

## Article

# Ultrasonic Pretreatment Combined with Microwave-Assisted Hydrodistillation for Extraction of Essential Oil from *Melaleuca bracteata* 'Revolution Gold' Leaves Scales Induced by Cellulase-Inorganic Salt and Its Anti-Fungal Activity

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**Abstract:** In order to further develop the commercial use of *Melaleuca bracteata* (F. Muell), this report studied the extraction of essential oil from *Melaleuca bracteata* (F. Muell) leaves using ultrasonic pretreatment, cellulase-inorganic salt soaked and combined with microwave-assisted hydrodistillation. To optimize the primary contributing parameters, the Box–Behnken design (BBD) was applied. The optimum yield of essential oil was 9.61 mL/kg DW at a microwave power of 510.77 W, lithium chloride dose of 63.56  $\mu\text{mol}$ , and microwave irradiation period of 46.97 min. The essential oil included a total of 41 compounds, and methyl eugenol (76.53%) and methyl cinnamate (12.62%) were the main compounds. The inhibitory impact was notable when the essential oil concentration was 1.6 mg/mL. Therefore, it has the potential to replace chemical pesticides. When the concentration of the essential oil solution was 1.6 mg/mL, the three pathogenic species of fungus (*Pseudocercospora psidii*, *Colletotrichum eriobotryae*, and *Colletotrichum siamense*) were greatly affected; at this dose, the fungus was unable to develop and its growth diameter was 0 mm. Additionally, the fungus's inhibition rate reached 100%.

**Keywords:** microwave-assisted hydrodistillation; leaves of *Melaleuca bracteata*; cellulase-inorganic salt; fungal inhibition effects



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## 1. Introduction

Fungal diseases are frequent in agricultural production and continue to affect the healthy growth of plants and products. *Pseudocercospora psidii* is an asexual ascomycetes fungus [1] that has been found to cause serious harm to crops and fruits such as eggplants [2], guavas [3], and bananas [4], causing leaf browning [5]. Anthrax is mainly caused by *Colletotrichum eriobotryae* and *Colletotrichum siamense*, which are difficult to remove [6,7]. They frequently parasitize *Ziziphus jujuba* Mill [8], chili [9], avocado [10], and other plant leaves and fruit surfaces.

With the growth of new plants, illnesses continue to be propagated and spread by rain, insects, and other means, culminating in an irreversible condition [11] and causing widespread sores and infections. Eventually, the fruit wilts, and the leaves wither, resulting in significant economic waste [12]. The most efficient technique to manage illnesses is to spray pesticides to kill hazardous germs. However, according to studies, these chemical synthetic pesticides promote fungal disease resistance [13], and pesticide residues and environmental pollution are major concerns caused by pesticide usage [14]. The primary drawbacks of synthetic pesticides include increased hidden hazards, unexpected effects, and incapacity to deal with disease outbreaks [15,16].

Natural plant essential oils (EOs) are hydrophobic and consist of a range of volatile chemicals [17]. The chemical makeup of essential oils is heavily influenced by geographical region, plant location, soil type, plant water content, and even extraction methods. Because essential oils are diverse, their fungal inhibition effect does not rely on a single mechanism but rather on a number of various routes and processes [18]. The synergistic effect of the chemicals in essential oils can cause leaking of the fungus's cell membranes, cytoplasmic material, and ions, causing the fungus to die. *Melaleuca bracteata* is an aromatic plant of the genus *Myrtle* native to Australia and is now widely dispersed around the world [19], including Indonesia, South Asia, and southern China [20,21].

There is a distinct perfume emanating from the branches and leaves of *Melaleuca bracteata*, and prior research has indicated that certain parts of the plant are abundant in fragrant essential oils. In addition to being utilized as a premium plant scent, *Melaleuca bracteata*'s branches and leaves are used to extract essential oils, which are today one of the most widely used and very valuable ingredients in Europe for the production of perfumes. Furthermore, it has recently been demonstrated that *Melaleuca bracteata* possesses a wide range of pharmacological activities both in vivo and in vitro, including effects on tumor inhibition [22], antibacterial [23], anti-inflammatory [24], skin disease, stroke, wound infection, and antiulcer [25]. Aromatic agents and fungal inhibition agents [24] are widely distributed in the leaves of *Melaleuca bracteata* (LMB). Martina B et al. researched the ecology and biology of essential oils and their fungal inhibition activity and found that essential oils have potential antifungal and antibacterial effects on organisms that harm plant or human health [26].

An energy-wasting traditional method for extracting essential oils is steam distillation; the stability and content of essential oils will also be impacted to differing degrees by the prolonged exposure to high temperatures. Research focus is currently being paid to a number of challenges, including energy conservation, environmental pollution, and global ecological warming (carbon emissions). Therefore, there is an urgent need for an extraction process with reduced resource consumption and higher ecological sustainability in order to ensure the production, composition, and activity of essential oils.

The plant cell wall is a low-permeability tissue. To improve the extraction impact of essential oils, the salting-out method is frequently employed as a pretreatment method for fresh raw materials. This approach will break the plant cell wall in the leaves and then increase the release of volatile chemicals [27]. Avelina Franco-Vega et al. used a salt-out method to pretreat leaves [28], which had the effect of destroying the plant cell wall, thus increasing the essential oil extraction rate. Zaizhi Liu et al. used a solvent-free microwave-assisted method to pretreat fresh leaves, which had the effect of increasing extraction efficiency [29]. The "cavitation effect", tissue fragmentation, solvent shuttle, and deoxidation are the mechanisms behind ultrasonic action. Using ultrasonic can increase essential oil yield quickly.

In order to pretreat raw materials collaboratively, green ultrasonic pretreatment and the addition of a cellulase-inorganic salt mixed aqueous solution can hasten the release of active ingredients in plant tissues, lower liquid surface tension, enhance the osmotic pressure of active ingredients in cells, and increase the dispersion of substances in the solution. The above method can be used to more correctly assess essential oil yield while preventing hydrosol production in high-temperature water, which can lead to loss of essential oil and water contamination. The solution containing metal ions may effectively decrease the boiling time, minimize the heating gradient, and accomplish the boiling effect in a quick period during the microwave radiation process. Although lithium chloride is a good extraction catalyst, it is rarely cited or employed [30]. We speculate that microwave irradiation of an aqueous solution containing inorganic salt can rapidly absorb microwave energy and hence decrease the extraction time. Furthermore, cellulase pretreatment may efficiently degrade the cell wall in plant leaves, increasing the degree of cell wall damage. HeLa Mahmoudi et al. employed enzymatic pretreatment to boost essential oil output [31].

The application of cellulase-inorganic salt solution for raw material treatment, as well as in LMB essential oil extraction, has not yet been documented. We believe that combining the exceedingly effective salting-out method with the selective enzyme catalysis method may yield unexpected results. Additionally, we used three distinct fungi (*Pseudocercospora psidii*, *Colletotrichum eriobotryae*, and *Colletotrichum siamense*) as research subjects and verified the LMB essential oil's inherent inhibitory impact on fungi, in the hopes of offering a useful guide for real applications.

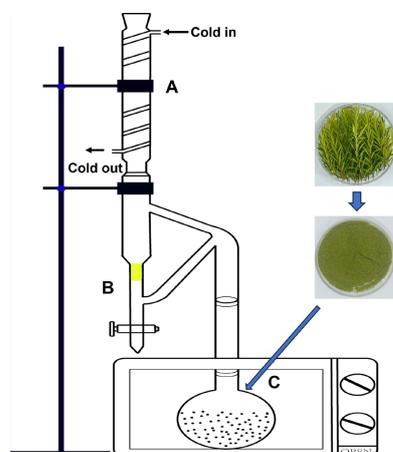
## 2. Material and Method

### 2.1. Materials

The fresh raw materials were gathered from Jiyang University (Meizhou, China), and the collected leaves are dried in a dry and ventilated place to remove moisture. We purchased anhydrous versions of dimethyl sulfoxone, anhydrous sodium sulfate, anhydrous sodium chloride, and anhydrous ethanol from Shandong Aowei Chemical Co., Ltd. in Jinan, China (purity  $\geq 98\%$ ). Tianjin Kemiou Chemical Reagent Co., Ltd. (Tianjin, China) supplied the chromatographically pure N-hexane that was used. *Pseudocercospora fungus*, *Colletotrichum siamense*, and *Colletotrichum eriobotryae* were obtained from the China Fungus Preservation Center.

### 2.2. Essential Oil Extraction

A desktop numerical control ultrasonic cleaner (KQ-100DE, Kunshan, China; 300 mm  $\times$  150 mm  $\times$  100 mm) was employed for the ultrasonic preprocessing. This apparatus had two Centro symmetric transducers with a 45 kHz operating frequency, 40–100 W of programmable power, and 20–100 °C of operating temperature. To determine the water content, we first dried fresh LMB for 24 h at a steady temperature and calculated water content. At the beginning, 100 g of fresh LMB was first ground into a 20-mesh powder and then placed in a flask with a round bottom. The raw material was then soaked in catalyst solution (with a liquid–solid ratio range of 3–12) between 0–75  $\mu\text{mol}$  (the ratio of cellulase to lithium chloride was 1:1); the aforementioned solution is treated in an ultrasonic environment with 100 W of power and 50 °C of temperature for 5 h. The essential oil was separated using microwave-assisted distillation under the following conditions: the microwave-assisted distillation device for extracting essential oil consisted of a P70D20N1P-G5 microwave oven (Galanz, Zhongshan, China), a reflux condenser, and a Clevenger extraction tube (Figure 1). The microwave radiation frequency of the oven was 2450 MHz, and its microwave irradiation power was continuously adjustable between 120–700 W. The reactor, a 1000 mL round-bottom flask placed in the inner chamber of the microwave oven, needed the Clevenger tube to pass through the top of the appliance and connect, with a solid–liquid ratio of 4–12 mL/g, extraction period of 10–60 min, and microwave irradiation power of 144–720 W.



**Figure 1.** Microwave-assisted extraction of essential oils (A: Circulating water condenser, B: Essential oil collector, and C: Round bottom flask-1000 mL).

### 2.3. Box–Behnken Design Optimizes the Extraction Rate of Essential Oils

To statistically optimize the essential oil yield, we use response surface methodology (Design-Expert 10) to examine the effects of each variable and level on the yield as well as the correlation between variable values and response values, including analysis of variance and fitting of quadratic regression equations.

### 2.4. Essential Oil Composition and Analysis

Gas chromatography–mass spectrometry was performed (Agilent-7000 GC/MS, Santa Clara, CA, USA), using a capillary column type, with a nonpolar capillary column (30 m × 0.25 mm × 320 μm) made of fused silica with dimethyl polysiloxane (containing 5% phenyl). A carrier gas of hydrogen was used to identify essential oil components, OV-101 0.1 mL of EO was dissolved in 500 times the volume of n-hexane, and the sample was injected at a column temperature range of 40–200 °C, a heating rate of 10 °C/min, and a column pressure of 0.8 kg/cm<sup>2</sup>, among other conditions. The temperature of gasification is 250 °C, and the electron energy is 70 eV. The compound identification was carried out through the library database and chemical composition analysis method for EO components, and data were analyzed and counted through hierarchical data packet inspection tools of chiplot [32].

### 2.5. Fungal Inhibition Activity

#### 2.5.1. Activation of Strains

To create the suspensions, we used a 6 mm diameter cake that was drilled into the center of the medium at a dense and uniform growth location. We then cultured the cake for 7 days at 28 °C in a constant temperature incubator, inoculated the activated fungal material into sterile PDB medium, and incubated the mixture for 48 h on a rotating shaking table at 28 °C (150 r/min).

#### 2.5.2. Fungal Inhibition of Essential Oil

In a 200 mL triangle bottle, sterilized PDA (89.5 mL) medium was added. The essential oil with the corresponding gradient (the essential oil's concentration range was 0.1–1.6 mg, dissolved in 0.5 mL DMSO) was then added to the PDA, mixed thoroughly, and poured into a petri dish with a diameter of 90 mm. PDA without essential oil was used as the negative control. The medium containing essential oil was created with a specific range of concentrations (the concentration range of essential oil in PDA: 0.1–1.6 mg/mL).

The 6 mm drug-sensitive paper was placed in the center of the sample plate (concentration: 0.1, 0.4, 0.8, 1.6, unit: mg/mL), and 10 μL fungal suspension was added to a tablet, which was sealed and cultivated in a 28 ± 1 °C electric thermostatic incubator. The procedure took 3 days. Tablets with varying amounts of essential oils served as experimental groups, while tablets without essential oils served as negative controls. The colony diameter was measured using the cross-crossing method, and the experiment was performed three times for each group, with the average value taken. The calculation method of the inhibition index:

$$II(\%) = \left(1 - \frac{Da}{Db}\right) \times 100\% \quad (1)$$

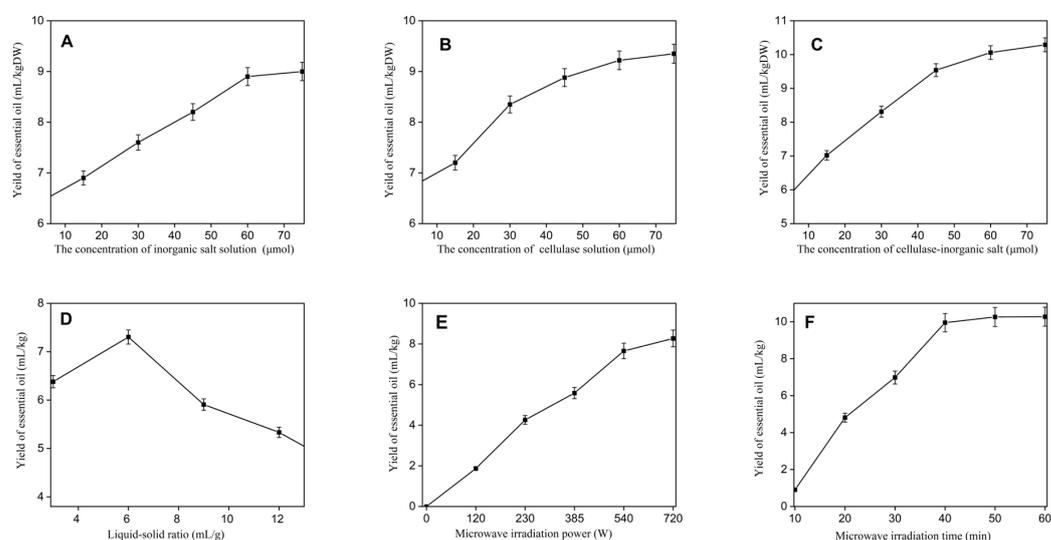
*II*: Inhibition index, *Da*: Fungal growth during treatment (mm), *Db*: Fungal growth in the control group (mm).

## 3. Results and Discussion

### 3.1. Influence of Various Factors on Extraction Rate

As indicated in Figure 2, raw materials were soaked and pretreated with a lithium chloride inorganic salt solution, cellulase solution, and cellulase-lithium chloride (1:1) combination solution (treatment temperature 50 °C, solution concentration range 5–75 μmol). The essential oil yield of all treated samples increased in a positive relationship with the concentration of catalyst. As demonstrated in Figure 2C, the yield of essential oil fol-

lowing cellulase-inorganic salt treatment was the highest of all samples, at 12.09 mL/kg DW (dry weight). The combined catalyst showed a stronger extraction effect than the other two samples, which had the greatest essential oil yields of 9.01 mL/kgDW and 9.35 mL/kgDW, respectively.



**Figure 2.** Effect of factors on the yield of essential oil. The above subgraphs show the effects of (A) inorganic salt concentration, (B) concentration of cellulase solution, (C) cellulase-inorganic salt concentration, (D) liquid-solid ratio, (E) microwave irradiation power and (F) microwave irradiation time on yield of essential oil.

We hypothesized that since cellulase and lithium chloride ions synergistically increased the damaging impact on the plant cell wall, essential oil components escaped, resulting in an increase in essential oil output. Lithium chloride is a metal catalyst that may improve the activity of cellulase. Chen et al. investigated the synergistic impact of metal ions and enzymes on cell wall flocculation [33]. A modest combination of cellulase and inorganic salts can act together to boost yields by raising the boiling point of water in plant cells and removing fiber networks in plant cell walls. Ultrasonic technology and combined water bath container-ultrasonic device approaches are the two forms of ultrasound-assisted surgery. To save energy, the ultrasonic probe can be positioned in close contact with the substance. There aren't many reports of combining normal ultrasound-assisted extraction technology with microwave technology to extract essential oils, despite the fact that this method is frequently employed in the process of essential oil separation.

We evaluated the effect of the liquid–solid ratio, microwave irradiation power, and microwave duration on essential oil production. Figure 2D depicts the impact of several liquid–solid ratios (3–12 mL/g) on the yield of three EOs. The essential oil output was 6.38 mL/kg DW when the liquid–solid ratio was 3. The essential oil yield was maximum when the liquid–solid ratio reached 6, reaching 7.30 mL/kg DW, and subsequently dropped as the liquid–solid ratio increased. As a result, a liquid–solid ratio of 6 is chosen as the ideal. It has been found that increasing the liquid–solid ratio decreases the mass transfer resistance and boosts the essential oil yield; however, increasing the liquid–solid ratio causes the forced loss of chemical components in essential oils. We assume that the increased liquid–solid ratio caused some of the essential oil to dissolve in the water, causing the extraction rate to vanish.

Figure 2E depicts the effect of various power ranges (230–700 W) on the experiment. The maximum essential oil obtained when the microwave power exceeds 700 W is 8.27 mL/kgDW. We chose the yield of essential oil produced at 540 W (7.65 mL/kg DW) as the level for the following experiment based on the regular connection between the cost and yield of essential oil. Microwave irradiation at specific levels can accelerate the extraction process and enhance the release of essential oils [34], but utilizing too much power might

result in fast temperature fluctuations that impact the yield and quality of essential oils. Figure 2F depicts the influence of the microwave irradiation time on the extraction rate. The increasing rate of essential oil yield is roughly the same once the extraction duration approaches 50 min. This phenomenon might be induced by uneven internal heating of the material over a short period of time. Long-term microwave irradiation causes osmotic pressure inside the material, and high-frequency vibration is produced as a result of water conduction between the materials during microwave irradiation. As a consequence of additional experimental adjustment, 50 min (10.27 mL/kgDW) was chosen. Since the yield in the trials is mostly unaffected by the liquid–solid ratio, we decided to base future optimization on the microwave power and time.

### 3.2. The Extraction Rate of Essential Oil Was Optimized by Response Surface

To create and refine the statistical analysis of EO yield, we investigated the ideal process parameters for each element in the test range. Through the optimized outcomes, we discovered that  $p = 0.0011$  was a significant value. The response surface data are in good agreement with the experimental results, and the C, AB, and BC items have no significant influence ( $p > 0.05$ ). The lack of fit term is 0.0917,  $R^2$  is 0.9466, CV is 5.45%, unknown factors had little impact on the experiment, the error was within the acceptable range, and the response surface data were in good agreement with the experimental results (Table 1). As a result, the examination of the response surface model yields the following theoretically optimum conclusion: the greatest production of essential oil was 9.73 mL/kg DW when all the indices achieved the optimal values, which occurred when the microwave irradiation power was 488.54 W, the microwave irradiation period was 48.21 min, and the lithium salt addition was 64.51  $\mu\text{mol}$ .

**Table 1.** Box–Behnken design with experimental value for total yield of EO, and analysis of variance (ANOVA) for response surface quadratic model.

Run	BBD Experiments				ANOVA							
	A (W)	B (min)	C ( $\mu\text{mol}$ )	$Y_{\text{EO}}$ (mL/kg)	Source of Variation	Sum of Squares	Degree of Freedom	Mean Square	F Value	p Value		
1	576	50	60	9.31	Model	24.59	9	2.73	13.78	0.0011 *		
2	576	50	60	9.82	A	1.44	1	1.44	7.29	0.0306		
3	576	50	60	9.73	B	3.51	1	3.51	17.71	0.0040		
4	720	60	60	6.98	C	0.061	1	0.06	0.31	0.5956		
5	720	40	60	7.92	AB	0	1	0.00	0.00	1.0000		
6	720	50	75	6.00	AC	4.41	1	4.41	22.25	0.0022		
7	576	40	45	8.11	BC	0.20	1	0.20	1.02	0.3458		
8	720	50	45	8.22	A <sup>2</sup>	2.78	1	2.78	14.02	0.0072		
9	432	50	45	7.63	B <sup>2</sup>	6.98	1	6.98	35.21	0.0006		
10	576	60	75	6.21	C <sup>2</sup>	3.70	1	3.70	18.67	0.0035		
11	576	50	60	9.90	Residual	1.39	7	0.19				
12	432	50	75	9.60	Lack of fit	1.07	3	0.36	4.45	0.0917		
13	576	60	45	6.90	Pure error	0.32	4	0.08				
14	432	60	60	7.16	Cor total	25.98	16					
15	576	40	75	8.36	Credibility analysis of the regression equations							
16	432	40	60	8.18	Index mark	SD	Mean	C.V.%	R <sup>2</sup>	Adjust R <sup>2</sup>	Predicted R <sup>2</sup>	AP
17	576	50	60	9.32	Y	0.4452	8.17	5.45	0.9466	0.8779	0.3232	9.7009

SD: standard deviation; AP: adequacy precision. \*: Significant.

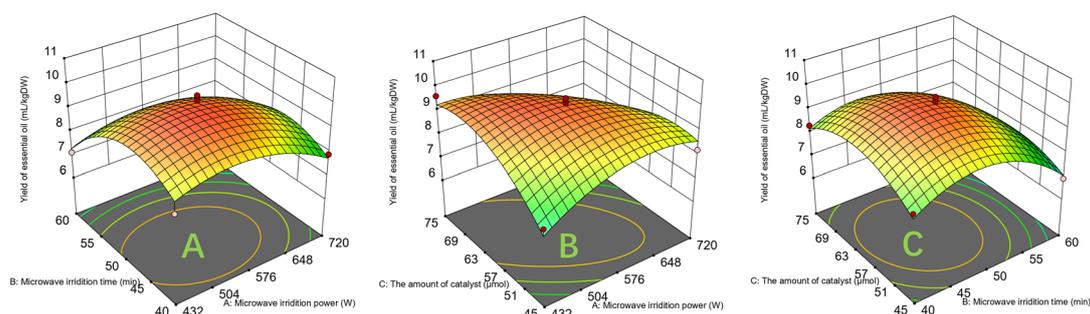
The regression equation of the response surface analysis fitted data is as follows:

$$Y = 9.6 - 0.425A - 0.6625B - 0.088C - 1.05AC - 0.225BC - 0.8125A^2 - 1.29B^2 - 0.9375C^2 \tag{2}$$

The 3D response surface, also known as the 3D fitting surface, is the response surface of the microwave irradiation duration, microwave irradiation power, and EO yield of the catalyst addition. In general, there was a favorable link between the yield of essential oil and the expansion of the microwave irradiation power and duration as well as the inclusion

of the catalyst. The yield of EO steadily increased as the microwave power and duration increased; however, as time increased, the yield of EO displayed a curve.

The maximum EO extraction rate (8.9 mL/kgDW), as shown in Figure 3A, occurs when the microwave period is 47 min and the microwave power is 480 W. The yield of EOs was influenced quadratically by the strength of the microwave irradiation and the quantity of catalyst supplied, as shown in Figure 3B. The EO yield was maximum (9.41 mL/kgDW) when the catalyst addition was 60  $\mu$ mol and the microwave power was 650 W. The interaction effects of catalyst addition quantity and microwave irradiation period on EO yield are shown in Figure 3C. The yield of EOs rose as the catalyst concentration increased between 46–60  $\mu$ mol; however, when the catalyst concentration grew as a percentage, the liquid–solid ratio continued to rise after reaching its ideal level. The essential oil extraction technique maintains a high temperature in the microwave oven by increasing the microwave power or microwave duration.



**Figure 3.** The essential oil was optimized by Box–Behnken design. (A) The interaction effects of microwave power and time on the yield of essential oil, (B) the interaction effects of catalyst addition amount and microwave power on the yield of essential oil, and (C) the interaction of catalyst addition amount and microwave irradiation time on essential oil yield).

### 3.3. Analysis of Essential Oil by Gas Chromatography–Mass Spectrometry

Forty-one compounds were discovered when the acquired essential oils were first dried and analyzed using GC–MS (Table 2). Methyl eugenol (76.53%) makes up most of the essential oil’s composition, followed by methyl cinnamate (12.62%). The compounds together make up the entire composition of 97.82%. Methyl eugenol, the primary ingredient in a spice recipe, is the principal component of essential oil. It is also a crucial raw material for the synthesis of chemical compounds. As a result, using GC–MS detection, the essential oil of LMB has the potential to be useful in food safety and clinical treatment [35].

We made a phylogenetic tree map for the classification of different components in essential oils (as is shown in Figure 4), which was categorized based on the composition of various chemical components through cluster analysis, in order to have a more thorough understanding of the composition classification of these compounds; it is easy to observe that alcohols, monoterpenoids, sesquiterpenoids, aldehydes, ketones, and esters make up essential oils. An amount of 13 sesquiterpenoids make up 2.49% of the overall composition of essential oils. Eight different types of alcohol made up 3.71% of the total. The overall oil content was made up of four ketones, which made up 76.63% of the oil content, and five aldehydes, which made up 0.16%. There are two monoterpenoids (0.04%), and three ester compounds (13.1%) were also present. Sesquiterpenoids and alcohols comprised the greatest number of species among all the components that were extracted (Table 2). Ketones (76.63%) and esters (13.1%) had the highest concentrations. Methyl eugenol is a significant natural flavoring ingredient with a variety of applications in the food and medicine field. In addition to having biological activities such as antioxidant and antibacterial properties, it also has a spicy and warm scent that is frequently utilized in flavoring agents and perfumes [36]. Methyl cinnamate is an ester molecule that is a typical food flavoring and perfume addition that provides a fresh fruit scent [37].

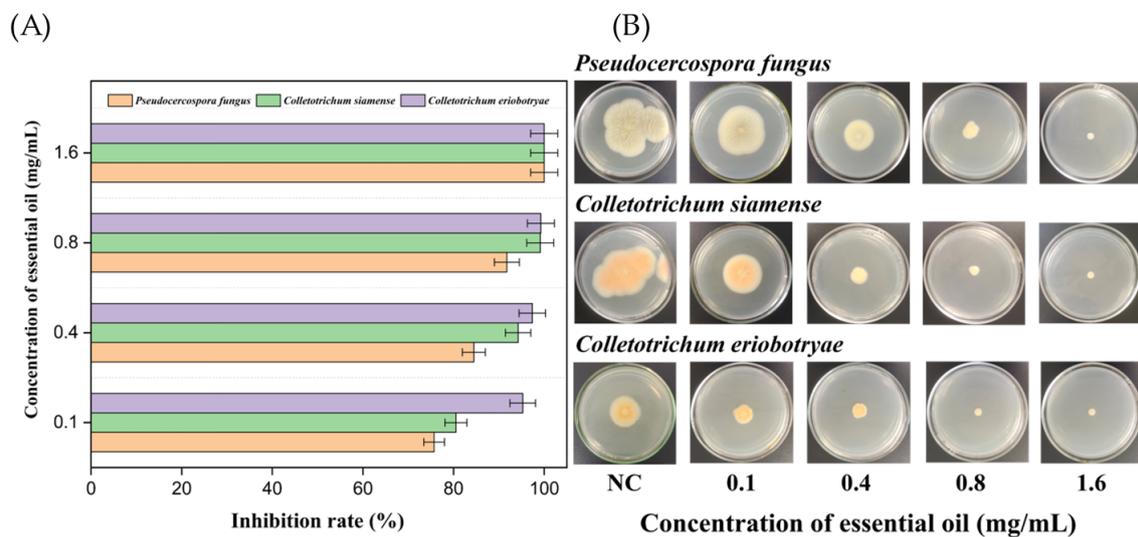


### 3.4. Comparison with Reference Techniques

The conventional approach for extracting essential oil from water using microwave-assisted hydrodistillation involves directly irradiating the material with a microwave; however, this method is not successful for homogenizing the material due to the absence of an ultrasonic device. By using a high-frequency microwave field instead of the conventional microwave heating extraction method, we are able to fully collide the material within the molecule under the action of the microwave, allowing electromagnetic energy to enter as heat energy. This improves the ability to extract material, speeds up temperature, and yields excellent extra essential oil. The extraction efficiency of this study is much greater than that of the conventional approach when compared to the two experimental procedures mentioned above.

### 3.5. Study on Antifungal Activity of Essential Oils

Methyl cinnamate also possesses anti-bacterial, anti-inflammatory, and other biological properties. Due to its distinct scent and specific biological activity, the essential oil of LMB has several potential uses in the culinary, taste, and medical industries, among other disciplines [38]. Figure 5 illustrates the impact of LMB essential oil on the development of fungal fluids. The essential oil clearly inhibited the growth of three different types of fungal fluids. The growing area of *Pseudocercospora psidii* increased with culture and growth time expansion. After 72 h of cultivation, *Pseudocercospora psidii* exposed to 0.1 mg/mL essential oil showed some inhibition compared to the blank control, with an inhibitory rate of 75.7%. Under the same conditions, the inhibitory activity of essential oil on *Colletotrichum* was the best, reaching 95.26%. The activity of inhibiting *Colletotrichum* reached 80.51%. The three fungi were significantly affected by the essential oil solution when the concentration was 1.6 mg/mL; at this dosage, the fungus could not live, and the inhibition rate reached 100%. The growth of the pathogens *Colletotrichum eriobotryae* and *Colletotrichum siamense* was considerably suppressed by the LMB essential oils and were more susceptible than *Pseudocercospora psidii*.



**Figure 5.** Inhibitory effect of essential oils on fungi. (A) Rate of inhibition; and (B) vary concentration of essential on the inhibit effects of fungi.

Methyleugenol has been shown to have potent fungal inhibition action in earlier research hence, it is hypothesized that these highly lipophilic compounds are the primary cause of inhibition activity in this work. The lipophilic characteristics of these aliphatic chemicals are sufficient to split and alter the structure of lipophilic lipids in mitochondria and the cell plasma membrane, causing leakage of the contents of fungal cells. This is how this inhibition mechanism works. The results of the inhibition studies indicate that the essential oil has tremendous potential for preventing plant infection and preserving fruits

and vegetables, as well as for use as a natural pesticide. In addition to variations in EO components, variations in fungal test procedures may affect the findings of essential oil inhibition activities in diverse research. After the fungal culture experiment, the fungus in the negative control (NC) environment expanded rapidly, and the inhibitory impact on the fungi gradually became stronger as the essential oil content increased (Figure 5B). The fungus entirely vanished at a dosage of 10 mg/mL essential oil. This capability can be connected to the quantity of hydrophobic chemical groups in the oil, which prevent fungal cells from retaining water and cause their demise. Additionally, the primary component of EO, methyl eugenol, may also be the key factor contributing to fungal mortality. EOs are frequently used to enhance the flavor and quality of food. This further suggests that EOs are regarded as safe by the US Food and Drug Administration when used as food additives. According to preliminary tests of antimicrobial activity and a study of the literature on ecological and biological features by Martina B et al., essential oils (EOs) may have antifungal effects on a range of organisms that may affect plants or human health substances. Numerous studies have demonstrated that applying plant components can lessen the severity of illnesses in a variety of crops, and using essential oils (EOs) is a useful strategy for managing bacteria and fungi.

#### 4. Conclusions

In this work, we described the ultrasonic pretreatment combined with microwave-assisted hydrodistillation for extraction of essential oil from LMB; three representative fungi were used as references to verify the functionality of EO. The response surface BBD test was used to optimize the extraction rate. The greatest production of essential oil was 9.73 mL/kg DW when all the indices achieved the optimal values, which occurred when the microwave irradiation power was 488.54 W, the microwave irradiation period was 48.21 min, and the lithium salt addition was 64.51  $\mu$ mol. Methyl eugenol, methyl cinnamate,  $\alpha$ -terpineol, etc., are the principal substances discovered by GC–MS. *Pseudocercospora psidii*, *Colletotrichum eriobotryae*, and *Colletotrichum siamense* were significantly inhibited by a 1.6 mg/mL essential oil solution, according to the researchers. This suggests that EO is a promising substitute for synthetic pesticides; however, toxicological testing is needed before using these natural products as a natural antibiotic to ensure that they do not harm humans or plants and to better understand the mechanisms by which essential oils affect organisms.

**Author Contributions:** Conceptualization, X.Z. and Y.H.; methodology, Y.H.; software, Y.H.; validation, X.Z., Y.H. and Z.L.; investigation, J.C.; resources, J.C.; data curation, F.Z.; writing—original draft preparation, X.Z. and F.Z.; writing—review and editing. All authors have read and agreed to the published version of the manuscript.

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**Conflicts of Interest:** The authors declare no conflict of interest.

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