



Article

Impact of Reaction Parameters and Water Matrices on the Removal of Organic Pollutants by TiO₂/LED and ZnO/LED Heterogeneous Photocatalysis Using 365 and 398 nm Radiation

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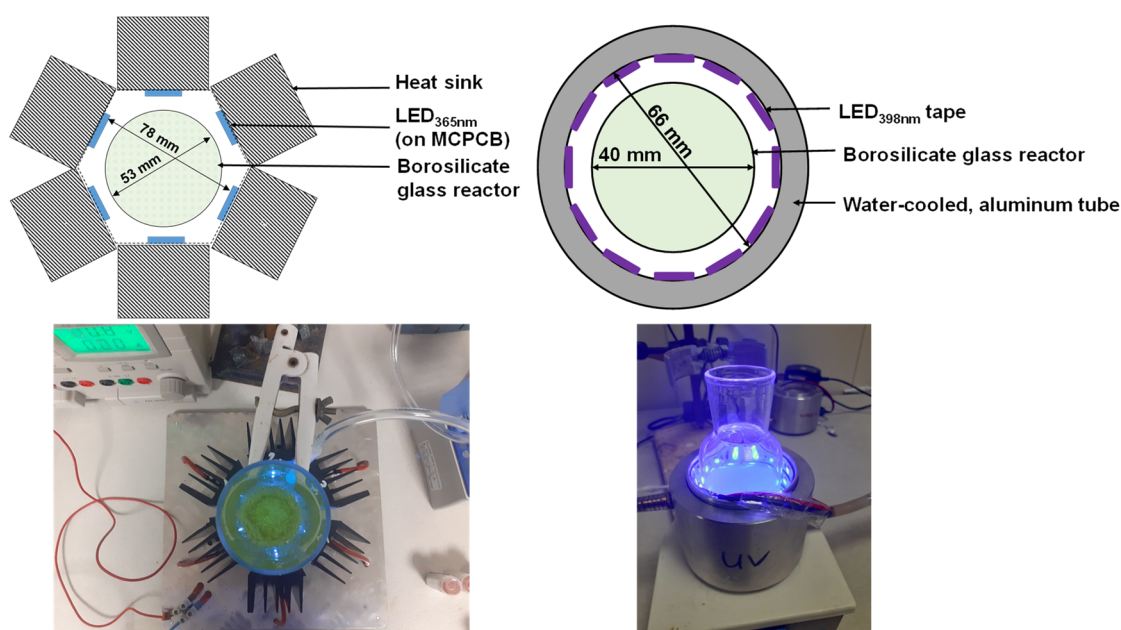


Figure S1. The schematic figures and photos of the photoreactors (left: LED_{365nm}, right: LED_{398nm}).

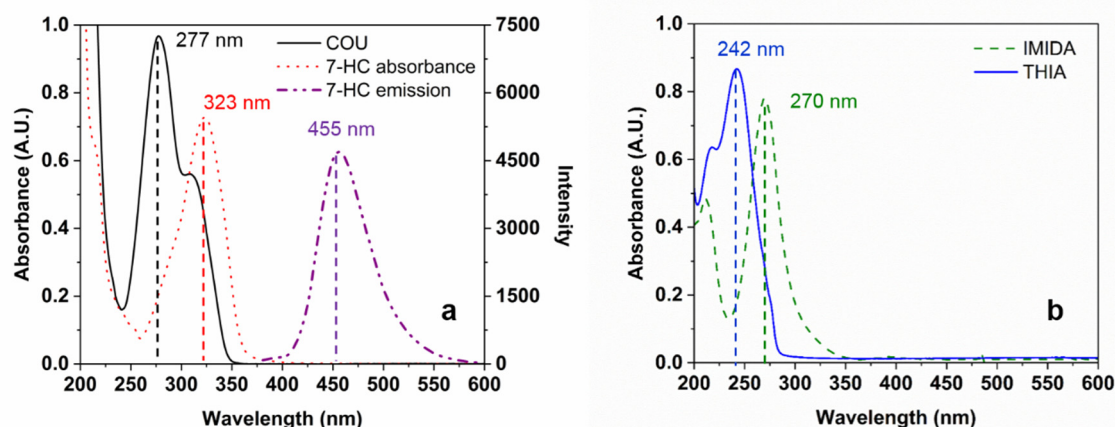


Figure S2. UV-Vis absorption and emission spectra of COU and 7-HC (a), and the UV-Vis absorption spectra of IMIDA and THIA (b).

Table S1. The list of used chemicals, their distributors and purity.

Chemical	Distributor	Purity
coumarin	Sigma Aldrich (St.Louis, USA)	99%
7-hydroxycoumarin	Sigma Aldrich (St.Louis, USA)	99%
imidacloprid	VWR (Radnor, USA)	98%
thiacloprid	Sigma Aldrich (St.Louis, USA)	99%
NaCl	VWR (Radnor, USA)	99%
NaHCO ₃	VWR (Radnor, USA)	99%
HCl	Sigma Aldrich (St.Louis, USA)	98%
NaOH	VWR (Radnor, USA)	99%
Catalase (bovine liver)	Sigma Aldrich (St.Louis, USA)	>55% protein content
Fe ₂ (SO ₄) ₃ × nH ₂ O	VWR (Radnor, USA)	98%
K ₂ C ₂ O ₄	Reanal (Budapest, Hungary)	98%
1,4-phenantroline	Sigma Aldrich (St.Louis, USA)	99%
TiO ₂	Acros Organics (Geel, Belgium)	99.5%
ZnO	Sigma Aldrich (St.Louis, USA)	80%
N ₂	Messer Hungary (Budapest, Hungary)	99.995%
Synthetic air	Messer Hungary (Budapest, Hungary)	Medical grade
MeOH	VWR (Radnor, USA)	99.8%
H ₂ O	Merck-Millipore (Burlington, USA)	ultrapure

Table S2. The parameters of the matrices.

Parameter	Tap Water	Biologically Treated Domestic Wastewater
pH	7.4	7.8
Conductivity ($\mu\text{S cm}^{-1}$)	627	1258
COD (mg dm^{-3})	4.2	24.4
$\text{NH}_4^+\text{-N}$ (mg dm^{-3})	<0.4	<0.4
NO_3^- (mg dm^{-3})	<0.7	3.37
Cl^- (mg dm^{-3})	8.75	120
TOC (mg dm^{-3})	0.79	6.9
Inorganic carbon * (mg dm^{-3})	73.4	103.4

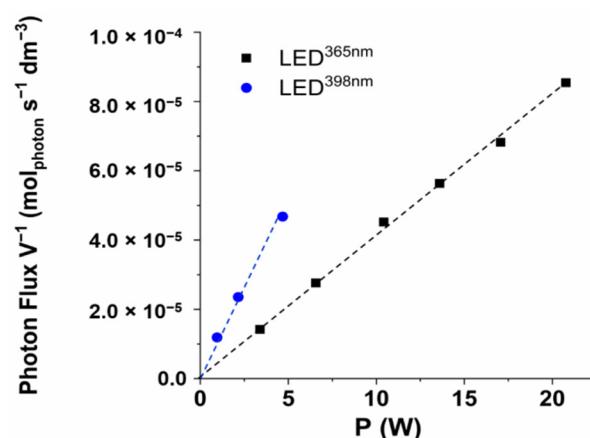
* Dissolved CO_2 , HCO_3^- and CO_3^{2-} .

Figure S3. The photon flux of the LEDs as a function of electric power input.

Table S3. The photon flux of the light sources and the calculated electric efficiencies.

Light Source	Photon Flux ($\text{mol}_{\text{photon}} \text{s}^{-1}$)	P_{electric} (W)	E_{photon}^* (eV)	P_{radiant} (W)	Electric Efficiency $P_{\text{radiant}}/P_{\text{electric}}$
LED ^{365nm}	2.83×10^{-6}	3.39	3.38	0.93	0.27
	5.52×10^{-6}	6.56	3.38	1.80	0.27
	1.13×10^{-5}	13.60	3.38	3.67	0.27
	1.71×10^{-5}	20.77	3.38	5.58	0.27
LED ^{398nm}	1.19×10^{-6}	0.96	3.12	0.36	0.37
	2.35×10^{-6}	2.16	3.12	0.71	0.33
	4.68×10^{-6}	4.68	3.12	1.41	0.30

* Nominal value calculated for λ_{max} .

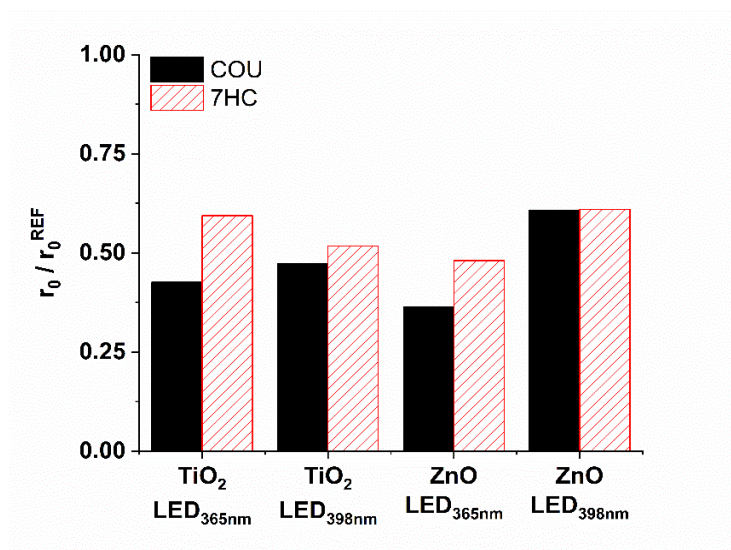


Figure S4. The effect of 5.0×10^{-3} M MeOH on the transformation rate of COU and the formation rate of 7-HC.

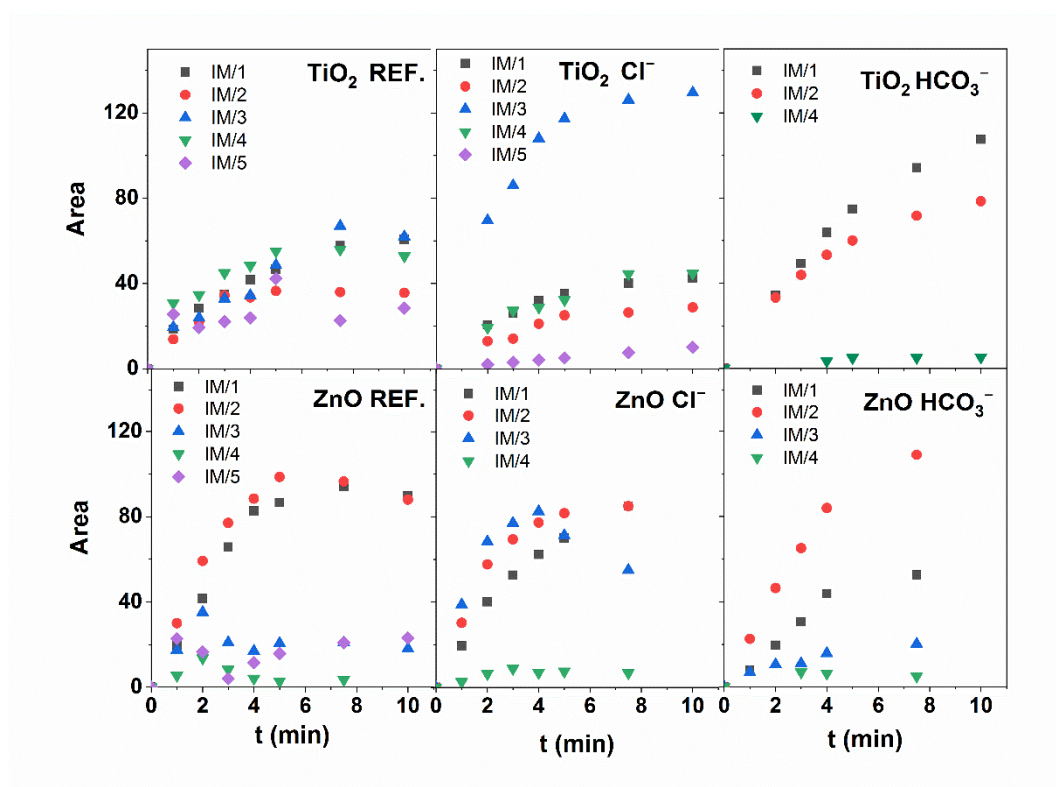


Figure S5. The effect of inorganic ions on the products of IMIDA detected by HPLC-DAD during treatment using LED_{365nm}.

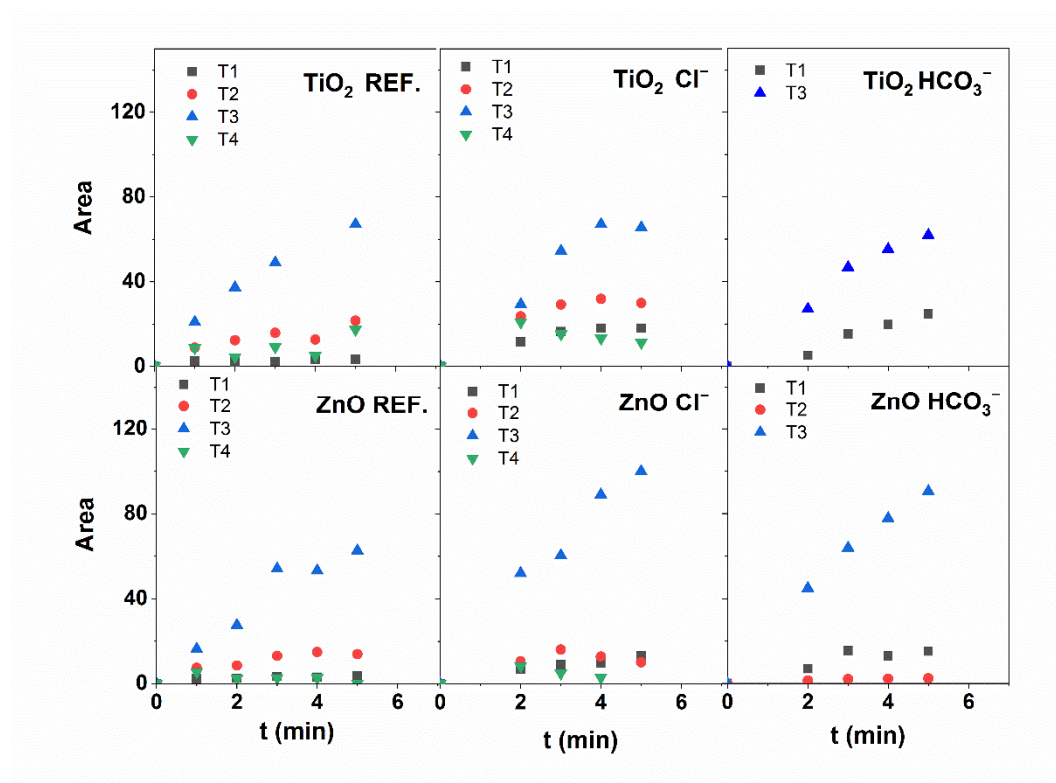


Figure S6. The effect of inorganic ions on the products of THIA detected by HPLC-DAD during treatment using LED_{365nm}.

Table S4. The inorganic standard electrode potentials of the ions and radical ions [105].

Half Reaction	Electrode Potential/V
$O_2(aq) + e^- \rightarrow O_2^{\bullet}$	−0.18
$O_2(aq) + H^+ + e^- \rightarrow HO_2^{\bullet}$	+0.10
$HO_2^{\bullet} + H^+ + e^- \rightarrow H_2O_2$	+1.46
$HO^{\bullet} + e^- + H^+ \rightarrow H_2O$	+2.730
$HO^{\bullet} + e^- \rightarrow OH^-$	+1.902
$Cl^{\bullet} + e^- \rightarrow Cl^-$	+2.432
$Cl_2^{\bullet-} + e^- \rightarrow 2 Cl^-$	+2.126
$HOCl + e^- \rightarrow ClOH^{\bullet-}$	+0.25
$ClOH^{\bullet-} + e^- \rightarrow Cl^- + OH^-$	+1.912
$ClOH^{\bullet-} + e^- + H^+ \rightarrow Cl^- + H_2O$	+2.740
$NO_3^{\bullet} + e^- \rightarrow NO_3^-$	+2.466
$NO_2^{\bullet} + e^- \rightarrow NO_2^-$	+1.04
$CO_2^{\bullet-} + H^+ + e^- \rightarrow CO_3^{2-}$	+1.52
$CO_3^{\bullet-} + e^- \rightarrow HCO_2^-$	+1.57