

*Supporting Information*

**Metalloporphyrin-based Metal-organic Frameworks for  
Photocatalytic Carbon Dioxide Reduction: The Influence of  
Metal Centers**

Qian Li, Keke Wang,\* Heyu Wang, Mengmeng Zhou, Bolin Zhou, Yanzhe Li, Qiang  
Li, Qin Wang, Hai-Min Shen, Yuanbin She\*

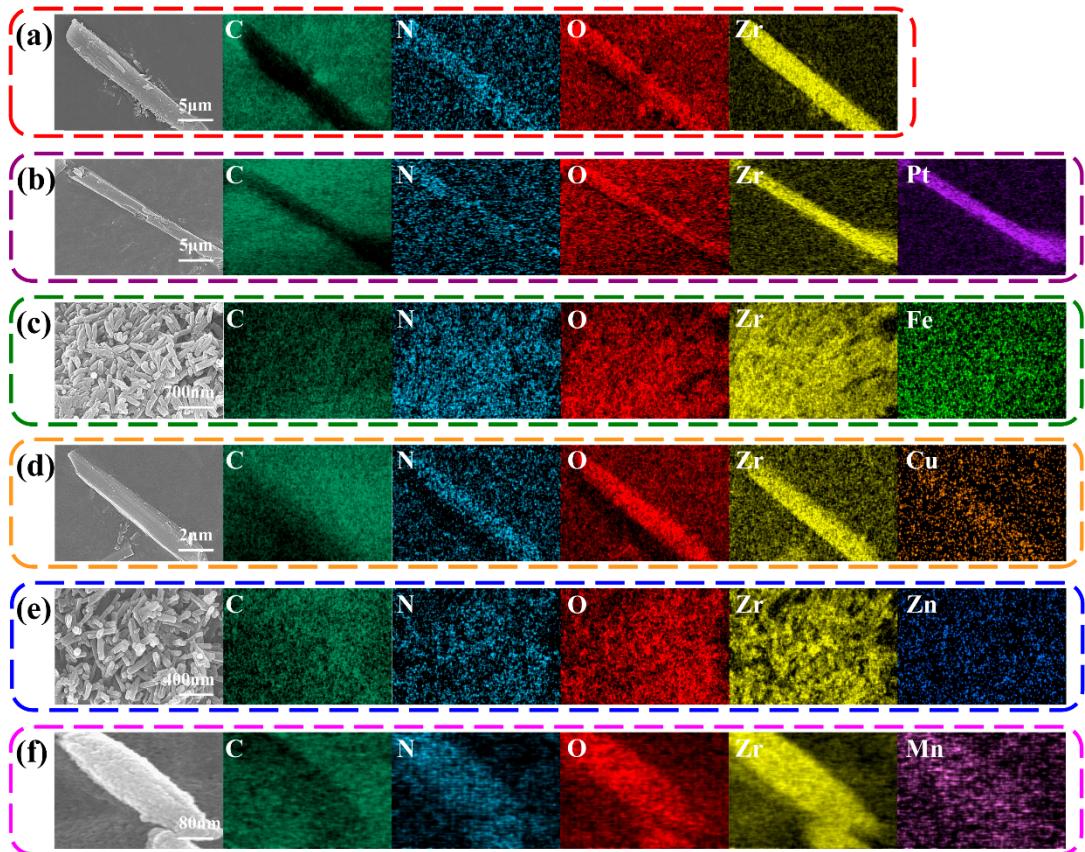


Figure S1. EDX-mapping images of (a) PCN-222(2H), (b) PCN-222(2H&Pt), (c) PCN-222(2H&Fe), (d) PCN-222(2H&Cu), (e) PCN-222(2H&Zn) and (f) PCN-222(2H&Mn), respectively.

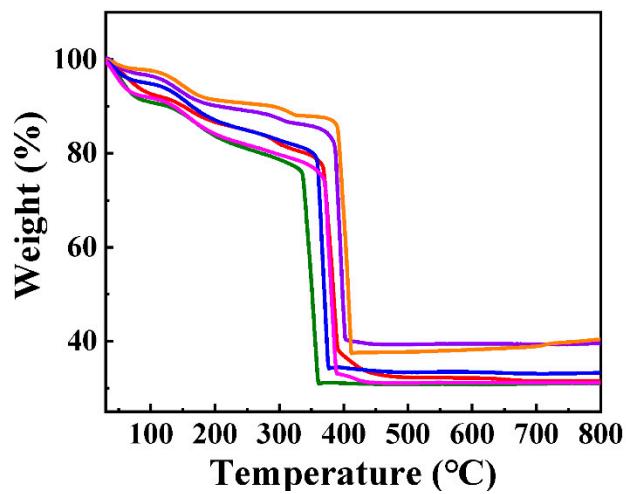


Figure S2. TGA curves of PCN-222(2H) and PCN-222(2H&M).

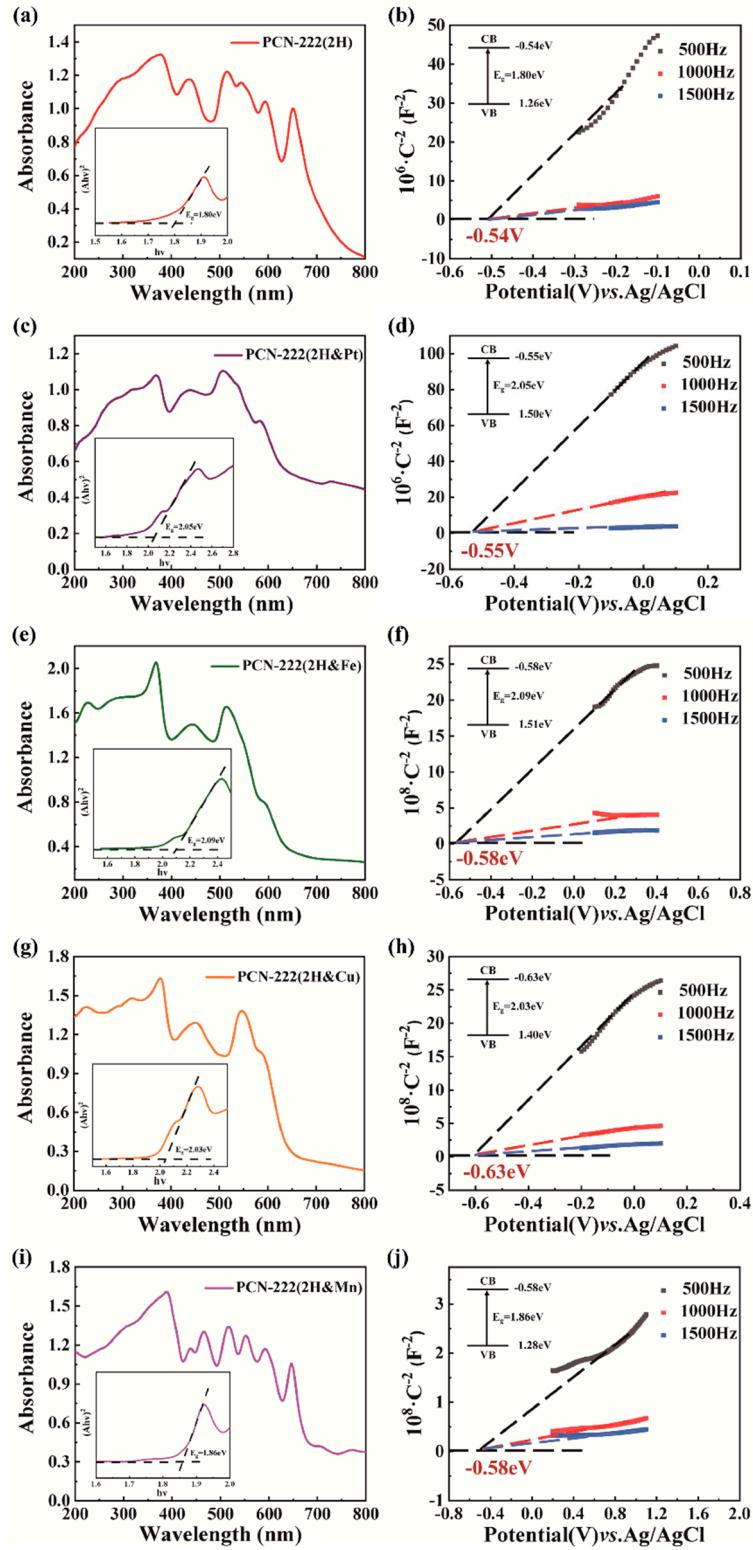


Figure S3. (a) UV-vis diffusion spectra and Tauc plot of PCN-222(2H), (b) Mott-Schottky plots of PCN-222(2H), (c) UV-vis diffusion spectra and Tauc plot of PCN-222(2H&Pt), (d) Mott-Schottky plots of PCN-222(2H&Pt), (e) UV-vis diffusion spectra and Tauc plot of PCN-222(2H&Fe), (f) Mott-Schottky plots of PCN-

222(2H&Fe), (g) UV-vis diffusion spectra and Tauc plot of PCN-222(2H&Cu), (h) Mott-Schottky plots of PCN-222(2H&Cu), (i) UV-vis diffusion spectra and Tauc plot of PCN-222(2H&Mn), and (j) Mott-Schottky plots of PCN-222(2H&Mn).

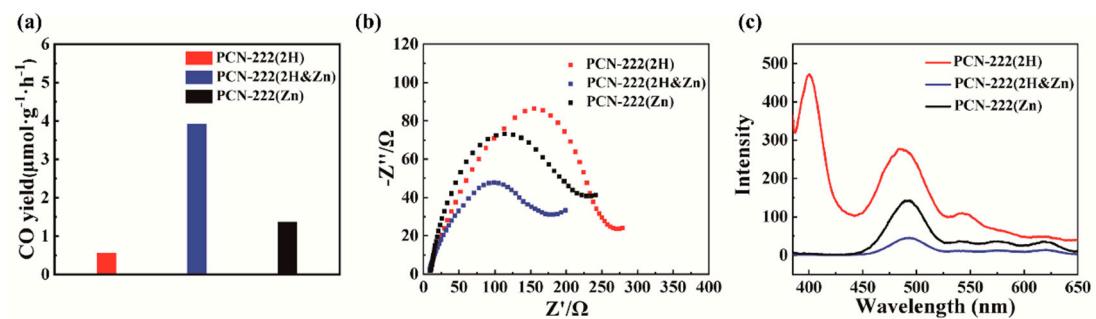


Figure S4. (a) The average CO yield, (b) EIS and (c) solid-state PL spectra excited at  $\lambda = 350$  nm of PCN-222(2H), PCN-222(2H&Zn) and PCN-222(Zn).

Table S1. Structural characteristics of PCN-222 materials

	$\text{SSA}/\text{m}^2 \cdot \text{g}^{-1}$	$V_{\text{total}}/\text{cm}^3 \cdot \text{g}^{-1}$	$V_{\text{micro}}$	$V_{\text{meso}}$	$V_{\text{meso}}/V_{\text{micro}}$
2H	1553	1.09	0.33	0.72	2.15
Zn	1350	0.85	0.25	0.57	2.29
Cu	1249	0.75	0.22	0.49	2.19
Pt	1126	0.68	0.23	0.43	1.92
Fe	1124	0.69	0.20	0.45	2.29
Mn	1071	0.45	0.10	0.29	2.86

Table S2. Summary of photocatalytic  $\text{CO}_2$  reduction performances of porphyrin-based MOFs materials in water or water vapor without the use of any organic solvent, photosensitizer, and sacrificial reagent.

Photocatalyst	Reaction condition	Products	References
PCN-224(Cu)	$\text{CO}_2, \text{H}_2\text{O}$	$\text{CO}: 3.717 \mu\text{mol}/(\text{g}\cdot\text{h})$ $\text{CH}_4: 1.357 \mu\text{mol}/(\text{g}\cdot\text{h})$	<i>ACS Sustainable Chem. Eng.</i> <b>2019</b> , 7, 15660
15%PCN-224(Cu)/ $\text{TiO}_2$	$\text{CO}_2, \text{H}_2\text{O}$	$\text{CO}: 37.21 \mu\text{mol}/(\text{g}\cdot\text{h})$ $\text{CH}_4: 0.2113 \mu\text{mol}/(\text{g}\cdot\text{h})$	<i>Chem. Eng.</i> <b>2019</b> , 7, 15660
$\text{TiO}_2$	$\text{CO}_2, \text{H}_2\text{O}$	$\text{CO}: 0.8183 \mu\text{mol}/(\text{g}\cdot\text{h})$	
PCN-601	$\text{CO}_2, \text{H}_2\text{O}$ vapor	$\text{CO}: 6.0 \mu\text{mol}/(\text{g}\cdot\text{h})$ $\text{CH}_4: 10.1 \mu\text{mol}/(\text{g}\cdot\text{h})$	
PCN-222	$\text{CO}_2, \text{H}_2\text{O}$ vapor	$\text{CO}: 5.5 \mu\text{mol}/(\text{g}\cdot\text{h})$ $\text{CH}_4: 3.5 \mu\text{mol}/(\text{g}\cdot\text{h})$	<i>J. Am. Chem. Soc.</i> <b>2020</b> , 142, 12515
$\text{Ni}_3\text{TCPP}$	$\text{CO}_2, \text{H}_2\text{O}$ vapor	$\text{CO}: 1.5 \mu\text{mol}/(\text{g}\cdot\text{h})$ $\text{CH}_4: 0.6 \mu\text{mol}/(\text{g}\cdot\text{h})$	
PCN-222(Cu)	$\text{CO}_2, \text{H}_2\text{O}$	$\text{CO}: \sim 0.2 \mu\text{mol}/(\text{g}\cdot\text{h})$	<i>Wuli Huaxue Xuebao.</i> <b>2020</b> , 36, 19050861-19050868
10%PCN-222(Cu)/ $\text{TiO}_2$	$\text{CO}_2, \text{H}_2\text{O}$	$\text{CO}: 13.24 \mu\text{mol}/(\text{g}\cdot\text{h})$ $\text{CH}_4: 1.73 \mu\text{mol}/(\text{g}\cdot\text{h})$	