

Supporting Information

Development of Structure-Property Relationships for Ammonium Transport Through Charged Organogels

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Figure S1. Photos of film as an organogel (left) and punch out (right).

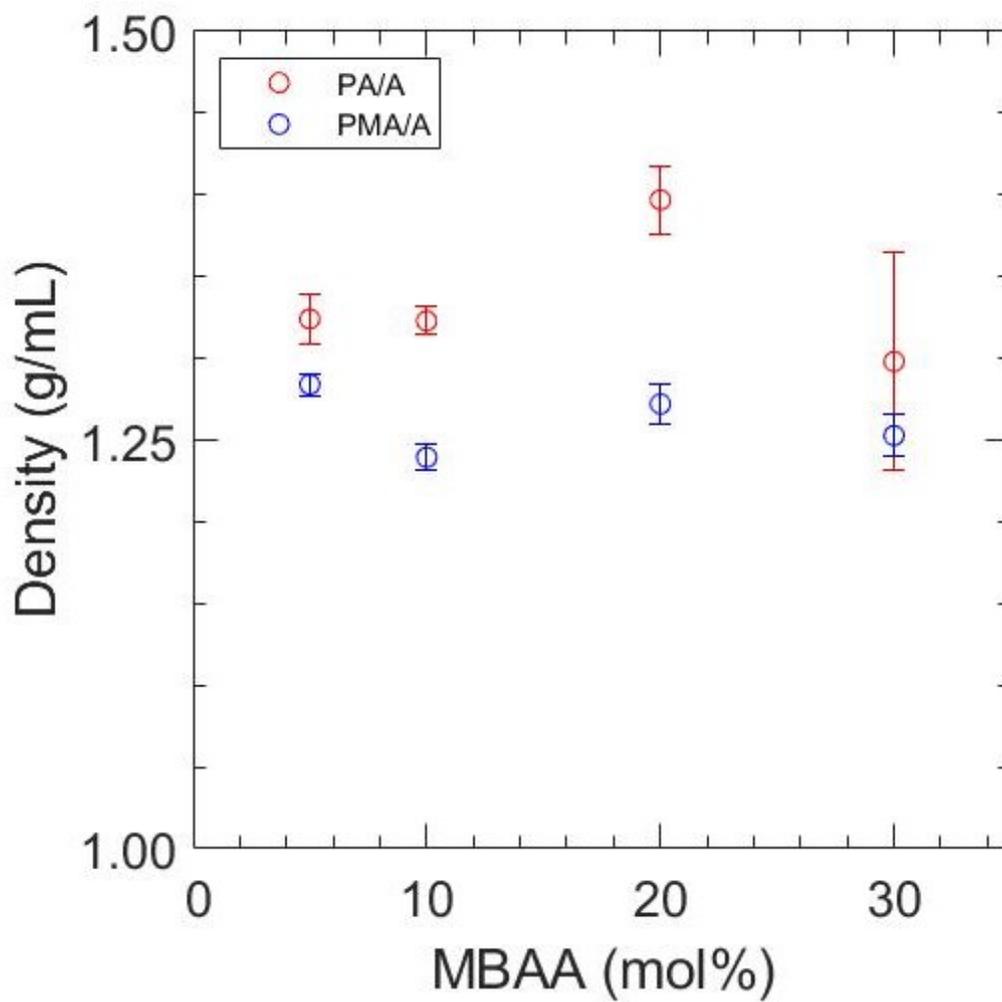


Figure S2. Film Densities for PA (red) and PMA (blue).

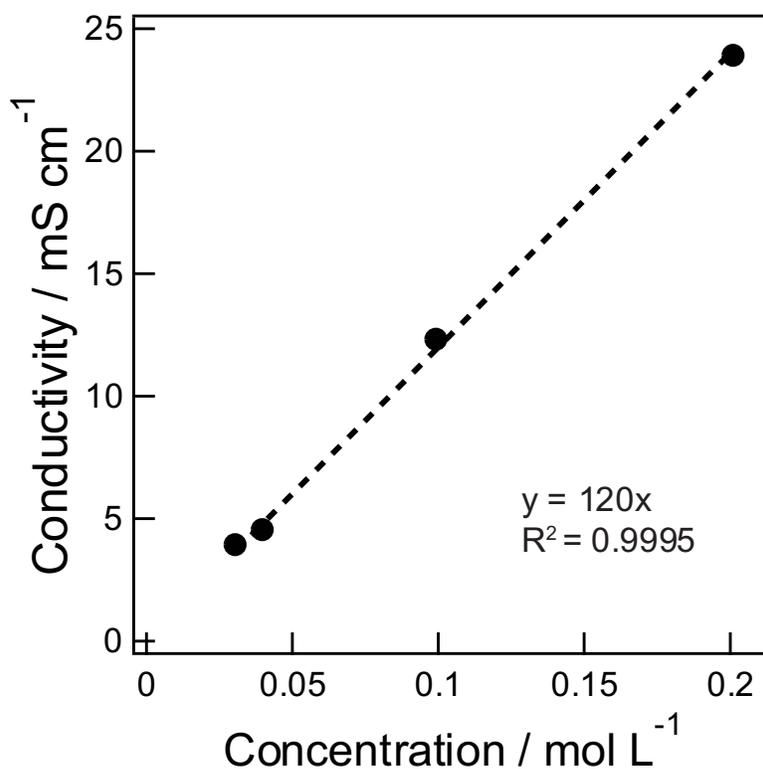


Figure S3. Calibration plot of measured conductivity versus concentration for aqueous ammonium chloride

Table S1. Water uptakes, film densities, and water volume fractions for PA and PMA films. Number in film name indicated MBAA content

Name	Water Uptake (%)	Film Density (g/mL)	Water Volume Fraction
PA-5	186 ± 8	1.32 ± 0.02	0.71 ± 0.01
PA-10	132 ± 5	1.323 ± 0.009	0.637 ± 0.007
PA-20	126.4 ± 0.8	1.40 ± 0.02	0.645 ± 0.003
PA-30	100 ± 8	1.30 ± 0.07	0.56 ± 0.03
PMA-5	95 ± 6	1.284 ± 0.007	0.55 ± 0.01
PMA-10	89 ± 9	1.24 ± 0.01	0.52 ± 0.02
PMA-20	68.9 ± 0.6	1.272 ± 0.008	0.468 ± 0.004
PMA-30	53 ± 2	1.25 ± 0.01	0.401 ± 0.008

Table S2. Ammonium permeability, solubility, and diffusivity for PA and PMA films.

Name	Permeability ($\times 10^{-7}$ cm s $^{-1}$)	Solubility	Diffusivity ($\times 10^{-7}$ cm s $^{-1}$)
PA-5	33 ± 2	0.502 ± 0.006	66 ± 3
PA-10	20.5 ± 0.8	0.477 ± 0.009	43 ± 2

PA-20	14 ± 2	0.40 ± 0.04	35 ± 5
PA-30	3.6 ± 0.1	0.34 ± 0.01	10.5 ± 0.5
PMA-5	12.3 ± 0.4	0.47 ± 0.04	26 ± 2
PMA-10	17 ± 1	0.34 ± 0.03	50 ± 6
PMA-20	6.8 ± 0.2	0.320 ± 0.002	21.1 ± 0.7
PMA-30	4.0 ± 0.4	0.38 ± 0.02	11 ± 1