



Article Machine Learning-Based Predictions for Half-Heusler Phases

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Abstract: Machine learning models (Support Vector Regression) were applied for predictions of several targets for 18-electron half-Heusler phases: a lattice parameter, a bulk modulus, a band gap, and a lattice thermal conductivity. The training subset, which consisted of 47 stable phases, was studied with the use of Density Functional Theory calculations with two Exchange-Correlation Functionals employed (GGA, MBJGGA). The predictors for machine learning models were defined among the basic properties of the elements. The most optimal combinations of predictors for each target were proposed and discussed. Root Mean Squared Errors obtained for the best combinations of predictors for the particular targets are as follows: 0.1 Å (lattice parameters), 11–12 GPa (bulk modulus), 0.22 eV (band gaps, GGA and MBJGGA), and 9–9.5 W/mK (lattice thermal conductivity). The final results of the predictions for a large set of 74 semiconducting half-Heusler compounds were disclosed and compared to the available literature and experimental data. The findings presented in this work encourage further studies with the use of combined machine learning and ab initio calculations.

Keywords: half-Heusler alloys; thermoelectric materials; machine learning; Support Vector Regression

1. Introduction

Half-Heusler (hH) phases continue to draw unwavering interest due to various possible applications, e.g., in optoelectronics [1,2], thermoelectric devices [1,3,4], and ferromagnets [5]. The stable hH compounds are limited to the 18 valence electron systems, i.e., ternary phases with stoichiometry of 1:1:1, *XYZ* (space group no $F\overline{4}3m$), where *Z* is a main group element, while *X* and *Y* are transition metals. Electronegativity determines the order of elements (*Z* is an anion) and their distribution in a unit cell, with Wyckoff positions X(1/4, 1/4, 1/4), Y(0, 0, 0), and Z(1/2, 1/2, 1/2).

Machine learning (ML) methods are nowadays widely used for cost-effective predictions of various parameters for half-Heusler (hH) phases, e.g., stability [6], atomic site preferences [7], band structures [8], lattice parameters [9], band gaps [10], lattice thermal conductivity [11–13], and spin polarization [14]. Furthermore, the lattice thermal conductivity was investigated for the double hH alloys [15], which exhibit relatively larger and more complex unit cells than those in hH materials. Such findings highly encourage further investigations of the possible ML support for ab initio methods and assure that the elemental properties can be successfully used for the predictions of structural and electronic properties of various intermetallics. Moreover, the ML approach is highly efficient when compared to the high-throughput DFT calculations. The ab initio studies may be very resource-consuming, due to the use of advanced Exchange-Correlation Functionals (XCF) such as hybrid functionals with an exact-exchange term. Expanding the database of potentially valuable hH systems may be accelerated thanks to a reasonable set of previously studied materials (training subset); further investigations may be optimized and supported via ML methods. It was already revealed that elemental features of ions may be sufficient input for predictions of various properties of hH alloys. Predictors proposed by Miyazaki et al. [12] were atomic mass and radius, density of a solid, molar volume, and Debye



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). temperature. Additionally, Zhang et al. [9] considered electronegativity as a key predictor for modeling of lattice parameters.

In this work, several targets for a large set of stable hH alloys are investigated: lattice parameter *a*, bulk modulus *B*, band gap E_{XCF} (where XCF = GGA, MBJ), and lattice thermal conductivity κ_L . These properties for the training subset are also explicitly calculated by employing the Density Functional Theory (DFT), and the results of the ML predictions and DFT calculations are compared. The ML approach used is Support Vector Regression (SVR) with Gaussian Radial Basis Function. Numerous sets of feature spaces are considered with the Valence Electron Count (VEC) and the first, second, and third ionization energies included. The selection of predictors is carefully studied and discussed. A satisfactory accordance between the predicted and available theoretical and experimental data for various hH phases is found. Furthermore, numerous novel materials are considered and characterized.

2. Computational Details

Compounds investigated in this paper were limited by the valence electron count of 18, which is crucial for stability of hH alloys [16]. DFT calculations were performed with the use of VASP [17–20] with spin–orbit coupling included. The cut-off energy of the plane–wave basis was set to 500 eV, and the XCF parameterizations employed were Generalized Gradient Approximation (GGA) [21] and modified Becke–Johnson GGA (MBJGGA) [22]. Lattice thermal conductivity κ_L was calculated following Slack's equation [23–25].

The general idea behind ML is to solve the regression issue and, therefore, to find a function that provides mapping from an input to a training sample target. Methods of solving the regression problem are numerous, based on the different mathematical models, e.g., Random Forest Regression [11], Multiple Linear Regression, and Boosted Decision Tree Regression [12]. SVR models perform the hyperplane analysis in the feature space, which maximizes the number of points that are inside the decision boundary line [26]. Different types of kernel functions enable the best model fit, depending on the data provided [27]. SVR was already shown to be successful in similar applications [28]. The methodology used in this work is depicted in Figure 1.



Figure 1. A flow chart of the ML investigation methodology, including the elemental features regarded and the SVR parameters summed up.

The elemental features for predictor sets were taken from the WebElements periodic table (University of Sheffield [29]). SVR methods implemented in sklearn library [30] with Gaussian Radial Basis Function [31] were used. SVR parameters, i.e., *C* and γ (responsible for the trade-off between regularization and fitting the training data and for the flexibility of the model), were determined as one and $1/(n * X_{VAR})$, where *n* is the number of features and X_{VAR} is a variation in the predictors subset. The feature scaling (Standard Scaler) was applied. The cross-validation of models tested was the Leave One Out (LOO) approach [32]. The quantitative accuracy of SVR models was determined by Root Mean Squared Error (RMSE) defined as $\sqrt{\frac{1}{n}\sum_{i=1}^{n}(a_i^{calc} - a_i^{pred})^2}$, where *a* is a studied quantity.

3. Results

The training subset of 47 18-valence electron hH phases considered in this work consists of 32 stable compounds (according to the Open Quantum Materials Database, OQMD [33]), as reported in the literature [34], and 15 systems with hull distance less or equal to 0.1 eV, which in many cases indicates stable phases (Aykol et al. [35]). The 32 stable compounds were already comprehensively investigated as candidates for thermoelectric materials [4]. The structural and electronic properties of 15 additional phases are gathered in Table 1. It is crucial to note that the further training of ML models is based on systems that are expected to be stable and feasible to be synthesized.

Table 1. The 18-valence electron hH systems with hull distance: 0 eV $\langle E_{HD} \leq$ 0.1 eV [33,35]. Structural and electronic parameters: lattice parameter *a* (Å), bulk modulus *B* (GPa), band gap with GGA parameterization E_{GGA} (eV), band gap with MBJGGA parameterization E_{MBJ} (eV), lattice thermal conductivity κ_L (W/mK).

Comp.	а	В	E _{GGA}	E _{MBJ}	κ _L
TiPdPb	6.328	103.88	0.352	0.324	55.57
TiPdGe	5.964	131.38	0.619	0.584	45.00
VRhGe	5.796	172.54	0.433	0.748	38.73
ZrCoAs	5.831	147.58	1.203	1.228	68.47
ZrRhAs	6.110	143.97	1.117	1.292	59.94
HfPdGe	6.142	133.71	0.552	0.506	66.63
HfRhAs	6.063	160.02	0.282	0.816	58.39
TaRhGe	5.973	185.13	1.044	1.026	64.25
VRuSb	6.044	165.64	0.189	0.631	42.96
ZrNiGe	5.893	141.57	0.679	0.654	74.56
ScPdAs	6.098	111.31	0.432	0.451	42.58
NbRuAs	5.961	183.13	0.337	0.506	53.07
TiNiPb	6.038	115.66	0.341	0.292	66.00
NbRuBi	6.307	151.77	0.383	0.557	57.56
LuNiAs	5.989	106.94	0.408	0.475	65.65

Some properties of the systems presented in Table 1 were already investigated in the literature, e.g., the cubic lattice parameters (a) of 5.964 and 6.327 Å were theoretically calculated for TiPdGe and TiPdPd, respectively [36]. The reported values of the bulk modulus (B) were 132.41 and 104.14 GPa, while band gaps (E_g) were 0.66 and 0.386 eV (PBE-GGA), for the Ge- and Pb-bearing systems. The results presented in this work for TiPdGe and TiPdPb are very similar to the literature data. Furthermore, the GGA predictions were also reported for VRhGe [37], indicating a of 5.73 Å, B of 186.1 GPa, and E_g of 0.34 eV. While the lattice parameter calculated here is almost the same, the bulk modulus and band gap are slightly different from the literature data. The discrepancy may be caused by the different XCF and DFT implementations used. Bendahma et al. [38] predicted a for HfPdGe of 11.62 Bohr, B of 137.33 GPa, and E_g of 0.55 eV, which are very close to the results obtained here. Interestingly, TaRhGe was experimentally synthesized in the TiNiSi-type hexagonal structure [39]. Theoretical studies of lattice parameters, bulk modulus, GGA, and MBJGGA band gaps for VRuSb led to the following values: 5.940 Å, 190.35 GPa, 0.22 eV, and 0.64 eV, respectively [40,41]. The arsenide ScPdAs was predicted as a phase with *a* of 6.14 Å, *B* of 106.90 GPa, and E_g of 0.44 eV [42]. The value of κ_L (35 W/mK at 300 K) calculated according to the modified Debye-Callaway model is very close to the results obtained for this system via Slack's formula. Nearly the same κ_L was also obtained for LuNiAs, which is clearly lower than the one presented in this work, although other properties obtained for this system are in good accordance. Calculations for TiNiPb presented by Hong et al. [43] revealed an *a* value which differs by 0.1 Å. The value of κ_L for ZrCoAs was estimated by Carrete et al. [11] to be 24.0 W/mK, whereas, for NiPbTi, it was 109 W/mK. Similar values of κ_L were obtained for BBeGa and BMgGa by Sun et al. with the Slack-derived methodology [24].

The number of hH materials gathered in Table 1 is limited by the assumption about the acceptable hull distance for stable phases [35]. One should also perceive the training subset as versatile enough to prepare the feature space related to compounds with different values of the targets desired, as pointed out by Tranås et al. [13]. For the training subset, the lattice parameters range from 5.496 (VFeAs) to 6.525 Å (ScPdBi), the bulk moduli range from 85.35 (ScPdBi) to 203.24 GPa (VIrGe), the GGA band gaps range from 0.071 (ScPdBi) to 1.300 eV (TiCoAs), the MBJGGA band gaps range from 0.121 (ScPdBi) to 1.361 eV (HfCoAs), and the lattice thermal conductivities range from 38.73 (VRhGe) to 113.52 W/mK (HfNiSn). Such wide ranges of structural and electronic properties indicate that multiple diverse ternary systems with various parameters were used for the model training. Finally, the various ions present in the training subset enable wide and reliable predictions for numerous ternary systems.

The feature space (Figure 1) used for the description of the hH systems consisted of the following elemental properties [29]:

- Atomic mass, u;
- Atomic radius, r;
- Molar volume, V;
- Density of solid, g;
- Electronegativity, n;
- Debye temperature, T;
- Thermal conductivity, k;
- I–III ionization energies, i(I), i(II), i(III);
- Valence Electron Count, VEC.

The parameters listed above were regarded for all ions with all atomic sites, i.e., denoting that parameter α was included in the feature space means that α_X , α_Y , and α_Z were included. This approach, well described by Miyazaki et al. [12], provides comprehensive insight about the distribution of the particular property in the whole set of ternary systems investigated.

The convergence of the SVR models as a function of the size of the training subset was determined based on RMSE, as presented in Figure 2 (learning curves with targets split into subplots (a) and (b) due to the order of magnitude). For lattice parameters and band gaps with both XCF regarded, one can note a clear decrease in RMSE and all 3 curves reach a plateau at the point of 40 records in the training set. Therefore, there may be no need to acquire more input data calculations for the training since the performance of the model may not significantly improve. RMSE for *a*, E_{GGA} , and E_{MBJ} improved from 0.235 to 0.105 Å, from 0.350 to 0.235 eV, and from 0.314 to 0.225 eV, respectively.



Figure 2. Learning curves of RMSE as a function of the size of the training subset: (**a**) lattice parameter (Å), bulk modulus (GPa), GGA band gap (eV), and (**b**) MBJGGA band gap (eV), lattice thermal conductivity κ_L (W/mK).

As depicted in Figure 2b, no clear plateau is observed in learning curves for *B* and κ_L . However, the value of RMSE obtained for *B* is already relatively low. Although a further extension of the training set is expected to decrease RMSE of κ_L , one may also consider that an arbitrary selection of the predictor set for this quantity is less favorable than in the cases of simple structural properties such as *a* or *B*. The decreases in RMSE between the minimal and 47 compound training sets are about 15 GPa for bulk modulus and 11 W/mK for lattice thermal conductivity. The final values of RMSE for *B* and κ_L are about 13 GPa and 20 W/mK, respectively, which seems to be a good accuracy, according to the order of magnitude of these parameters. The training set of 47 hH alloys is, therefore, large enough to derive reliable and repeatable results for further predictions.

Contrary to the five-fold cross-validation used by Miyazaki et al. [12], the Leave One Out (LOO) cross-validation approach was used in this work. This choice was motivated by the LOO efficiency within the Support Vector Machine family [44,45]. In this method, only one record among the whole feature space set is determined as the value to be predicted and compared with the actual result. Iteration of such a method over all the records in the feature space enables achieving a constant RMSE among the models for a particular set of predictors. Depending on the combinations of predictors used, RMSE may further vary. In general, one may expect that a lower RMSE would characterize a better model. However, in fact, low RMSE does not unequivocally determine the best combination of predictors. The particular subset of parameters in each combination of the elemental features may yield better predictions for some phases and worse results for others. The fact that RMSE is an average is worth considering in many cases.

Numerous ML band gap predictions for organic [46] and inorganic solids [28,47], MXenes [48], perovskites [49], and other systems [50–52] were carefully discussed in the literature. However, there are still few reports on band gap predictions for hH phases [10]. Olsthoorn et al. [46] obtained the lowest RMSE of 0.519 eV, whereas the results of Lee et al. [28] exhibited RMSE of 0.24 eV. Finally, Choudhary et al. [10] disclosed RMSE of 0.286 eV, which is also larger than the RMSE obtained in our research (about 0.22 eV).

For each target, over 2000 combinations of the elemental predictors were tested, from 1-element feature spaces (e.g., atomic radii of *XYZ* ions: r_X , r_Y , r_Z) to the most complex feature space consisting of 11 predictors for each ion in the particular phase. The summaries of RMSE as a function of the number of elemental predictors are depicted in Figure 3. As one may expect, the discrepancy between the ML- and DFT-derived results is maximal for one predictor feature spaces due to oversimplification. While the extension of the predictor set generally improves the model performance, the best sets are based on 3–4 elemental features, which are characteristic of particular targets. This effect is related to the fact that bigger predictor sets include many features that may be unsuitable for SVR in each case.



Figure 3. Histograms for all the possible combinations of eleven elemental predictors with their RMSE for (a) lattice parameter (Å), (b) bulk modulus (GPa), (c) GGA-derived band gap (eV), (d) MBJGGA-derived band gap (eV), and (e) lattice thermal conductivity κ_L (W/mK).

The scatter plots (ML-predicted vs. DFT-calculated) for three favorable combinations of predictors are depicted for all targets in Figure 4. The Debye temperature and thermal conductivity occurred most often in the best RMSE combinations for the target of lattice parameter. An analysis of three combinations of predictors with the lowest and comparable RMSE of 0.1 Å provides different distributions of the predictions among the 47 systems, as presented in Figure 3a. The RMSE values among the predictor sets selected here are

very close to each other. The four-predictor set [V, i(II), u, T] seems to be the best approach for the majority of compounds in the training set. However, in some cases, e.g., for VIrGe ($a_{calc} = 5.818$ Å), this set leads to a worse prediction when compared to those of [i(II), i(III), T] and [i(II), u, T]. The selection of the best predictor set is, therefore, a challenging task, which requires a careful analysis of the data presented in scatter plots.



Figure 4. The comparisons of the predicted and calculated values for 47 systems creating the training subset for (**a**) lattice parameter (Å), (**b**) bulk modulus (GPa), (**c**) GGA-derived band gap (eV), (**d**) MBJGGA-derived band gap (eV), and (**e**) lattice thermal conductivity κ_L (W/mK).

The best models for bulk modulus are strongly connected with the third ionization energy. The predictor set with the lowest RMSE for bulk modulus (Figure 4b) is [i(II), i(III), T]. Merging the additional predictors to this feature space resulted in a slight increase in RMSE (0.32 GPa for additional [g, k] and 0.42 GPa for additional [u]). Enriched feature spaces neither provide better single predictions for the boundary cases nor yield lower RMSE.

For E_{GGA} (Figure 4c) and E_{MBJ} (Figure 4d), the second and third ionization energies of the constituent elements are present in sets with the lowest RMSE. The model based on [i(II), i(III), VEC, T] seems to perform slightly better than the other ones. Reasonable results for E_{GGA} require VEC, which illustrates the importance of optimum distribution of the valence electrons in the hH alloys for insights into electronic structure of such systems. Similar findings are revealed for the best combinations of predictors for E_{MBJ} (Figure 4d). In addition to the ionization energies, VEC, thermal conductivity, and ionic radius are favorable as the third predictor.

In the case of the lattice thermal conductivity (Figure 4e), the volume and atomic mass were regarded as the best predictors for SVR models. Combining or replacing them with the atomic radius or Debye temperature leads to higher RMSE. Models for this quantity do not require any of [i(II), i(III)], a fact which is related to Slack's formula.

After analyses of the best predictor sets, the favorable ones for particular targets are as follows:

• *a*: [V, i(II), u, T];

- B: [i(II), i(III), T];
- *E_{GGA}*: [i(II), i(III), VEC, T];
- *E_{MBJ}*: [r, k, i(II), i(III), VEC];
- κ_L: [V, u].

One shall notice that the most common parameters are the ionization energies. The significance of the second and third ionization energies for various targets may be useful for further ML modeling for similar intermetallics.

The overall performance of SVR models is limited not only by the size of the training set and variation of elements, but also by the characteristic features of materials that exhibit extremal values of targets. As depicted in Figure 4b, the error in predictions for relatively low values of bulk modulus reaches 40 GPa, which is high when compared with RMSE of 12 GPa. The electronic structures of hH systems with narrow or relatively wide band gaps are also difficult to predict with ML methods. Namely, band structures of bisimides and antimonides are unique, and many hH systems are on the edge of a transition between the very narrow E_g and topological insulator. The training set of semiconducting materials could be extended with phases with zero or negative band gaps. This finding may be an interesting direction for further ML-based studies on E_g in intermetallics. A similar issue is found for predictions of κ_L , i.e., the most interesting values of κ_L for thermoelectrics are low. The materials with low κ_L are rare and difficult to predict even based on relatively large training sets because the SVR models are dominated by average properties of materials.

The final SVR predictions for 74 hH phases, which were not studied here with the DFT-based calculations, are gathered in Table 2. The structural and electronic properties of some alloys among this set were already investigated in the literature. Adetunji et al. [53] reported some properties of the cubic HfNiGe phase, i.e., *a* of 5.861 Å, *B* of 143.1 GPa, and E_g of 0.61 eV. These three values are very close to our predictions and are consistent with the RMSE of the related SVR models. Interestingly, the value of E_g obtained here for NbRhGe is strongly underestimated when compared to the literature data [54], in spite of the fact that several novel germanides were included in the training set.

Table 2. Predicted values of the lattice parameter *a* (Å), bulk modulus *B* (GPa), band gap with GGA parametrization E_{GGA} (eV), band gap with MBJGGA parametrization E_{MBJ} (eV), and lattice thermal conductivity κ_L (W/mK).

Comp.	а	В	E _{GGA}	E _{MBJ}	\mathbf{K}_L	Comp.	а	В	E _{GGA}	E _{MBJ}	\mathbf{K}_L
HfNiGe	5.893	151.53	0.74	0.70	79.52	TiNiSn	5.918	124.13	0.43	0.49	62.46
HfRhBi	6.331	128.63	0.64	0.76	86.65	TiPdSn	6.20	114.38	0.43	0.46	57.02
HfNiPb	6.175	129.51	0.46	0.48	94.83	TiPtPb	6.342	128.13	0.57	0.60	72.91
HfRhSb	6.284	138.78	0.65	0.78	104.5	TiCoSb	5.843	135.62	0.92	0.92	61.96
HfPdPb	6.353	124.08	0.39	0.39	92.69	TiRhSb	6.110	135.24	0.66	0.83	59.22
HfPtSn	6.348	143.19	0.69	0.82	95.28	TiIrAs	5.952	171.78	0.72	0.94	51.43
HfPtPb	6.369	138.14	0.62	0.70	92.82	TiIrBi	6.258	144.95	0.63	0.82	67.39
HfCoSb	6.028	134.71	0.97	0.96	101.62	VCoSn	5.813	150.09	0.57	0.82	50.27
LuNiSb	6.142	118.59	0.45	0.53	86.67	VRhSn	6.018	152.80	0.41	0.68	42.65
LuPdSb	6.300	118.32	0.39	0.42	90.92	VIrSn	6.052	171.75	0.38	0.65	55.34
LuNiBi	6.185	117.78	0.47	0.57	83.54	VRuAs	5.844	176.39	0.24	0.73	37.81
NbRhGe	5.942	176.72	0.78	0.88	53.10	VOsSb	6.051	163.10	0.32	0.60	59.10
NbOsAs	5.992	173.71	0.42	0.59	57.95	VRuBi	6.183	154.00	0.27	0.63	46.24
NbOsBi	6.319	153.33	0.46	0.53	73.17	VOsAs	5.854	173.33	0.32	0.65	44.98
NbRhSn	6.195	152.96	0.62	0.78	70.38	VOsBi	6.200	155.11	0.36	0.60	60.37
NbRhPb	6.317	142.94	0.58	0.67	67.46	VCoPb	6.004	142.22	0.53	0.62	56.36
NbIrPb	6.317	162.40	0.58	0.64	79.63	VRhPb	6.175	143.60	0.40	0.59	47.93
NbCoSn	5.940	150.13	0.81	0.92	74.59	VIrPb	6.192	161.41	0.41	0.61	62.40

Comp.	a	В	E _{GGA}	E _{MBI}	KL	Comp.	a	В	E _{GGA}	E _{MBI}	KL
MhCaDh	6 104	141.06	0.72	0.72	72.06	VEaPi	6.052	150.49	0.41	0.64	E1 61
NDCOPD	6.104	141.06	0.75	0.75	75.20	Vredi	6.052	130.40	0.41	0.64	51.61
NbOsSb	6.228	162.70	0.43	0.52	83.18	YN1As	6.058	124.92	0.53	0.54	58.54
NbFeSb	5.932	160.15	0.62	0.74	74.12	YNiSb	6.157	121.03	0.49	0.55	66.32
NbFeBi	6.144	145.50	0.62	0.70	66.30	YNiBi	6.209	120.66	0.48	0.57	67.77
ScNiAs	6.048	126.16	0.71	0.69	58.06	YPdAs	6.111	121.33	0.42	0.49	49.56
ScNiSb	6.172	111.29	0.60	0.59	66.83	YPdSb	6.290	118.01	0.35	0.45	65.94
ScNiBi	6.239	106.29	0.56	0.58	68.60	YPtSb	6.276	128.57	0.51	0.59	69.09
ScPdSb	6.348	100.79	0.34	0.34	66.58	ZrNiSn	6.072	123.54	0.48	0.52	85.14
ScPtSb	6.333	120.26	0.58	0.64	69.62	ZrNiPb	6.177	119.30	0.42	0.46	82.71
TaRhSn	6.131	162.94	0.82	0.89	87.23	ZrPdGe	6.149	130.13	0.63	0.55	62.95
TaRhPb	6.206	155.14	0.75	0.76	84.45	ZrPdPb	6.455	110.42	0.39	0.38	82.90
TaIrGe	6.015	189.33	0.80	0.85	69.30	ZrPtSn	6.389	130.93	0.71	0.84	89.55
TaIrSn	6.159	172.9	0.75	0.79	89.15	ZrCoSb	6.014	127.17	1.00	0.93	87.32
TaIrPb	6.212	164.16	0.71	0.70	86.42	ZrCoBi	6.155	119.47	0.91	0.84	78.01
TaRuSb	6.108	173.42	0.46	0.64	93.97	ZrRhSb	6.307	126.79	0.76	0.82	92.14
TaRuBi	6.195	161.39	0.49	0.61	79.00	ZrRhBi	6.384	118.40	0.71	0.77	74.73
TaOsSb	6.165	164.87	0.51	0.59	94.12	ZrIrAs	6.110	160.96	0.85	0.94	66.41
TaCoPb	6.086	153.94	0.89	0.82	88.35	ZrIrSb	6.304	147.57	0.76	0.82	96.04
TaFeBi	6.110	156.66	0.76	0.78	84.33	ZrIrBi	6.360	138.67	0.72	0.77	83.56

Table 2. Cont.

The electronic structure of ScNiAs was investigated by Jaishi et al. [55], revealing *a* of 5.84 Å and E_g values of 0.48 (GGA) and 0.52 eV (MBJGGA). The lattice parameter predicted here is larger by 0.2 Å than the DFT result, whereas the E_g values are close to the literature data. One may connect this fact with the presence of numerous arsenides present in our training set.

Previous theoretical investigations for HfRhBi revealed a, B, and E_g of 6.41 Å, 127.62 GPa, and 0.17 eV, respectively [56]. The thermal conductivity of HfRhBi was reported as up to 40 W/mK at 300 K (significantly smaller than our predictions). Kangsabanik et al. [56] also reported similar results for ZrIrBi and ZrRhBi, disclosing κ_L (W/mK) of up to 55 and 30 W/mK, respectively. The discrepancy between our predictions and the literature data for κ_L is caused mainly by the methodology used for calculations (Slack's formula vs. phonon spectra, etc.). There is also a significant discrepancy between the band gaps, i.e., predicted 0.71 eV versus calculated 0.26 eV and predicted 0.77 eV versus calculated 1.02 eV for ZrIrBi and ZrRhBi, respectively. One may explain this issue by considering that the band structures of these materials, the fact that the number of bisimides in our training set is limited, and the performance of our SVR approach is poor for narrow band gap materials. The reported *a* for TiIrBi ranged from 6.309 to 6.358 Å, depending on the alloy model [57], which is in good accordance with the value predicted here. The range of reported *B* was from 104.4 up to 123.7 GPa, which is slightly lower than our result. However, the GGA- and MBJGGA-derived band gaps (0.56 and 0.87 eV, respectively) are in very good accordance with the predictions presented here.

The analysis for ScMSb (M = Ni, Pd, and Pt) indicates very good accordance between the DFT-derived a and B and the SVR [58]. However, the SVR-predicted values of a for LuPdSb, YNiSb, and YPdSb differ from the calculated ones by 0.2–0.3 Å, which are clearly higher values than the RMSE of the model. Furthermore, the predicted band gaps for antimonides, ScMSb and (Lu;Y)(Ni;Pd)Sb, are strongly overestimated when compared with the calculated ones. This effect is connected with the characteristic features of electronic structures of hH antimonides and bisimides [58–62]. Bands at Valence Band Maximum (VBM) in these systems exhibit very small effective mass, which is not present in systems in our training set. Amont the hH antimonides, NbFeSb is the most studied material. The experimental lattice parameter value (5.949 Å [63]) and the calculated value (5.968 Å [64]) are in very good accordance with our predictions. The E_g of 0.51 eV is also close to the predicted value of 0.62 eV, whereas the reported values of κ_L are lower than that predicted here.

The SVR predictions for NbCoSn are in good agreement with the reported data [65]. The lattice thermal conductivity of CoSbZr was calculated by Carrete et al. [11] to be 25 W/mK, while κ_L of 109 W/mK was calculated for FeNbP and NiPbTi. Furthermore, the calculated κ_L for RuAsV (23.5 W/mK) with the declared standard deviation of 13% is in good accordance with our predicted value for VRuAs (37.81 W/mK).

4. Conclusions

In summary, all SVR models considered in this work lead to very good predictions for lattice parameters and bulk modulus. The ML-derived values of band gaps and lattice thermal conductivity may be strongly affected by specific electronic structures among the phases regarded, e.g., the direct–indirect type of the band gap and the spectrum of carrier concentration. The significance of the second and third ionization energies of the elements in the process of predicting different targets of complex alloys, i.e., lattice parameter, bulk modulus, and band gap (GGA and MBJGGA), is an important insight into the selection of feature spaces for ML support of the ab initio calculations. The predictions obtained here for 74 hH alloys are in good accordance with the available literature data, especially for the well-known NbFeSb compound, which indicates good predictive power of SVR models.

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