

Supplementary Materials

Novel Photoluminescence and Optical Thermometry of Solvothermally derived Tetragonal ZrO₂:Ti⁴⁺,Eu³⁺ Nanocrystals

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Table S1. The assignment of the observed emission lines for Eu³⁺ in the t-ZrO₂ NCs and fluorescence decay constants yielded by bi-exponential fitting

Eu ³⁺ type	⁵ D ₀ → ⁷ F ₁	⁵ D ₀ → ⁷ F ₂	⁵ D ₀ → ⁷ F ₃	⁵ D ₀ → ⁷ F ₄	Fluorescence decay constants (ms)
	magnetic dipole transition (nm)	forced dipole transition (nm)	electric forbidden transition by the electric dipole and magnetic dipole (nm)	forced dipole transition (nm)	
Eu(I)	592.2	607.0, 634.0	628.0, 652.0, 660.0	714.8	$\tau_1 \sim 6.23$, $\tau_2 \sim 2.54$
Eu(II)	591.6, 597.2	614.0, 631.2	626.4, 656.2	713.0	$\tau_1 \sim 6.23$, $\tau_2 \sim 2.54$

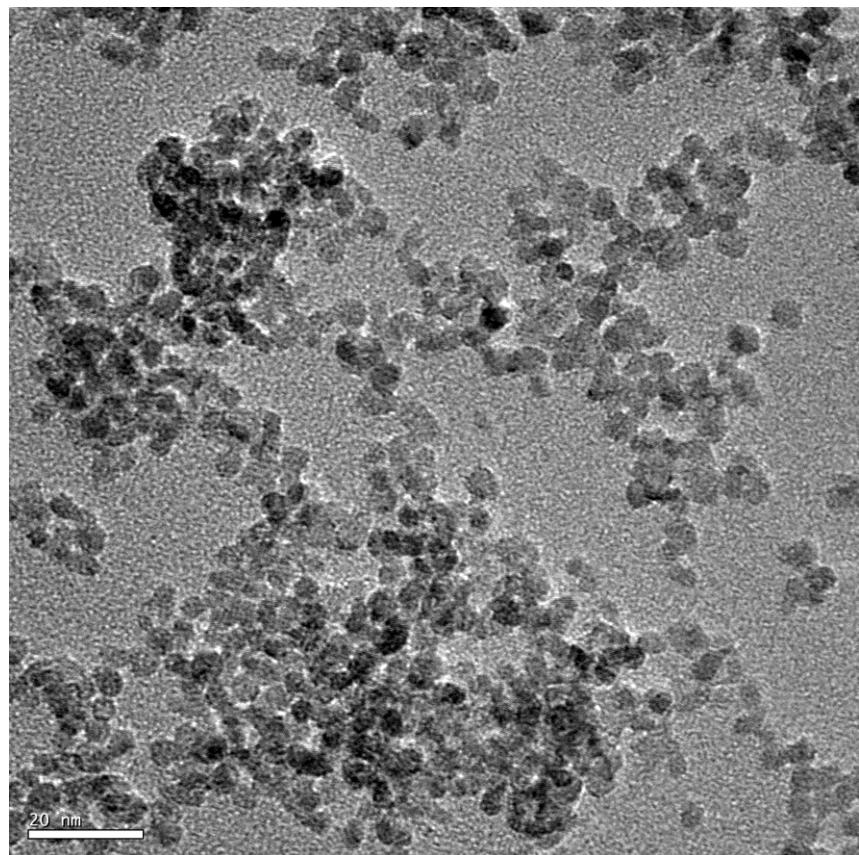


Figure S1. TEM image of t-ZrO₂:4% Ti⁴⁺,1% Eu³⁺ NCs.

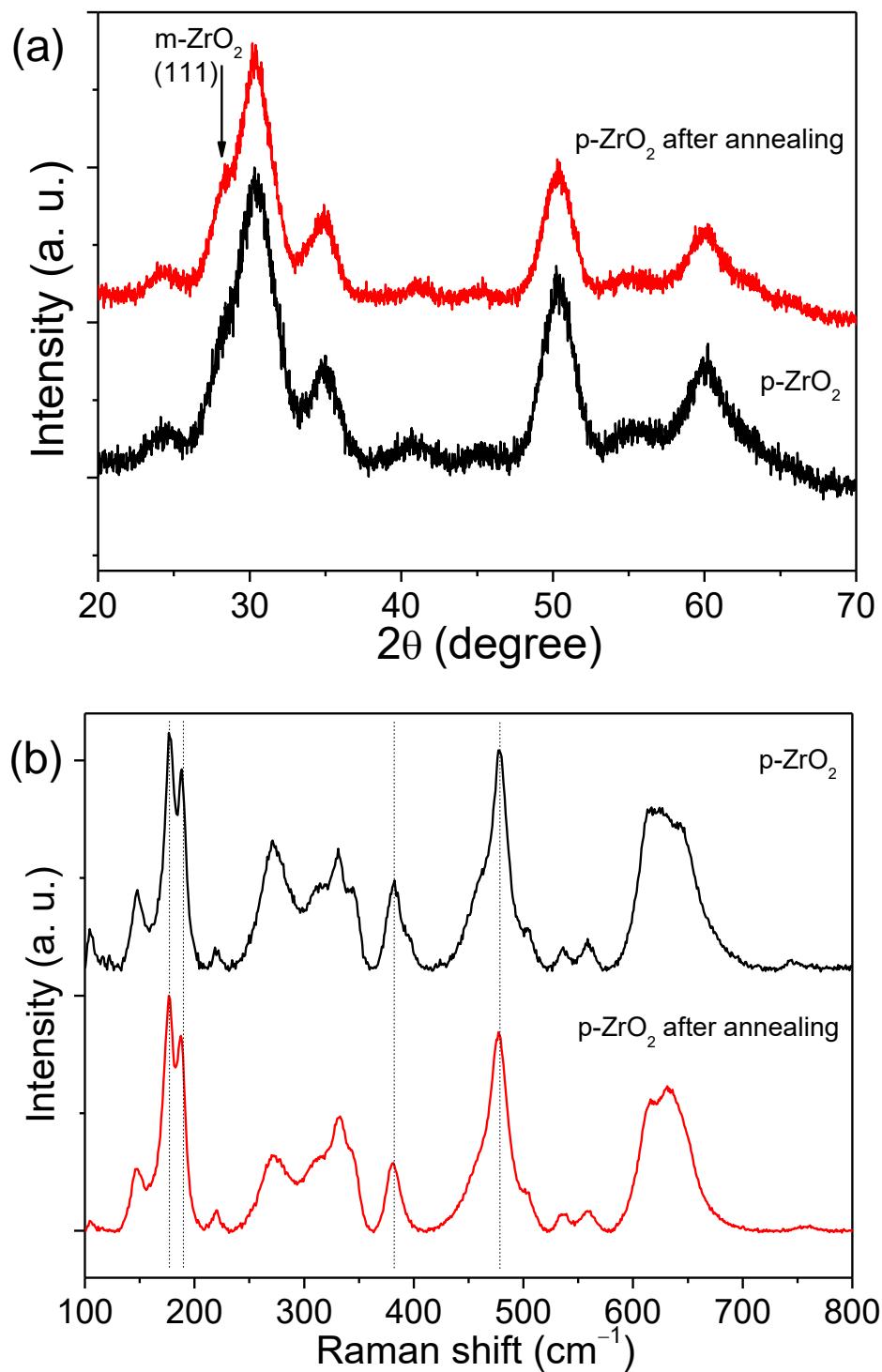


Figure S2. **(a)** Powder XRD patterns and **(b)** Raman spectra of $p\text{-ZrO}_2$ NCs before and after annealing at 400°C ; the dash lines in **(b)** indicate the modes of $m\text{-ZrO}_2$.

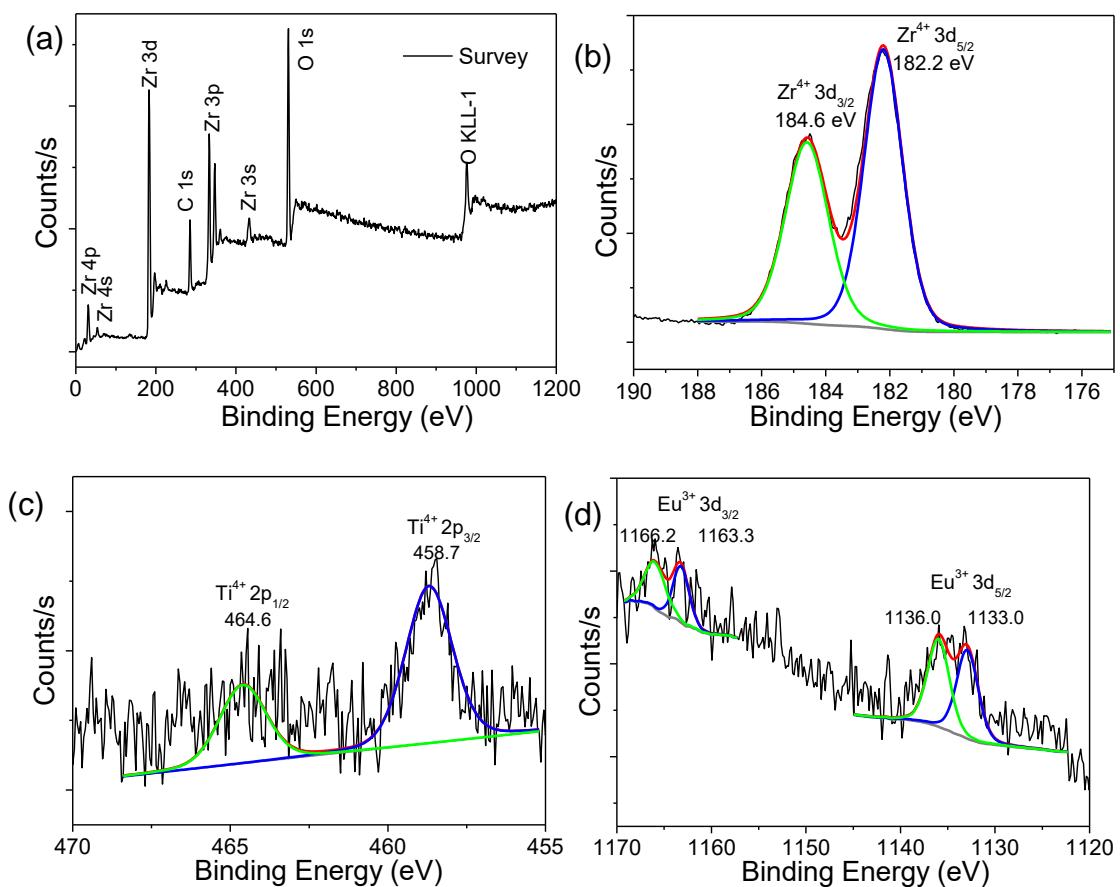


Figure S3. (a) XPS survey spectrum of t-ZrO₂:4%Ti⁴⁺,1%Eu³⁺ NCs and XPS high resolution spectra of (b) Zr 3d, (c) Ti 2p and (d) Eu 3d in t-ZrO₂:4%Ti⁴⁺,1%Eu³⁺ NCs.

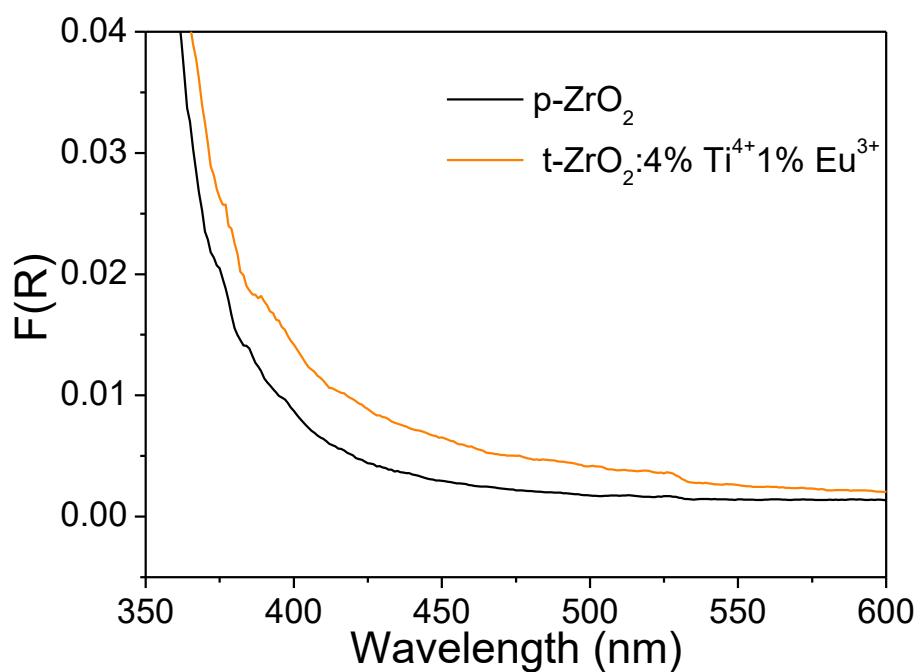


Figure S4. The enlarged diffuse reflection-UV-visible spectra of the as-synthesized p-ZrO₂ and t-ZrO₂:4% Ti⁴⁺, 1% Eu³⁺ NCs in the F(R) range of 0.0–0.04.

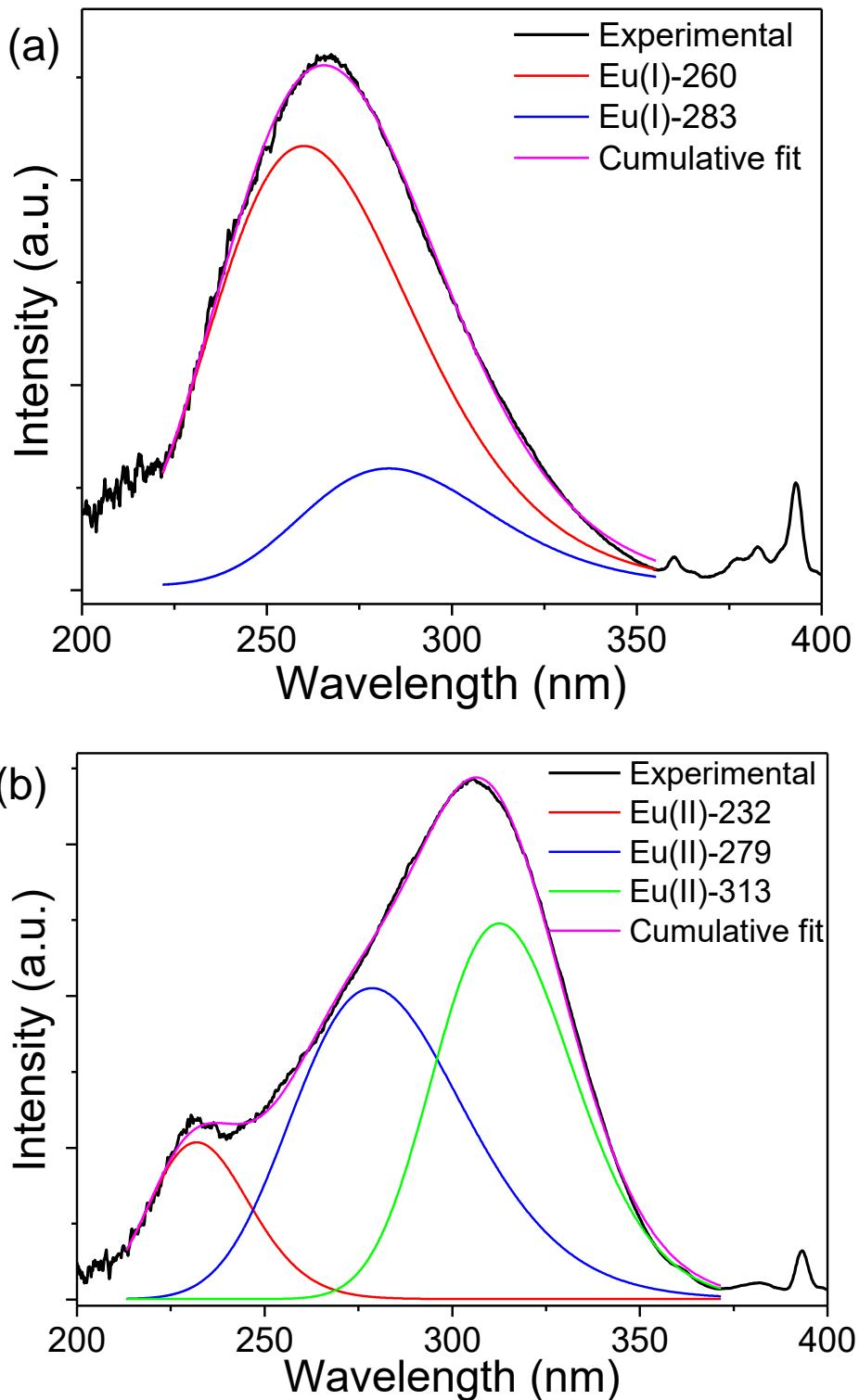


Figure S5. PL excitation spectra of t-ZrO₂:1%Eu³⁺ NCs by monitoring the relative maxima of Eu(I) ((a), $\lambda_{\text{em}}=607$) and Eu(II) ((b), $\lambda_{\text{em}}= 614$ nm) and their spectral decomposition of the corresponding broadband with Gauss typed peaks.

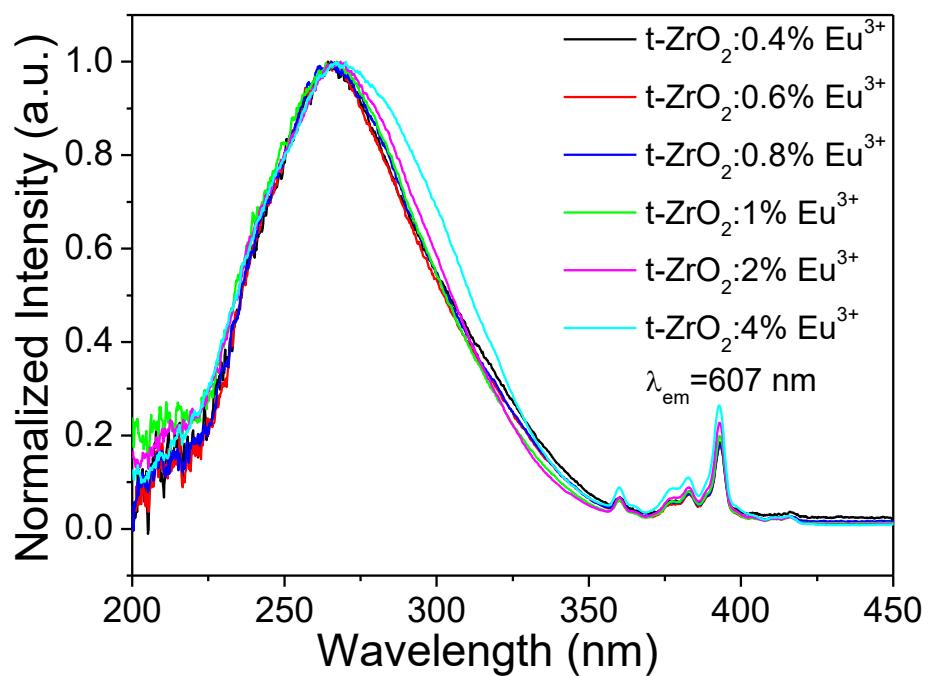


Figure S6. Normalized PL excitation spectra of t-ZrO₂:xEu³⁺ ($x=0.4\%$, 0.6% , 0.8% , 1% , 2% and 4%) NCs by monitoring the relative maxima of Eu(I) ($\lambda_{\text{em}}=607$).

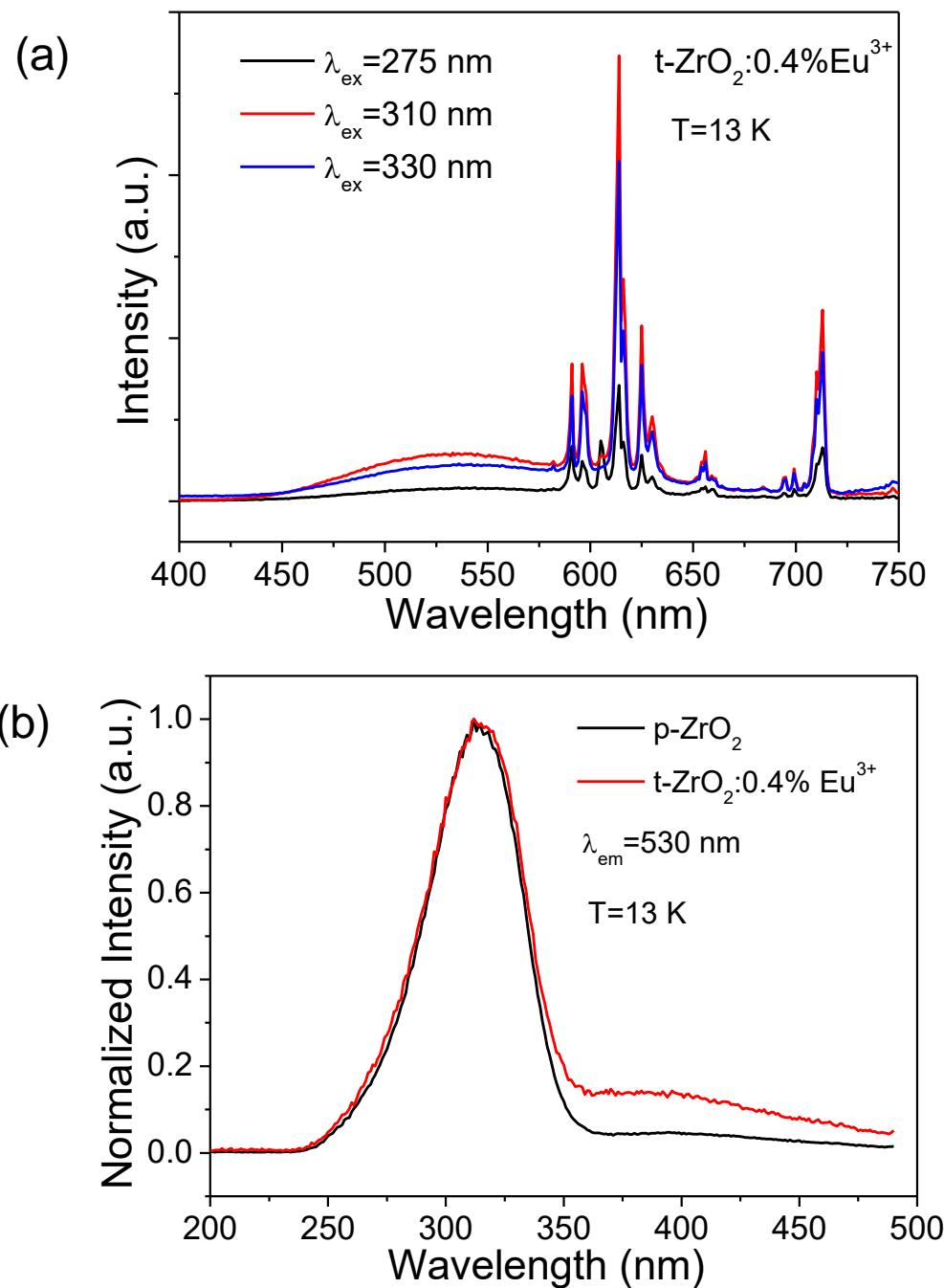


Figure S7. (a) The bright broadband PL emissions of CT ($\text{Ti}^{3+} \rightarrow \text{O}^-$) transitions in t-ZrO₂:0.4%Eu³⁺ NCs upon excitation into titanate groups at 13 K, together with the spectral superimposition of stronger sharp line emissions of Eu³⁺ due to nonradiative energy transfer; (b) Normalized PL excitation spectra of p-ZrO₂ and t-ZrO₂:0.4%Eu³⁺ NCs upon monitoring the peak of the broadband emissions at 530 nm.

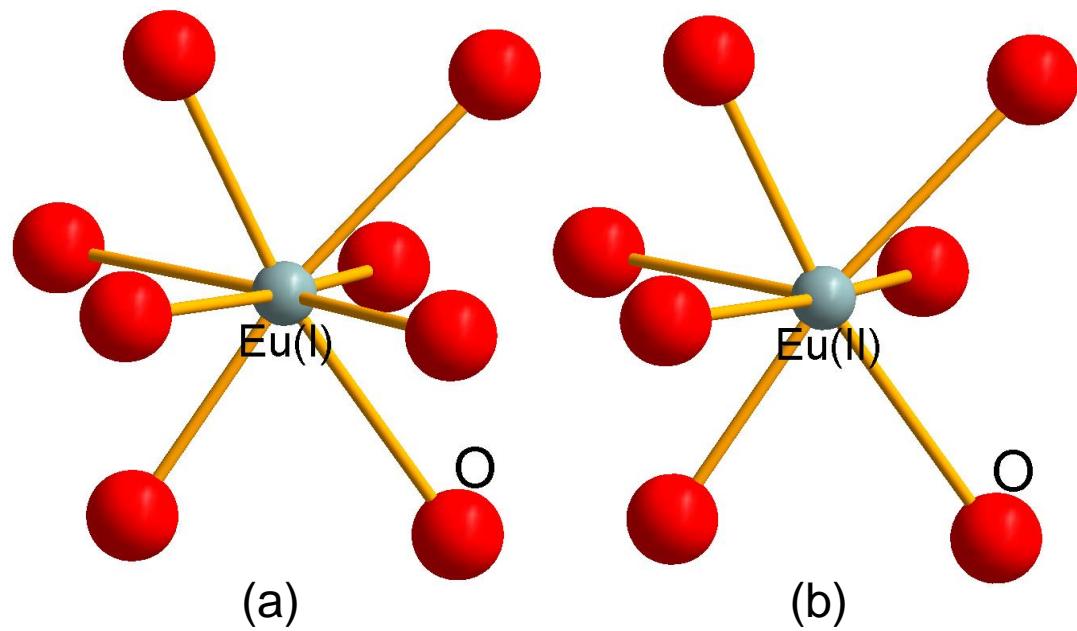


Figure S8. The scheme of oxygen coordination spheres of Eu(I) (a) and Eu (II) (b) in t-ZrO₂ NCs.

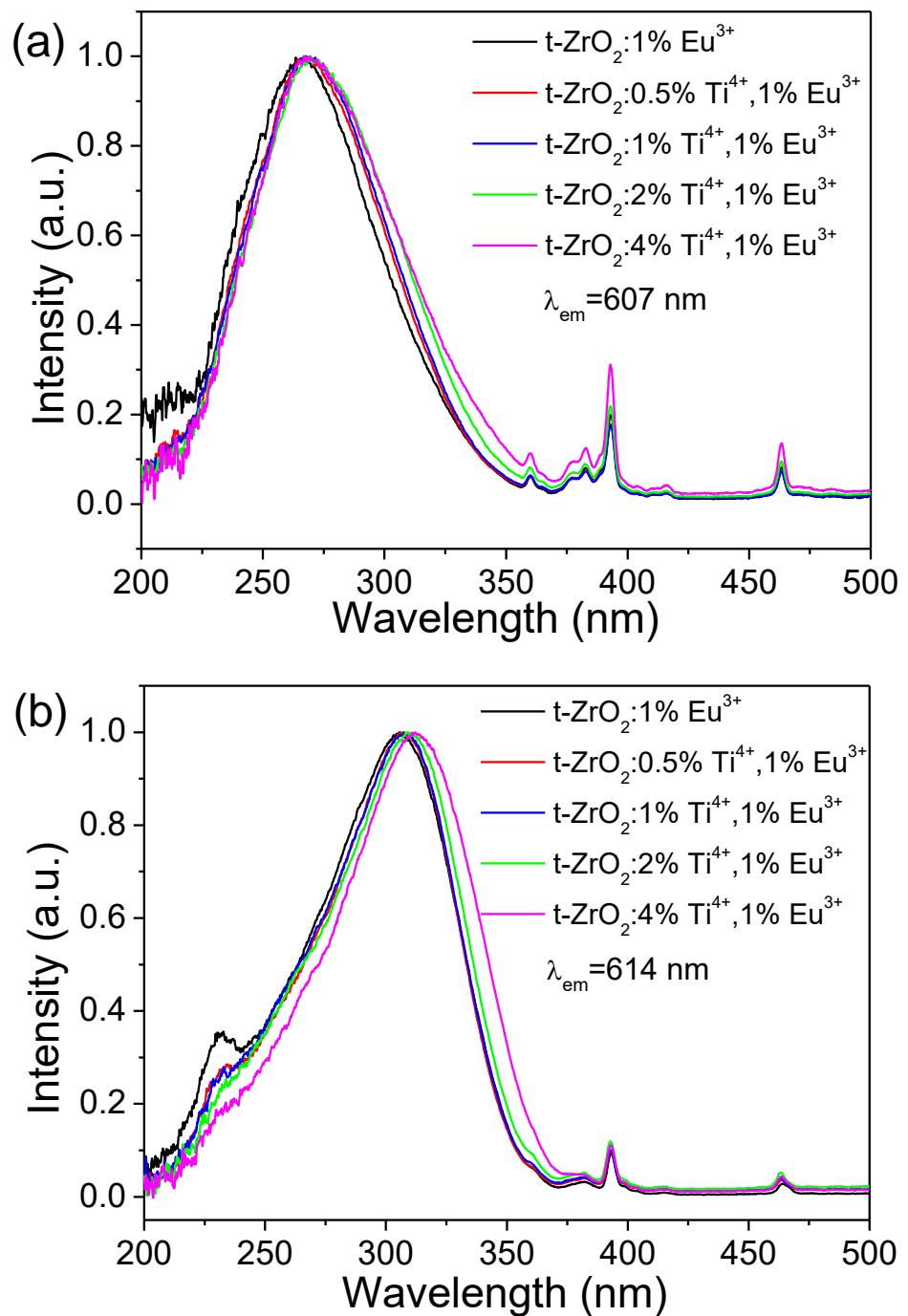


Figure S9. Normalized PL excitation spectra of t-ZrO₂: x Ti⁴⁺, 1% Eu³⁺ ($x=0, 0.5\%, 1\%, 2\%$ and 4%) NCs by monitoring the relative maxima of Eu(I) ((a), $\lambda_{\text{em}}=607$) and Eu(II) ((b), $\lambda_{\text{em}}=614$ nm).

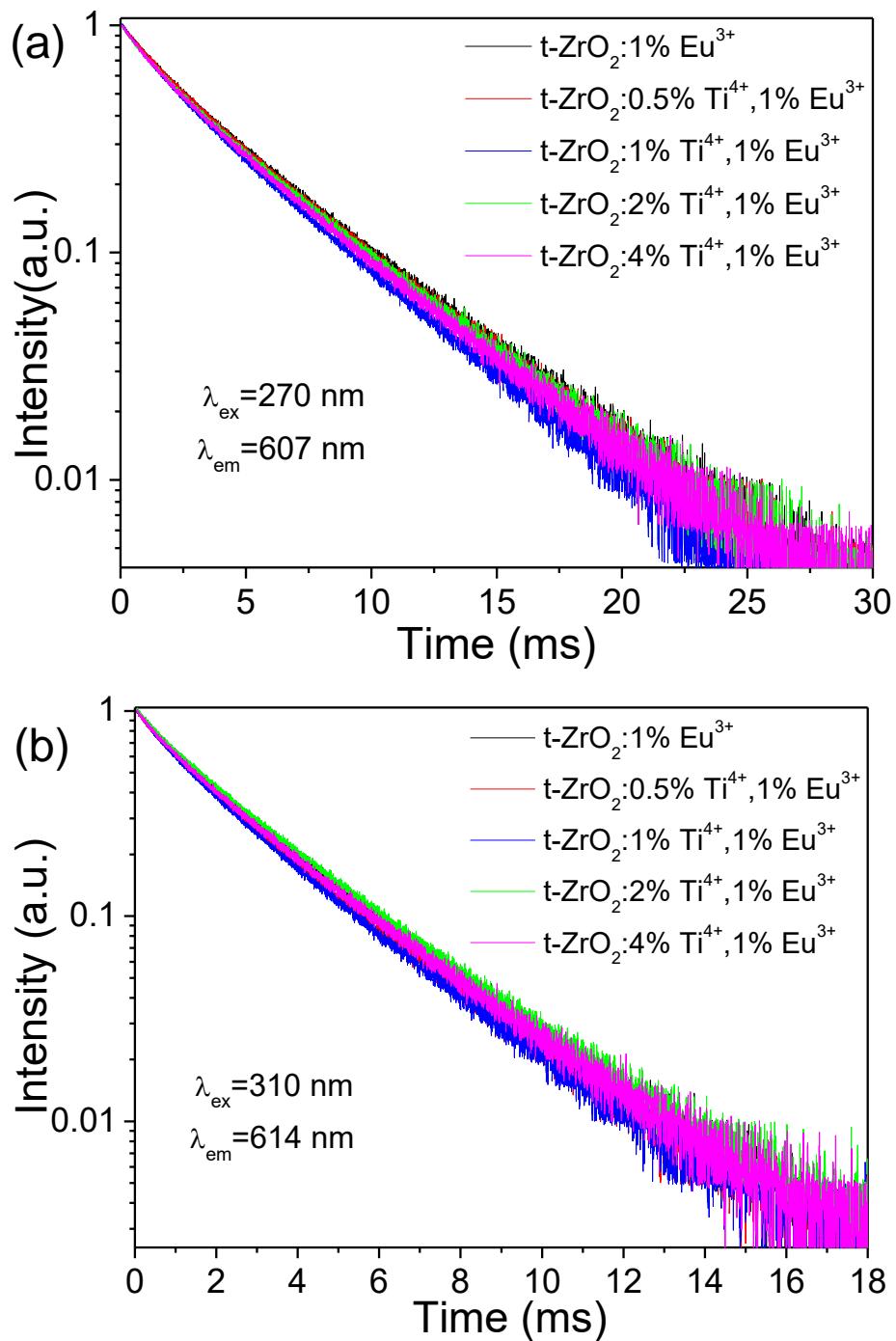


Figure S10. Fluorescence decay traces of t-ZrO₂:xTi⁴⁺,1%Eu³⁺ ($x=0, 0.5\%, 1\%, 2\%$ and 4%) NCs by monitoring the relative maxima of Eu(I) ((a), $\lambda_{\text{ex}}=270 \text{ nm}$, $\lambda_{\text{em}}=607 \text{ nm}$) and Eu(II) ((b), $\lambda_{\text{ex}}=310 \text{ nm}$, $\lambda_{\text{em}}=614 \text{ nm}$).

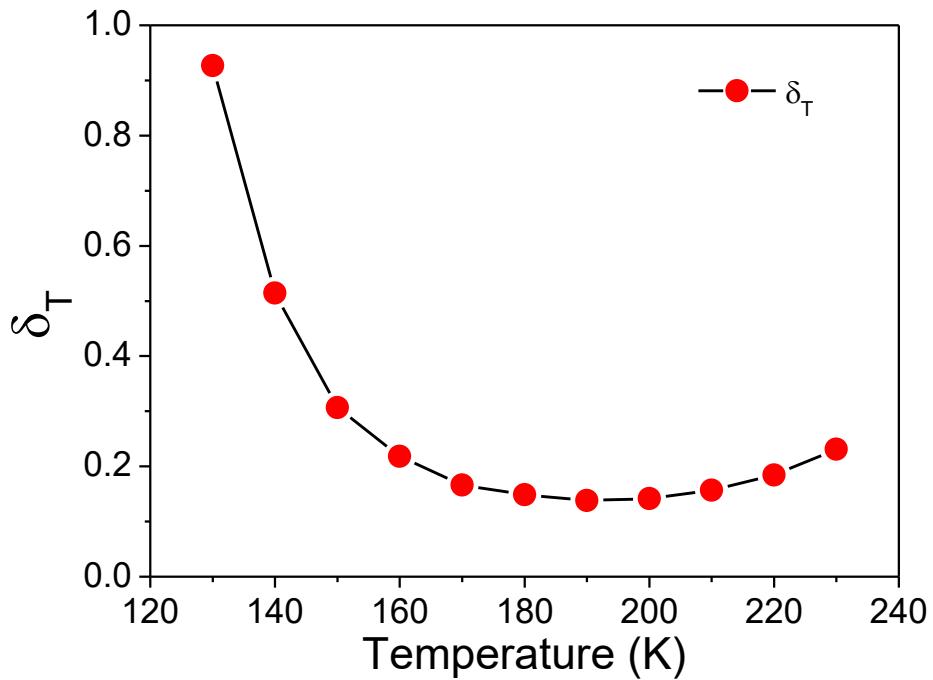


Figure S11. Temperature resolution δ_T of t-ZrO₂:0.4%Eu³⁺ NCs temperature sensing material in the temperature range from 130 to 230 K.

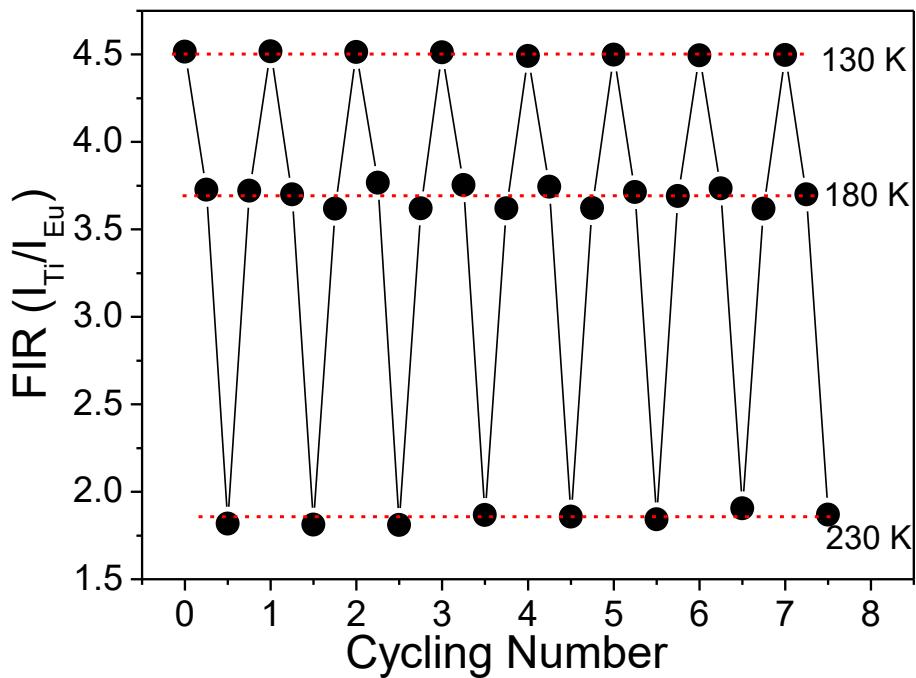


Figure S12. Temperature-induced switching of the FIR between the titanate groups signals (I_{Ti}) (450–580 nm) and Eu³⁺singals (690–725 nm) (I_{Eu}) (alternating between 130 K and 230 K). The dashed line is drawn to guide the eyes.