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# Analysis and Risk Assessment of PM<sub>2.5</sub>-Bound PAHs in a Comparison of Indoor and Outdoor Environments in a Middle School: A Case Study in Beijing, China

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**Abstract:** People spend most of their time in indoors and, as a result, indoor air quality has become an issue of increasing concern. Due to the use of coal and heavy transportation in Beijing, China, concentrations of polycyclic aromatic hydrocarbons (PAHs) bound to PM<sub>2.5</sub> have risen and caused concerns about health risk, both outdoors and indoors. This study carried out quantitative investigation of PM<sub>2.5</sub>-bound PAHs in middle school classrooms and estimated the health risk to adolescents. According to the results, indoor PM<sub>2.5</sub> concentrations ranged from 20.9 µg/m<sup>3</sup> to 257.6 µg/m<sup>3</sup>, indoor PAH concentrations ranged from 8.0 ng/m<sup>3</sup> to 83.0 ng/m<sup>3</sup>, and both were statistically correlated with outdoor concentrations. Results of diagnostic ratios (DR) and the PMF (positive matrix factorization) model indicated that coal combustion was the main source of PAHs in the classroom environment. The average value of incremental lifetime cancer risk (ILCR) was estimated to be  $1.49 \times 10^{-6}$ , which indicated a potential health risk to students according to USEPA standards. Predictions showed that by 2021–2022, the risk will be reduced to an acceptable level. Results of this study could provide useful information for air pollution control in Beijing and proposing targeted solution against indoor air pollution.

**Keywords:** PM<sub>2.5</sub>-bound PAHs; indoor air quality; risk assessment; middle school students; Beijing

## 1. Introduction

Due to an increasing population, vehicles, and industrial activities, severe air pollution has occurred and has drawn attention in recent years in China, especially in its capital, Beijing. Based on previous research [1–3], the main pollutant that causes this problem is PM<sub>2.5</sub>. PM<sub>2.5</sub> is believed to cause numerous adverse effects on the environment and human health [4]. According to the “Beijing Ecology and Environment Statement 2019”, data published by the Beijing Municipal Ecology and Environmental Bureau [5], the average annual concentration of PM<sub>2.5</sub> in Beijing in 2019 was 42 µg/m<sup>3</sup>, while the guideline is 35 µg/m<sup>3</sup>. A considerable number of policies and regulations have been carried out, and the number of days when the air quality level was above heavy pollution from 2013 to 2019 for each year was 58, 45, 42, 38, 22, 13 and 4 days, respectively. Despite the trend of air quality getting better, heavy pollution weather occurred sometimes.

The spatial topography and climatic conditions of Beijing have a significant connection with the air pollution in this city [6]. There are several mountains locating in the western, northern and northeastern area of Beijing, while the land is flat in the southeast region. Furthermore, the concentration of air

pollutants decreases when wind speed increases. Moreover, lots of tall buildings have been built in recent decades, thus, the diffusion of air pollutants is blocked [7]. On this basis, a large population and complex industrial structure make this problem even worse [6].

Since PM<sub>2.5</sub> is able to penetrate and deposit deep in the tracheobronchial and alveolar regions in the human body and have systemic effects [8], it can cause various health problems, such as respiratory disease, the development of lung cancer and increased daily mortality and morbidity [9].

Previous studies showed that people spend approximately 80%–90% of their time indoors everyday [13], thus, indoor air quality is crucial to people's health. However, several studies have found that the indoor PM<sub>2.5</sub> level is even worse than that of the outside environment [14–16]. PAHs bound to PM<sub>2.5</sub> have a great effect on the indoor air quality, which is impacted by various sources, such as cooking, smoking, domestic chemicals, and outdoor sources [17]. To the best of our knowledge, the vast majority of research about indoor air pollution focused on public buildings [18,19], especially residential communities [20,21], and comparatively few have studied classroom environments. Chinese middle-school students spend nearly a half a day or more time in the classroom every day, so the classroom environment is important for student health.

Among all organic chemicals contained in PM<sub>2.5</sub>, polycyclic aromatic hydrocarbons (PAHs) are of great concern because they have potential risks to human health [10]. PAHs are relatively ubiquitous chemicals in the environment [11]. One of the PAHs, benzo(a)pyrene (BaP), is a strong carcinogen and, as such, it is used as an indicator in risk assessments regarding exposure to PAHs (EC, 2005). PAHs can be generated from natural sources such as volcanic eruptions and forest fires, but its main source is anthropogenic emissions, including the combustion of fossil fuel and biomass, automobile exhaust, and industrial activities, etc. [12].

Studies found that the fine particulate matter concentrations observed in classrooms were higher than those being observed in residences and commercial buildings [17,22]. At the same time, there are few typical indoor sources such as smoking and cooking in classroom environments [23], leading to challenges for source tracking. Because most students are in the period of growth and development, they are relatively susceptible to pollutants when compared to adults [24]. Therefore, it is necessary to study pollution levels and health risk of PM<sub>2.5</sub>-bound PAHs in classrooms. Furthermore, China has not carried out indoor PM<sub>2.5</sub> standards as in the case of some western countries and, hence, this area needs further consideration.

The PAHs in air have gas and particle phases. Gas-phase PAHs mostly have less than four rings in their chemical structure, while particle-phase PAHs mostly have more than four rings. According to their toxicity [25], high molecular weight (HMW) PAHs with five or six rings, such as BaP, are more dangerous to human health. Thus, PAHs in the particle phase are much more dangerous than those in the gas phase. In this study, we collected PAHs bound to PM<sub>2.5</sub> for analysis.

In our study, samples were all collected from a middle school in Beijing. Possible sources of PAHs bound to PM<sub>2.5</sub> were identified by applying diagnostic ratio and the positive matrix factorization (PMF) model. Assessment of students' health risk caused by indoor PAHs was performed by the incremental lifetime cancer risk (ILCR) model, combined with Monte Carlo simulation to reduce the uncertainty of results. Potential health risks to students were pointed out in the classroom environment, which should not be ignored in the long term. The findings of this study aim to broaden our understanding of air pollution in specific enclosed environments like the classroom. We also analyzed the main factor that contributed to the risk assessment. The results are also expected to provide useful information for researchers conducting analyses of health risks and help decision-makers in regard to environmental policy-making and control measures.

## 2. Materials and Methods

### 2.1. Sampling Site

In order to ensure that the sampling points were representative, we chose a middle school in the Xicheng District. The middle school is near a road that has a large traffic flow. This district is in the center of Beijing and has typical surroundings, such as main roads, a residential area, and a mall, with no construction sites or factories. In order to use this sample site as a representative school, both indoor and outdoor sampling points were established. The indoor sampling point was set as a classroom on the third floor of the main three-story teaching building, while the outdoor point was set as the roof of the same teaching building with a similar height. Indoor and outdoor sampling were operated simultaneously to be comparable.

### 2.2. Sample Collection

Sampling was conducted from October 2016 to March 2017. PM<sub>2.5</sub> samples were collected across three seasons: (1) autumn, October to November 2016; (2) winter, January 2017 and (3) spring, March 2017. According to the ambient PM<sub>2.5</sub> level calendar [5], most heavy pollution occurred during the period from October to March of the following year. The sampling dates were random, and we tried to conduct the sampling as much as possible. Several samples were collected in winter vacation to show the differences between school time and long vacations. Furthermore, autumn samples were collected before Beijing's central heating period, while winter and spring samples were collected within the central heating period.

A total of 54 samples were taken. The sampling was carried out using the TH-150 series intelligent total suspended particle flow sampler (Wuhan Tianhong Instrument Co., Ltd. in Wuhan, China). This instrument can collect PM<sub>2.5</sub>, PM<sub>10</sub> or TSP (total suspended particulate) using cutters of different sizes. In this study, we used the cutter for collecting PM<sub>2.5</sub>. The sampling procedure referred to the national standards of China (HJ637-2013). The sampler operated at a flow rate of 100 L/min. Each sample was collected by the quartz filter for 24 h. Before sampling, the quartz filters were prebaked in muffle oven at 500 °C for 5 h, as this removes the target organic compounds. The sample filters were stabilized in a temperature-stable and humidity-controlled incubator before and after collection for 24 h. Each filter was weighed at least three times using an electronic analytical balance before and after sampling to ensure variance among the measurement results for each sample at <10 µg. All the filters were analyzed by ultrasonic extraction–liquid chromatography using Waters UPLC H-class. Blank tests were carried out to ensure the reliability of the sampling and testing process. The result showed that no target PAHs were found in the blank tests.

According to the results, 12 PAHs were identified: phenanthrene (Phe), anthracene (Ant), fluoranthene (FL), pyrene (Pyr), benzo(a)anthracene (BaA), chrysene (Chr), benzo(b)fluoranthene (BbF), benzo(k)fluoranthene (BkF), benzo(a)pyrene (BaP), dibenzo(a,h)anthracene (DBA), benzo(ghi)perylene (BghiP), and indeno(1,2,3-cd)pyrene (IcdP).

### 2.3. Source Apportionment

Atmospheric polycyclic aromatic hydrocarbons (PAHs) are the result of comprehensive emissions from a variety of sources. Studies have been experimenting with different methods to determine the main source of PAH pollution. Principal component analysis (PCA), cluster analysis, and diagnostic ratios (DR) are usually applied [26]. The diagnostic ratios method is based on differences in PAH composition and relative concentrations resulting from different combustion modes. DR method can identify the source of pollution according to the ratio of different PAHs, and is widely used and very convenient for the analysis of possible sources of PAHs. In this study, the DR method was applied to carry out source apportionment.

The main indicators used in the DR method are InP/(InP + BghiP), BaA/(BaA + Chr), Flu/(Flu + Pyr) and BaP/BghiP. According to these values, the possible sources of the collected PAHs can be found. The values we found in previous research are shown in Table 1 [26].

**Table 1.** Values of the diagnostic ratios (DR) method from different polycyclic aromatic hydrocarbon (PAH) resources.

Sources	IcdP/(IcdP + Bghip)	BaA/(BaA + Chr)	Flu/(Flu + Pyr)	BaP/BghiP
Cigarette	0.55–0.60	0.31	0.51–0.59	1.75
Cooking	0.4	0.54	0.69	1.61
Coal combustion	0.38–0.52	0.50–0.55	0.68–0.74	0.9–6.6
Traffic	0.42	0.39	0.6–0.7	0.3–0.78
Fuel combustion	0.18	0.39–0.60	0.38–0.4	0.3–0.4
Diesel combustion	0.37	0.15–0.26	0.6–0.7	0.46–0.81

To make the source apportionment more precisely, the PMF model was used. USEPA PMF v5.0 was used by many researchers, and after many improvements, the source apportionment results conducted by PMF were reliable.

#### 2.4. Exposure Assessment

In this study, a BaP-equivalent concentration ( $BaP_{eq}$ ) was used to evaluate the toxicity of PM<sub>2.5</sub>-bound PAHs, which can be expressed by the toxic equivalent factor (TEFs) of PAHs based on benzo [a] pyrene (BaP). The equation is as follows:

$$BaP_{eq} = C_i \times TEF_i \quad (1)$$

where  $C_i$  is the concentration of PAH species. Table 2 showed the TEF value [25,27] and average concentration of PAHs during our observation.

**Table 2.** PAH species, toxic equivalent factors (TEFs) and concentrations.

PAH Species	Chemical Formula	Rings	MW * Groups	TEFs	Average Concentration(ng/m <sup>3</sup> )	
					Indoor	Outdoor
Phe	C <sub>14</sub> H <sub>10</sub>	3	LMW	0.001	2.34	1.39
Ant	C <sub>14</sub> H <sub>10</sub>	3	LMW	0.01	0.36	0.18
FL	C <sub>16</sub> H <sub>10</sub>	4	MMW	0.001	3.35	3.10
Pyr	C <sub>16</sub> H <sub>10</sub>	4	MMW	0.001	2.24	2.01
BaA	C <sub>18</sub> H <sub>12</sub>	4	MMW	0.1	2.07	2.84
Chr	C <sub>18</sub> H <sub>12</sub>	4	MMW	0.01	2.75	4.08
Bbf	C <sub>20</sub> H <sub>12</sub>	5	HMW	0.1	4.34	5.27
BkF	C <sub>20</sub> H <sub>12</sub>	5	HMW	0.1	1.61	2.01
BaP	C <sub>20</sub> H <sub>12</sub>	5	HMW	1	3.06	3.67
DBA	C <sub>22</sub> H <sub>14</sub>	5	HMW	5	1.14	1.32
BghiP	C <sub>22</sub> H <sub>12</sub>	6	HMW	0.1	3.18	3.84
IcdP	C <sub>22</sub> H <sub>12</sub>	6	HMW	0.1	3.39	4.03

\* molecular weight (LMW= low molecular weight; MMW = medium molecular weight; HMW = high molecular weight).

The USEPA assumes that carcinogenic risk is believed to be a non-threshold effect, since even some small influences on the individual cell can result in abnormal cell proliferation and a clinical disease. People may face potential health risks due to exposure to the pollutants even at a relatively low concentration.

PAH exposure to the human body is mainly through inhalation and skin (although inhalation is much more dangerous than skin exposure). PAHs bound to PM<sub>2.5</sub> can enter the human body via respiration and present a potential carcinogenic risk. The carcinogenic risk of PAHs by respiratory exposure can be estimated using the incremental lifetime cancer risk (ILCR) model [28,29].

The value of risk can be calculated as a unitless parameter,  $R$ , indicating carcinogenic incidence over the normal level. The equation is as follows:

$$RI = \frac{CI \cdot \left( CSFi \cdot \left( \frac{BW}{70} \right)^{\frac{1}{3}} \right) \cdot IRi \cdot EF \cdot ED}{BW \cdot AT} \cdot cf \quad (2)$$

where  $R_i$  indicates the incremental lifetime cancer risk due to PAH exposure (dimensionless);  $C_i$  denotes the total concentration of *BaP<sub>eq</sub>* (ng/m<sup>3</sup>);  $CSF$  represents the inhalation cancer slope factor of BaP (kg·day·mg<sup>-1</sup>);  $BW$  is body weight (kg);  $IR_i$  is inhalation rate (m<sup>3</sup>·day<sup>-1</sup>);  $EF$  is exposure frequency (day·year<sup>-1</sup>);  $ED$  is exposure duration (year);  $AT$  = averaging time (70 years for carcinogens) (US EPA 1991); and  $cf$  is the conversion factor (10<sup>-6</sup>).

To reduce uncertainty of parameters in the equation, the Monte Carlo simulation was also conducted for both probability analysis and uncertainty assessment, where parameters were expressed by a series of data based on probability distribution. The result of the Monte Carlo simulation was a distribution value rather than specific data.

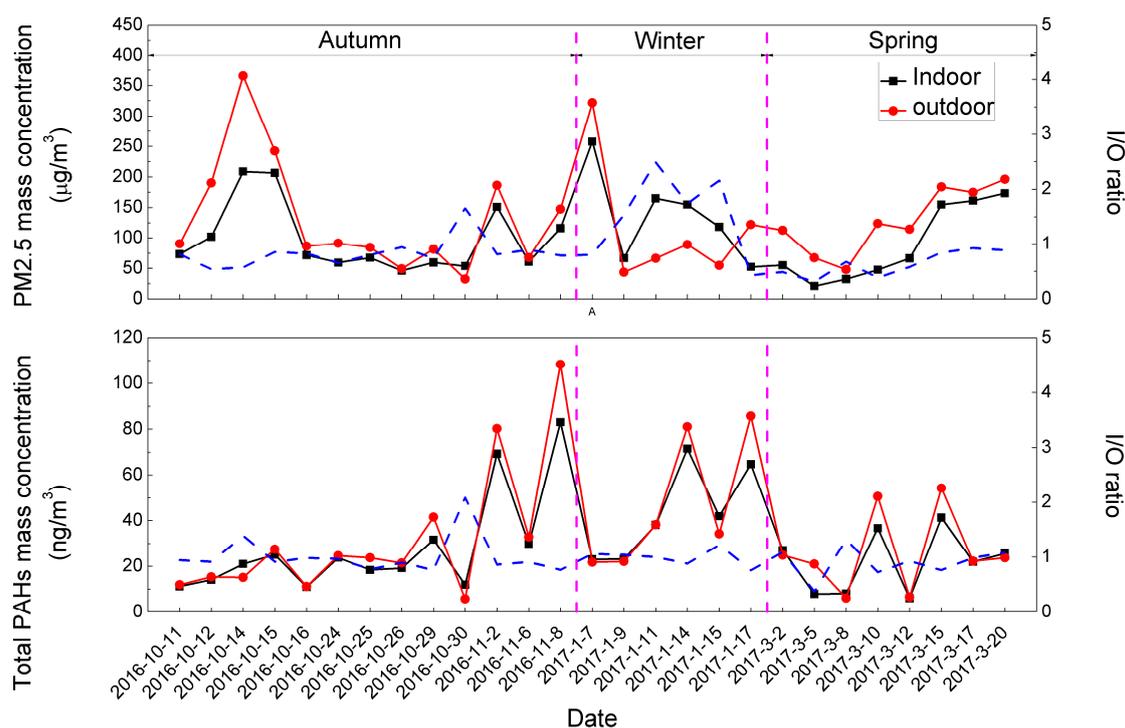
USEPA regards  $1.0 \times 10^{-6}$  as acceptable for risk level (USEPA, 2001), which suggests one in one million might suffer from cancer due to pollutant exposure in the sampling area. If the assessment result is less than  $1.0 \times 10^{-6}$ , the risk level is acceptable; if the result is more than  $1.0 \times 10^{-6}$  but less than  $1.0 \times 10^{-4}$ , there is potential risk; and if the result is higher than  $1.0 \times 10^{-4}$ , the risk level is serious and further protective management should be carried out.

### 3. Results

#### 3.1. Characteristics of the Pollutants in the Sample Site

Figure 1 shows the variation levels of PM<sub>2.5</sub> and PAH mass concentration in indoor and outdoor observations from our samples. We conducted observations from autumn 2016 to spring 2017 and collected 54 samples in total. The average concentration of PM<sub>2.5</sub> in the classroom was 103.7 µg/m<sup>3</sup> (range: 20.9–257.6 µg/m<sup>3</sup>), while the average concentration of ambient PM<sub>2.5</sub> outdoors was 131.8 µg/m<sup>3</sup> (range: 32.4–365.2 µg/m<sup>3</sup>). According to the Environmental Air Quality Standard (GB 3095-2012) published by the Ministry of Environmental Ecology of China, the standard for PM<sub>2.5</sub> daily average concentration in a residential area is 75 µg/m<sup>3</sup>. During the sampling period, 44.4% of the indoor and 70.4% of outdoor PM<sub>2.5</sub> samples exceeded the national standard. The outdoor PM<sub>2.5</sub> concentration during the most polluted period was almost 11 times higher than the concentration during the most ideal period.

Although compared to the concentrations of outdoor air, indoors is a relatively safe environment, there was still a statistically significant correlation between PM<sub>2.5</sub> concentrations in the indoor and outdoor environments. In the autumn and spring sampling period, the Pearson correlation coefficient between the indoor and outdoor PM<sub>2.5</sub> concentration was 0.90 ( $p < 0.01$ ). Hence, outdoor PM<sub>2.5</sub> concentrations have a great impact on air quality indoors, and, in most cases, were higher than those of indoors. However, during the winter sampling period, the correlation between the indoor and outdoor PM<sub>2.5</sub> was relatively weak, with a correlation coefficient of 0.73 ( $p < 0.01$ ), and observations from 9 January to 15 January showed that indoor PM<sub>2.5</sub> concentrations actually exceeded outdoor concentrations significantly. This time period was during the school's winter vacation. The concentration of PAHs indoors did not rise to the corresponding level, and this was most likely related to the dust accumulation in the indoors environment during long vacations.



**Figure 1.** Time series and indoor/outdoor (I/O) ratio of PM<sub>2.5</sub> and total PAH concentrations in indoor/outdoor air.

High concentration of fine particulates in indoor environments have been reported in several investigations. Chithra et al. [23] found that in Indian classrooms, concentrations of PM<sub>10</sub> and PM<sub>2.5</sub> were observed to be  $136.1 \pm 60.43$  and  $36.30 \pm 14.98$   $\mu\text{g}/\text{m}^3$ , respectively. Indoor PM concentrations were higher than outdoor PM levels, with the concentration of PM<sub>10</sub> being around two times higher than that of the outdoors. Chen et al. [30] conducted investigations in three indoor environments in Beijing and found that average concentrations of PM<sub>2.5</sub> in a dormitory, office, and residential house were  $52.1$   $\mu\text{g}/\text{m}^3$ ,  $55.3$   $\mu\text{g}/\text{m}^3$ ,  $35.5$   $\mu\text{g}/\text{m}^3$ , respectively. Compared with these studies, the classroom pollution in our study was much higher than other indoor environments such as offices and homes.

In this study, the average mass concentration of PAHs bound to PM<sub>2.5</sub> in the outdoors during the entire sampling period was  $33.7$   $\text{ng}/\text{m}^3$ , ranging from  $6.5$   $\text{ng}/\text{m}^3$  to  $108.4$   $\text{ng}/\text{m}^3$ . The indoor PAH concentration showed the same trend as that of the outdoors and the average value was  $29.8$   $\text{ng}/\text{m}^3$ , ranging from  $8.0$   $\text{ng}/\text{m}^3$  to  $83.0$   $\text{ng}/\text{m}^3$ . Tables 3 and 4 show the concentration data of PM<sub>2.5</sub> and total PAHs in this study.

Compared to PM<sub>2.5</sub>, PAHs showed more obvious seasonal trends, as during the cold sampling period the concentration rose and during the warm period it declined. The difference in concentration was significant. Generally, the effect of the meteorological conditions can lead to high PAH concentrations during the winter season, such as low atmospheric temperature affecting the distribution of PAH gas and particles, low mixing height reducing PAH dispersion and so on, as several studies have reported [31–33]. Furthermore, Beijing's central heating might contribute significantly to the pollutants according to the seasonal variation.

Consistent with PM<sub>2.5</sub>, there was an apparent correlation between indoor and outdoor PAHs with a correlation coefficient of  $0.97$  ( $p < 0.01$ ), which suggested that outdoor pollution was a major source of indoor air pollutants.

**Table 3.** Mass concentrations of PM<sub>2.5</sub> during the sampling period.

Number	Date	C <sub>in-PM2.5</sub> ( $\mu\text{g}/\text{m}^3$ )	C <sub>out-PM2.5</sub> ( $\mu\text{g}/\text{m}^3$ )	Number	Date	C <sub>in-PM2.5</sub> ( $\mu\text{g}/\text{m}^3$ )	C <sub>out-PM2.5</sub> ( $\mu\text{g}/\text{m}^3$ )
1	10–11	73.16	90.31	15	1–9	66.42	43.35
2	10–12	101.88	190.12	16	1–11	164.88	66.34
3	10–14	208.72	365.21	17	1–14	154.74	88.98
4	10–15	206.45	242.46	18	1–15	118.18	54.78
5	10–16	71.32	86.43	19	1–17	52.27	121.90
6	10–24	59.21	91.06	20	3–2	54.97	112.34
7	10–25	67.32	84.40	21	3–5	20.98	66.91
8	10–26	46.14	48.97	22	3–8	32.37	47.97
9	10–29	59.41	81.17	23	3–10	47.53	123.84
10	10–30	53.60	32.44	24	3–12	66.31	114.33
11	11–2	150.96	186.20	25	3–15	154.72	183.83
12	11–6	60.50	67.73	26	3–17	161.16	174.87
13	11–8	116.00	147.09	27	3–20	173.24	196.06
14	1–7	257.64	321.88				

**Table 4.** Mass concentrations of total PAHs during sampling period.

Number	Date	C <sub>in-PAHs</sub> ( $\text{ng}/\text{m}^3$ )	C <sub>out-PAHs</sub> ( $\text{ng}/\text{m}^3$ )	Number	Date	C <sub>in-PAHs</sub> ( $\text{ng}/\text{m}^3$ )	C <sub>out-PAHs</sub> ( $\text{ng}/\text{m}^3$ )
1	10–11	11.22	11.89	15	1–9	23.04	22.09
2	10–12	14.03	15.31	16	1–11	38.06	38.15
3	10–14	20.98	15.10	17	1–14	71.47	81.10
4	10–15	25.00	27.44	18	1–15	41.88	34.13
5	10–16	10.94	11.11	19	1–17	64.73	85.67
6	10–24	23.72	24.58	20	3–2	27.06	24.79
7	10–25	18.38	23.69	21	3–5	7.86	20.88
8	10–26	19.15	21.34	22	3–8	8.05	6.07
9	10–29	31.63	41.52	23	3–10	36.65	50.58
10	10–30	11.92	5.72	24	3–12	6.05	6.54
11	11–2	69.28	80.22	25	3–15	41.27	54.23
12	11–6	29.80	32.77	26	3–17	21.91	22.14
13	11–8	83.03	108.41	27	3–20	25.54	23.65
14	1–7	23.00	21.78				

The blue dotted line in Figure 1 shows the ratio of indoor PM<sub>2.5</sub> or PAH concentration to outdoors (I/O ratio) [34]. The I/O ratio directly represents the relationship between indoor and outdoor concentrations, which is easy to understand and widely used [35].

Air continues to circulate from the outdoor environment to indoors through voids in windows and doors, and human activities also cause air circulation, both affecting indoor air quality [21]. In our study, the I/O ratio was used to describe the relationship between indoor and outdoor pollution states. Most I/O ratios of PM<sub>2.5</sub> and PAHs were lower than 1; the average values were 0.82 and 0.98, respectively.

Investigations into the I/O ratio in different environments have been conducted in previous papers. For example, Zhou et al. [21] reported that the I/O ratio of residential houses in Tianjin varied from 0.67 to 0.89 with a temporal delay in the outdoor PM<sub>2.5</sub> concentration. Chen et al. [35] reviewed 77 studies including 4000 homes and found that indoor particle emission rates, building structures, and air exchange rates had considerable impacts on the I/O ratio. As for PAHs in the school environment, few studies have been carried out. Wang et al. [36] reviewed over 16 studies from different areas, and found that the I/O ratio of PM<sub>2.5</sub> ranged from 0.50 to 0.95, and the I/O ratio of PAH ranged from 0.43 to 0.93. According to Long and Sarnat [37], if the I/O ratio was larger than 1.15, an indoor source would be present. From this point, in this study, the main sources of PM<sub>2.5</sub> come from outdoors,

for example, vehicle exhausts, biomass combustion, etc. Some results showed PAH I/O ratios higher than 1, indicating that there might be indoor PAH sources.

Descriptive statistics on the analyzed 12 PAH species both indoor and outdoor are presented in Figure 2. According to correlation analysis by SPSS, there was a statistically strong positive correlation between indoor and outdoor PAH species sequence, with a correlation coefficient of 0.86 ( $p < 0.01$ ), indicating the significant impacts of outdoor PAHs on indoors.

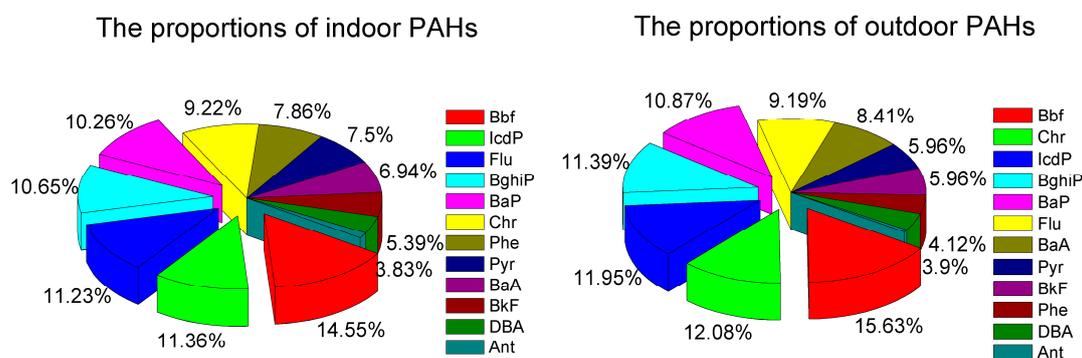


Figure 2. The proportion of PAHs in indoor/outdoor air.

The most abundant species (top five) observed in indoor and outdoor environment during sampling period were quite similar: Bbf ( $4.34 \text{ ng/m}^3$ ), IcdP ( $3.39 \text{ ng/m}^3$ ), Flu ( $3.35 \text{ ng/m}^3$ ), BghiP ( $3.18 \text{ ng/m}^3$ ) and BaP ( $3.06 \text{ ng/m}^3$ ) for the indoor site, accounting for 58.1% of total PAHs, and BbF ( $5.27 \text{ ng/m}^3$ ), Chr ( $4.08 \text{ ng/m}^3$ ), IcdP ( $4.03 \text{ ng/m}^3$ ), BghiP ( $3.84 \text{ ng/m}^3$ ) and BaP ( $3.67 \text{ ng/m}^3$ ) for the outdoor site, accounting for 61.9% of total PAHs.

Because of the high carcinogenicity and atmosphere concentration of BaP, the risk assessment of inhalational PAHs is usually based on BaP environment level. In most of European countries, the target annual average concentration of BaP in inhalational PAHs is set to  $0.70\text{--}1.30 \text{ ng/m}^3$  [38].

This observation in the classroom showed the indoor average BaP level during the whole sampling period was  $3.06 \text{ ng/m}^3$  ( $0.25\text{--}13.36 \text{ ng/m}^3$ ). The mean level of BaP in the classroom in autumn was  $2.88 \text{ ng/m}^3$ , in winter was  $4.75 \text{ ng/m}^3$ , and in spring was  $2.09 \text{ ng/m}^3$ , suggesting that the BaP in PM<sub>2.5</sub> in the classroom posed risks to health during the three sampling seasons.

Both indoor and outdoor PAH concentrations with different molecular weights showed consistent concentration trends of HMW > MMW > LMW. MMW- and HMW-PAHs made up 79.7% of the total PAHs for indoor sites and 86.1% for outdoor sites. As several studies reported, this phenomenon may be due to the fact that PAHs with different molecular weights exist in different phases [39]. Over 90% of LMW-PAHs were in the gas phase because of their higher vapor pressure and higher volatility. Meanwhile, over 90% of HMW-PAHs were in the particle phase because of their lower vapor pressure. However, the concentration of different PAH species in the indoor and outdoor environments varied with different molecular weights. In the indoor sites, LMW-PAH species occurred at higher concentrations than in the outdoors, which was consistent with previous research results [40–44].

Although indoor PAHs were mostly from outdoors as previously found, this result suggests the potential for indoor LMW-PAH sources as well [30]. The major possible sources in the school classroom environments are human activity, classroom furnishings, painting and the use of electronic equipment, such as PCs or projectors [45,46].

In addition, because of high volatility, most LMW-PAHs exist in the gas phase. In the outdoor environment, they are more likely to diffuse with air flow. However, despite the volatilization of LMW-PAHs in the indoor environment, poor air mobility of the indoor environment decreases their diffusion, which leads to an accumulation of LMW-PAHs. On the contrary, all of HMW-PAH species in outdoors exceeded indoors concentration, which could be attributed to the fact that HMW-PAHs are mainly from automobile exhausts [47,48].

Obvious seasonal variations of PAH species are presented in pie charts in Figure 3. In general, for both indoors and outdoors, the highest concentrations occurred in winter, with lower levels in autumn, and the lowest in spring.

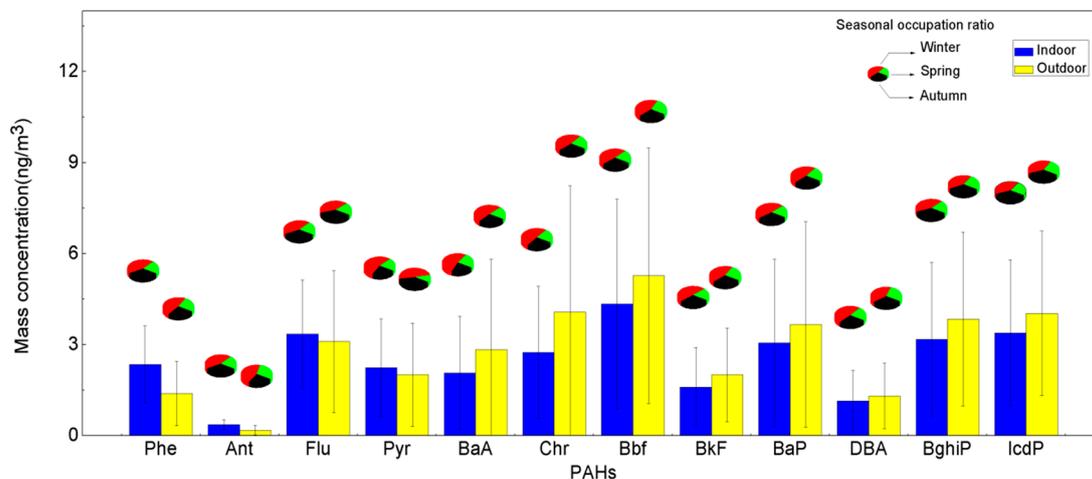


Figure 3. Seasonal variations of PAHs in indoor/outdoor air.

This was consistent with the results reported by most of the previous studies [49–51]. Increasing concentrations of particulate PAHs in the winter could be attributed to several factors. On the one hand, meteorological conditions in winter such as low ambient temperature and solar radiation result in a relatively larger occupation of PAHs partitioned to the particle phase [52,53]. On the other hand, coal combustion and biomass burning for heating increase significantly during winter, and both are major contributors of PM<sub>2.5</sub>-bound PAHs [54].

### 3.2. Source Apportionment

Based on our determination that outdoor pollution is a major source PAHs in classrooms, the next step was to analyze the PAH sources.

The diagnostic ratio method was used in this study for preliminary identification of possible indoor PAH sources, including IcdP/(IcdP + BghiP), BaA/(BaA + Chr), Flu/(Flu + Pyr) and BaP/BghiP. Figure 4a,b shows the scatter plot for BaP/BghiP vs. Flu/(Flu + Pyr) and IcdP/(IcdP + BghiP) vs. BaA/(BaA + Chr) with different seasons, respectively. As shown in Figure 4, for indoor PAHs, automobile exhausts, fossil fuel combustion and biomass combustion source were the major contributors. The BaP/BghiP ratios of all samples were higher than 0.6, suggesting a significant vehicle atmospheric environment, and, especially in autumn, petroleum played a domination role according to the BaP/BghiP ratio [55,56]. The Flu/(Flu + Pyr) ratio can further identify diesel vehicle sources (0.6–0.7), indicating a mixed-traffic source of indoor PAHs. The ratios of BaA/(BaA + Chr) in most samples higher than 0.35 suggest that the combustion is also a primary source [30]. The value of IcdP/(IcdP + BghiP) lower than 0.5 was regarded as an indicator of a petroleum source, while exceed 0.5 was distinguished as a coal, grass and wood combustion source [57]. In addition, ratios of Flu/(Flu + Pyr) around 0.5 suggested a liquid fossil fuel source, while Flu/(Flu + Pyr) > 0.5 was the mark of coal and biomass combustion [58,59].

For further investigation of PAH source apportionment and contribution, the PMF model analysis was conducted, and the primary sources' contributions to the total indoor PAHs are as shown in Figure 5. There were five identified PAH sources.

Factor 1 was highly loaded on Pyr, which is the typical tracers of coal combustion [60], thus, it was interpreted as a coal combustion emission source. Factor 2 was mainly defined by HMW-PAHs, such as BaP, BkF, BghiP, and IcdP, which are related to vehicle emissions [57]. Therefore, factor 2 was identified as a traffic source, which was mainly automobile exhausts. Factor 3 was mainly LWM-PAHs, such as Phe and Ant, which indicate an indoor source of electronic devices [61]; thus, factor 3 is an indoor

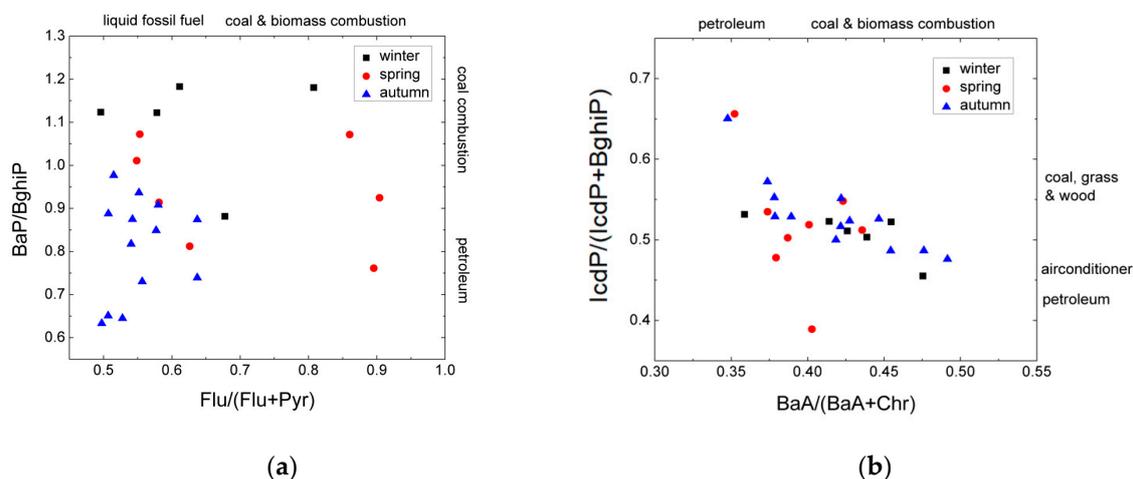
source. Factor 4 was mainly BaA, Chr, BbF, BkF and BaP, i.e., PAHs that have a close relationship with family cooking and natural gas combustion, which is commonly used in the Chinese kitchen [62]. Thus, factor 4 was a cooking source of mainly natural gas combustion and biomass combustion. Factor 5 was mainly Flu and Pyr; these two PAHs are connected with industrial exhausts and, thus, factor 5 is an industrial source [63,64].

Furthermore, as shown in Figure 6, different sources contribute unequally to the sample site, as coal combustion contributed the most and the indoor source the least.

### 3.3. Health Risk Assessment

The concentration of PAHs in the indoor environment was beyond the safe level. Based on the method and simulations introduced in Section 2.4, the mean  $\Sigma[\text{BaP}]_{\text{eq}}$  in the classroom for the whole sampling period was  $10.27 \text{ ng/m}^3$ , higher than the standard value  $1 \text{ ng/m}^3$  provided by the European Commission [65]. Health risk threats of classroom atmosphere are becoming a huge concern to adolescents. The high level of  $\text{BaP}_{\text{eq}}$  found in this study may be due to the high concentrations of PAHs that have a relatively heavier molecular weight (especially 5 ring PAHs) and are mainly attributed to outdoor sources, including automobile exhausts and coal combustion.

Due to physiological differences, individuals varied in body weight, respiration rate, etc., leading to uncertainty in the assessment of health risks. In our study, Monte Carlo simulations were performed to make the estimation more accurate. Parameters of individual body weight and respiration rate were set as lognormal distribution using Monte Carlo simulation, and 100,000 performances were carried out with Crystal Ball software. Figure 7 shows the results. The mean value of ILCR was estimated to be  $1.49 \times 10^{-6}$ , and highly sensitive individuals (10% of total) have a cancer risk of  $2.47 \times 10^{-6}$ . It is stated in USEPA (2009) standards that less than 1 in 1 million ( $10^{-6}$ ) is an acceptable value, while over 1 in 10,000 ( $10^{-4}$ ) is a serious risk level. In our research, more than half of adolescents faced an indoor health risk exceeding the acceptable risk level ( $10^{-6}$ ), but below the serious risk level. The classroom atmospheric environment has potential health risks for most middle school students.



**Figure 4.** Results of diagnostic ratio method. (a)  $\text{BaP}/\text{BghiP}$  and  $\text{Flu}/(\text{Flu} + \text{Pyr})$  (b)  $\text{IcdP}/(\text{IcdP} + \text{BghiP})$  and  $\text{BaA}/(\text{BaA} + \text{Chr})$ .



Figure 5. Results of positive matrix factorization.

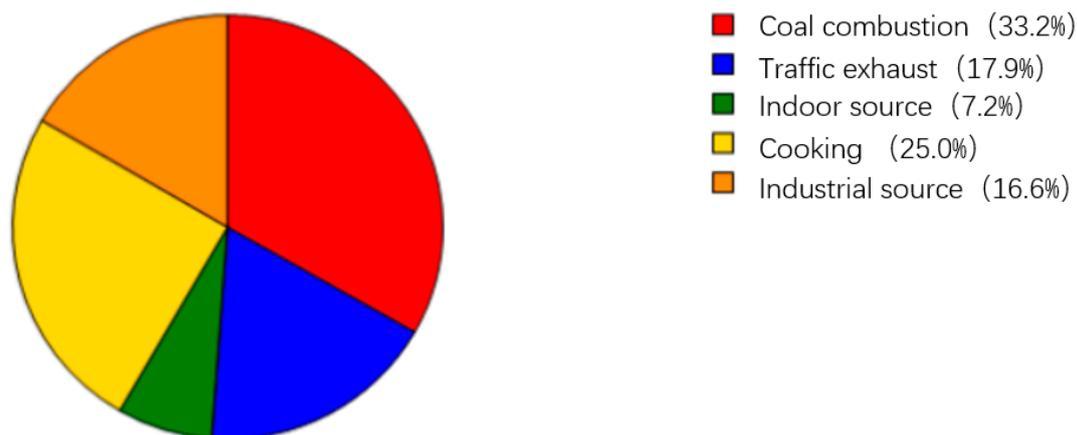


Figure 6. PAH contribution of different sources.

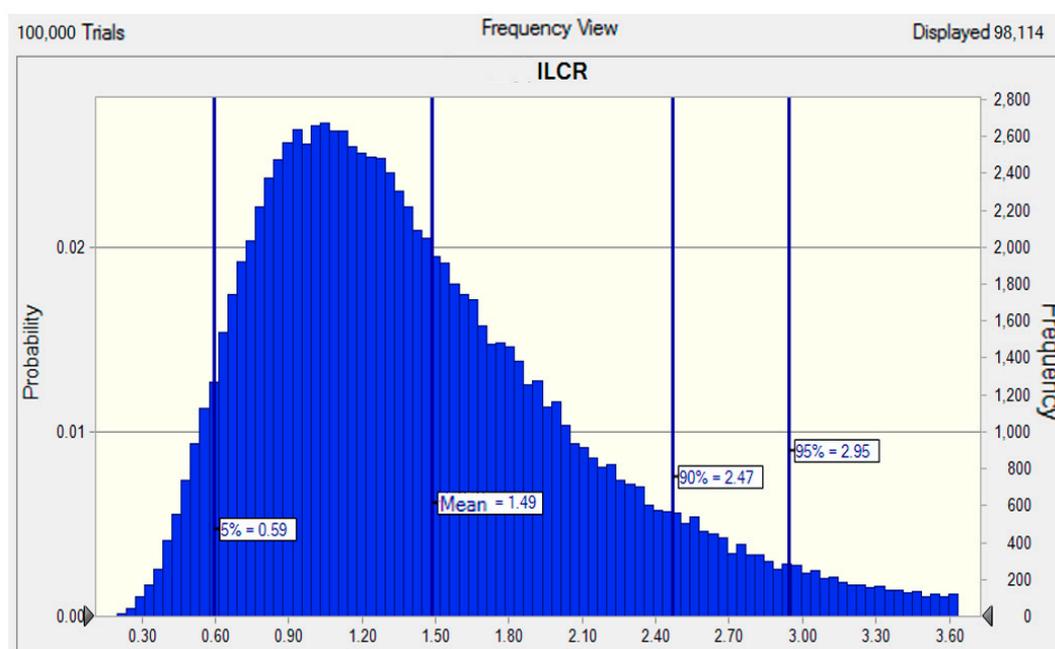


Figure 7. Probability distributions of incremental lifetime cancer risks (ILCR).

There are previous studies on indoor PAH exposure risk. The average excess total inhalation cancer risk for nine communities in Xi'an was  $7.6 \times 10^{-6}$  [20]. The average inhalation ILCRs in Taiyuan was  $1.76 \times 10^{-6}$  for children and  $7.95 \times 10^{-6}$  for adults, respectively [66]. As for worldwide research, few studies have been carried out on the school environment. Reported carcinogenic risks associated with exposure to PAHs can only be found in Asia and Europe [67]. The risks were higher at Asian schools than at European schools ( $1.3 \times 10^{-6}$ ~ $5.4 \times 10^{-5}$  indoors and  $1.3 \times 10^{-6}$ ~ $6.1 \times 10^{-5}$  outdoors in Asia versus  $5.9 \times 10^{-9}$  to  $1.1 \times 10^{-8}$  indoors and  $1.1 \times 10^{-8}$  to  $2.2 \times 10^{-8}$  outdoors in Europe).

To predict the risk trend, this study used environment monitoring data to conduct further predictions. As above-mentioned, the PM<sub>2.5</sub> concentration level showed a linear relationship with this ILCR risk, which we use as a black box model instead. This means that ILCR risk in this study shows the same variation pattern as PM<sub>2.5</sub> concentration. We found that the monitoring data of Xicheng district had a good correlation with the sampling data that we collected,  $R = 0.785$ . Then, we found that the PM<sub>2.5</sub> monitoring level has an approximately linear changing trend in recent years, as  $y = -8.5321x + 17272$ ,  $R^2 = 0.9546$ . In this prediction equation,  $x$  is the year, and  $y$  is the PM<sub>2.5</sub>

monitoring level. Thus, using the above data, we assume that the ILCR risk can be reduced to the acceptable level in the year 2021 or 2022. Further verification should be carried out if possible.

#### 4. Conclusions and Discussion

Quantitative analysis of PM<sub>2.5</sub>-bound PAHs in classrooms was carried out in this study to broaden our understanding of atmospheric pollution in enclosed environments. Our results suggest that the classroom is a not-so-safe environment to adolescents. During the sampling period, the average concentration of PM<sub>2.5</sub> in the classroom was 103.7  $\mu\text{g}/\text{m}^3$  (range: 20.9–257.6  $\mu\text{g}/\text{m}^3$ ), and the average value of the indoor PAH concentration was 29.8  $\text{ng}/\text{m}^3$  (range: 8.0–83.0  $\text{ng}/\text{m}^3$ ), suggesting poor air quality in the classroom environment. Notably, it deteriorated in the winter season with higher PM<sub>2.5</sub> concentration indoors. A statistically significant correlation between indoor and outdoor PM<sub>2.5</sub>-bound PAHs indicated that outdoor pollution was a primary source of indoor air pollutants. As PMF results showed, coal combustion, residential cooking and traffic exhausts were the primary sources of indoor PAHs. Due to heavy transportation and the special resource utilization mode in Beijing, these outdoor sources exerted a considerable influence on indoor atmosphere. Thus, to improve indoor air quality measures such as energy conservation, implementation of green energy, appropriate vehicle control policies and effective emission controls are needed.

This study also proposed estimation of students' health risk caused by indoor PAHs, and potential cancer risks to students were found according to ILCR results. As cancer risk assessment considers the lifetime period, exposure characteristics may vary among individuals. For example, students' behaviors could lead to different results, or less pollutants would be accumulated if the classroom was regularly cleaned.

There are several potential complete pathways for exposure to PAHs, including inhalation, skin exposure and dietary intake. Since it is impossible to carry out measurement of the total amount of PAHs in the classroom and estimate exposure via all pathways, we could only focus on the inhalation risk. However, exposure by the other pathways is likely minor when compared to inhalation. Thus, the health risk we calculated above is relatively reasonable and can be an instruction index for public health management.

Although we applied the ILCR model to this study, some limitations of the ILCR model can cause uncertainty. For example, PAH concentration is a constant value in the model, and as PAHs' adverse health effects are not fully understood, the calculation factors are not completely accurate. In addition, our study used only one monitoring point in the classroom and outdoors, and this may lead to uncertainty in the final result. Research in more schools in other areas of Beijing would help summarize the pollution profile in the whole city of Beijing. Furthermore, due to the limited conditions, we did not obtain abundant samples to conduct the analysis. A long-term continuous sampling would make the dataset more reliable and provide more convincing results.

However, despite these uncertainties, great merits of this study should not be neglected. The results reveal some potential risks that are often likely to be neglected. Although classrooms appear to be relatively safe and clean environments, there are also health threats to individuals, and some insurance actions have to be taken to ensure that the quality of indoor air is maintained to an acceptable standard.

This work can improve knowledge regarding studies on health risk assessments of PAH inhalation exposure, filling information gaps regarding enclosed environments in China. The results of this study are also expected to provide useful information to support the policy-making of pollution control management. For example, limiting the number of cars in Beijing in more effective ways, banning high-emission cars and recommending electric cars. Meanwhile, students should protect themselves from pollutants, such as by using a face mask when the air quality is low. Schools also should guarantee a safe environment for students and take measures to prevent pollutants from affecting students' health.

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