

Figure S1. Comparison of the ^1H -NMR spectra of NQ, CQ and HQ Ligands with the amine (quinoline-3-carbohydrazide, PQ).

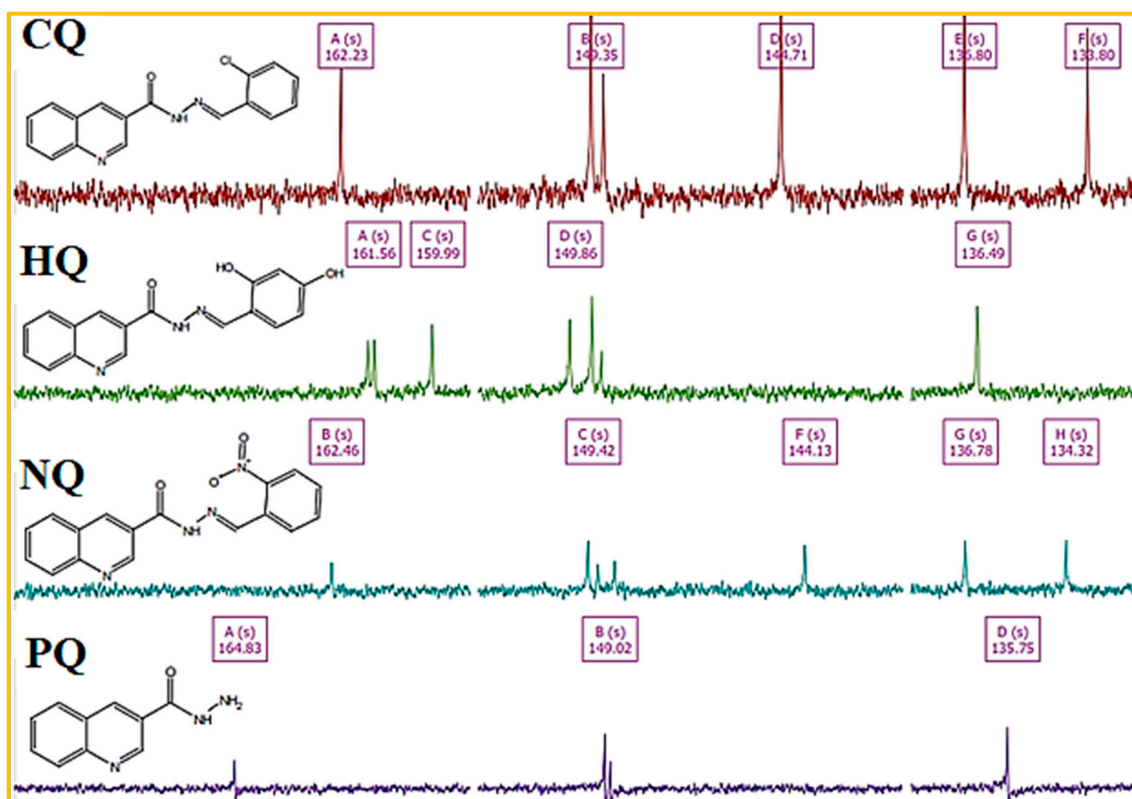


Figure S2. Comparison of the ^{13}C -NMR spectra of NQ, CQ and HQ Ligands with the amine (quinoline-3-carbohydrazide, PQ)

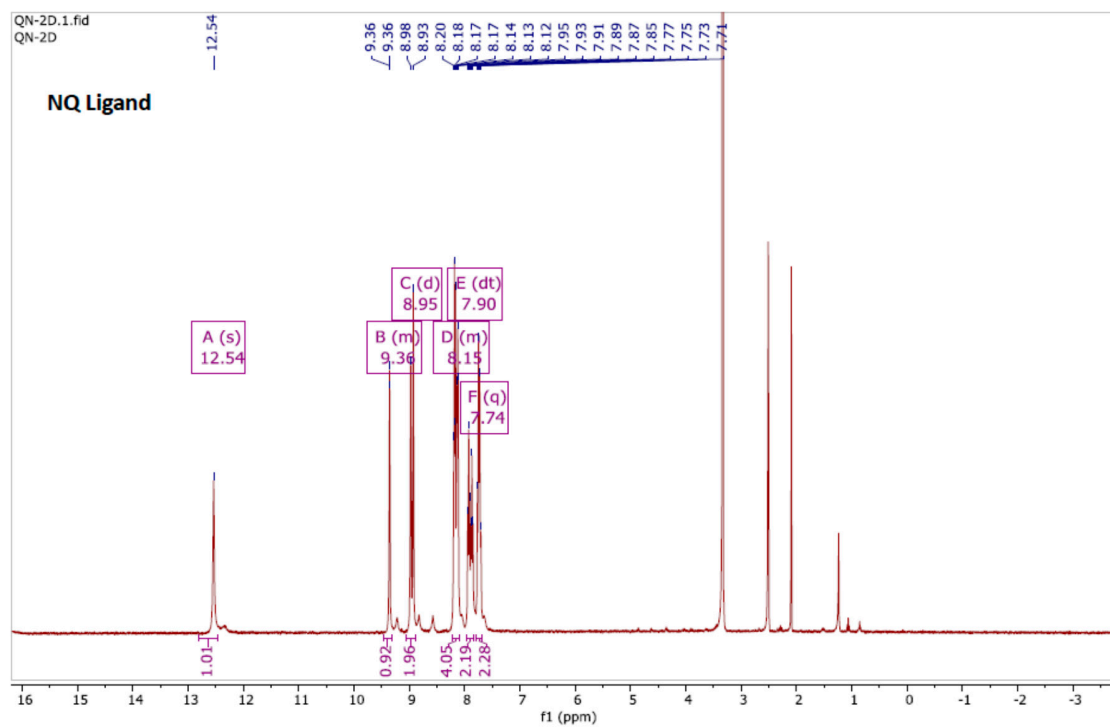


Figure S3. ^1H -NMR spectrum of NQ ligand.

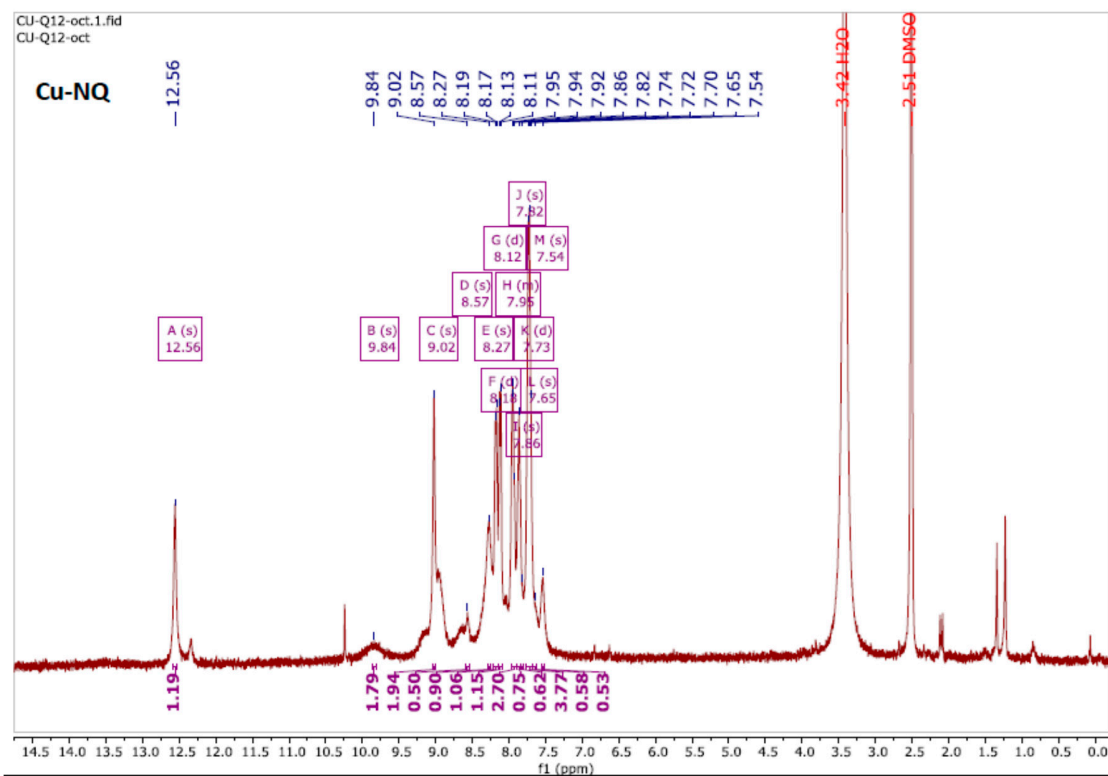


Figure S4. ^1H -NMR spectrum of Cu-NQ complex.

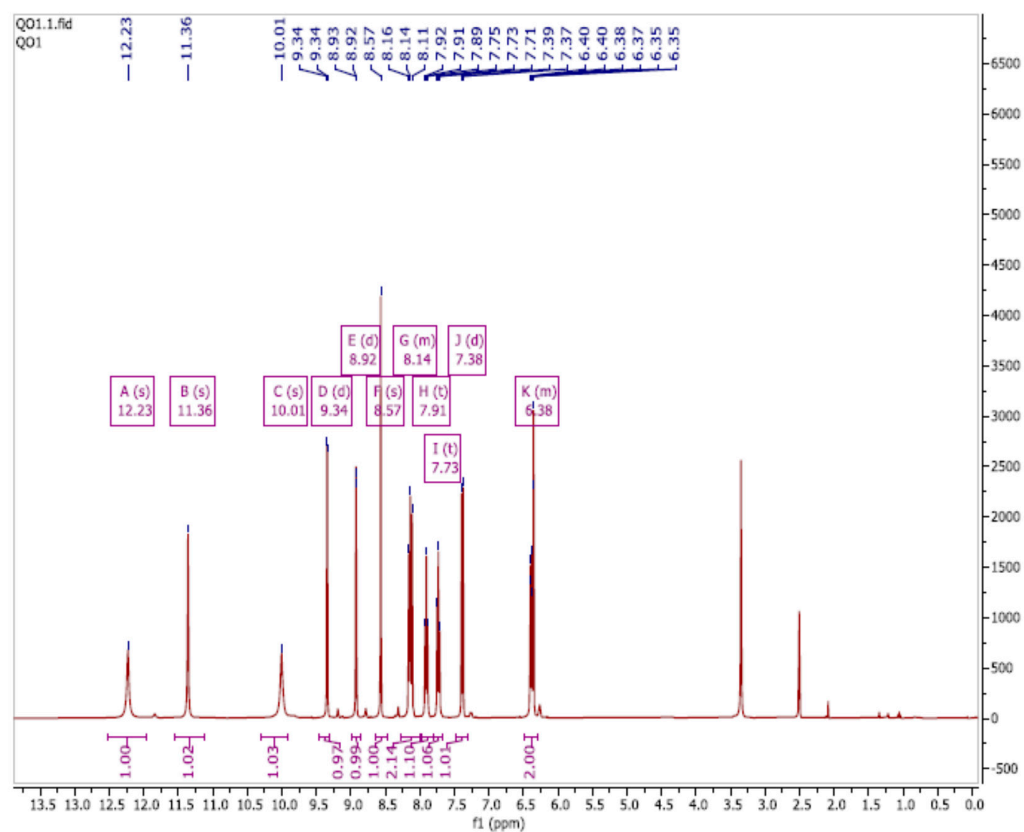


Figure S5. ^1H -NMR spectrum of HQ ligand.

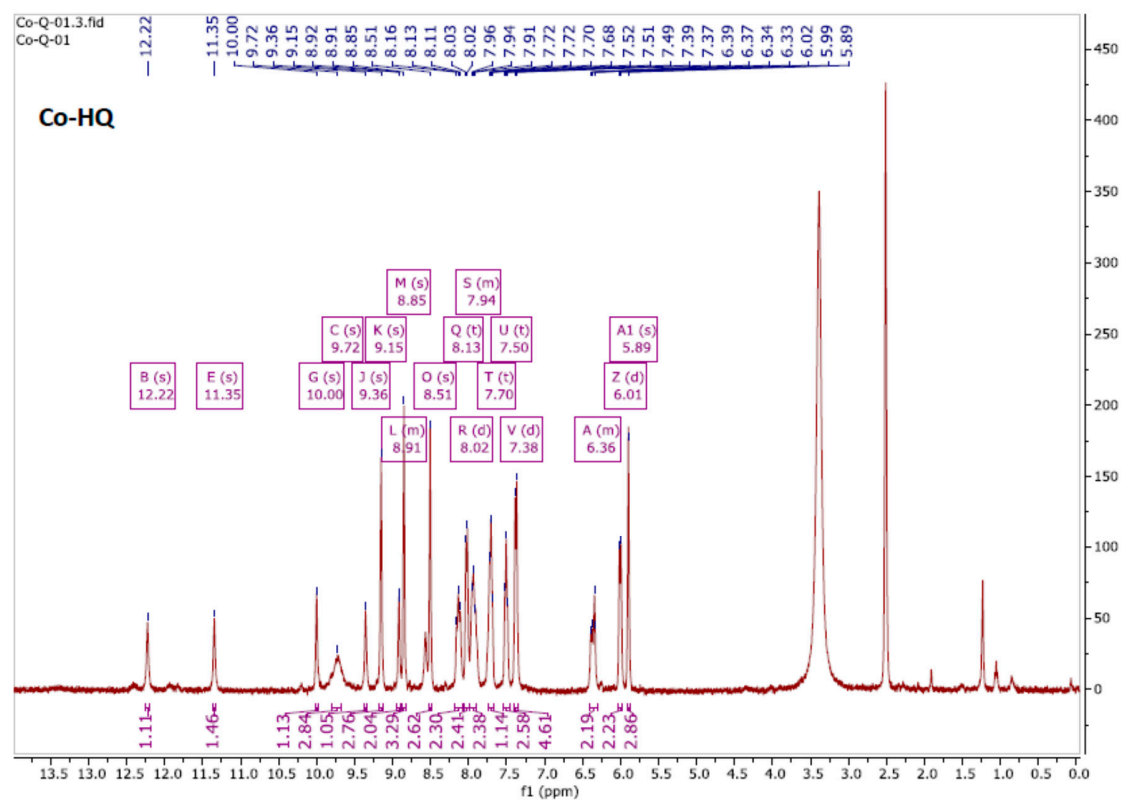


Figure S6. ^1H -NMR spectrum of Co-HQ complex.

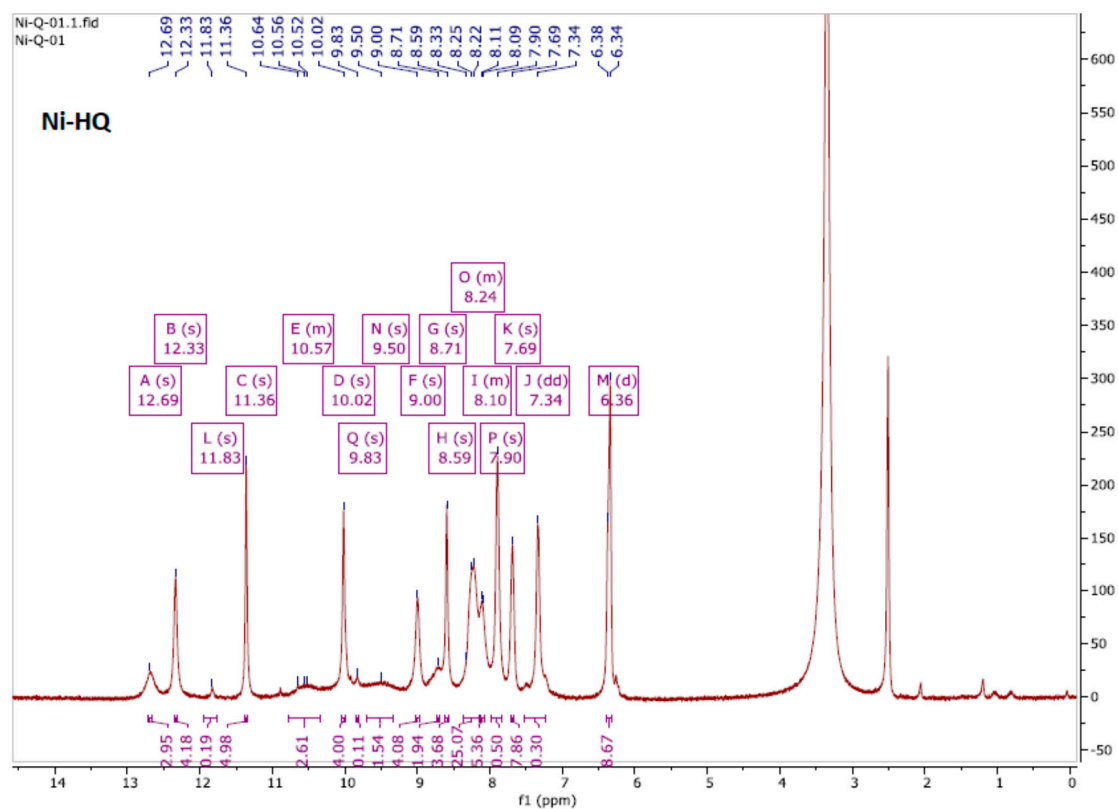


Figure S7. ^1H -NMR spectrum of Ni-HQ complex.

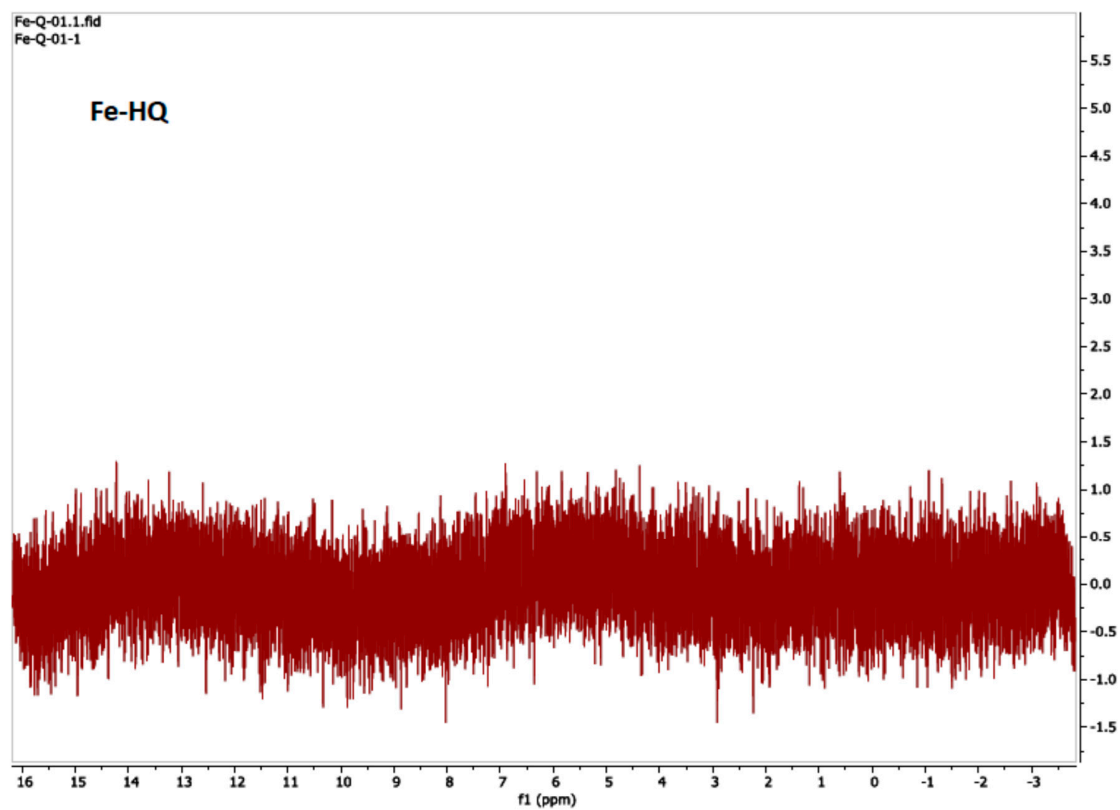


Figure S8. ^1H -NMR spectrum of Fe-HQ complex.

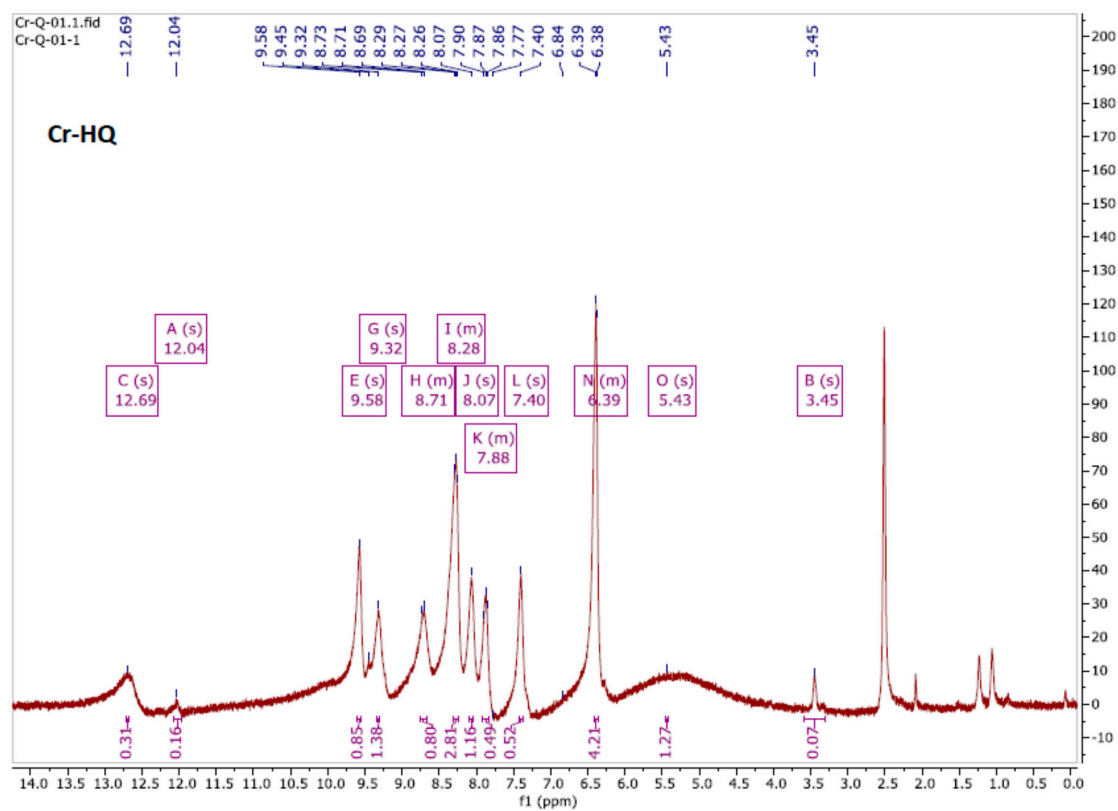


Figure S9. ^1H -NMR spectrum of Cr-HQ complex.

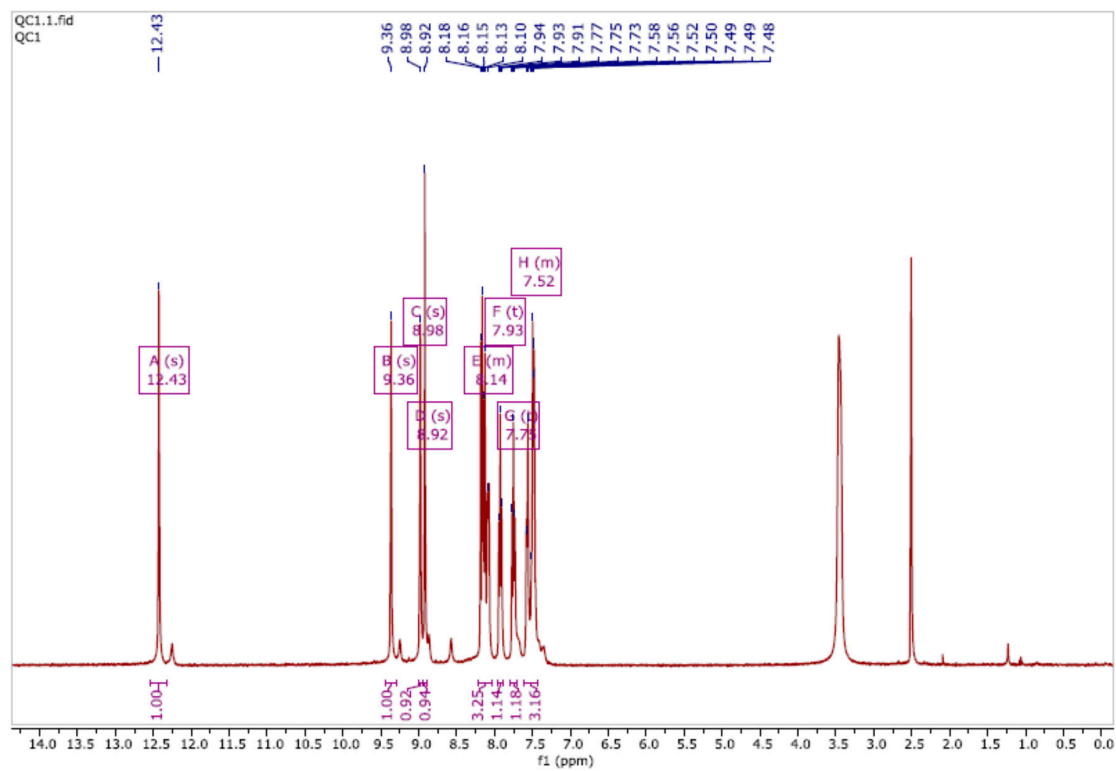


Figure S10. ^1H -NMR spectrum of CQ ligand.

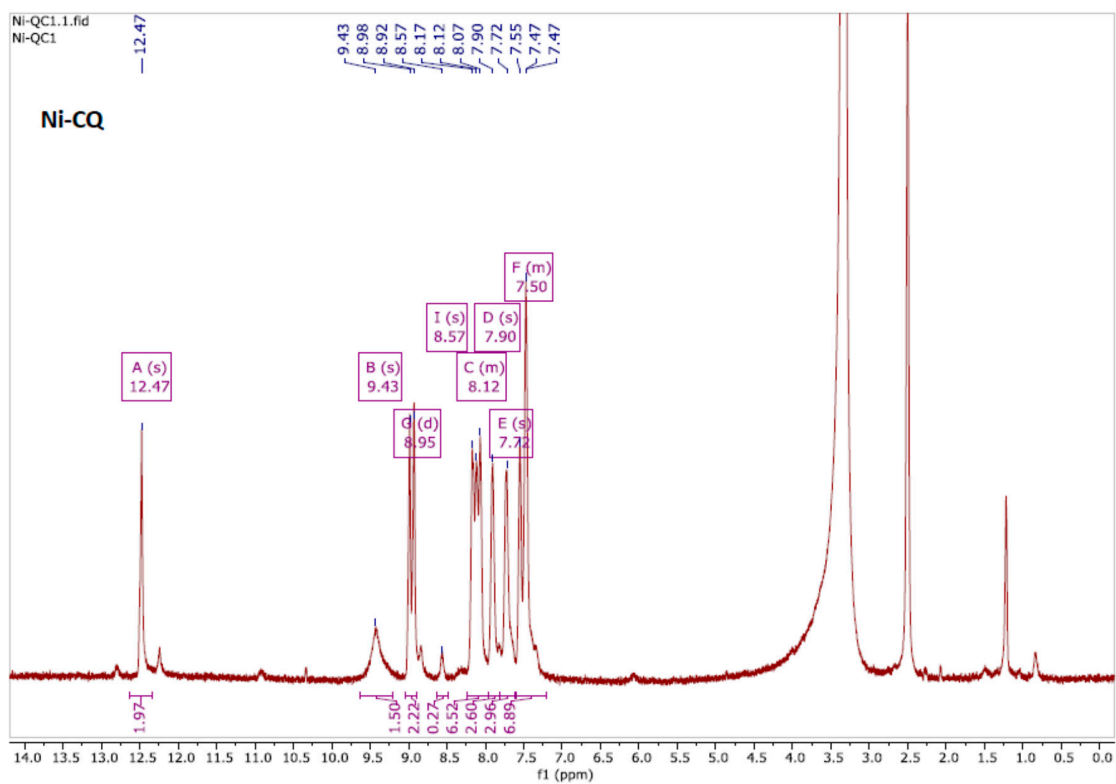


Figure S11. ^1H -NMR spectrum of Ni-CQ complex.

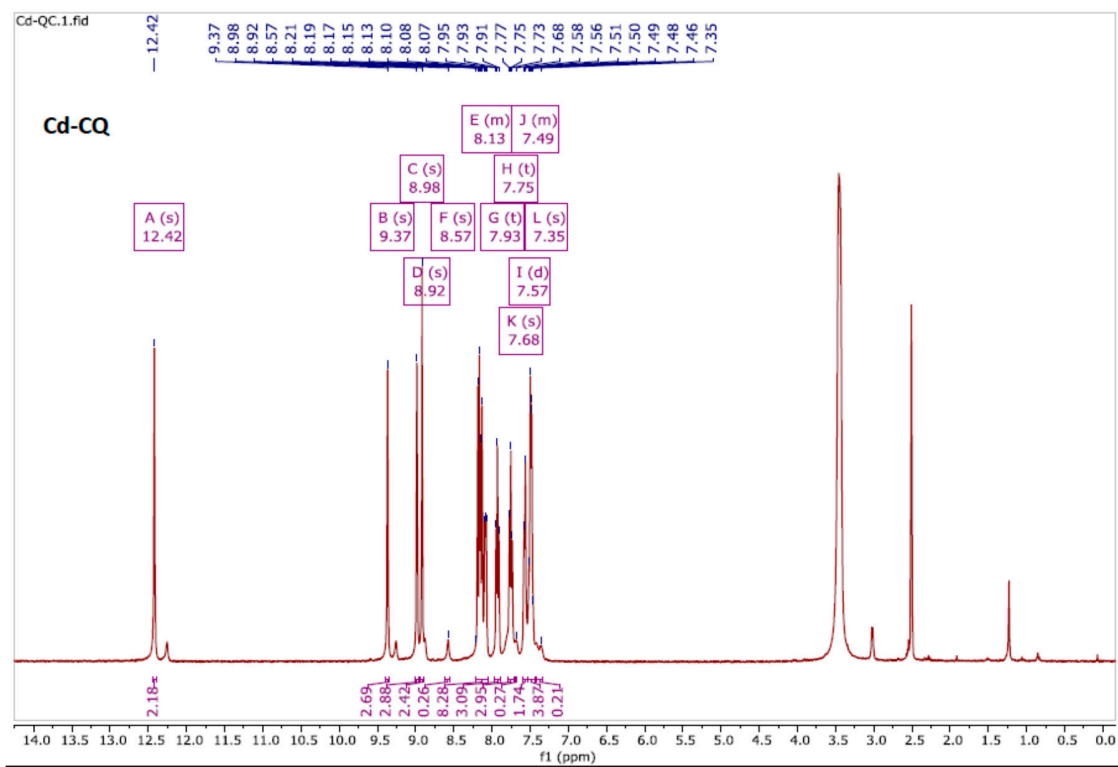


Figure S12. ^1H -NMR spectrum of Cd-CQ complex.

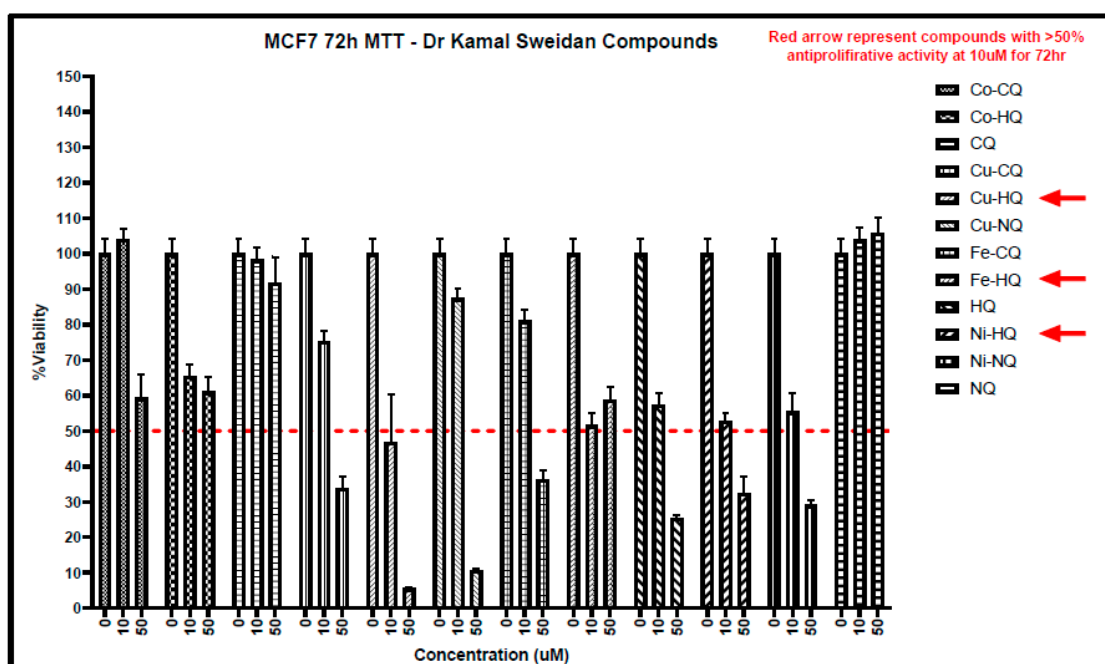


Figure S13. The 50% inhibitory concentration (IC₅₀) values for tested compounds treatment in MCF7 cell lines. Experiments were carried out for 72 hours treatment duration. Experiments were run in triplicates for at least three independent trials (n=6). Concentrations are expressed in micromolar. NA: not applicable; SD: standard deviation; h: hour; μ M: micromolar.

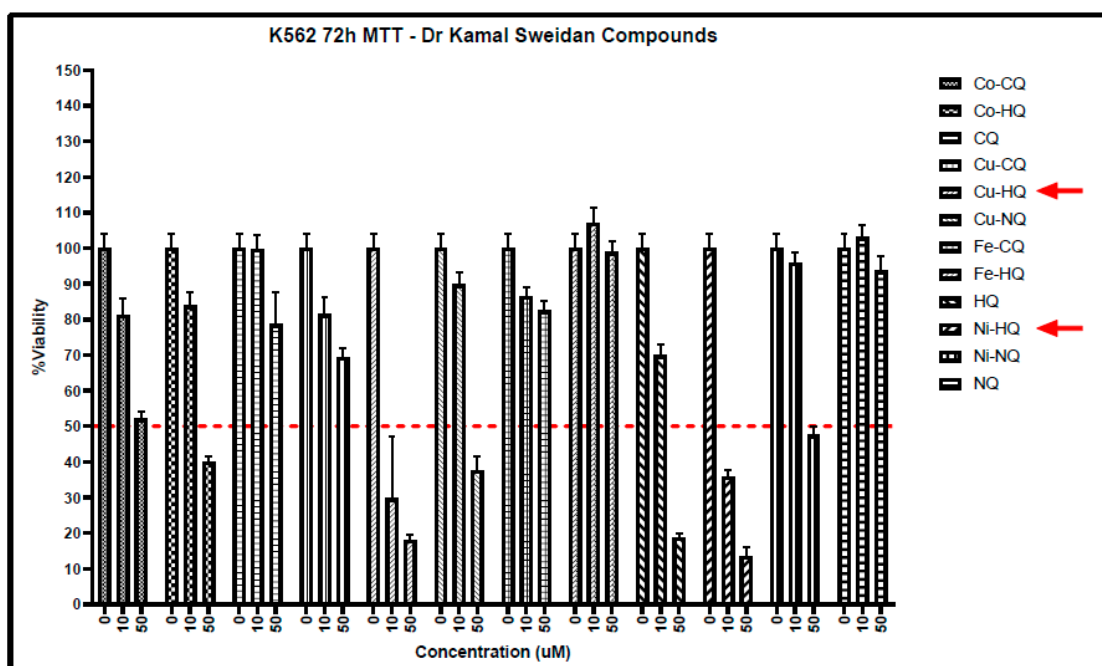


Figure S14. The 50% inhibitory concentration (IC₅₀) values for tested compounds treatment in K562 cell lines. Experiments were carried out for 72 hours treatment duration. Experiments were run in triplicates for at least three independent trials (n=6). Concentrations are expressed in micromolar. NA: not applicable; SD: standard deviation; h: hour; μ M: micromolar.

Table S1. Crystal data and structure refinement for CQ·H₂O.

| | | |
|-----------------------------------|---|----------|
| Identification code | 2112coqc | |
| Empirical formula | C ₁₇ H ₁₄ ClN ₃ O ₂ | |
| Formula weight | 327.76 | |
| Temperature | 190(2) K | |
| Wavelength | 1.54184 Å | |
| Crystal system | Orthorhombic | |
| Space group | <i>Pbca</i> | |
| Unit cell dimensions | a = 12.7287(6) Å | α = 90°. |
| | b = 12.9558(5) Å | β = 90°. |
| | c = 18.8331(12) Å | γ = 90°. |
| Volume | 3105.8(3) Å ³ | |
| Z | 8 | |
| Density (calculated) | 1.402 Mg/m ³ | |
| Absorption coefficient | 2.294 mm ⁻¹ | |
| F(000) | 1360 | |
| Crystal size | 0.300 x 0.200 x 0.100 mm ³ | |
| Theta range for data collection | 4.696 to 61.648°. | |
| Index ranges | -14 ≤ h ≤ 13, -13 ≤ k ≤ 14, -21 ≤ l ≤ 21 | |
| Reflections collected | 8638 | |
| Independent reflections | 2409 [R(int) = 0.0536] | |
| Completeness to theta = 61.648° | 99.5 % | |
| Absorption correction | Semi-empirical from equivalents | |
| Max. and min. transmission | 1 and 0.899 | |
| Refinement method | Full-matrix least-squares on F ² | |
| Data / restraints / parameters | 2409 / 0 / 209 | |
| Goodness-of-fit on F ² | 1.090 | |
| Final R indices [I > 2σ(I)] | R1 = 0.0495, wR2 = 0.1271 | |
| R indices (all data) | R1 = 0.0659, wR2 = 0.1383 | |
| Extinction coefficient | n/a | |
| Largest diff. peak and hole | 0.381 and -0.361 e.Å ⁻³ | |

Table S2. Bond lengths [Å] and angles [°] for CQ·H₂O.

| | |
|-----------------|----------|
| C(1)-N(1) | 1.316(4) |
| C(1)-C(2) | 1.422(4) |
| C(2)-C(3) | 1.371(4) |
| C(2)-C(10) | 1.492(4) |
| C(3)-C(4) | 1.410(4) |
| C(4)-C(9) | 1.406(4) |
| C(4)-C(5) | 1.425(4) |
| C(5)-C(6) | 1.364(4) |
| C(6)-C(7) | 1.407(5) |
| C(7)-C(8) | 1.363(5) |
| C(8)-C(9) | 1.421(4) |
| C(9)-N(1) | 1.379(4) |
| C(10)-O(1) | 1.230(3) |
| C(10)-N(2) | 1.350(4) |
| C(11)-N(3) | 1.276(4) |
| C(11)-C(12) | 1.466(4) |
| C(12)-C(13) | 1.391(4) |
| C(12)-C(17) | 1.402(4) |
| C(13)-C(14) | 1.385(5) |
| C(13)-Cl(1) | 1.747(3) |
| C(14)-C(15) | 1.379(5) |
| C(15)-C(16) | 1.386(5) |
| C(16)-C(17) | 1.372(5) |
| N(2)-N(3) | 1.378(3) |
| | |
| N(1)-C(1)-C(2) | 123.9(3) |
| C(3)-C(2)-C(1) | 117.8(3) |
| C(3)-C(2)-C(10) | 117.9(2) |
| C(1)-C(2)-C(10) | 124.2(2) |
| C(2)-C(3)-C(4) | 120.1(3) |
| C(9)-C(4)-C(3) | 118.2(2) |
| C(9)-C(4)-C(5) | 118.8(3) |
| C(3)-C(4)-C(5) | 123.0(3) |
| C(6)-C(5)-C(4) | 120.3(3) |

| | |
|-------------------|----------|
| C(5)-C(6)-C(7) | 120.3(3) |
| C(8)-C(7)-C(6) | 121.1(3) |
| C(7)-C(8)-C(9) | 119.6(3) |
| N(1)-C(9)-C(4) | 121.8(3) |
| N(1)-C(9)-C(8) | 118.4(3) |
| C(4)-C(9)-C(8) | 119.8(3) |
| O(1)-C(10)-N(2) | 122.1(3) |
| O(1)-C(10)-C(2) | 120.5(2) |
| N(2)-C(10)-C(2) | 117.4(2) |
| N(3)-C(11)-C(12) | 119.3(3) |
| C(13)-C(12)-C(17) | 117.2(3) |
| C(13)-C(12)-C(11) | 121.8(3) |
| C(17)-C(12)-C(11) | 121.1(3) |
| C(14)-C(13)-C(12) | 122.1(3) |
| C(14)-C(13)-Cl(1) | 117.4(3) |
| C(12)-C(13)-Cl(1) | 120.4(2) |
| C(15)-C(14)-C(13) | 119.0(3) |
| C(14)-C(15)-C(16) | 120.3(3) |
| C(17)-C(16)-C(15) | 120.1(3) |
| C(16)-C(17)-C(12) | 121.3(3) |
| C(1)-N(1)-C(9) | 118.2(2) |
| C(10)-N(2)-N(3) | 118.0(2) |
| C(11)-N(3)-N(2) | 115.9(2) |

Table S3. Torsion angles [°] for CQ·H₂O.

| | |
|-------------------------|-----------|
| N(1)-C(1)-C(2)-C(3) | 1.0(4) |
| N(1)-C(1)-C(2)-C(10) | -178.2(3) |
| C(1)-C(2)-C(3)-C(4) | -0.9(4) |
| C(10)-C(2)-C(3)-C(4) | 178.3(2) |
| C(2)-C(3)-C(4)-C(9) | 0.2(4) |
| C(2)-C(3)-C(4)-C(5) | -179.4(3) |
| C(9)-C(4)-C(5)-C(6) | -0.2(4) |
| C(3)-C(4)-C(5)-C(6) | 179.5(3) |
| C(4)-C(5)-C(6)-C(7) | 0.1(5) |
| C(5)-C(6)-C(7)-C(8) | 0.1(5) |
| C(6)-C(7)-C(8)-C(9) | -0.2(5) |
| C(3)-C(4)-C(9)-N(1) | 0.5(4) |
| C(5)-C(4)-C(9)-N(1) | -179.8(3) |
| C(3)-C(4)-C(9)-C(8) | -179.6(3) |
| C(5)-C(4)-C(9)-C(8) | 0.1(4) |
| C(7)-C(8)-C(9)-N(1) | -180.0(3) |
| C(7)-C(8)-C(9)-C(4) | 0.1(4) |
| C(3)-C(2)-C(10)-O(1) | -6.0(4) |
| C(1)-C(2)-C(10)-O(1) | 173.2(3) |
| C(3)-C(2)-C(10)-N(2) | 174.4(2) |
| C(1)-C(2)-C(10)-N(2) | -6.5(4) |
| N(3)-C(11)-C(12)-C(13) | 175.7(3) |
| N(3)-C(11)-C(12)-C(17) | -5.0(4) |
| C(17)-C(12)-C(13)-C(14) | -0.1(4) |
| C(11)-C(12)-C(13)-C(14) | 179.3(3) |
| C(17)-C(12)-C(13)-Cl(1) | 179.1(2) |
| C(11)-C(12)-C(13)-Cl(1) | -1.5(4) |
| C(12)-C(13)-C(14)-C(15) | 0.2(5) |
| Cl(1)-C(13)-C(14)-C(15) | -179.1(3) |
| C(13)-C(14)-C(15)-C(16) | 0.2(5) |
| C(14)-C(15)-C(16)-C(17) | -0.6(5) |
| C(15)-C(16)-C(17)-C(12) | 0.6(5) |
| C(13)-C(12)-C(17)-C(16) | -0.3(5) |
| C(11)-C(12)-C(17)-C(16) | -179.7(3) |

| | |
|-----------------------|-----------|
| C(2)-C(1)-N(1)-C(9) | -0.3(4) |
| C(4)-C(9)-N(1)-C(1) | -0.5(4) |
| C(8)-C(9)-N(1)-C(1) | 179.6(3) |
| O(1)-C(10)-N(2)-N(3) | 1.2(4) |
| C(2)-C(10)-N(2)-N(3) | -179.2(2) |
| C(12)-C(11)-N(3)-N(2) | 179.9(2) |
| C(10)-N(2)-N(3)-C(11) | 177.8(3) |

Table S4. Hydrogen bonds for CQ·H₂O [Å and °].

| D-H...A | d(D-H) | d(H...A) | d(D...A) | <(DHA) |
|---------------------|--------|----------|----------|--------|
| C(1)-H(1)...O(2) | 0.95 | 2.42 | 3.356(4) | 168.8 |
| C(11)-H(11)...O(2) | 0.95 | 2.49 | 3.284(4) | 141.2 |
| N(2)-H(2)...O(2) | 0.88 | 2.03 | 2.877(3) | 162.1 |
| O(2)-H(2A)...O(1)#1 | 0.81 | 2.08 | 2.862(3) | 162.8 |
| O(2)-H(2B)...N(1)#2 | 0.81 | 2.06 | 2.846(3) | 166.6 |

Symmetry transformations used to generate equivalent atoms:

#1 $x-1/2, -y+3/2, -z+1$ #2 $-x+1, -y+1, -z+1$

Table S5. Crystal data and structure refinement for NQ·H₂O.

| | | |
|---|---|-------------------------|
| Identification code | 2110niqn | |
| Empirical formula | C ₁₇ H ₁₄ N ₄ O ₄ | |
| Formula weight | 338.32 | |
| Temperature | 190(2) K | |
| Wavelength | 0.71073 Å | |
| Crystal system | Monoclinic | |
| Space group | <i>P</i> 2 ₁ / <i>c</i> | |
| Unit cell dimensions | <i>a</i> = 9.7148(13) Å | α = 90°. |
| | <i>b</i> = 12.6343(18) Å | β = 101.402(13)°. |
| | <i>c</i> = 12.9992(14) Å | γ = 90°. |
| Volume | 1564.0(4) Å ³ | |
| <i>Z</i> | 4 | |
| Density (calculated) | 1.437 Mg/m ³ | |
| Absorption coefficient | 0.106 mm ⁻¹ | |
| <i>F</i> (000) | 704 | |
| Crystal size | 0.400 x 0.200 x 0.030 mm ³ | |
| Theta range for data collection | 3.329 to 25.000°. | |
| Index ranges | -11 ≤ <i>h</i> ≤ 11, -15 ≤ <i>k</i> ≤ 12, -10 ≤ <i>l</i> ≤ 15 | |
| Reflections collected | 6436 | |
| Independent reflections | 2750 [<i>R</i> (int) = 0.0471] | |
| Completeness to theta = 25.000° | 99.8 % | |
| Refinement method | Full-matrix least-squares on <i>F</i> ² | |
| Data / restraints / parameters | 2750 / 0 / 226 | |
| Goodness-of-fit on <i>F</i> ² | 1.121 | |
| Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)] | <i>R</i> 1 = 0.0649, <i>wR</i> 2 = 0.1114 | |
| <i>R</i> indices (all data) | <i>R</i> 1 = 0.1059, <i>wR</i> 2 = 0.1267 | |
| Extinction coefficient | n/a | |
| Largest diff. peak and hole | 0.219 and -0.214 e.Å ⁻³ | |

Table S6. Bond lengths [Å] and angles [°] for NQ·H₂O.

| | |
|-----------------|----------|
| C(1)-N(1) | 1.323(3) |
| C(1)-C(2) | 1.413(4) |
| C(2)-C(3) | 1.364(4) |
| C(2)-C(10) | 1.502(3) |
| C(3)-C(4) | 1.411(3) |
| C(4)-C(9) | 1.412(4) |
| C(4)-C(5) | 1.418(4) |
| C(5)-C(6) | 1.372(4) |
| C(6)-C(7) | 1.396(4) |
| C(7)-C(8) | 1.364(4) |
| C(8)-C(9) | 1.415(3) |
| C(9)-N(1) | 1.379(3) |
| C(10)-O(1) | 1.232(3) |
| C(10)-N(2) | 1.354(3) |
| C(11)-N(3) | 1.277(3) |
| C(11)-C(12) | 1.475(3) |
| C(12)-C(17) | 1.393(4) |
| C(12)-C(13) | 1.395(4) |
| C(13)-C(14) | 1.382(4) |
| C(13)-N(4) | 1.472(4) |
| C(14)-C(15) | 1.372(5) |
| C(15)-C(16) | 1.384(4) |
| C(16)-C(17) | 1.380(4) |
| N(2)-N(3) | 1.386(3) |
| N(4)-O(3) | 1.226(3) |
| N(4)-O(2) | 1.238(3) |
| | |
| N(1)-C(1)-C(2) | 123.7(3) |
| C(3)-C(2)-C(1) | 118.6(2) |
| C(3)-C(2)-C(10) | 117.7(3) |
| C(1)-C(2)-C(10) | 123.8(2) |
| C(2)-C(3)-C(4) | 120.1(3) |
| C(3)-C(4)-C(9) | 117.6(2) |
| C(3)-C(4)-C(5) | 123.2(3) |

| | |
|-------------------|----------|
| C(9)-C(4)-C(5) | 119.2(2) |
| C(6)-C(5)-C(4) | 119.6(3) |
| C(5)-C(6)-C(7) | 120.9(3) |
| C(8)-C(7)-C(6) | 121.0(3) |
| C(7)-C(8)-C(9) | 119.7(3) |
| N(1)-C(9)-C(4) | 122.2(2) |
| N(1)-C(9)-C(8) | 118.2(3) |
| C(4)-C(9)-C(8) | 119.6(3) |
| O(1)-C(10)-N(2) | 122.6(2) |
| O(1)-C(10)-C(2) | 120.4(3) |
| N(2)-C(10)-C(2) | 117.0(3) |
| N(3)-C(11)-C(12) | 118.8(3) |
| C(17)-C(12)-C(13) | 116.3(2) |
| C(17)-C(12)-C(11) | 120.6(3) |
| C(13)-C(12)-C(11) | 123.1(3) |
| C(14)-C(13)-C(12) | 123.1(3) |
| C(14)-C(13)-N(4) | 116.1(3) |
| C(12)-C(13)-N(4) | 120.7(2) |
| C(15)-C(14)-C(13) | 118.5(3) |
| C(14)-C(15)-C(16) | 120.6(3) |
| C(17)-C(16)-C(15) | 119.9(3) |
| C(16)-C(17)-C(12) | 121.6(3) |
| C(1)-N(1)-C(9) | 117.8(3) |
| C(10)-N(2)-N(3) | 117.8(2) |
| C(11)-N(3)-N(2) | 114.5(2) |
| O(3)-N(4)-O(2) | 123.7(3) |
| O(3)-N(4)-C(13) | 118.6(2) |
| O(2)-N(4)-C(13) | 117.6(2) |

Table S7. Torsion angles [°] for NQ·H₂O.

| | |
|-------------------------|-----------|
| N(1)-C(1)-C(2)-C(3) | -0.6(4) |
| N(1)-C(1)-C(2)-C(10) | 179.2(2) |
| C(1)-C(2)-C(3)-C(4) | 0.7(4) |
| C(10)-C(2)-C(3)-C(4) | -179.1(2) |
| C(2)-C(3)-C(4)-C(9) | -0.3(4) |
| C(2)-C(3)-C(4)-C(5) | -179.1(2) |
| C(3)-C(4)-C(5)-C(6) | 179.3(2) |
| C(9)-C(4)-C(5)-C(6) | 0.5(4) |
| C(4)-C(5)-C(6)-C(7) | 0.6(4) |
| C(5)-C(6)-C(7)-C(8) | -0.7(4) |
| C(6)-C(7)-C(8)-C(9) | -0.4(4) |
| C(3)-C(4)-C(9)-N(1) | -0.4(4) |
| C(5)-C(4)-C(9)-N(1) | 178.5(2) |
| C(3)-C(4)-C(9)-C(8) | 179.6(2) |
| C(5)-C(4)-C(9)-C(8) | -1.6(4) |
| C(7)-C(8)-C(9)-N(1) | -178.5(2) |
| C(7)-C(8)-C(9)-C(4) | 1.5(4) |
| C(3)-C(2)-C(10)-O(1) | 4.8(4) |
| C(1)-C(2)-C(10)-O(1) | -175.0(3) |
| C(3)-C(2)-C(10)-N(2) | -175.0(2) |
| C(1)-C(2)-C(10)-N(2) | 5.2(4) |
| N(3)-C(11)-C(12)-C(17) | 16.5(4) |
| N(3)-C(11)-C(12)-C(13) | -166.2(2) |
| C(17)-C(12)-C(13)-C(14) | 2.5(4) |
| C(11)-C(12)-C(13)-C(14) | -174.9(3) |
| C(17)-C(12)-C(13)-N(4) | -175.0(2) |
| C(11)-C(12)-C(13)-N(4) | 7.6(4) |
| C(12)-C(13)-C(14)-C(15) | -3.0(4) |
| N(4)-C(13)-C(14)-C(15) | 174.6(3) |
| C(13)-C(14)-C(15)-C(16) | 1.0(5) |
| C(14)-C(15)-C(16)-C(17) | 1.2(5) |
| C(15)-C(16)-C(17)-C(12) | -1.7(4) |
| C(13)-C(12)-C(17)-C(16) | -0.1(4) |
| C(11)-C(12)-C(17)-C(16) | 177.4(3) |

| | |
|-----------------------|-----------|
| C(2)-C(1)-N(1)-C(9) | -0.1(4) |
| C(4)-C(9)-N(1)-C(1) | 0.6(4) |
| C(8)-C(9)-N(1)-C(1) | -179.4(2) |
| O(1)-C(10)-N(2)-N(3) | -3.5(4) |
| C(2)-C(10)-N(2)-N(3) | 176.3(2) |
| C(12)-C(11)-N(3)-N(2) | -179.8(2) |
| C(10)-N(2)-N(3)-C(11) | 178.7(2) |
| C(14)-C(13)-N(4)-O(3) | 31.3(4) |
| C(12)-C(13)-N(4)-O(3) | -151.1(3) |
| C(14)-C(13)-N(4)-O(2) | -145.1(2) |
| C(12)-C(13)-N(4)-O(2) | 32.5(4) |

Table S8. Hydrogen bonds for NQ·H₂O [Å and °].

| D-H...A | d(D-H) | d(H...A) | d(D...A) | <(DHA) |
|----------------------|--------|----------|----------|--------|
| C(1)-H(1)...O(4)#1 | 0.95 | 2.33 | 3.276(3) | 170.9 |
| C(8)-H(8)...O(2)#2 | 0.95 | 2.46 | 3.352(4) | 156.6 |
| C(11)-H(11)...O(4)#1 | 0.95 | 2.52 | 3.323(3) | 142.2 |
| N(2)-H(2)...O(4)#1 | 0.88 | 2.02 | 2.884(3) | 165.4 |
| O(4)-H(4A)...N(1)#3 | 0.82 | 2.05 | 2.861(3) | 168.2 |
| O(4)-H(4B)...O(1) | 0.82 | 2.03 | 2.830(3) | 164.7 |

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,y+1/2,-z+3/2 #2 -x+1,-y+2,-z+1 #3 x,-y+3/2,z+1/2