

## **Supporting Information**

### ***Separating electronic from steric effects in ethene/α-olefin copolymerization: A case study on octahedral [ONNO] Zr-catalysts***

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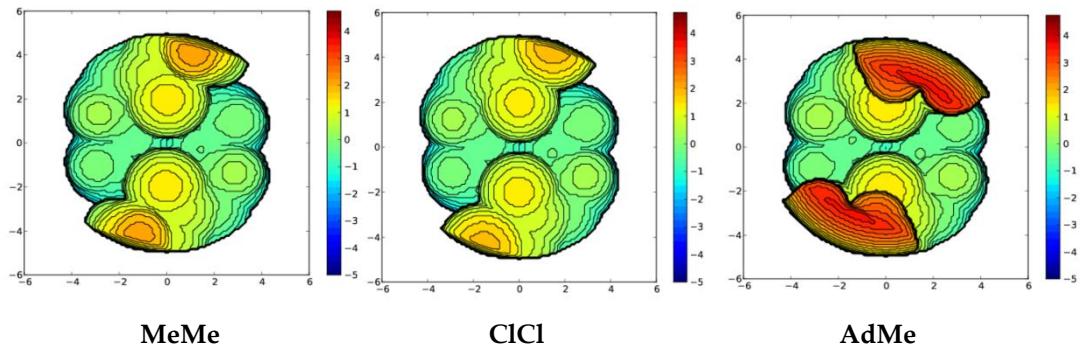
- **Polymerization results**

**Table S1.** Experimental results for catalysts **1** in ethene/butene (*E/B*) copolymerization.

Entry	Catalyst	<i>p(E), bar</i>	<i>n(Zr), μmol</i>	<i>X<sub>i</sub></i> <sup>(a)</sup>	<i>X<sub>f</sub></i> <sup>(a)</sup>	<i>R<sub>p</sub></i> <sup>(b)</sup>	<i>[B]<sub>cop</sub></i> <sup>(c)</sup> mol%	<i>10<sup>-2</sup> × P<sub>n</sub></i> <sup>(d)</sup>	<i>r<sub>E</sub></i>	<i>r<sub>B</sub></i>	<i>r<sub>E</sub>r<sub>B</sub></i>
1	<b>MeMe</b>	1.1	38	0.0460	0.0464	0.5	24	1.0	61(6)	0.011(1)	0.7(1)
2		0.55	40	0.0225	0.0241	1.0	39	0.91	59(6)	0.013(1)	0.7(1)
3	<b>MeCl</b>	1.1	37	0.0472	0.0526	7	32	0.53	42(4)	0.021(2)	0.9(2)
4		0.55	38	0.0238	0.0278	19	53	0.50	34(3)	0.028(3)	1.0(2)
5	<b>ClMe</b>	1.1	6.0	0.0543	0.0584	31	37	0.24	33(3)	0.037(3)	1.2(2)
6		0.55	5.7	0.0200	0.0220	15	59	0.27	33(3)	0.030(3)	1.0(2)
7	<b>ClCl</b>	1.1	4.1	0.0552	0.0610	360	43	0.22	22(2)	0.041(4)	0.9(2)
8		0.55	3.9	0.0233	0.0295	281	71	0.26	22(2)	0.068(7)	1.5(3)

In toluene (150mL); 25 °C; *p*(B) = 1.1 bar; cocatalyst = MAO/TBP ([Al] = 5.4 × 10<sup>-2</sup> M). E = ethene; B = butene. <sup>(a)</sup> Initial (*i*) and final (*f*) comonomer feeding ratio ([E]/[B], see also Materials and Methods section); <sup>(b)</sup> Productivity in kg(copolymer) × mol(Zr)<sup>-1</sup> × [C<sub>n</sub>H<sub>2n</sub>]<sup>-1</sup> × h<sup>-1</sup>; <sup>(c)</sup> butene content in the copolymer, determined by <sup>13</sup>C NMR; <sup>(d)</sup> average copolymerization degree determined by <sup>1</sup>H NMR based on the concentration of chain end groups.

- Buried volume analysis



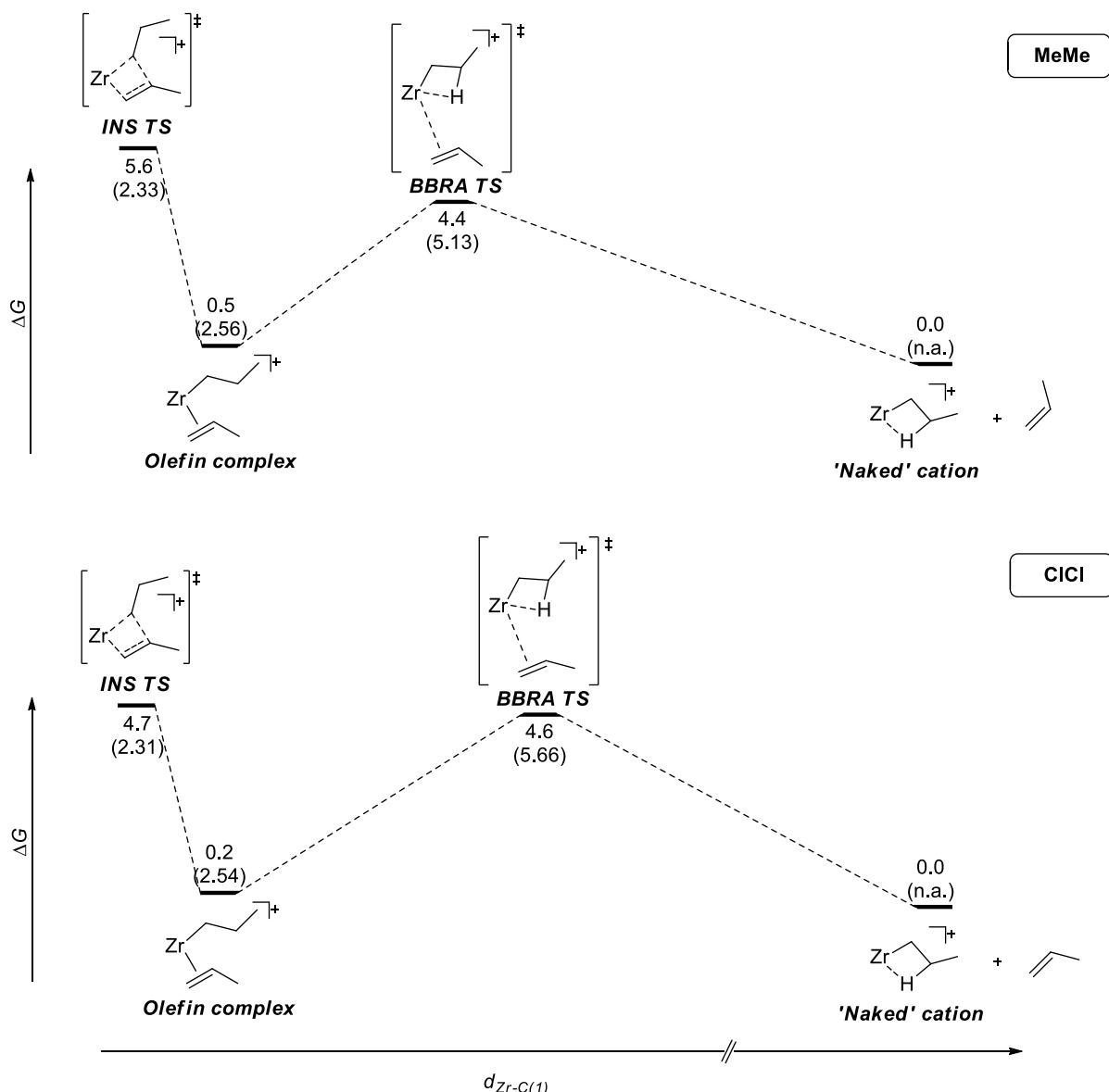
**Figure S1.** Steric heat maps from buried volume analysis for catalysts **MeMe** and **ClCl** (see also Table 3). **AdMe** given for comparison

- Electronic charges

**Table S6.** Comparison between DFT estimated  $Q_{\text{ZrCl}_2}$ , experimental  $r$ -parameters for ethene/butene copolymerization and corresponding  $\Delta\Delta G^\ddagger$  (in kcal/mol).

Entry	Catalyst	$Q_{\text{ZrCl}_2}$	$r_E$	$r_B$	$\Delta\Delta G^\ddagger_{(\text{EB-EE})}$	$\Delta\Delta G^\ddagger_{(\text{BE-BB})}$
1	<b>MeMe</b>	0.874	60	0.012	2.4	-2.6
2	<b>MeCl</b>	0.882	38	0.025	2.2	-2.2
3	<b>ClMe</b>	0.895	33	0.034	2.1	-2.0
4	<b>ClCl</b>	0.904	22	0.06	1.8	-1.7

- Calculated reaction profiles for propene enchainment after ethene



**Figure S2.** Energies profiles for propene enchainment after ethene predicted by DFT for **MeMe** and **CICl**. INS = insertion, BBRA = backbone rearrangement. Numbers are relative Gibbs free energy differences in kcal/mol; numbers in brackets are distances between the Zr center and the C(1) carbon of ethene in Å ( $d_{Zr-C(1)}$ ). A Zr-*n*Pr group is used as model for the Zr-polymeryl after ethene insertion (see Materials and Methods).

- Final energies, entropy and enthalpy corrections in Hartree

Name	Formula	RI-TPSSTPSS/DZ			M06-2X(PCM)/TZ	
		Energy	ZPE	EnthalpyCorr (T=298 K, p=1.0 atm)	EntropyCorr (T=298 K, p=1.0 atm)	Energy
MeMe_ZrCl2	C22H30Cl2N2O2Zr	-2084.361224	0.477314	0.510633	0.097763	-2083.894133
MeCl_ZrCl2	C20H24Cl4N2O2Zr	-2924.962856	0.404720	0.436752	0.093948	-2924.495354
ClMe_ZrCl2	C20H24Cl4N2O2Zr	-2924.956744	0.404516	0.436729	0.095355	-2924.489575
ClCl_ZrCl2	C18H18Cl6N2O2Zr	-3765.554608	0.331533	0.362585	0.092789	-3765.086479
ethene	C2H4	-78.605619	0.050223	0.054225	0.024892	-78.571775
propene	C3H6	-117.933593	0.078307	0.083377	0.030094	-117.883416
MeMe_nPr	C25H37N2O2Zr	-1282.106400	0.566008	0.600650	0.099172	-1281.645679
MeMe_ethylene_nPr	C27H41N2O2Zr	-1360.723808	0.618557	0.657048	0.107795	-1360.226474
MeMe_propene_nPr	C28H43N2O2Zr	-1400.052269	0.645664	0.685861	0.111962	-1399.541741
MeMe_ethylene_nPr_BBRA_TS	C27H41N2O2Zr	-1360.714884	0.616755	0.655883	0.113782	-1360.219590
MeMe_ethylene_nPr_INS_TS	C27H41N2O2Zr	-1360.723311	0.619147	0.656602	0.105638	-1360.225028
MeMe_propene_nPr_BBRA_TS	C28H43N2O2Zr	-1400.043321	0.644616	0.684973	0.117252	-1399.531010
MeMe_propene_nPr_INS_TS	C28H43N2O2Zr	-1400.047796	0.647068	0.685808	0.108139	-1399.536178
MeMe_ethylene_iBu_BBRA_TS	C28H43N2O2Zr	-1400.038550	0.644204	0.684735	0.116505	-1399.530246
MeMe_ethylene_iBu_INS_TS_A	C28H43N2O2Zr	-1400.045024	0.646559	0.685332	0.107834	-1399.534608
MeMe_ethylene_iBu_INS_TS_B	C28H43N2O2Zr	-1400.043060	0.646499	0.685301	0.108207	-1399.531223
MeMe_propene_iBu_BBRA_TS	C29H45N2O2Zr	-1439.366693	0.672107	0.713804	0.118956	-1438.841650
MeMe_propene_iBu_INS_TS_A	C29H45N2O2Zr	-1439.369398	0.674395	0.714475	0.110302	-1438.845527
MeMe_propene_iBu_INS_TS_B	C29H45N2O2Zr	-1439.366465	0.674487	0.714550	0.110697	-1438.842006
ClCl_nPr	C21H25Cl4N2O2Zr	-2963.289346	0.420107	0.452528	0.095541	-2962.832557
ClCl_ethylene_nPr	C23H29Cl4N2O2Zr	-3041.909112	0.472799	0.509033	0.103821	-3041.414210
ClCl_propene_nPr	C24H31Cl4N2O2Zr	-3081.238086	0.499872	0.537787	0.107650	-3080.729588
ClCl_ethylene_nPr_BBRA_TS	C23H29Cl4N2O2Zr	-3041.897581	0.470831	0.507723	0.109791	-3041.405952
ClCl_ethylene_nPr_INS_TS	C23H29Cl4N2O2Zr	-3041.908850	0.473239	0.508483	0.101481	-3041.413409
ClCl_propene_nPr_BBRA_TS	C24H31Cl4N2O2Zr	-3081.226092	0.498684	0.536829	0.113516	-3080.717690
ClCl_propene_nPr_INS_TS	C24H31Cl4N2O2Zr	-3081.233760	0.501232	0.537736	0.104123	-3080.724799
ClCl_ethylene_iBu_BBRA_TS	C24H31Cl4N2O2Zr	-3081.221779	0.498307	0.536565	0.112165	-3080.717024
ClCl_ethylene_iBu_INS_TS_A	C24H31Cl4N2O2Zr	-3081.230867	0.500619	0.537195	0.103939	-3080.722936
ClCl_ethylene_iBu_INS_TS_B	C24H31Cl4N2O2Zr	-3081.228788	0.500535	0.537182	0.104653	-3080.719601

- **Full Gaussian citation**

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