

Supplementary Materials: Naphthyl-Containing Organophosphonate Derivatives of Keggin-Type Polyoxotungstates

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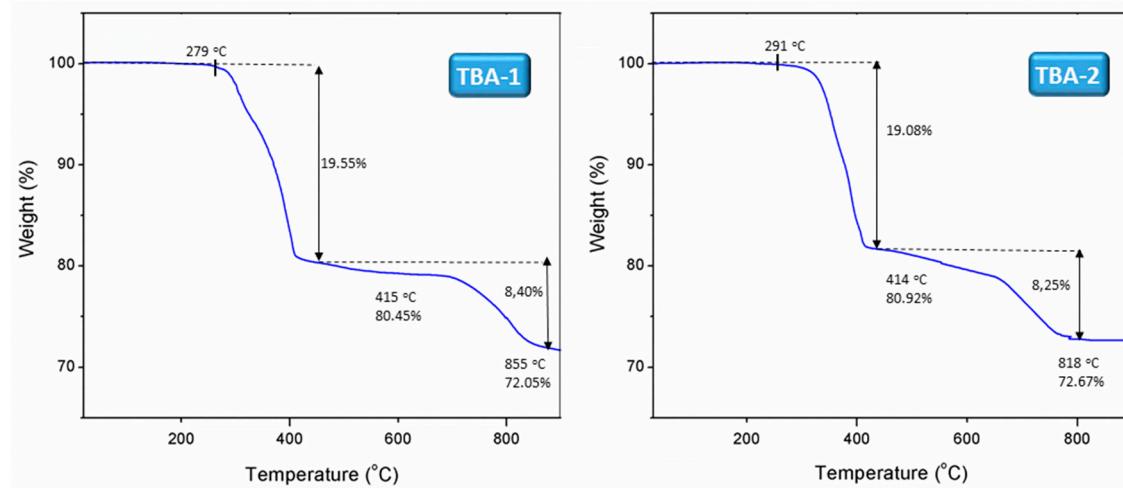


Figure S1. TGA curves for TBA-1 and TBA-2.

Table S1. Bond lengths (\AA) and angles ($^\circ$) for the P atoms in TBA-1 and TBA-2.

Atom	P–C	P–O _{POM1}	P–O _{POM2}	P–O _T	<P–C–C>
TBA-1					
P1	1.80(2)	1.514(16)	1.520(13)	1.533(17)	114.1(16)
P2	1.88(2)	1.498(15)	1.535(18)	1.51(2)	109(3)
P21	1.83(3)	1.522(17)	1.528(14)	1.530(18)	109.4(19)
P22	1.88(3)	1.562(17)	1.570(19)	1.499(18)	108(2)
TBA-2					
P1	1.81(5)	1.56(2)	1.585(16)	1.504(18)	109(3)
P2	1.75(4)	1.548(16)	1.532(14)	1.479(17)	109(2)
P21	1.77(3)	1.547(19)	1.526(14)	1.507(16)	107(2)
P22	1.76(4)	1.56(2)	1.535(16)	1.485(17)	111(3)

Table S2. Geometrical parameters (\AA , $^\circ$) for the intermolecular C–H \cdots O hydrogen bonds involving 1-naphthylmethylphosphonate groups and O_{POM} atoms in **TBA-1** and **TBA-2**.

D–H \cdots A	H \cdots A	D \cdots A	$\langle \text{D–H} \cdots \text{A} \rangle$
TBA-1			
C8–H8 \cdots O223	2.57	3.30(3)	133
C8–H8 \cdots O229	2.52	3.39(3)	152
C9–H9 \cdots O207	2.55	3.48(4)	169
C29–H29 \cdots O209 ⁱ	2.42	2.966(17)	116
C208–H208 \cdots O14 ⁱⁱ	2.66	3.30(3)	126
C208–H208 \cdots O15 ⁱⁱ	2.60	3.37(4)	138
C208–H208 \cdots O20 ⁱⁱ	2.97	3.58(3)	123
C209–H209 \cdots O3 ⁱⁱ	2.57	3.50(4)	167
C209–H209 \cdots O14 ⁱⁱ	2.98	3.50(3)	116
C229–H229 \cdots O11 ⁱⁱⁱ	2.54	3.02(5)	112
TBA-2			
C9–H9 \cdots O211 ^{iv}	2.35	2.98(4)	124
C28–H28 \cdots O214	2.70	3.388(16)	129
C28–H28 \cdots O215	2.79	3.553(16)	138
C28–H28 \cdots O220	2.96	3.599(15)	125
C29–H29 \cdots O203	2.44	3.38(3)	167
C208–H208 \cdots O14 ⁱⁱ	2.66	3.364(15)	131
C208–H208 \cdots O15 ⁱⁱ	2.91	3.569(15)	127
C208–H208 \cdots O20 ⁱⁱ	2.78	3.529(16)	137
C209–H209 \cdots O1 ⁱⁱ	2.46	3.387(16)	165
C229–H229 \cdots O10 ^v	2.23	2.87(4)	124

Symmetry codes: (i) $-x, 3 - y, 1 - z$; (ii) $x, -1 + y, z$; (iii) $1 - x, 3 - y, 1 - z$; (iv) $2 - x, -y, 1 - z$; (v) $1 - x, -y, 1 - z$.

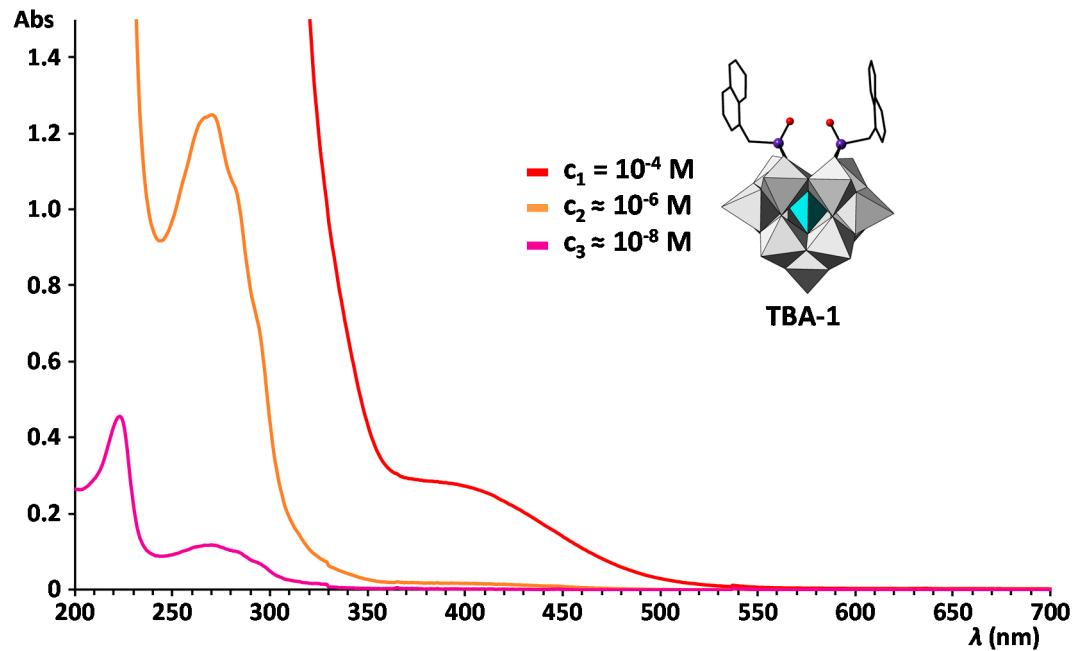


Figure S2. UV–Vis spectra of acetonitrile solutions of **TBA-1** with different concentrations.

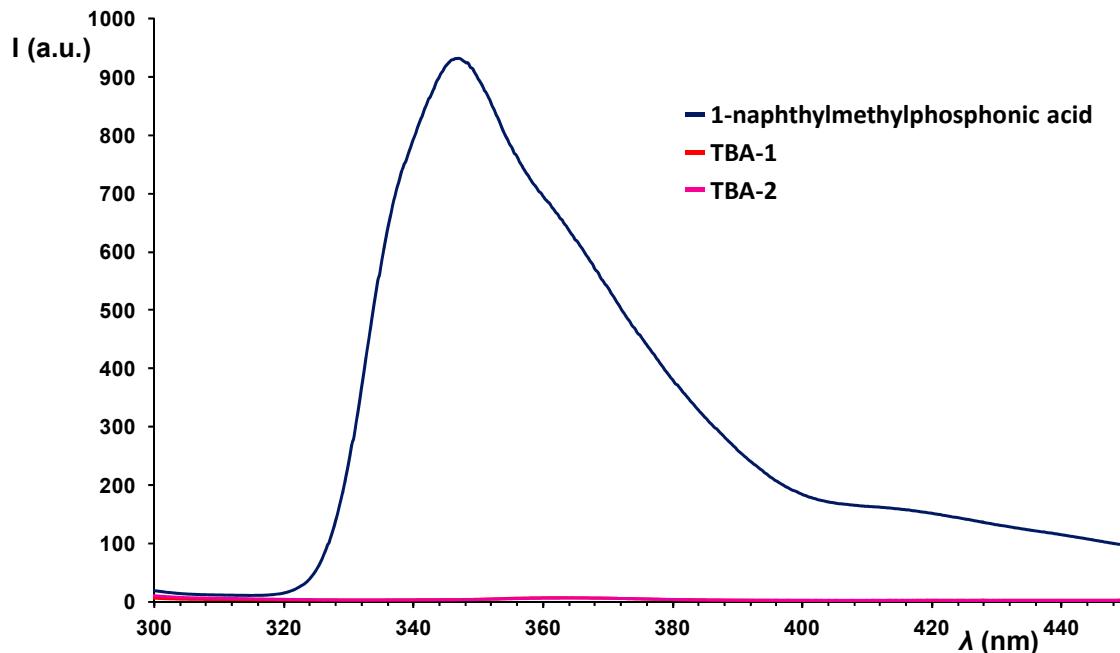


Figure S3. Fluorescence emission spectra of solid samples of **TBA-1**, **TBA-2** and 1-naphthylmethylphosphonic acid in the 300–450 nm region ($\lambda_{\text{exc}} = 283 \text{ nm}$).

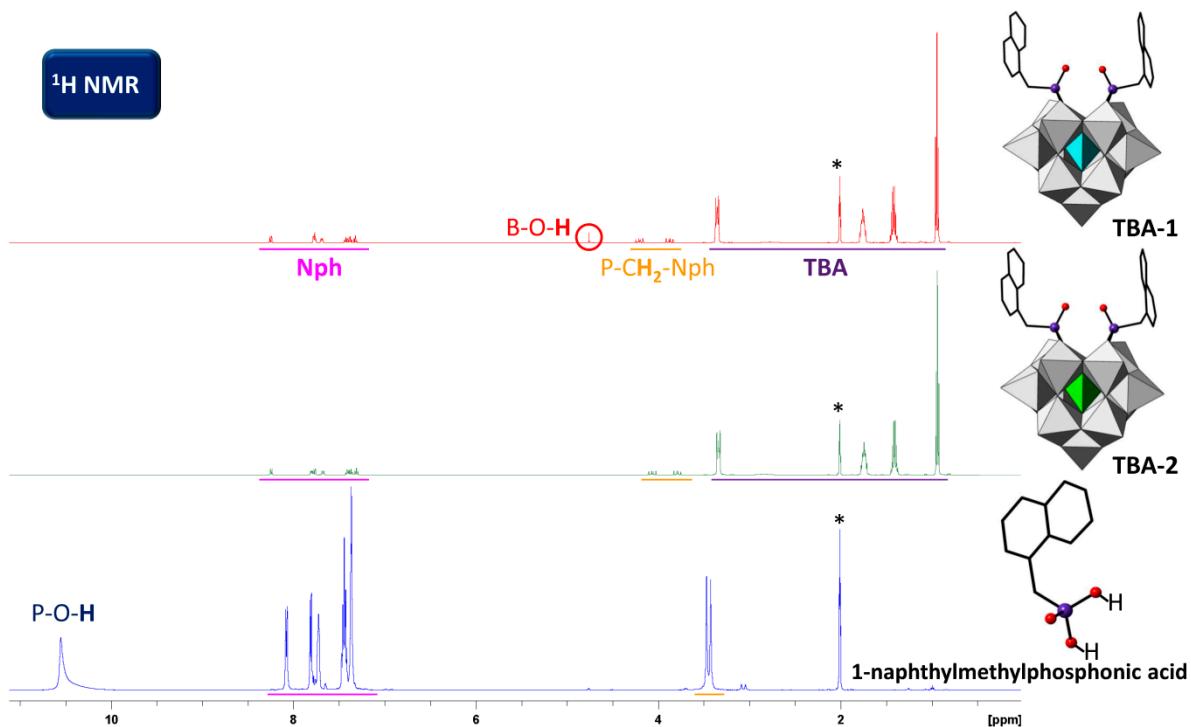


Figure S4. ^1H -NMR spectra of **TBA-1** and **TBA-2** in acetone- d_6 compared with that of the commercial 1-naphthylmethylphosphonic acid. The signals labeled as * correspond to the non-deuterated acetone.

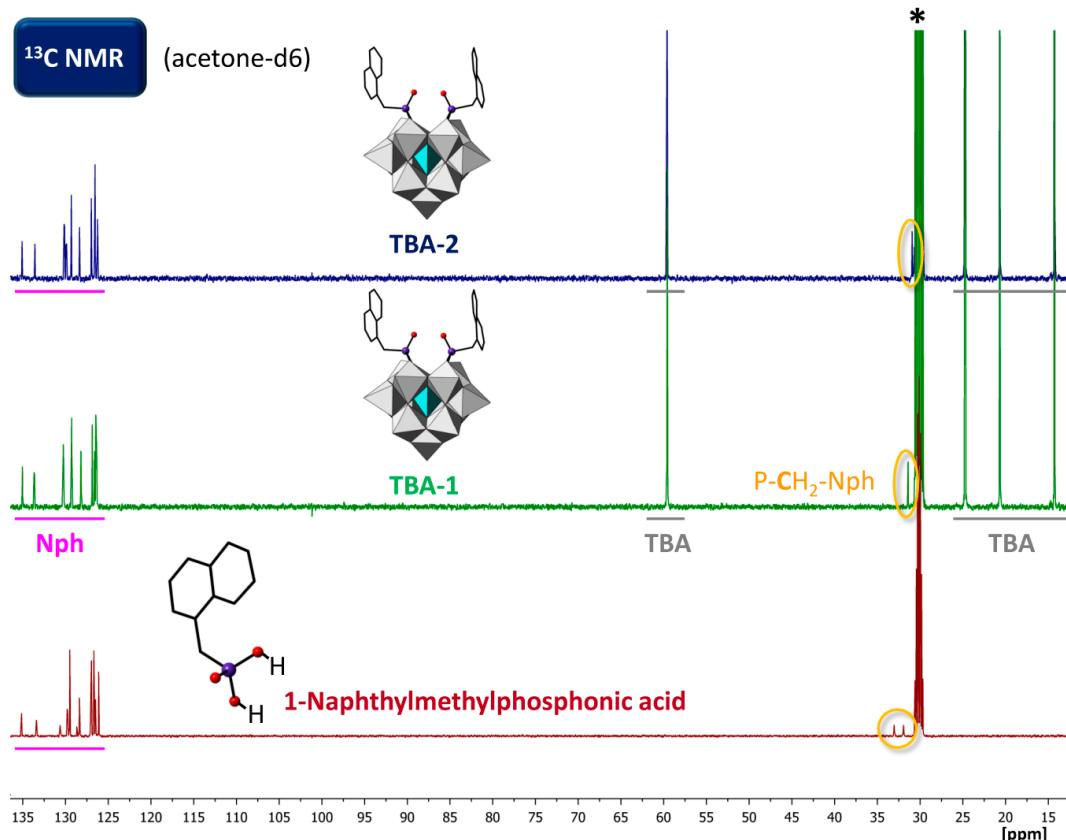


Figure S5. ^{13}C -NMR spectra of **TBA-1** and **TBA-2** in acetone- d_6 compared with that of the commercial 1-naphthylmethylphosphonic acid. The signals labeled as * correspond to the solvent.

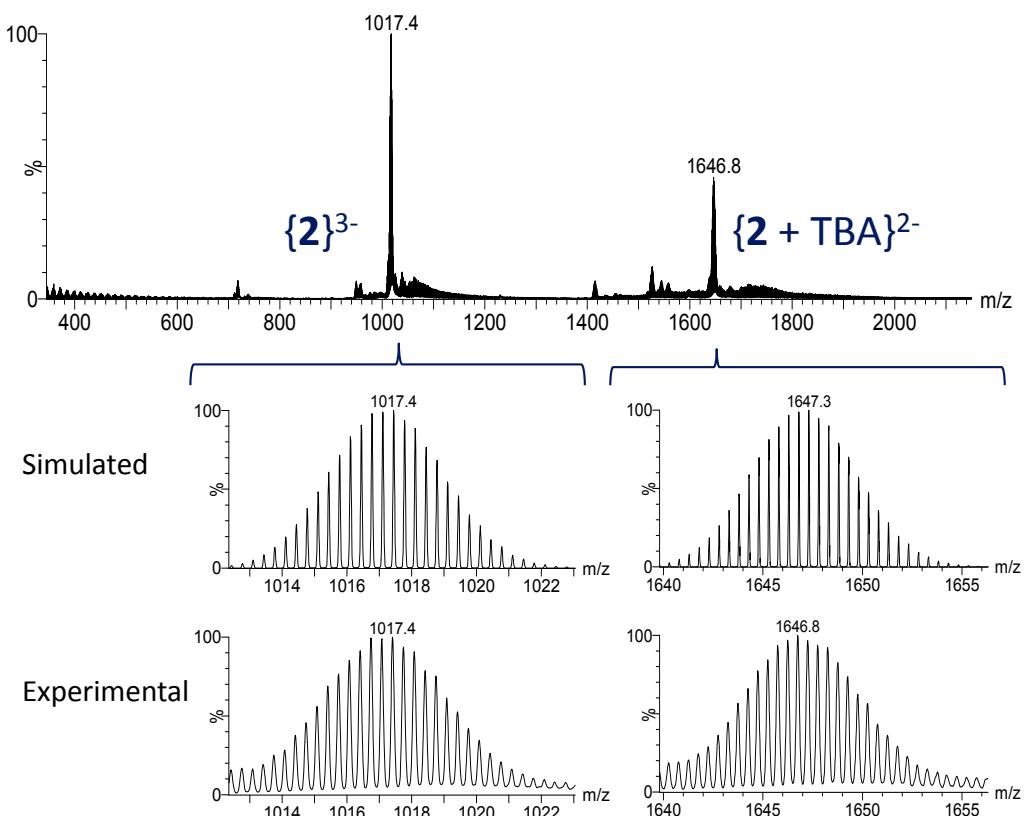


Figure S6. Negative ESI mass spectrum of a CH_3CN solution of **TBA-2** ($U_c = 15$ V) and comparison of the signals corresponding to the species $\{2\}^{3-} = [\text{H}(\text{C}_{11}\text{H}_9\text{PO})_2(\text{SiW}_{11}\text{O}_{39})]^{3-}$ and $\{2 + \text{TBA}\}^{2-} = \{(\text{C}_{16}\text{H}_{36}\text{N})[\text{H}(\text{C}_{11}\text{H}_9\text{PO})_2(\text{SiW}_{11}\text{O}_{39})]\}^{2-}$ with the simulated isotopic patterns.