

*Supplementary Information*

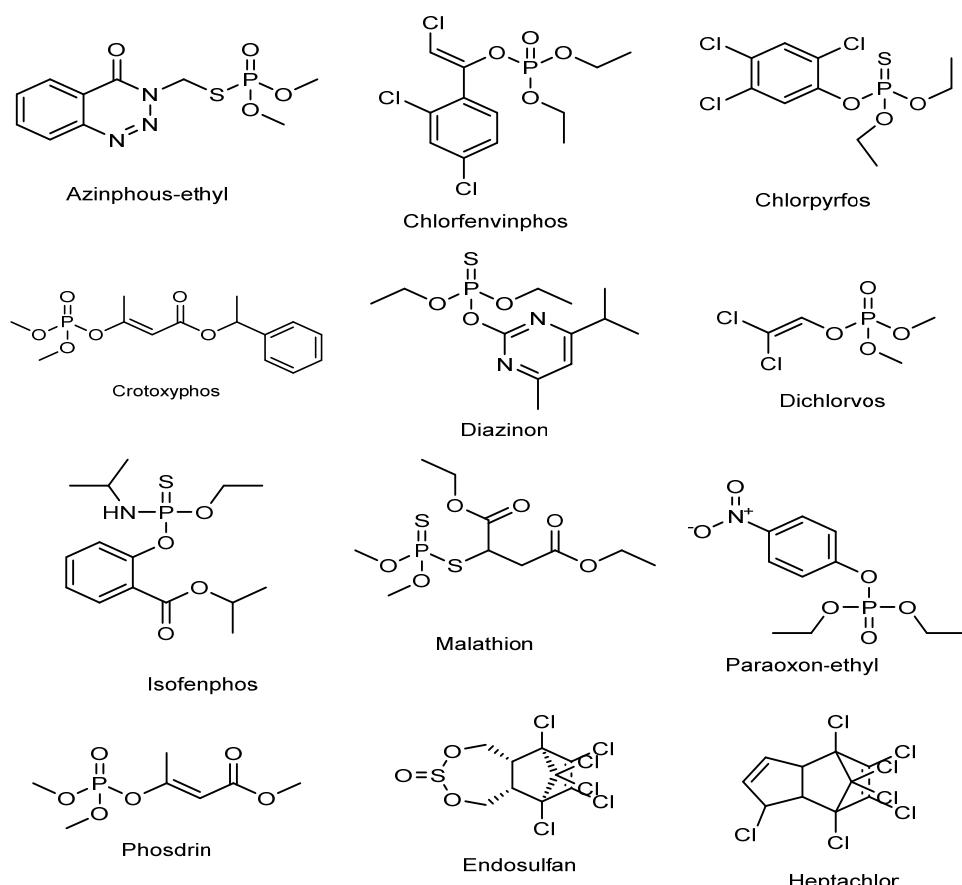
# New Terbium Complex as a Luminescent Sensor for the Highly Selective Detection of Malathion in Water Samples

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**Scheme S1.** The chemical structures of the organophosphorus pesticides.

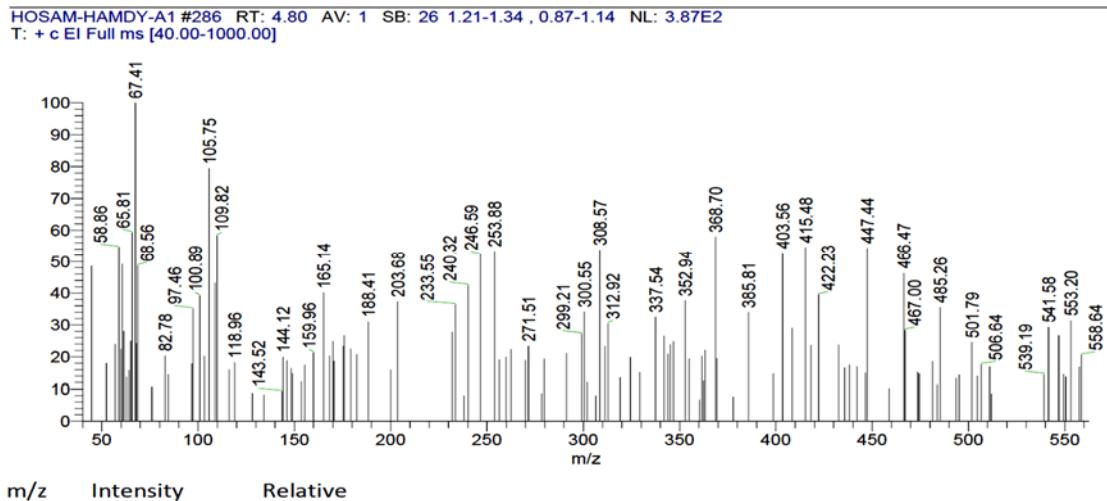


Figure S1. mass spectrum of the ligand.

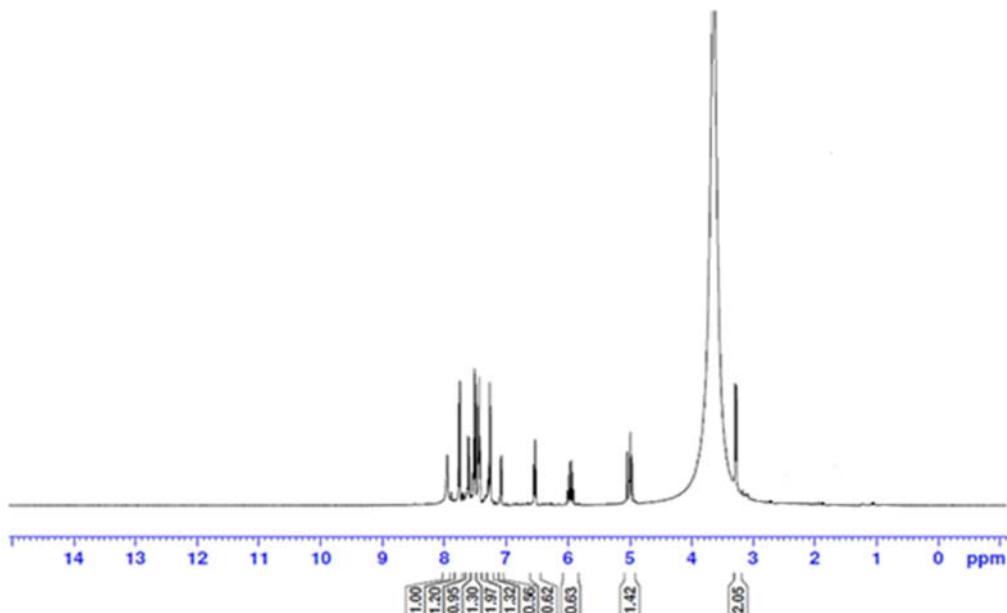
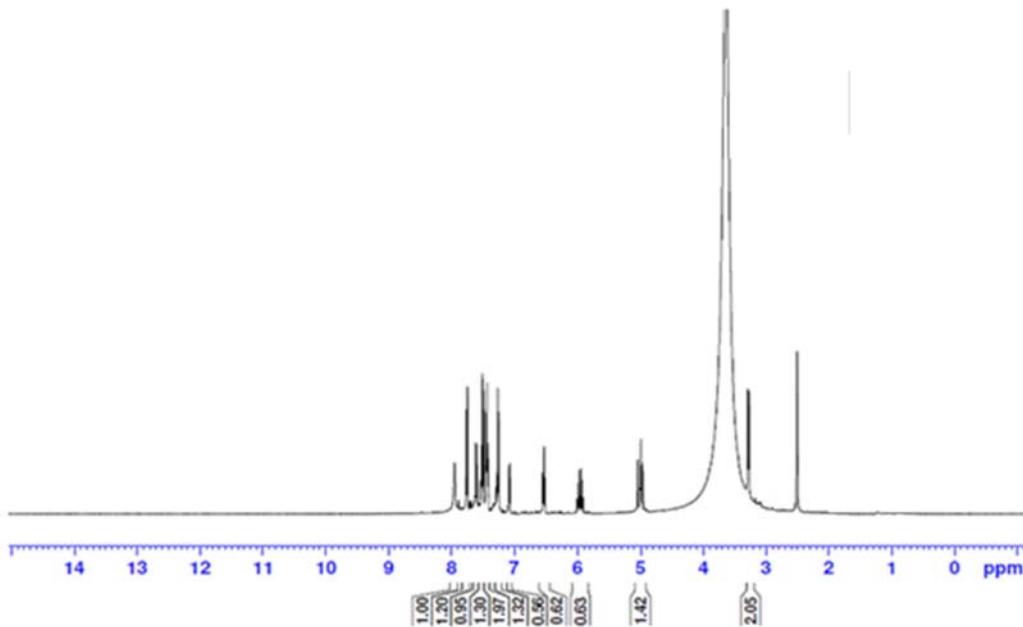
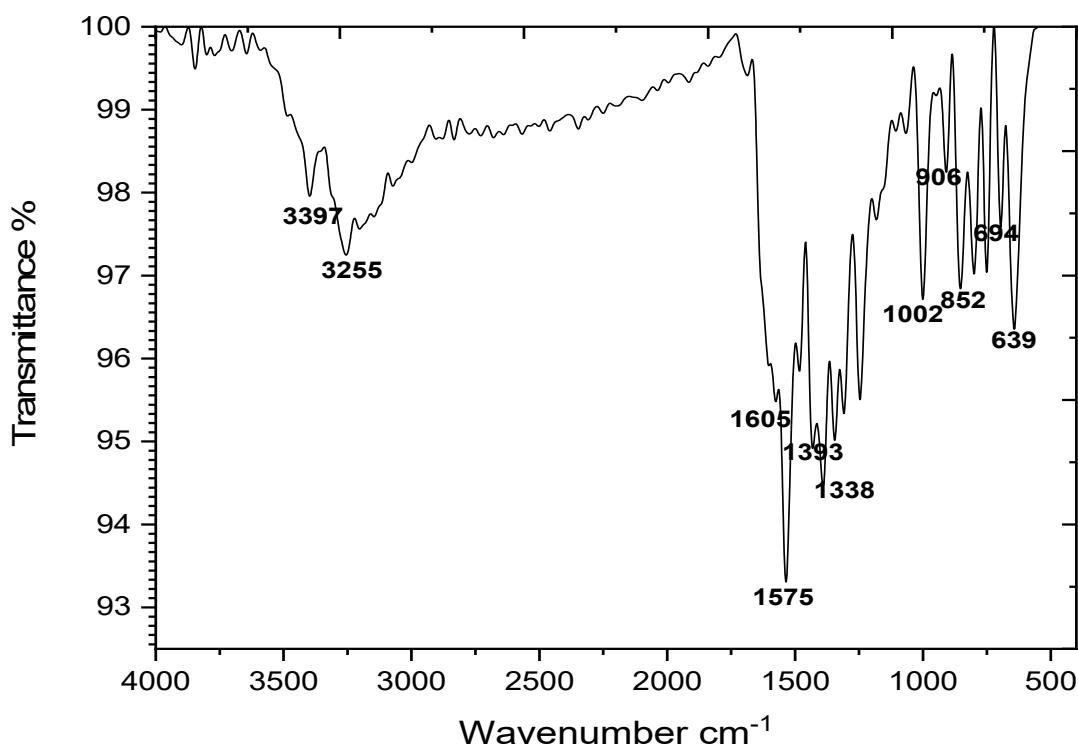


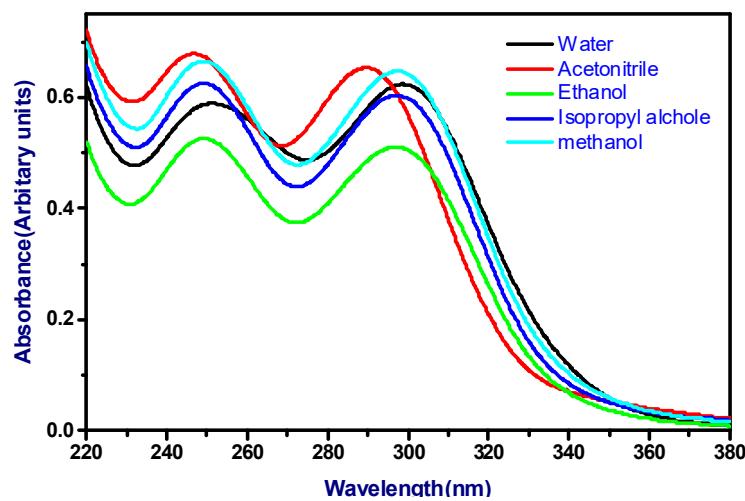
Figure S2.  $^1\text{H}$ NMR of the ligand in  $\text{DMSO-d}_6$ .



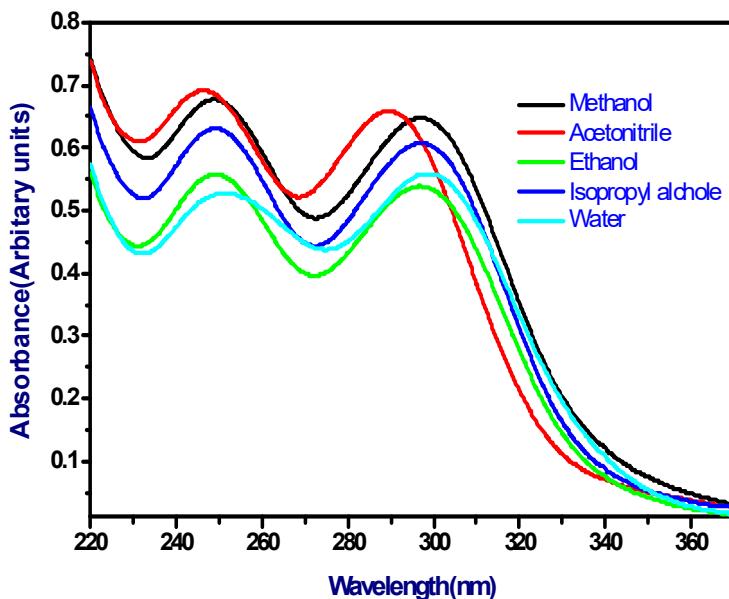
**Figure S3.** <sup>1</sup>HNMR of the ligand in DMSO-d<sub>6</sub> + D<sub>2</sub>O.



**Figure S4.** IR spectrum of the ligand.

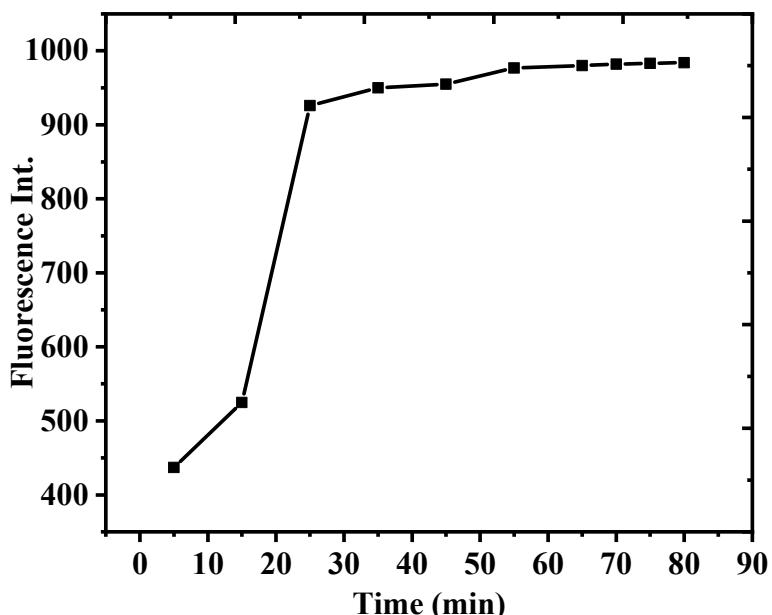


**Figure S5.** UV-spectra of  $5 \times 10^{-5}$  M H<sub>2</sub>DBAZ in different solvents at room temperature.

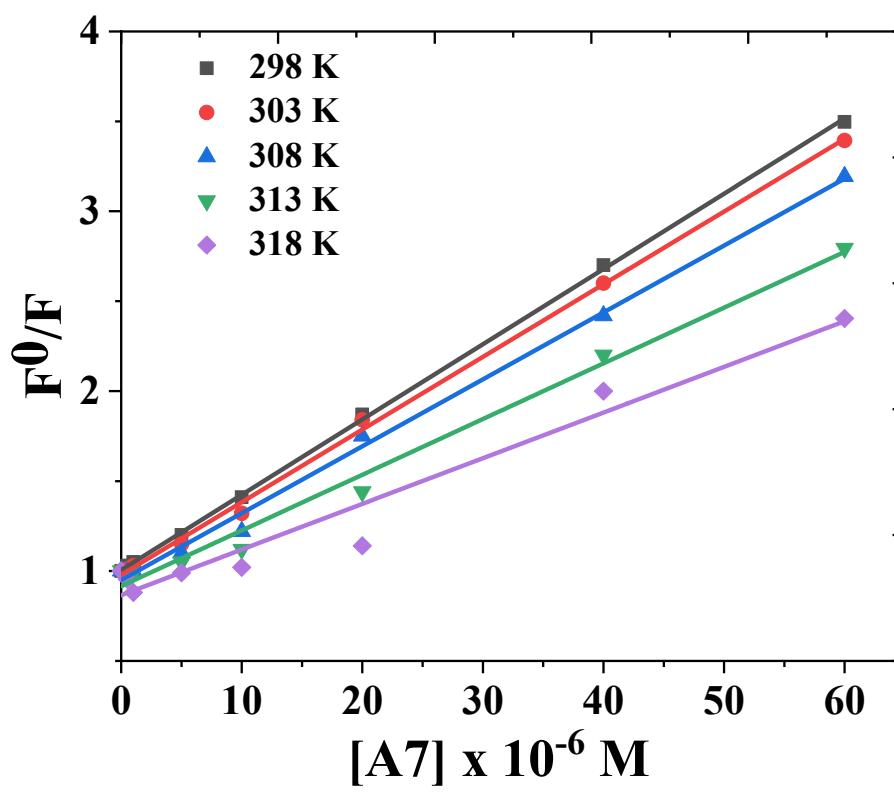


**Figure S6.** UV-spectra of  $5 \times 10^{-5}$  M Tb(III)-DBAZ complex in different solvents at room temperature.

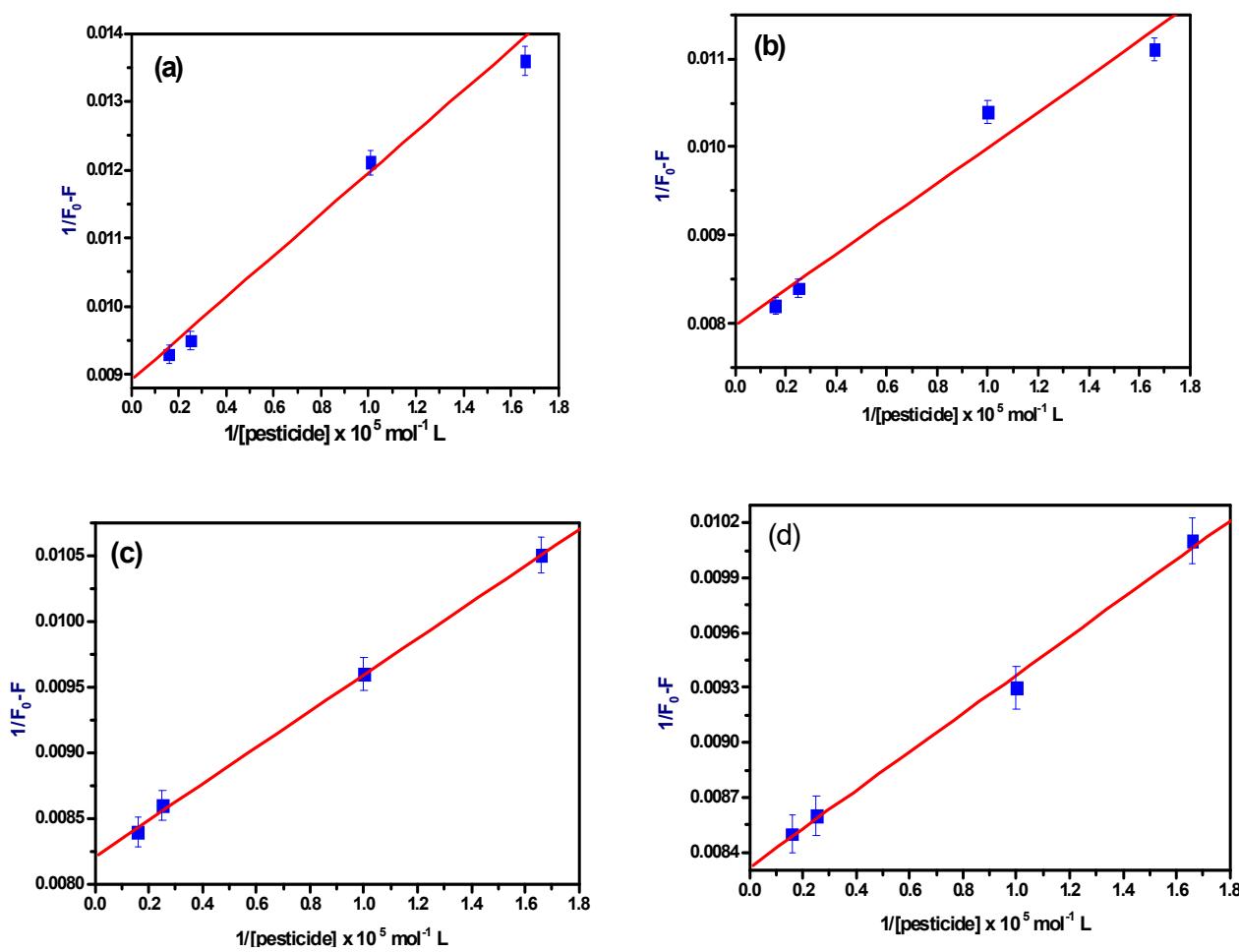
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**Figure S7.** Effect of time on the luminescence intensity of  $5 \times 10^{-5}$  M of Tb(III)-(DBAZ) complex in methanol medium, sensitivity high,  $\lambda_{\text{ex}} = 360$  nm, at 25 °C.



**Figure S8.**  $F^0/F$  against  $[A7]$  for Malathion upon its interaction with  $5 \times 10^{-5}$  M of Tb(III)-(DBAZ) complex in methanol at different temperatures,  $\lambda_{\text{ex}} = 360$  nm.



**Figure S9.**  $1/F_0 - F$  against  $1/\text{[pesticide]}$  for Malathion (A7) upon its interaction with  $5 \times 10^{-5} \text{ M}$  of Tb(III)-DBAZ at different temperatures, (a) at 303 K (b) at 308 K (c) at 313 K (d) at 318 K.