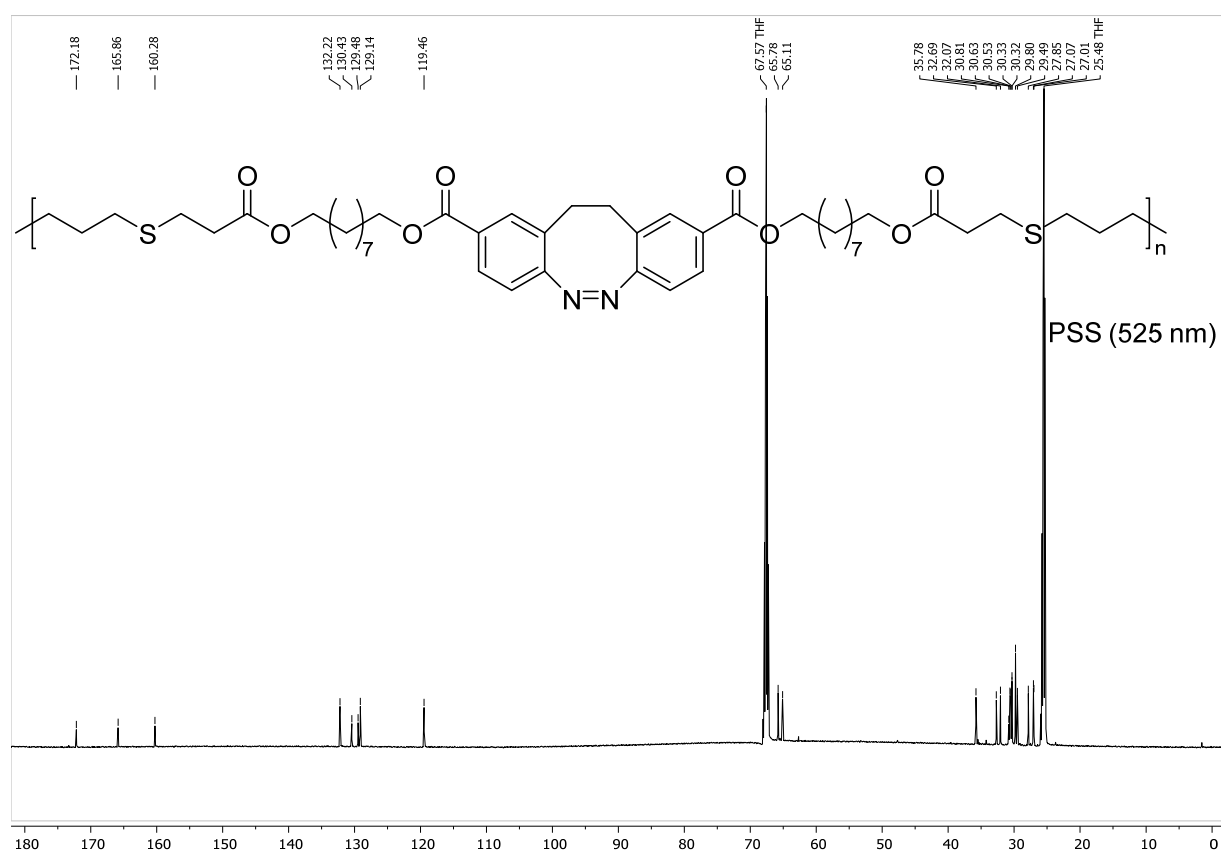


**(Z)-(11,12-Dihydrodibenzo[*c,g*][1,2]diazocine-2,9-diyl)bis(methylene) diacrylate (M2)**

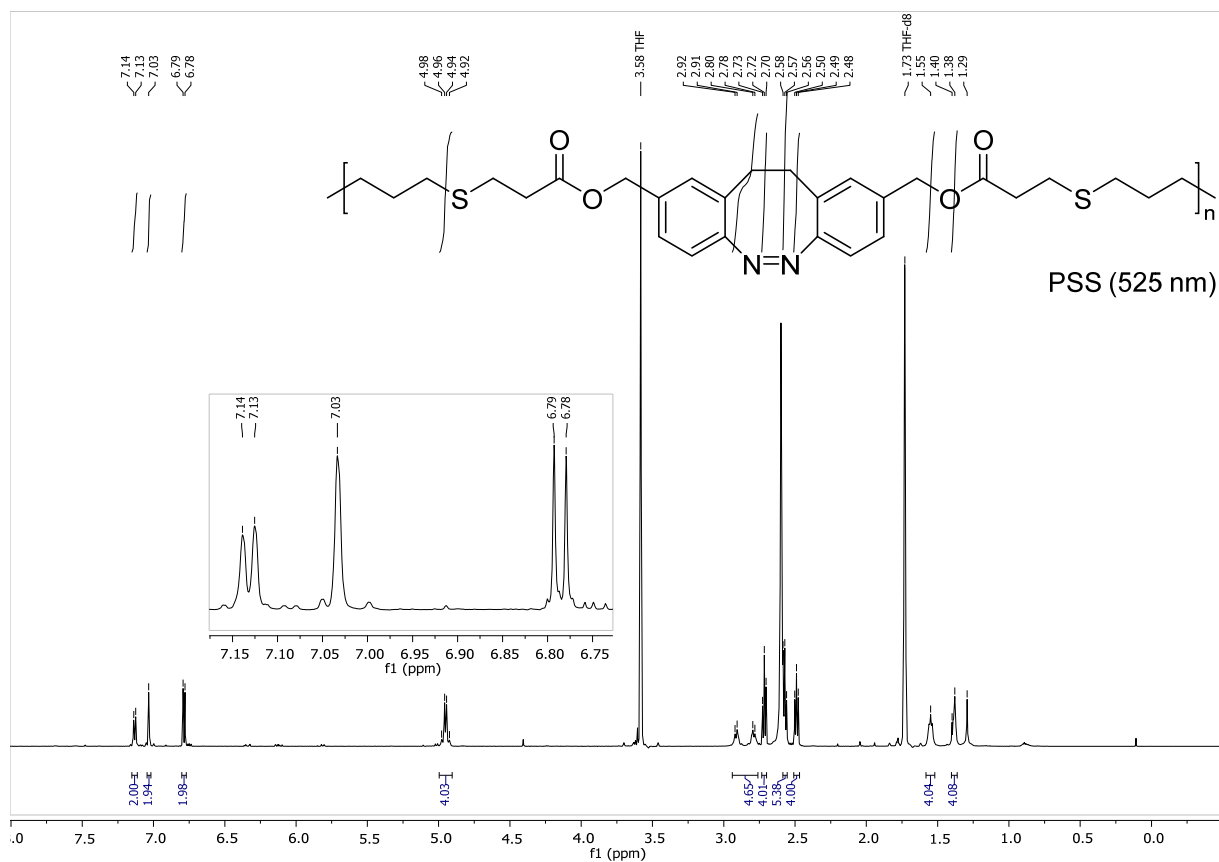


**Poly[3,3'-hexane-1,6-diylbis(sulfanediyl) bis(propionyloxynonyl) (Z)-(11,12-dihydrodibenzo[c,g][1,2]diazocine-2,9-dicarboxylate) (P1)**

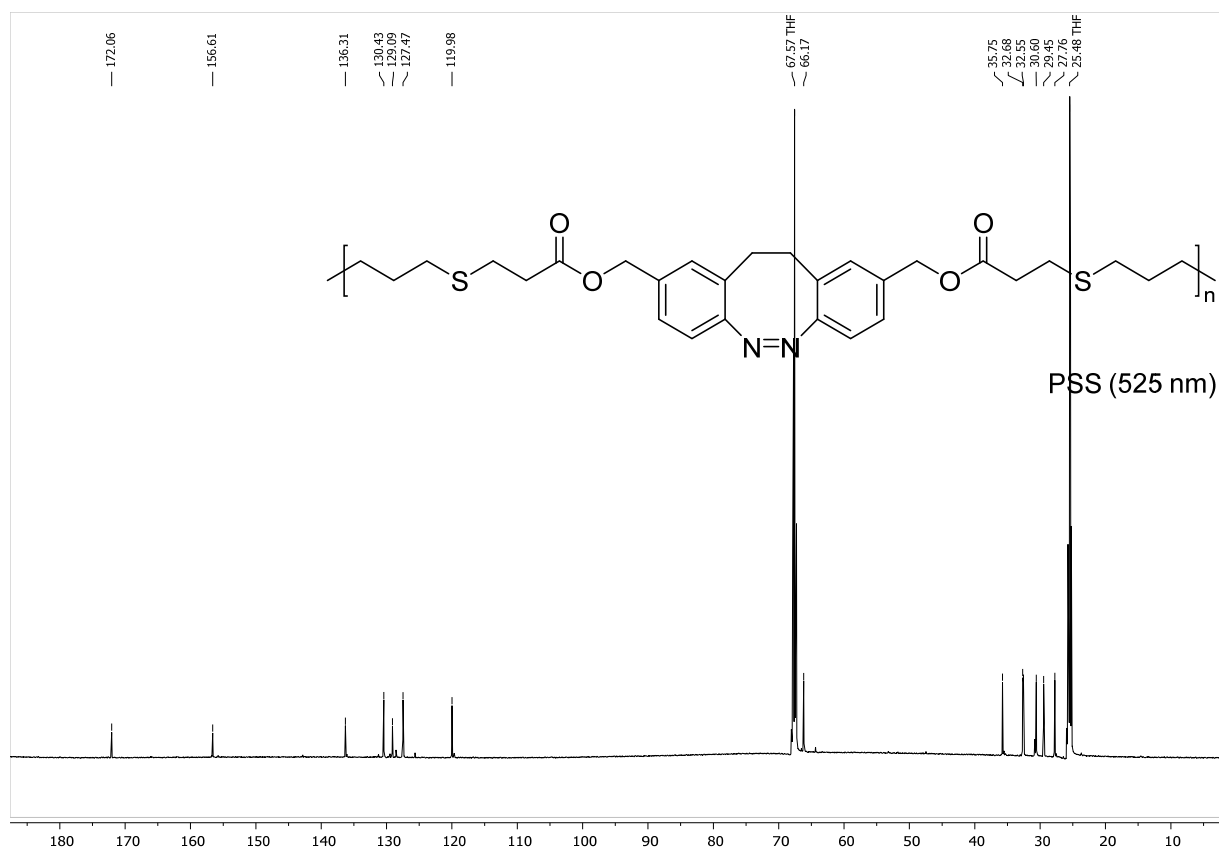
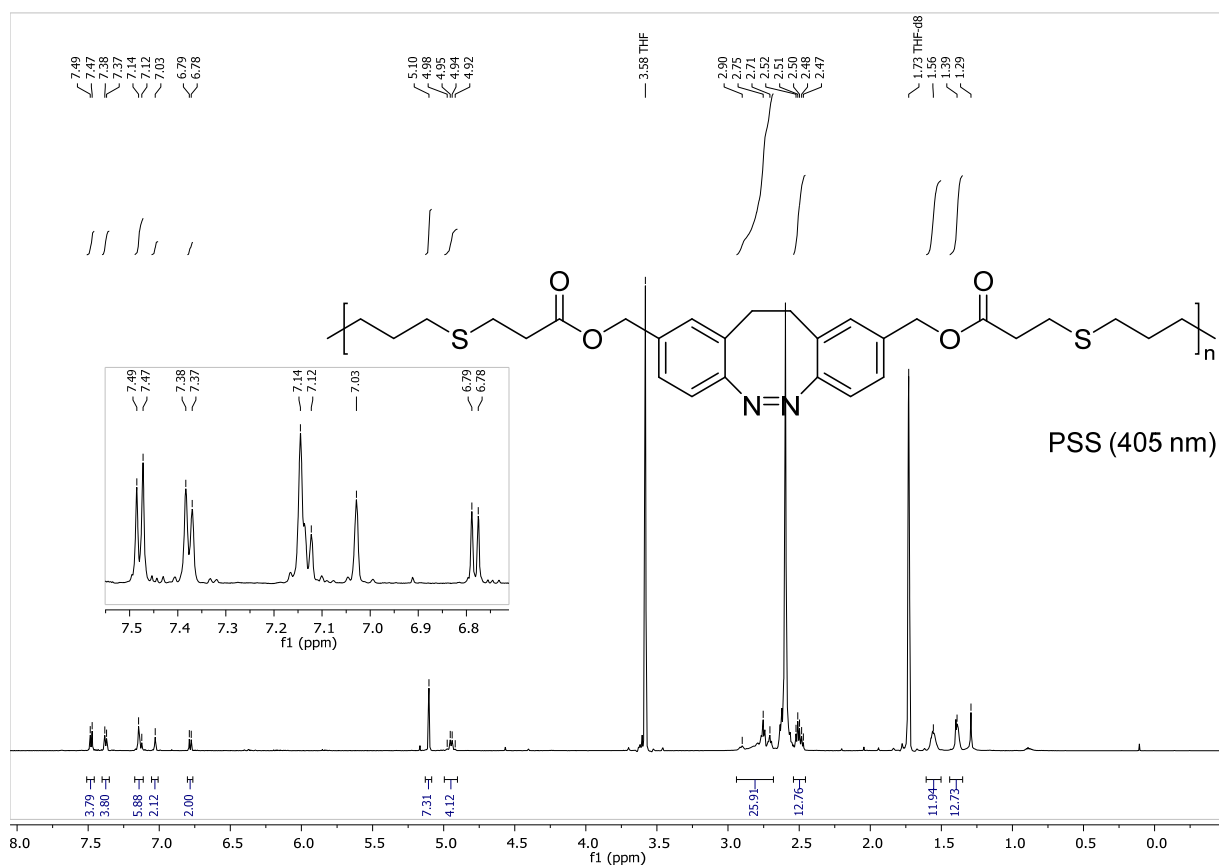




**Poly[(Z)-(11,12-dihydrodibenzo[c,g][1,2]diazocine-2,9-dimethyl-3,3'-(hexane-1,6-diylbis(sulfanediyl))dipropionate]**  
**(P2)**

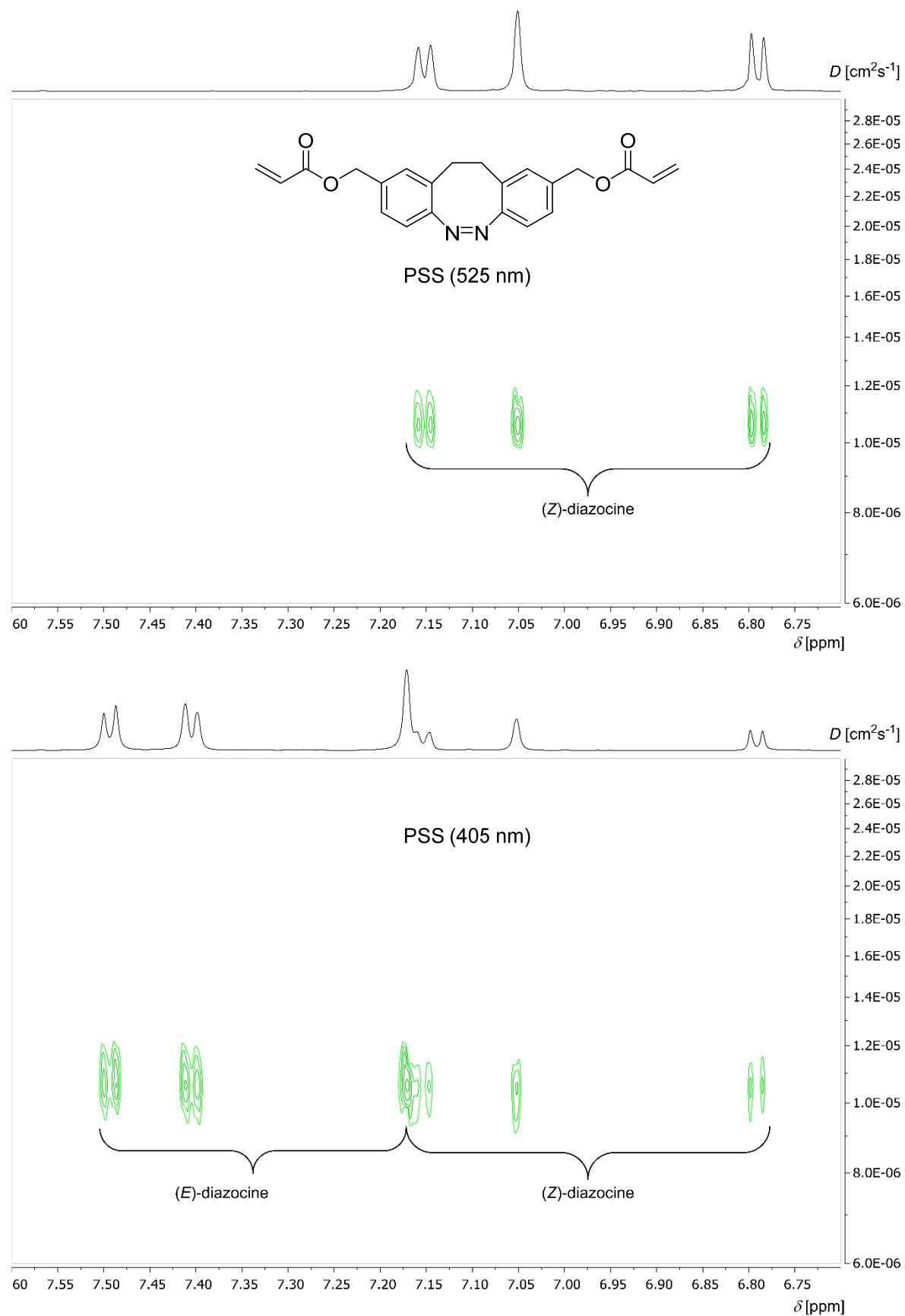






<sup>1</sup>H DOSY NMR Spectra of M2 and P2

## (Z)-(11,12-Dihydrodibenzo[c,g][1,2]diazocine-2,9-diyl)bis(methylene) diacrylate (M2)



**Poly[(Z)-(11,12-dihydrodibenzo[c,g][1,2]diazocine-2,9-dimethyl-3,3'-(hexane-1,6-diylbis(sulfanediyl))dipropionate]**  
**(P2)**

