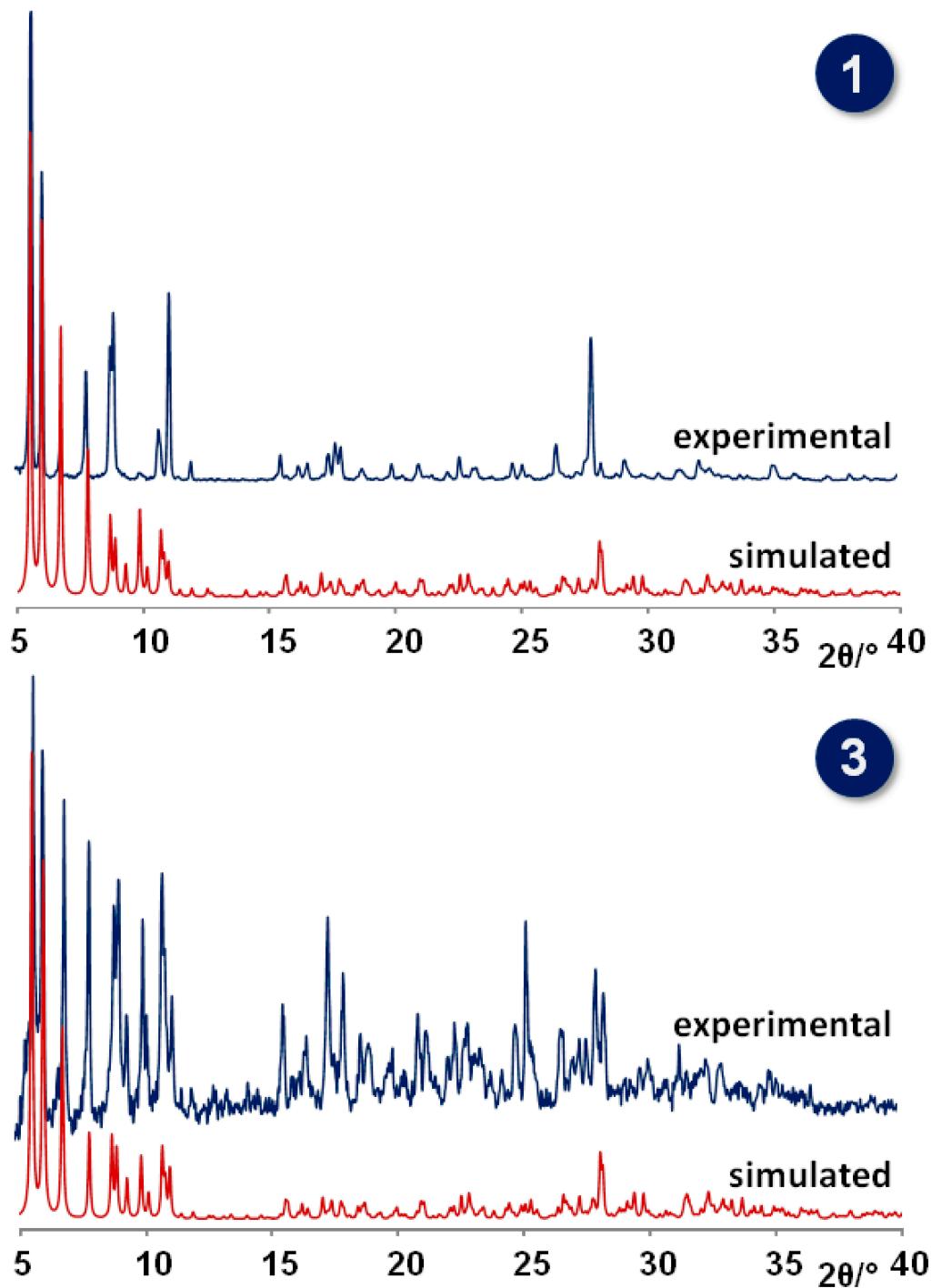
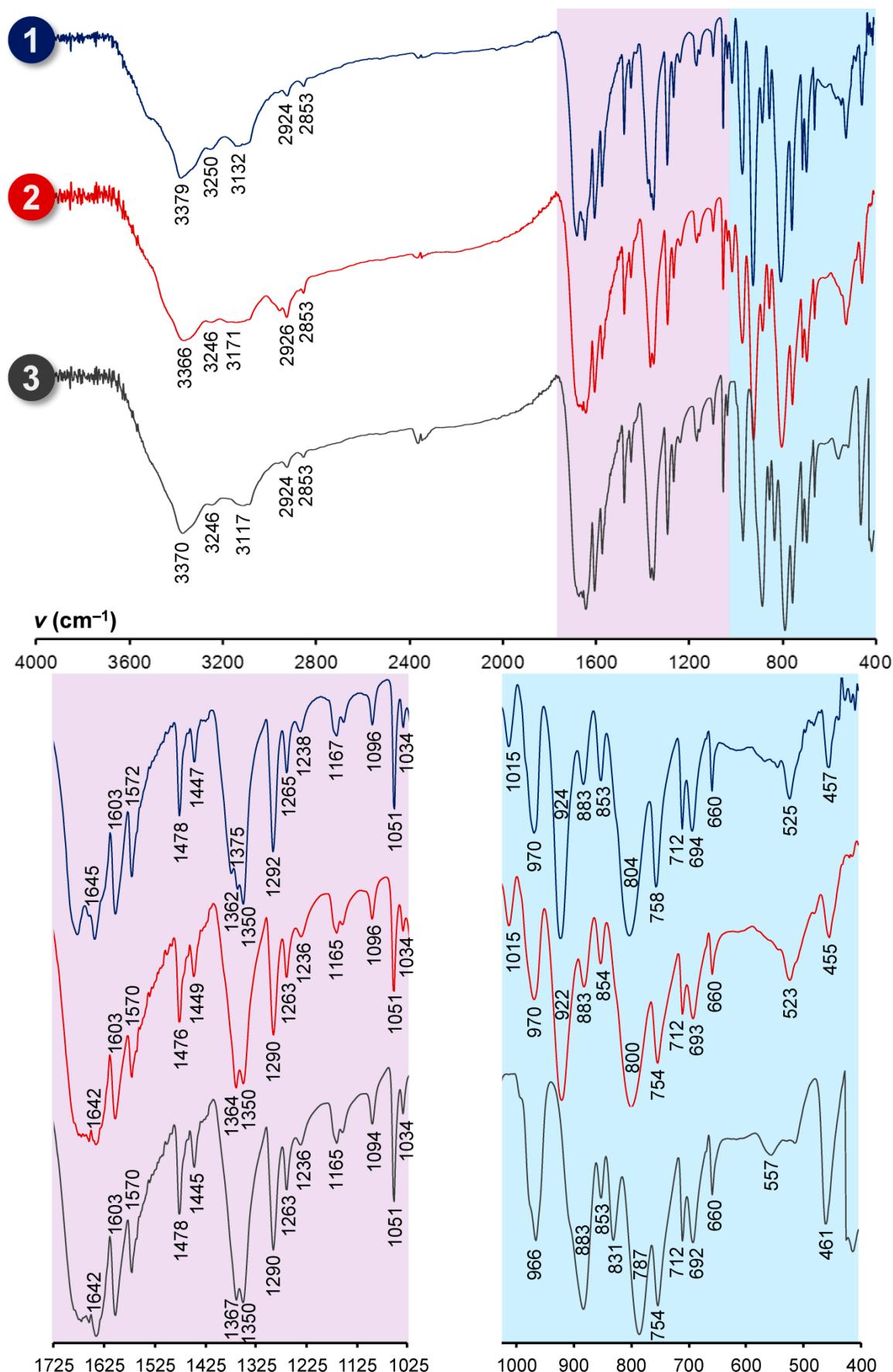


## Supplementary Information



**Figure S1.** Experimental powder X-ray diffraction patterns of compounds **1** and **3** compared to those simulated from single-crystal X-ray diffraction data.



**Figure S2.** Infrared spectra of compounds 1–3 with details of the metalorganic and inorganic regions (framed in violet and blue, respectively).

**Table S1.** Ranges of bond lengths, (mean) values and selected interatomic distances ( $\text{\AA}$ ) in the Keggin clusters of **1–3** and their anhydrous derivatives **1a** and **3a** compared to those of the DFT-optimized species (opt)<sup>a</sup>.

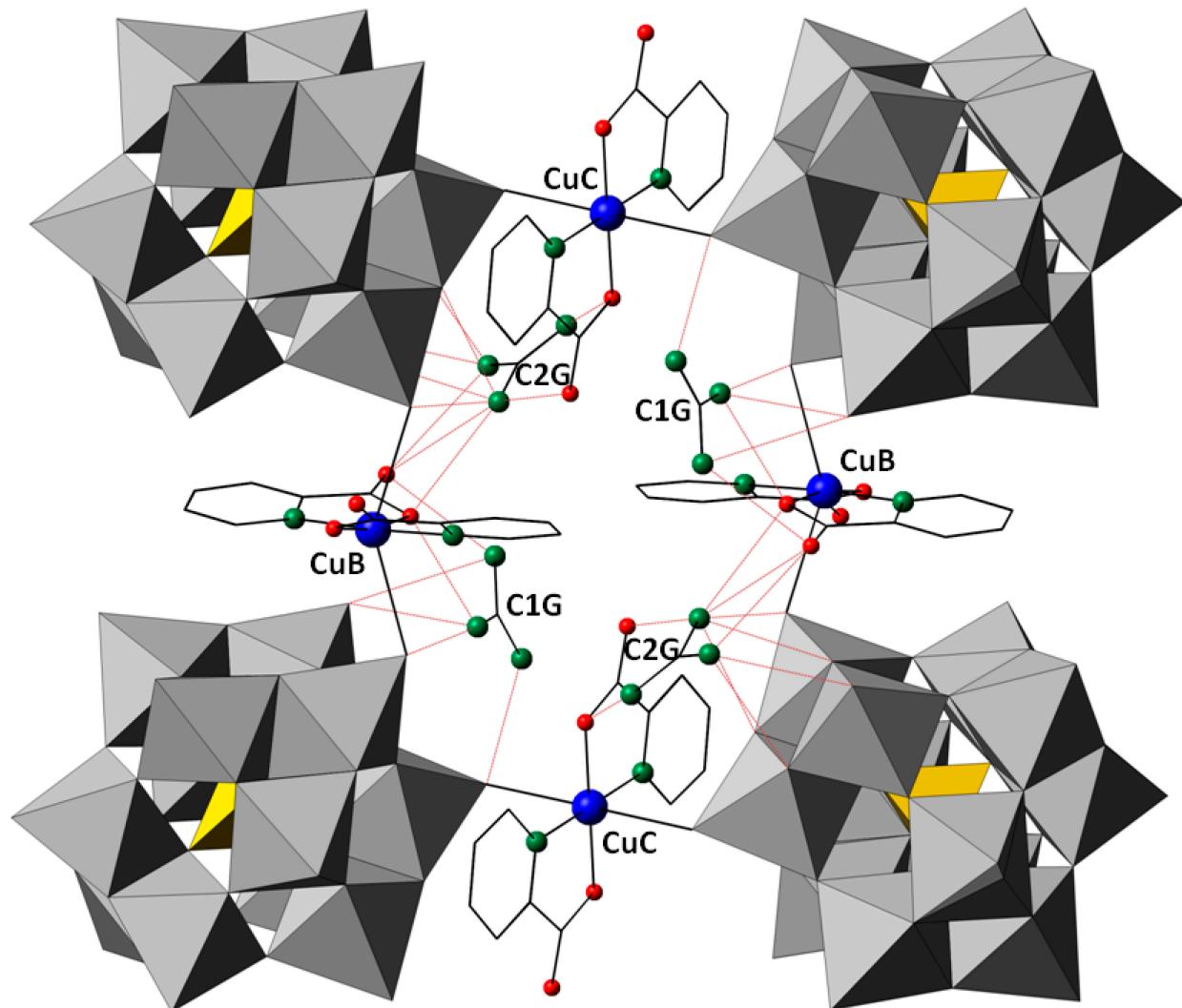
| $[\text{SiW}_{12}\text{O}_{40}]^{4-}$ | <b>1</b>                       | <b>1a</b>                      | <b>2</b>                     | <b>opt</b>   |
|---------------------------------------|--------------------------------|--------------------------------|------------------------------|--------------|
| <b>W–O<sub>c</sub></b>                | 2.316(12)–2.462(12)<br>(2.383) | 2.350(12)–2.478(12)<br>(2.398) | 2.26(3)–2.53(4)<br>(2.38)    | 2.325        |
| <b>W–O<sub>b</sub></b>                | 1.690(15)–2.108(13)<br>(1.926) | 1.741(12)–2.067(13)<br>(1.920) | 1.59(4)–2.20(5)<br>(1.92)    | 1.916, 1.937 |
| <b>W–O<sub>t</sub></b>                | 1.629(14)–1.773(16)<br>(1.702) | 1.631(16)–1.748(15)<br>(1.693) | 1.58(5)–1.81(6)<br>(1.70)    | 1.743        |
| <b>X–O<sub>c</sub></b>                | 1.569(12)–1.683(12)<br>(1.625) | 1.591(11)–1.686(12)<br>(1.629) | 1.54(4)–1.74(4)<br>(1.64)    | 1.667        |
| <b>X···W</b>                          | 3.516(1)–3.541(1)<br>(3.529)   | 3.525(1)–3.538(1)<br>(3.532)   | 3.498(6)–3.545(6)<br>(3.526) | 3.588        |
| <b>W···W<sub>trans</sub></b>          | 7.032(1)–7.082(1)<br>(7.058)   | 7.045(1)–7.077(1)<br>(7.064)   | 7.042(1)–7.061(2)<br>(7.052) | 7.172        |
| $[\text{GeW}_{12}\text{O}_{40}]^{4-}$ | <b>3</b>                       | <b>3a</b>                      |                              | <b>opt</b>   |
| <b>W–O<sub>c</sub></b>                | 2.290(12)–2.425(13)<br>(2.354) | 2.28(2)–2.44(2)<br>(2.34)      |                              | 2.345        |
| <b>W–O<sub>b</sub></b>                | 1.733(13)–2.123(13)<br>(1.939) | 1.72(2)–2.13(3)<br>(1.92)      |                              | 1.928, 1.940 |
| <b>W–O<sub>t</sub></b>                | 1.653(15)–1.727(17)<br>(1.691) | 1.51(3)–1.83(3)<br>(1.70)      |                              | 1.740        |
| <b>X–O<sub>c</sub></b>                | 1.679(12)–1.777(12)<br>(1.728) | 1.70(2)–1.83(2)<br>(1.75)      |                              | 1.738        |
| <b>X···W</b>                          | 3.527(1)–3.548(1)<br>(3.539)   | 3.530(1)–3.549(1)<br>(3.538)   |                              | 3.592        |
| <b>W···W<sub>trans</sub></b>          | 7.053(1)–7.096(1)<br>(7.077)   | 7.060(1)–7.097(1)<br>(7.076)   |                              | 7.180        |

<sup>a</sup> San Felices, L.; Vitoria, P.; Gutiérrez-Zorrilla, J.M.; Lezama, L.; Reinoso, S. *Inorg. Chem.* **2006**, *45*, 7748–7757.

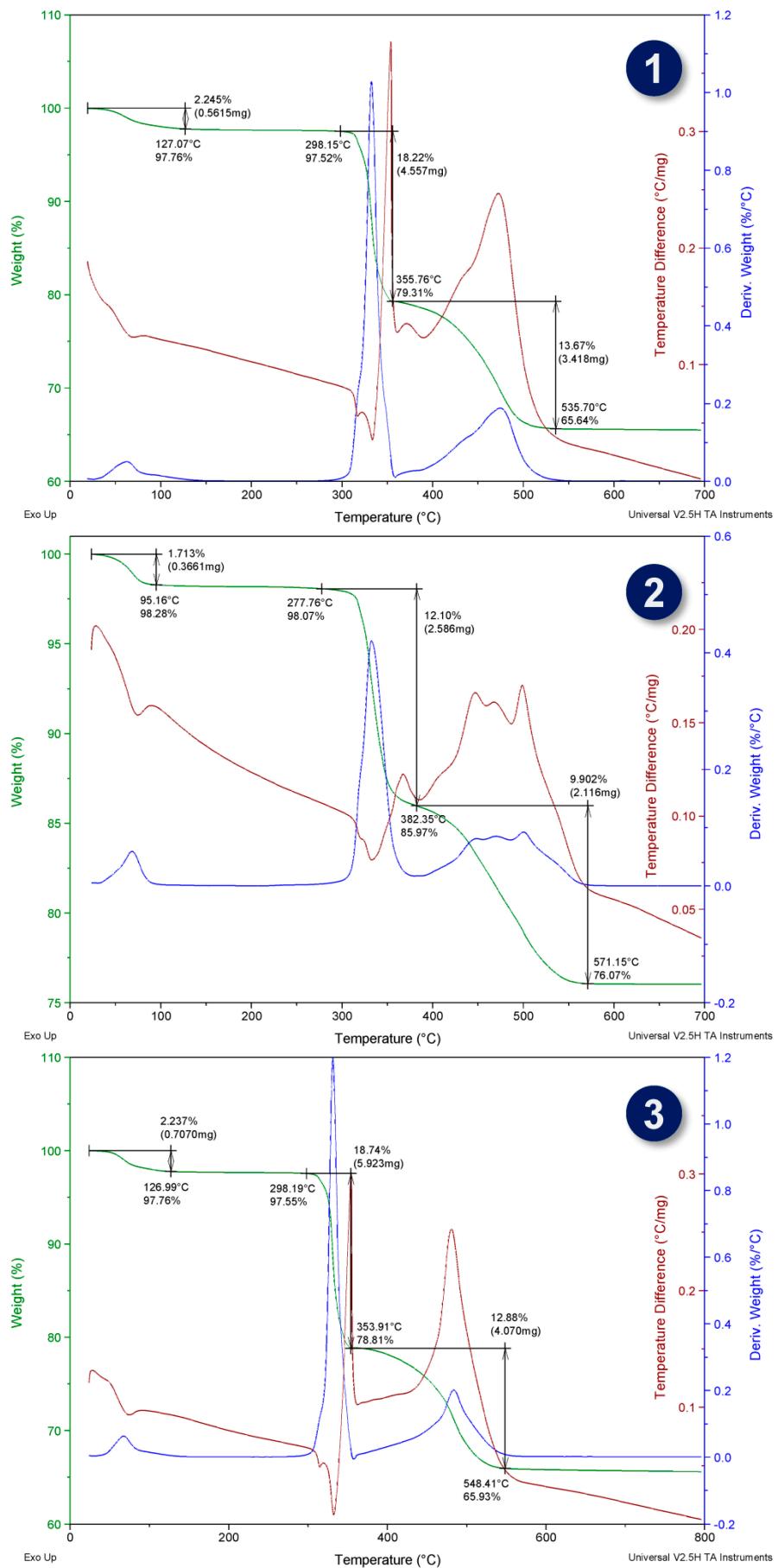
**Table S2.** Donor···acceptor distances ( $\text{\AA}$ ) for the N–H···O and O–H···O hydrogen bonds in **1** and **3** and their anhydrous derivatives **1a** and **3a**.

| Donor···Acceptor           | <b>1</b>  | <b>1a</b>         | <b>3</b>  | <b>3a</b>       |
|----------------------------|-----------|-------------------|-----------|-----------------|
| N1G···O1C <sup>i</sup>     | 3.059(14) | 3.213(13)         | 3.066(13) | 3.20(3)         |
| N1G···O2C <sup>i</sup>     | 3.366(16) | 3.479(14)         | 3.362(15) | 3.49(3)         |
| N1G···O4/O4Z <sup>ii</sup> | –         | 2.941(19)/2.90(2) | –         | 2.85(4)/2.91(4) |
| N1G···O4w <sup>i</sup>     | 2.875(15) | –                 | 2.882(15) | –               |
| N2G···O2C                  | 2.822(14) | 2.832(14)         | 2.814(14) | 2.83(3)         |
| N2G···O1A                  | 3.322(12) | 3.195(12)         | 3.335(12) | 3.17(2)         |
| N2G···O2A                  | 3.042(12) | 3.057(13)         | 3.058(13) | 3.10(3)         |
| N3G···O2A                  | 3.059(13) | 2.916(13)         | 3.075(14) | 2.96(3)         |
| N3G···O2B <sup>iii</sup>   | 3.070(14) | 2.871(12)         | 3.084(13) | 2.85(3)         |
| N3G···O3w <sup>ii</sup>    | 3.29(3)   | –                 | 3.26(3)   | –               |
| N4G···O3A                  | 3.068(11) | 3.134(12)         | 3.077(12) | 3.10(2)         |
| N4G···O4A                  | 3.444(13) | 3.495(13)         | 3.435(13) | 3.48(2)         |
| N4G···O4C <sup>iii</sup>   | 2.865(13) | 2.863(13)         | 2.876(12) | 2.89(3)         |
| N5G···O3B <sup>iv</sup>    | 2.963(12) | 2.885(12)         | 2.968(12) | 2.93(2)         |
| N5G···O4A                  | 2.847(12) | 2.800(12)         | 2.842(13) | 2.82(2)         |
| N6G···O4B <sup>iv</sup>    | 2.856(13) | 2.852(11)         | 2.861(12) | 2.88(2)         |
| N6G···O3C <sup>iii</sup>   | 2.956(12) | 2.982(12)         | 2.960(12) | 2.94(2)         |
| N6G···O4C <sup>iii</sup>   | 3.435(13) | 3.351(13)         | 3.435(13) | 3.31(3)         |
| O1w···O1w <sup>v</sup>     | 2.763(13) | –                 | 2.761(12) | –               |
| O1w···O2w <sup>v</sup>     | 2.724(17) | –                 | 2.727(15) | –               |
| O1w···O3w                  | 2.94(2)   | –                 | 2.95(2)   | –               |
| O2w···O5w                  | 2.98(5)   | –                 | 3.14(4)   | –               |
| O2w···O6                   | 2.87(2)   | –                 | 2.91(2)   | –               |
| O2w···O2A <sup>vi</sup>    | 2.746(17) | –                 | 2.731(17) | –               |
| O4w···O2B <sup>vii</sup>   | 2.720(13) | –                 | 2.723(13) | –               |
| O4w···O4Z <sup>viii</sup>  | 2.79(2)   | –                 | 2.82(2)   | –               |
| O4w···O12Z                 | 2.822(19) | –                 | 2.820(17) | –               |

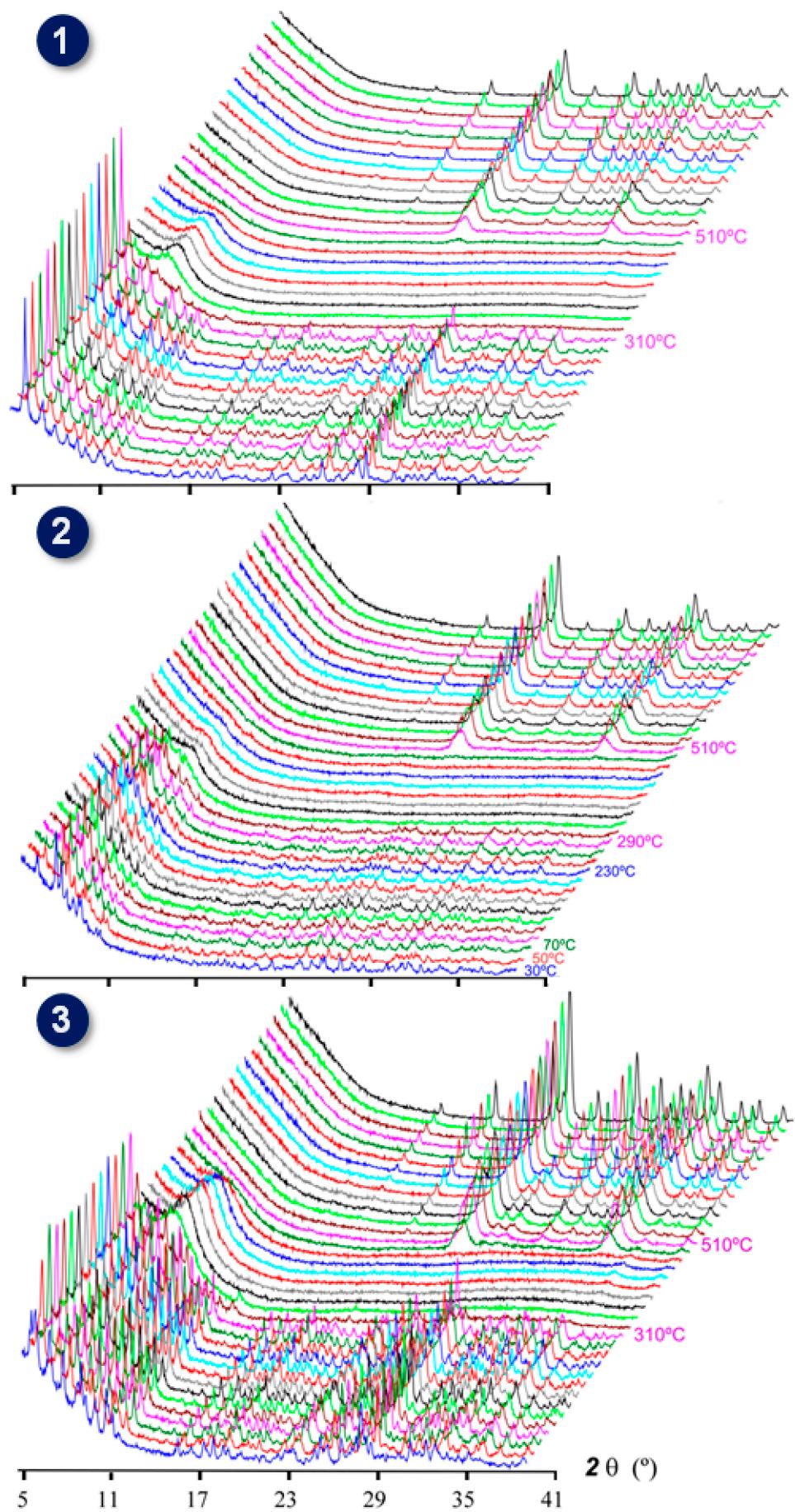
Note: Symmetry Codes: (i)  $-x, -y, -1-z$ ; (ii)  $x, y, -1+z$ ; (iii)  $-x, 1-y, -z$ ; (iv)  $1-x, 1-y, 1-z$ ; (v)  $-x, 1-y, 1-z$ ; (vi)  $-x, 1-y, -z$ ; (vii)  $x, -1+y, -1+z$ ; (viii)  $-x, -y, -z$ .



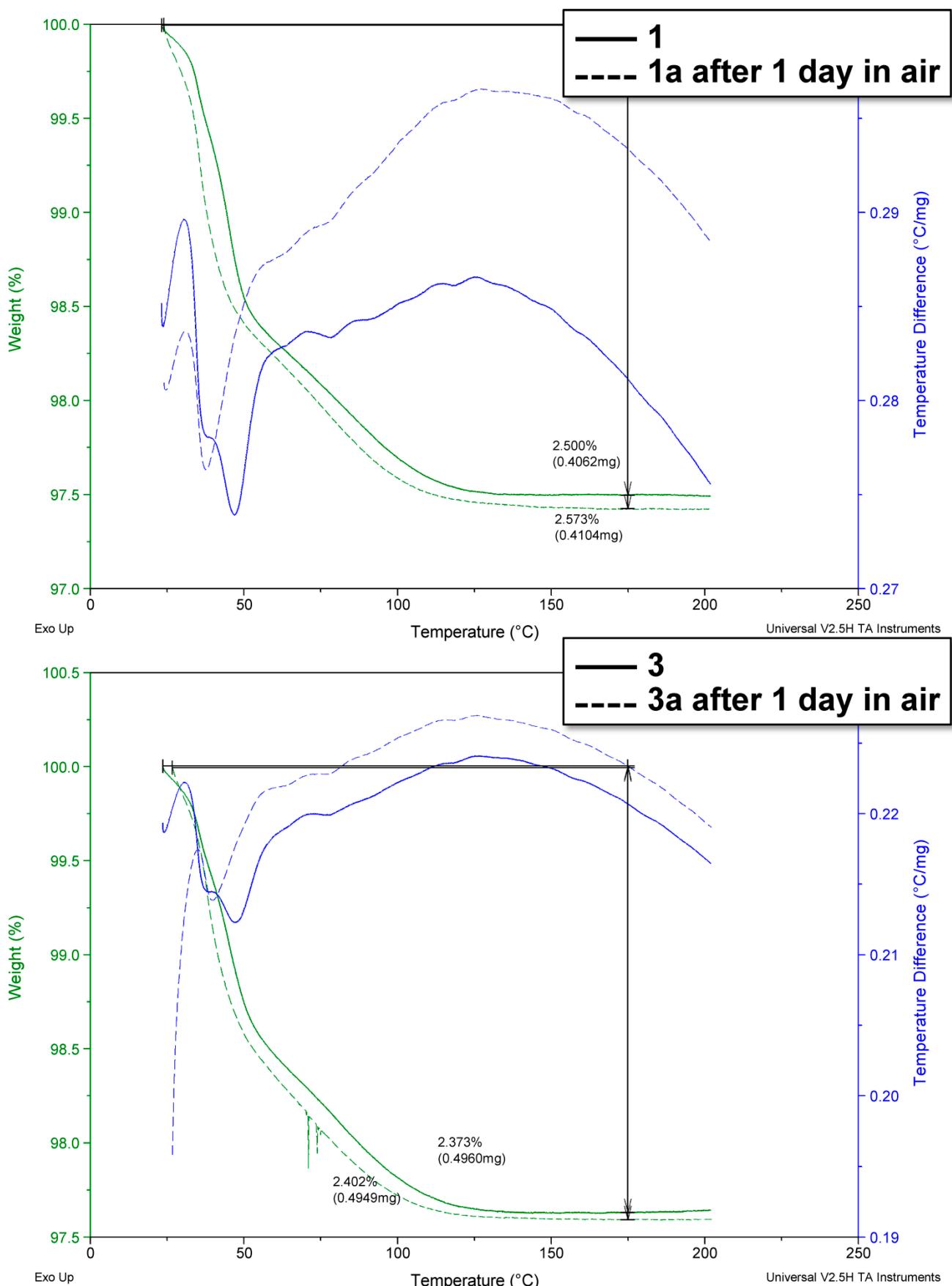
**Figure S3.** Detail of the N–H···O hydrogen bonds (dotted maroon lines) established by the guanidinium cations C1G and C2G hosted in the rectangular spaces inside the double-chained backbone of the  $\left[\{\text{SiW}_{12}\text{O}_{40}\}\{\text{Cu}(\text{pic})_2\}_{1.5}\right]^{4n-}$  polymer of **2**.



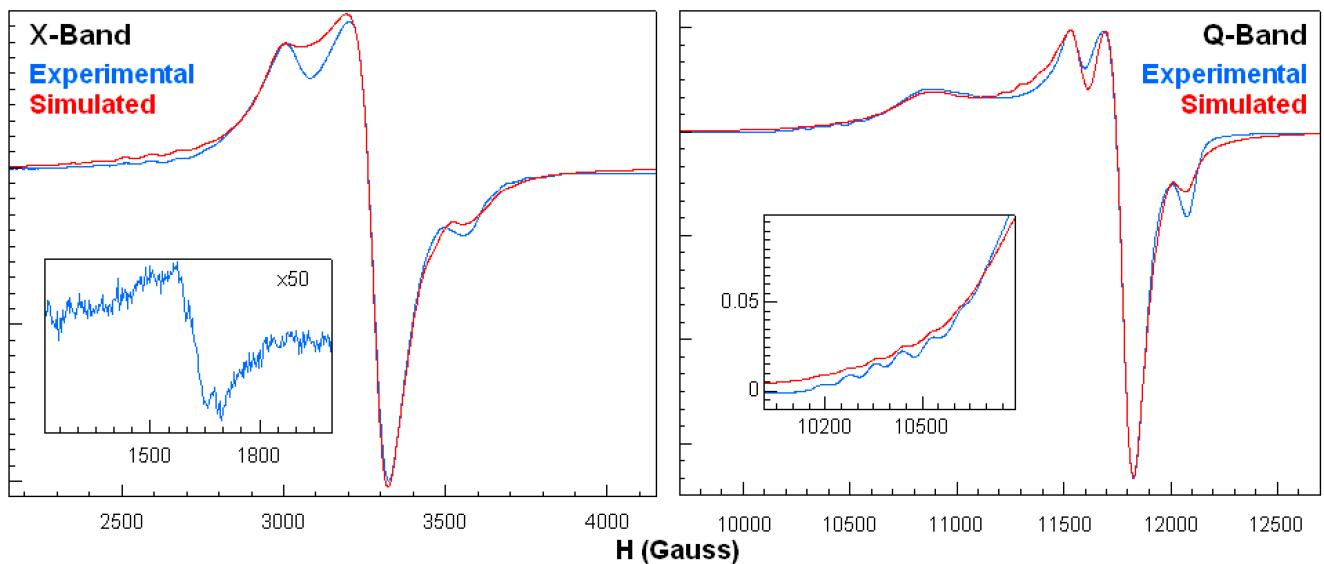
**Figure S4.** TGA/DTA curves of compounds **1–3**.



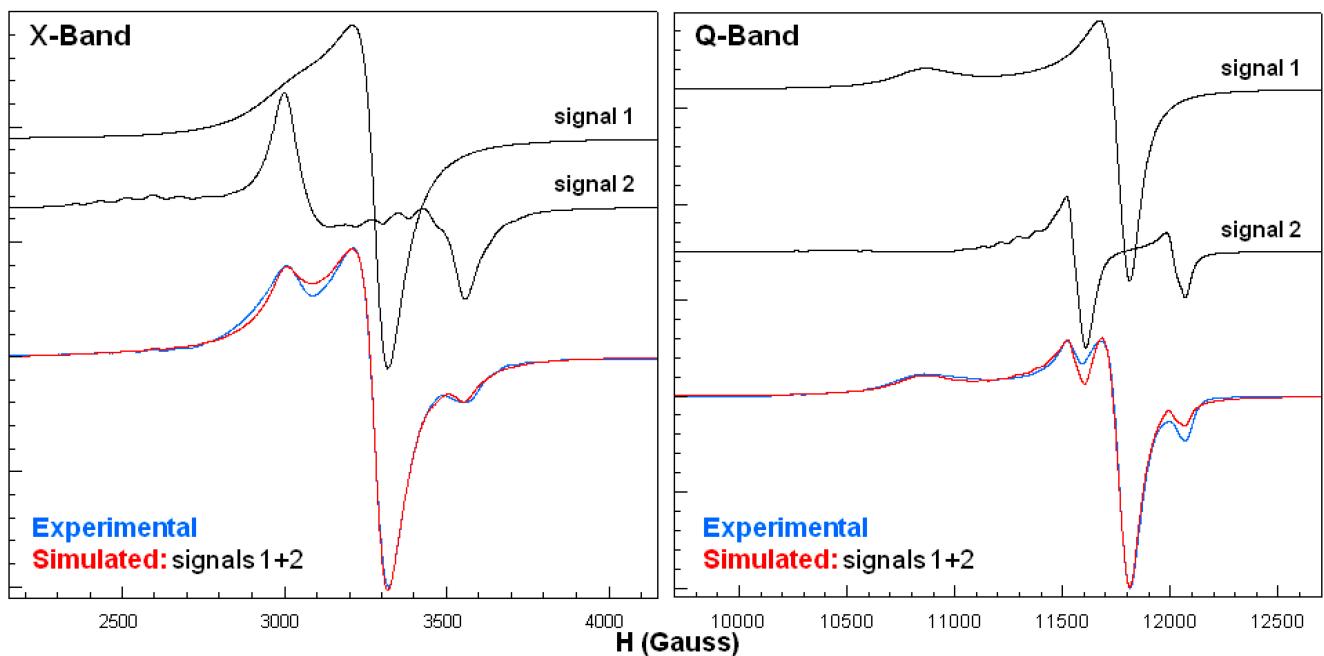
**Figure S5.** Variable temperature powder X-ray diffraction patterns of compounds **1–3**.



**Figure S6.** Comparative TGA/DTA curves for the dehydration of compounds **1/3** and the resulting phases **1a/3a** after being exposed to air for 1 day.



**Figure S7.** Experimental and simulated X-band ( $\nu = 9.49$  GHz) and Q-band ( $\nu = 34.05$  GHz) EPR spectra of **3** at room temperature.



**Figure S8.** Individual contributions used to simulate the experimental X-band and Q-band EPR spectra of **1** at room temperature.

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