



## Supplementary Materials **The effect of buffers on weak acid uptake by vesicles**

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**Figure S1.** pH titration of carboxyfluorescein (CF) at 22°C and 4°C. 1  $\mu$ M CF in a 95 mM KCl solution buffered with 5 mM beta alanine, 5mM MES, 5 mM TRIS and 5 mM CAPS. The samples are titrated by HCl addition. The added volumes produce negligible dilution. Emission spectra (slit-width 2.5 nm) are recorded on a fluorescence spectrophotometer with temperature control (HITACHI F2700; Tokyo, Japan). The excitation wavelength is equal to 480 nm (slit-width 2.5 nm). Integration of the spectra from 515 nm to 650 nm mimics the detection of the stopped-flow device that is equipped with a 515 nm longpass filter. The intensities are plotted against the respective pH for 22°C (red) and 4°C (blue). The solid lines represent fits of a function of the form  $(pH) = I_0 + \Delta I/(1 + 10^{pK_{CF}-pH})$  to the data.  $I_0$  and  $\Delta I$  are device dependent with  $pK_{CF}$  being equal to 6.45 (22°C) or 6.54 (4°C). That is, the temperature induced shift of  $dpK_{CF}$ .amounts to -0.005 K<sup>-1</sup>.



**Figure S2.** Proton permeability  $P_{H^+}$  is too small to significantly affect intravesicular acidification. Two fits of *I*<sub>theor</sub> (black and gay lines) to every *I*<sub>exp</sub> (colored lines) in the presence of 10 mM MES (Figure 4, middle panel) are displayed. The black line is obtained by assuming  $P_{H^+} = 3.5 \cdot 10^{-5}$  cm/s. The gray line dashed line assumes  $P_{H^+} = 3.5 \cdot 10^{-4}$  cm/s.



**Figure S3.** Parameter sensitivity to variation of  $P_{H^+}$ . Results of the numerical calculations performed for Figure S2 are shown: solid lines for  $P_{H^+} = 3.5 \cdot 10^{-5}$  cm/s 3; dashed lines for  $P_{H^+} = 3.5 \cdot 10^{-4}$  cm/s.



**Figure S4. Temperature dependence of acid uptake into DOPC vesicles.** Formic acid influx induced intravesicular acidification depends on temperature as indicated by CF fluorescence intensity. The traces have been used for calculation of Figure 7 in the main text. For experimental conditions see lower panel of Error! Reference source not found. Maintaining an invariant 20 mM sodium formate gradient, the temperature was incremented in 2°C steps from 18°C to 28°C. The normalized traces are shifted by 0.1 arbitrary units for displaying purposes. *I*theor (black lines) is fitted to *I*<sub>exp</sub> (colored lines) to obtain *P*<sub>m</sub>. In addition, the exponential function  $I(t) = I_0 + \Delta I \cdot \exp(-t/\tau)$  is also fitted to the data (grey line) to obtain the exponential time constant  $\tau$ . *P*<sub>m</sub> and  $\tau$  are plotted in Figure 7 of the main text.



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