



Review

Air Ambulance: Antimicrobial Power of Bacterial Volatiles

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Abstract: We are currently facing an antimicrobial resistance crisis, which means that a lot of bacterial pathogens have developed resistance to common antibiotics. Hence, novel and innovative solutions are urgently needed to combat resistant human pathogens. A new source of antimicrobial compounds could be bacterial volatiles. Volatiles are ubiquitous produced, chemically divers and playing essential roles in intra- and interspecies interactions like communication and antimicrobial defense. In the last years, an increasing number of studies showed bioactivities of bacterial volatiles, including antibacterial, antifungal and anti-oomycete activities, indicating bacterial volatiles as an exciting source for novel antimicrobial compounds. In this review we introduce the chemical diversity of bacterial volatiles, their antimicrobial activities and methods for testing this activity. Concluding, we discuss the possibility of using antimicrobial volatiles to antagonize the antimicrobial resistance crisis.

Keywords: volatile organic compounds; volatiles; chemical ecology; metabolomics; antibacterial; antifungal; antibiotics; antimicrobial resistance crisis



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1. Antimicrobial Resistance: A Global Crisis

Many would argue that the discovery of antibiotics changed the world of medicine. Penicillin is often reported as the first antibiotic available to the public, whereas it was actually sulfamidochrysoidine [1,2]. In contrast to penicillin, sulfamidochrysoidine, which was sold under the trade name Prontosil, was toxic for humans and disappeared quickly from the market and history books [1]. However, with the introduction of penicillin in the 1940s, antibiotics have saved millions of lives and the subsequent years are often referred as the "golden age" of antibiotics due to the discovery of numerous novel classes [3]. In fact, in the end of the 1960s around 24 novel classes of antibiotics were introduced to the market [2], but since the 1970s only three classes, namely pseudomonic acids [4,5], oxazolidiones [6] and lipopeptides [7] have been introduced to the market. One of the latest promising antibiotics is the peptide teixobactin, which due to its highly conserved targets is unlikely to induce resistance but is still not available on the market [8]. Interestingly, Alexander Fleming understood the fragility of the powerful tool antibiotic and warned at his Nobel Prize speech in 1945: "The time may come when penicillin can be bought by anyone in the shops. Then, there is the danger that the ignorant man may easily underdose himself and by exposing his microbes to non-lethal quantities of the drug make them resistant" [9]. In his speech, Fleming indicated the risk of antimicrobial resistance—an issue that we are facing today.

How is it possible that, despite Alexander Fleming's warning, resistant pathogens could become a global issue? In human medicine antibiotics are often overused as well as misused. That means that antibiotics are prescribed as a prophylactic or the actual pathogen is not identified before prescription. Furthermore, patients may take antibiotics without referring to a doctor. Additionally, the massive and prophylactic use of antibiotics

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in agriculture causes resistance which may be transferred to humans. Globalization makes the spread of resistant pathogens very easy and the hurdles for (economically driven) pharmaceutical companies are very high. However, these are only the main reasons and have already been reviewed in detail [10]. This misuse of antibiotics has led to the development of numerous antibiotic-resistant pathogens resulting in an antimicrobial resistance crisis on a global scale [10-12]. There are already at least 700,000 deaths caused every year by drug-resistant pathogens globally [13] and scientists predict that by 2050 antibiotic resistance could be responsible for over 10,000,000 deaths per year [14]. Moreover, the global COVID-19 pandemic has caused an increased use of antibiotics due to bacterial co-infections or prophylactic treatment with antibiotics to avoid those co-infections [15]. In order to address this crisis, we need to explore alternatives to classical antibiotics. Numerous pharmaceutical options to counter the antimicrobial resistance crisis are under discussion [16,17]. One approach is to discover novel sources for antimicrobial compounds that can be developed into future treatments. In recent years, more and more studies have reported volatiles with antimicrobial activity which indicates that volatiles might play an important role in countering the antimicrobial resistance crisis [18–21]. For example, the new volatile antibiotics albaflavenone and pentalenolactone produced by Streptomyces coelicolor and Streptomyces avermitilis, respectively were discovered [22].

In this review we introduce the chemical characteristics of volatiles, provide an overview of antimicrobial volatiles of bacterial origin and the main methods used for testing their antimicrobial activity. Concluding, we discuss the potential role of volatiles as a novel class of antimicrobials.

2. Biochemistry of Volatiles: Diverse and Diffusive

Volatiles form a chemical class of molecules that all have one characteristic in common: their high vapor pressure at ambient temperatures [23]. Additionally, volatiles are characterized by their low molecular weights of a maximum of 200–500 Dalton, low boiling points and often lipophilic moieties [23–25]. Volatiles significantly differ from soluble compounds in one key characteristic: they do not depend on solvents. Although many volatiles are nonpolar showing low solubility in water due to their restricted number of functional groups, this solubility is sufficient to allow dissemination into the water phase. Hence, volatiles spread fast in both the gas and water phase [26]. Via the gaseous phase, volatiles can spread in highly complex ecosystems such as soil [27], insect nests [28] and spider nests [29] and can fulfill functions such as communication [30] and antimicrobial defense [31] which cannot be performed by solvents due to the lack of effective spreading.

Volatiles belong to diverse chemical classes such as hydrocarbons, aromates, alcohols, aldehydes, acids, esters, amines and thiols [23]. Bacterial volatiles in particular were for the first time systemically reviewed by Schulz and Dickschat in 2007 describing in detail the biosynthesis of common volatiles like fatty acids or sulfur-compounds which are produced by most bacteria but also rare volatiles like halogenated compounds [32]. Volatile biosynthesis is usually based on pyruvate and therefore takes place in primary metabolism (Figure 1) [24,33]. Pyruvate can be directly metabolized to short acids or alcohols [33]. Bacillus spp. for example, which are well known for producing volatiles as we will discuss later in this review, produces mainly 2,3-butanediol and acetoin via fermentation [34].

Under aerobic conditions pyruvate will be metabolized to acetyl-CoA and can enter the fatty acid anabolism, citric acid cycle or be converted to terpenes. The fatty acid metabolism results mainly in alkanes, alkenes, aliphatic alcohols and ketones. The β -oxidation with acetyl-CoA results only in fatty acids with even numbers, for odd-chain fatty acids propionyl-CoA replaces one acetyl-CoA in the final step [33]. Typical volatile products from the fatty acid pathway are aldehydes such as nonanal, ketones such as nonan-2-one or fatty acids such as nonanoic acid [26].

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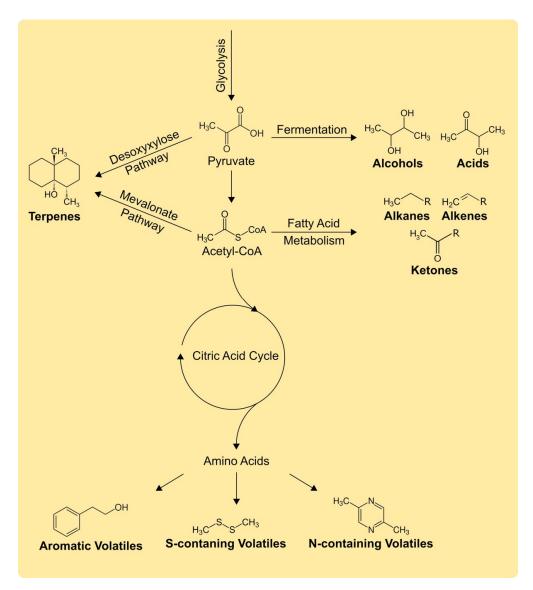


Figure 1. Overview of main biochemical pathways for the production of bacterial volatiles. The chemical structures show representative examples: alcohols (2,3-butanediol), acids (acetoin), alkanes (general structure), alkenes (general structure), ketones (general structure), terpenes (geosmin), aromatic volatiles (2-phenylethanol), S-containing volatiles (dimethyl disulfide) and N-containing volatiles (2,5-dimethylpyrazine). Details are described in the main text.

When acetyl-CoA enters the citric acid cycle it is metabolized to the precursors of most amino acids, which act again as precursors for aromatic, nitrogen-containing and sulfur-containing volatiles [24]. Aromatic volatiles are metabolized based on aromatic amino acids or directly via the shikimate pathway. 2-Phenylethanol for example, a common aromatic volatile compound, can be metabolized based on the amino acid phenylalanine [32,33]. Pyrazines are usually based on amino acids due to their nitrogen-containing aromatic ring. Additionally, sulfur-containing volatiles such as dimethyl disulfide and dimethyl trisulfide which are produced by most bacteria are based on methionine [32].

Another important group of volatiles are terpenes, which are well known to be present in essential oils [35] but in recent years have also been discovered frequently in bacterial volatile blends [36]. Terpenes are synthesized via the mevalonate or desoxyxylulose pathway [26]. The mevalonate pathway starts with acetyl-CoA from the glycolysis and was for a long time assumed to be the only way to biosynthesize isopentenyl diphosphate and dimethylallyl diphosphate, the precursors of terpenes. However, the desoxyxylulose

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pathway starts with pyruvate, the precursor of acetyl-CoA [37,38]. The sesquiterpene geosmin, which is produced by actinomycetes, myxobacteria and cyanobacteria has a characteristic soil-like smell [32,33]. Interestingly, different geosmin synthases were found in actinobacteria compared to myxobacteria and cyanobacteria [36].

Alongside organic volatiles, bacteria produce inorganic volatiles such as hydrogen sulfide, hydrogen cyanide, nitric oxide or ammonia [36]. Ammonia for example is produced in high amounts by *Streptomyces* spp. and is produced within amino acid catabolism [39,40]. Moreover, ammonia was shown to be antimicrobial against Gram-positive and -negative bacteria and can act therefore as a long-distance (several centimeters) antibiotic [39].

3. Bacterial Antimicrobial Volatiles: An Overview

Recently, an increasing number of studies have revealed individual volatiles or volatile blends of bacterial origin with antimicrobial activities [18]. These studies indicate that the antimicrobial potential of volatiles is as diverse as their biochemistry (Table 1). A bulk of the investigated volatiles were reported for their antifungal activities and cause for reduced hyphal extension and/or hyphal biomass as well as spore germination. For example, the volatile blends produced by Paenibacillus polymyxa Sb3-1 and Bacillus velezensis I3 were shown for their antifungal activities [41,42]. However, several studies also reported bacteria that produce antibacterial [43] or anti-oomycete volatiles [44]. Beyond that, some volatiles were reported for their broad antimicrobial spectrum. For example, the volatile 2,5-bis(1-methylethyl)-pyrazine produced by Paenibacillus sp. AD87 revealed a broad-spectrum activity against a range of human and plant pathogens. The volatile inhibited the bacterial pathogens Escherichia coli and Staphylococcus aureus, the fungal pathogens Fusarium culmorum and Rhizoctonia solani as well as the yeast Candida albicans [21,45]. At the same time 2,5-bis(1-methylethyl)-pyrazine showed very low toxicity on mammalian cells [45]. Another example of volatiles with broad spectrum antimicrobial activity is γ -Butyrolactones, active against fungi, yeasts, and bacteria [46].

Table 1. Overview of recent (2017–2021) studies showing the antimicrobial activity of bacterial volatiles. The studies are ordered alphabetically by the volatile producer's name. Only studies that trapped the antimicrobial volatiles in the gas phase and/or showed the antimicrobial effect via the gas phase are listed. blend = The volatile blend might be analyzed in the cited study but only the antimicrobial activity of the blend was tested. f = antifungal, b = antibacterial, o = anti-oomycete.

Volatile Producer	Volatile(s)	Bioactivity	Reference
Bacillus amyloliquefaciens CPA-8	blend 1,3-pentadiene thiophene acetoine	f	[47]
Bacillus amyloliquefaciens DA12	blend	f	[48]
Bacillus amyloliquefaciens FZB42	blend 1,2-benzisothiazol-3(2H)-one benzaldehyde other	b	[49]
Bacillus amyloliquefaciens L3 other	blend 2-heptanone 2-ethyl-1-hexanol 2-nonanone other	f	[50]
Bacillus artrophaeus LSSC22	blend 1,2-benzisothiazol-3(2 <i>H</i>)-one other	b	[49]
Bacillus cereus CHP20	blend	О	[44]
Bacillus megaterium KU143	blend	f	[51]

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Table 1. Cont.

Volatile Producer	Volatile(s)	Bioactivity	Reference
Bacillus pumilus TM-R	blend	f	[52]
Bacillus siamensis LZ88	blend	f	[53]
Bacillus (diverse spp.)	blend	f	[54]
Bacillus (diverse spp.)	blend 2-undecanone benzothiazole 1,3-butadiene N,N-dimethyldodecylamine other	f	[55]
Bacillus sp. BO53	blend	b	[43]
Bacillus sp. D13	blend	b	[56]
Bacillus sp. TM-I-3	blend	f	[57]
Bacillus subtilis CHP14	blend	О	[44]
Bacillus subtilis FA26	blend benzaldehyde nonanal benzothiazole acetophenone	b	[58]
Bacillus subtilis M29	blend 1-butanol acetic acid butyl ester 1-heptylene-4-alcohol 3-methyl-3-hexanol other	f	[59]
Bacillus velezensis BUZ-14	blend diacetyl benzaldehyde isoamyl alcohol other	f	[41]
Bacillus velezensis G341	blend	f	[60]
Bacillus velezensis I3	blend	f	[41]
bacterial community	blend	f	[31]
Cronobacter muytjensii JZ38	blend	О	[61]
Frigoribacterium endophyticum CHP33	blend	О	[44]
Microbacterium testaceum KU313	blend	f	[51]
Paenibacillus sp. AD87	2,5-bis(1-methylethyl)-pyrazine	b, f	[21]
Paenibacillus polymyxa Sb3-1	blend	f	[42]
Proteus mirabilis 04	blend	b	[62]
Pseudoalteromonas sp. GA327	blend	b	[43]
Pseudomonas chlororaphis subsp. aurantiaca KNU17Pc1	blend	f	[63]
Pseudomonas chlororaphis subsp. aureofaciens SPS-41	blend 3-methyl-1-butanol phenylethyl alcohol 2-methyl-1-butanol other	f	[64]
Pseudomonas protegens AS15	blend	f	[51]

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 Table 1. Cont.

Volatile Producer	Volatile(s)	Bioactivity	Reference
Pseudomonas protegens CHA0	dimethyl trisulfide 2-ethylhexanol ammonium hydroxide phenol acetophenone 1,3-diphenylpropane 3-phenylpropiophenone	f	[65]
Pseudomonas putida BP25	blend 2-ethyl-5-methylpyrazine	f	[66]
Pseudomonas putida BP25	2,5-dimethyl pyrazine 2-methyl pyrazine 2-ethyl-5-methyl pyrazine 2-ethyl-3,6-dimethyl pyrazine dimethyl trisulfide	b, f, o	[67]
Pseudomonas stutzeri E25	blend dimethyl disulfide	f	[68]
Sphingobacterium multivorum Bel3-4	blend	f	[54]
Stenotrophomonas maltophilia CR71	blend dimethyl disulfide	f	[68]
Stenotrophomonas maltophilia (TD1 and GH1-5)	blend	f	[54]
Streptomyces alboflavus TD-1	blend anisole dimethyl trisulfide β-pinene benzenamine 1,5-cyclooctadiene	f	[69]
Streptomyces fimicarius BWL-H1	phenylethyl alcohol ethyl phenylacetate methyl anthranilate α-copaene caryophyllene methyl salicylate 4-ethylphenol	o	[70]
Streptomyces lavendulae SPS-33	blend 2-methyl-1-butanol 3-methyl-1-butanol pyridine phenylethyl alcohol other	f	[71]
Streptomyces sp. MBT11	blend	b	[39]
Streptomyces venezuelae (ATCC 15439)	blend	b	[39]
Streptomyces yanglinensis 3–10	blend methyl 2-methylbutyrate 2-phenylethanol β-caryophyllene	f, o f f f	[72]
Xenorhabdus szentirmaii PAM 25	blend	f	[73]

3.1. Sulfur-Containing Volatiles

Often, described antimicrobial volatiles of bacterial origin are alcohols, pyrazines and sulfides (Table 1). Especially sulfur-containing volatiles such as dimethyl sulfide, dimethyl disulfide and dimethyl trisulfide are often reported because they are commonly produced

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by bacteria and apparently have strong antimicrobial activities [26]. For example, dimethyl disulfide has antibacterial potential, as it revealed bacteriostatic effects against the two plant pathogens *Agrobacterium tumefaciens* and *Agrobacterium vitis* [74]. However, the volatile is also known for its antifungal activity. Currently, dimethyl disulfide is used as a novel fumigant (PALADIN®) to target soil-borne plant pathogens [25]. The chemically related volatile dimethyl trisulfide significantly inhibited the growth of three human pathogens *Serratia marcescens, Escherichia coli* and *Staphylococcus aureus* [75]. As well as linear sulfurcontaining compounds, scientists also report from aromatic sulfur-containing compounds. Gotor-Vila reported thiophene in the volatilome of *Bacillus amyloliquefaciens* CPA-8 and showed its antifungal activity [47].

3.2. Bacillus and Streptomyces as Volatile-Producers

Most likely, all bacterial genera produce volatiles, but *Bacillus* species especially are often reported to produce volatiles with antimicrobial potential (Table 1). For example, volatiles emitted by Bacillus amyloliquefaciens FZB42 including benzaldehyde, 1,2-benzisothiazol-3(2H)-one and 1,3-butadiene showed strong inhibitory activities against *Ralstonia solanacearum*, a bacterial plant pathogen causing wilt disease [49]. Other studies report the same from Bacillus amyloliquefaciens strains but with antifungal activities [47,48,50]. Alongside Bacillus, Streptomyces species were also often investigated because the genus is well known for its antimicrobial potential, including volatiles [39,76]. For example, terpenoid volatiles are abundantly emitted by *Streptomyces* species and pose interesting antimicrobial properties. The soil isolate, Streptomyces albidoflavus, was shown to produce a sesquiterpene, namely albaflavenone, with antibacterial properties [77]. Lately, albaflavenone was isolated from other Streptomyces species and fungi [78,79]. Another sesquiterpene compound with antibacterial activity is dihydro- β -agarofuran, produced by *Streptomyces* species [80]. The antimicrobial volatile pentalenolactone emitted by Streptomyces roseogriseus was discovered to possess antibacterial activity against Gram-positive and Gram-negative bacteria. Furthermore, anisole, emitted by Streptomyces albulus, was reported to inhibit the growth of fungal plant pathogens Sclerotinia sclerotiorum and Fusarium oxysporum [81].

3.3. Co-Cultivation and Volatile Blends

Some studies revealed that the co-cultivation of different microbial strains can influence the metabolism of bacteria (Table 2). For example, Paenibacillus sp. AD87 was shown to produce the antimicrobial volatile 2,5-bis(1-methylethyl)-pyrazine when cultivated alone. After co-cultivation of Paenibacillus sp. AD87 together with the phylogenetically different strain Burkholderia sp. AD24, the headspace concentration of 2,5-bis(1-methylethyl)-pyrazine was increased [21]. Due to the growth inhibiting effect of 2,5-bis(1-methylethyl)-pyrazine on Burkholderia sp. AD24, it is likely that the volatile production of Paenibacillus sp. AD87 was increased as a response to the competition. As well as the increased production of the pyrazine, the co-cultivation resulted in a changed gene expression in both bacteria. For example, Burkholderia sp. AD24 showed an increased expression of a type IV secretion system gene which is involved in virulence. Paenibacillus sp. AD87 showed increased expression of genes involved in antibiotic resistance. Another study cultivated five bacterial strains together, among others likewise *Paenibacillus* sp. AD87 and *Burkholderia* sp. AD24 [82]. The analysis of the collective volatile blend of the five bacteria revealed among others 2,5-bis(1-methylethyl)-pyrazine as well. Interestingly, the pyrazine was only found in the collective volatile blend but neither in the volatilome of Paenibacillus sp. AD87 nor in one of the others indicating production activation by the co-cultivation. Rybakova et al. co-cultivated the bacterium Paenibacillus polymyxa Sb3-1 with the fungus Verticillium longisporum EVL43 resulting in several up- and downregulations of the volatile blends of both strains, including volatiles that are most likely related to antimicrobial defense [42]. Furthermore, the corporate production of volatiles was shown under lab conditions [83]. By co-cultivation of Serratia plymuthica 4Rx13 and Staphylococcus delphini without physical contact they produced corporately the volatiles schleiferon A and B. Separately, none of

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the bacteria was able to produce those products. Furthermore, Abis et al. analyzed the relation between microbial diversity and volatile emission in general [84]. Interestingly, they could show that a reduced microbial diversity in soil correlates with an increased volatile emission and a smaller number of released volatiles. They discussed that these findings might be caused by a bacterial volatile absorption. However, it is certain that the microbial diversity and community influence the volatile blend in an ecosystem, even when the detailed relations still remaining unknown.

Table 2. Examples of bacterial volatiles that were upregulated or downregulated in co-cultures. blend = The volatile blend might be analyzed in the cited study but only the antimicrobial activity of the blend was tested. f = antifungal, b = antibacterial, o = anti-oomycete, na = not analyzed.

Co-Culture	Volatile(s)	Bioactivity	Reference
Burkholderia sp. AD24 Paenibacillus sp. AD87	2,5-bis(1-methylethyl)-pyrazine	b, f	[21]
Burkholderia sp. AD024 Paenibacillus sp. AD087 Dyella sp. AD056 Janthinobacterium sp. AD080 Pseudomonas sp. AD021	2,5-bis(1-methylethyl)-pyrazine other	na	[82]
Chryseobacterium sp. AD48 Tsukamurella sp. AD106	blend dimethyl trisulfide other	b, f, o b	[75]
Janthinobacterium sp. AD80 Dyella sp. AD56	blend dimethyl trisulfide other	f, o b	[75]
Paenibacillus polymyxa Sb3-1 Verticillium longisporum EVL43	blend trans-2,2,4,5-tetramethyl-1,3-dioxolane 1-butanol other	f na na	[42]
Serratia plymuthica 4Rx13 Staphylococcus delphini	schleiferon A and B	na	[83]

Mixtures of volatiles may result in increased antimicrobial activities compared to single volatiles. For example, a mix of four monoterpenes (γ -terpinene, 1S- α -pinene, β -pinene and β -myrcene) revealed strong antibacterial activity against the pathogenic bacteria *Escherichia coli* and *Staphylococcus aureus* [85]. However, as single compounds they revealed little or no antimicrobial activity. Furthermore, for a number of fungal and bacterial isolates, antimicrobial activities of their volatile blend are reported, but the compounds responsible for this activity remained unknown (Table 1). It is plausible that not a single volatile but a mix of compounds is responsible for this activity.

3.4. Modes of Action and Abiotic Factors

Although, for many microbial volatiles, powerful antimicrobial activities have been reported, little is known about modes of action of these molecules on the target organisms. Some microbial volatiles can interfere with well-known bacterial chemical communication systems like N-acylhomoserine lactones' (AHLs) quorum-sensing. Bacteria use AHLs' quorum sensing to regulate certain phenotype expressions, such as biofilm formation, virulence factor expression, motility and others. Many lactones (10-methylundec-2-en-4-olide, 10-methylundec-2-en-3-olide, 10-methyldodecan-5-olide, others) positively or negatively influenced the quorum-sensing bacterial communication. This influence could be due to the structural similarity between lactones and AHLs. However, other classes of volatiles such as dimethyl disulfide could also impact bacterial quorum sensing communication by significantly suppressing the transcription of AHLs synthase genes [86]. The above discussed 2,5-bis(1-methylethyl)-pyrazine resulted in more direct

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damages which depend interestingly on the concentration. At high levels, the volatile resulted in DNA damage, whereas at low levels cell-wall damages were observed [45]. Likewise, another study investigating the volatile blend activity of a *Streptomyces* species against the oomycete plant pathogen *Peronophythora litchi* showed, among others, cell-wall damages [70]. A study investigating the antimicrobial activity of the monoterpenes linally acetate, menthol and thymol indicates that those volatiles modify the membrane permeability which causes leakage of intracellular material [87]. Furthermore, it is likely that the altered membrane allows the volatiles to enter the cells and might cause further damages. Nevertheless, detailed information about the modes of actions of antimicrobial volatiles are lacking yet and need further investigation. Moreover, detailed information about volatile concentrations are mostly lacking which makes clear statements about the ecological relevance of often challenging volatiles [88].

Additionally, the emission of microbial volatiles is influenced by various abiotic factors such as nutrient availability, temperature and pH. For example, a nutrient-poor growing condition triggered higher levels of terpene emission at an early growth stage of the fungal isolate *Fusarium culmorum* [82]. It is plausible that such compounds have strong antimicrobial activity and are important for the producing organisms to survive under competitive interactions. Another study showed changes in antifungal activity of the volatile blend of *Bacillus amyloliquefaciens* CPA-8 when cultivated on different media [47]. Similarly, the production of antifungal volatiles by the mycophagous soil bacterium *Collimonas* was strongly influenced by different nutrient conditions [89].

4. Antimicrobial Activity of Volatiles: The Testing Methods

The approaches to test the antimicrobial activity of volatiles can be divided into two main categories: indirect and direct (Figure 2). In indirect approaches the volatiles need to diffuse through the gas phase (usually air) to influence the test organisms. In contrast, direct approaches allow the volatiles to make contact with the test organisms via the liquid phase (usually water) and do not need to diffuse through the gas phase.

One of the most common indirect approaches is the use of two-chamber Petri dishes (Figure 2A) [27,75]. Two-chamber Petri dishes contain a physical wall to divide the inside space into two chambers, making it possible to fill each side with a different media. One side can be inoculated with volatile producing organism or pure volatiles, whereas the other side is inoculated with test organism. After a common incubation, the growth of the test organism is analyzed, e.g., by counting colony forming units or analyzing the growth area. The Petri dishes are commercially available with two, three, or four chambers allowing combination of several organisms or pure compounds with each other. In contrast, the double plate approach is based on standard Petri dishes (Figure 2B) [30]. The bottom and top parts can be filled with different media, whereas one part is inoculated with the volatile producing organisms or pure volatiles and the other part with test organisms. The growth of the test organisms under exposure of the volatiles can be analyzed similarly to the two-chamber Petri dish approach. The vial approach is based on vials filled with solutions containing volatiles (Figure 2C) [90]. The lid contains filter paper with a defined inoculum of test organisms and is exposed to the volatiles before it is incubated in liquid medium to analyze the growth of the test organism. Recently, the AntiBio Vol approach was published (Figure 2D) [91]. Defined biofilms in a 24-well plate were placed upside-down on a second 24-well plate filled with solutions containing volatiles. After common incubation the biofilm is transferred to fresh, liquid broth, incubated shortly and the biomass is analyzed.

In contrast to indirect approaches, direct approaches do not compel the volatiles to actually diffuse through the gas phase, which comes with the advantage of better control of the volatile concentrations but the drawback that those approaches are not suitable with many volatiles due to their frequent lipophilic moiety, which may require the use of organic solvents [23]. The agar diffusion test is widely used to test the antimicrobial activity of pure compounds on solid media (Figure 2E) [46,92]. Cotton discs prepared with pure compounds are placed on agar plates that were inoculated with test organisms. The test

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compounds diffuse through the agar and may result in a zone of inhibition (ZOI) around the cotton disc. The application of different concentrations of test compounds can be used to determine the minimal inhibitory concentration (MIC). However, a more common approach to investigate the MIC is the two-fold dilution approach which is based on liquid media (Figure 2F). For the approach a defined concentration of the test compound in liquid broth is prepared and subsequently several times 1:2 diluted [93]. Often, this approach is performed using 96-well plates to reach high throughputs with little material usage.

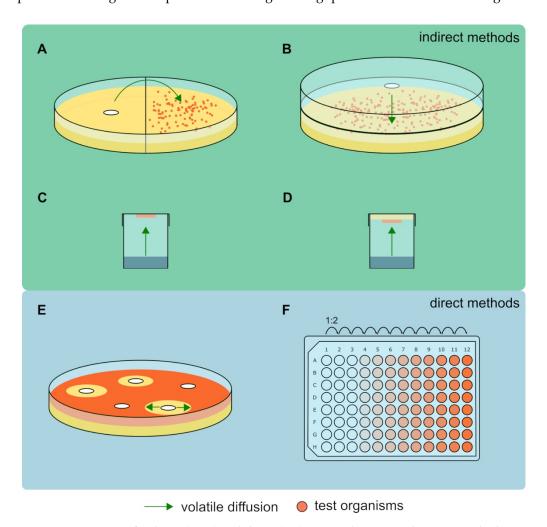


Figure 2. Overview of indirect (**A**–**D**) and direct (**E**,**F**) approaches to test the antimicrobial activity of volatiles. Indirect approaches such as the two-chamber Petri dish (**A**), double plate (**B**), vial (**C**) and AntiBio Vol approach (**D**) compel the volatiles to diffuse through the gas phase to reach the test organisms. In contrast, direct approaches such as the agar diffusion approach (**E**) and the minimal inhibitory test (**F**) allow direct contact between the volatiles and test organisms. Details are described in the main text.

5. Concluding Remarks and Future Perspectives

Recent studies clearly demonstrate the ability of various bacteria to produce antimicrobial volatiles that inhibited the growth of either human or plant pathogens, indicating their antibiotic potential and possible application in agriculture and medicine. The demand for new approaches and compounds is high in both agriculture (due to an EU ban of many chemical pesticides) and healthcare (due to antibiotic resistance and side-effects). As the research of antimicrobial volatiles is a newly developing field, there are still many novel volatile compounds to be discovered and chemically characterized. Individual compounds might be commonly found in many often unrelated strains, while others are restricted only

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to a certain group of strains. Usually, mixtures of compounds are released with widely varying concentrations. Yet, their effects on other organisms and their biosynthesis need to be investigated in more detail in the future. Furthermore, the effects of antimicrobial volatiles need to be evaluated on non-target beneficial (micro)organisms. Many fundamental questions about the modes of action of antimicrobial volatiles and possible resistances remain unanswered and need to be investigated in order to advance our basic knowledge in this research field.

Alongside the treatment of pathogens with volatiles alone, another approach is to combine volatiles with common antibiotics. For example, when exposed to the volatile blend of a *Streptomyces* species, *Bacillus subtilis* showed increased sensitivity against several antibiotics [39]. Interestingly, other studies indicate opposite effects. *Bacillus subtilis* exposed to triethylamine showed reduced sensitivity to tetracycline [94]. However, a selective combination of volatiles and common antibiotics may be a successful tool against resistant bacteria in future.

Volatiles may be also used for fast and reliable detection of pathogen infections as pathogens emit volatiles as well. For example, by analyzing the volatile profile of plant pathogenic fungi and oomycetes, we revealed that each isolate emits a specific blend of volatiles [24]. Another study investigated the breath volatile profile of swine infected with Influenza A, resulting in volatiles that could be related to the infection [95]. Other studies worked with human cells lines (lung epithelium) and combined those with *Pseudomonas aeruginosa* causing ventilator-associated pneumonia showing likewise volatiles that could potentially be used as biomarkers [96,97]. *Staphylococcus aureus* is a common bacterium infecting children with cystic fibrosis and a recent study detected this pathogen in cystic fibrosis patients using breath volatile profiles [98].

In agriculture as well volatiles can provide information of early pathogen infractions. Volatiles emitted during the infection of apple plants by bacterial pathogens *Erwinia amylovora* or *Pseudomonas syringae* pv. *syringae* were studied by gas chromatography-mass spectrometry and proton transfer reaction-mass spectrometry. Infected plants showed a disease-specific emission of volatiles, including several bioactive compounds, such as hexenal isomers and 2,3-butanediol [99]. Those approaches are non-invasive, fast, reliable and have the potential to avoid the prophylactic and wrongly use of antibiotics.

Finally, the potential use of volatiles as antimicrobials is often criticized because of their physicochemical properties. In fact, numerous volatiles are liquid at room temperature if not solid and could therefore be solved in appropriate solvents. Furthermore, Avalos suggested the inhaling of volatiles as possible therapy in future [18]. We are convinced that innovative and novel approaches are needed, and antimicrobial volatiles could be a future solution.

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