

Supplementary Materials: Mechanochemical Induced Structure Transformations in Lithium Titanates: A Detailed PXRD and ${}^6\text{Li}$ MAS NMR Study

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Formation of α - Li_2TiO_3 from LiOH and TiO_2

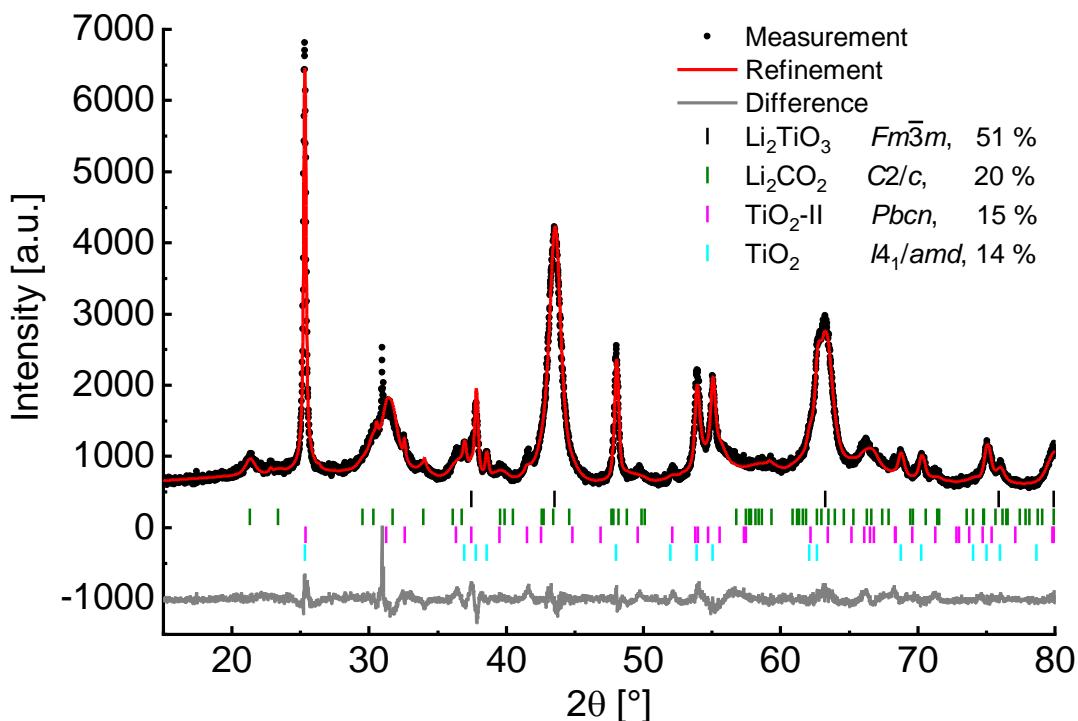


Figure S1: Rietveld refinement of the milling product from LiOH and anatase (ZrO_2 tools, 600 rpm, 6h). Li_2TiO_3 in SG $Fm\bar{3}m$ with $a = 4.1555(2)$ Å, $V = 71.76(1)$ Å³, number of reflections = 8 and $R_{\text{Bragg}} = 0.99\%$. Refinement parameters: number of independent parameters = 46, $R_{wp} = 5.41\%$, $R_{exp} = 3.30\%$, $GOF = 1.64$.

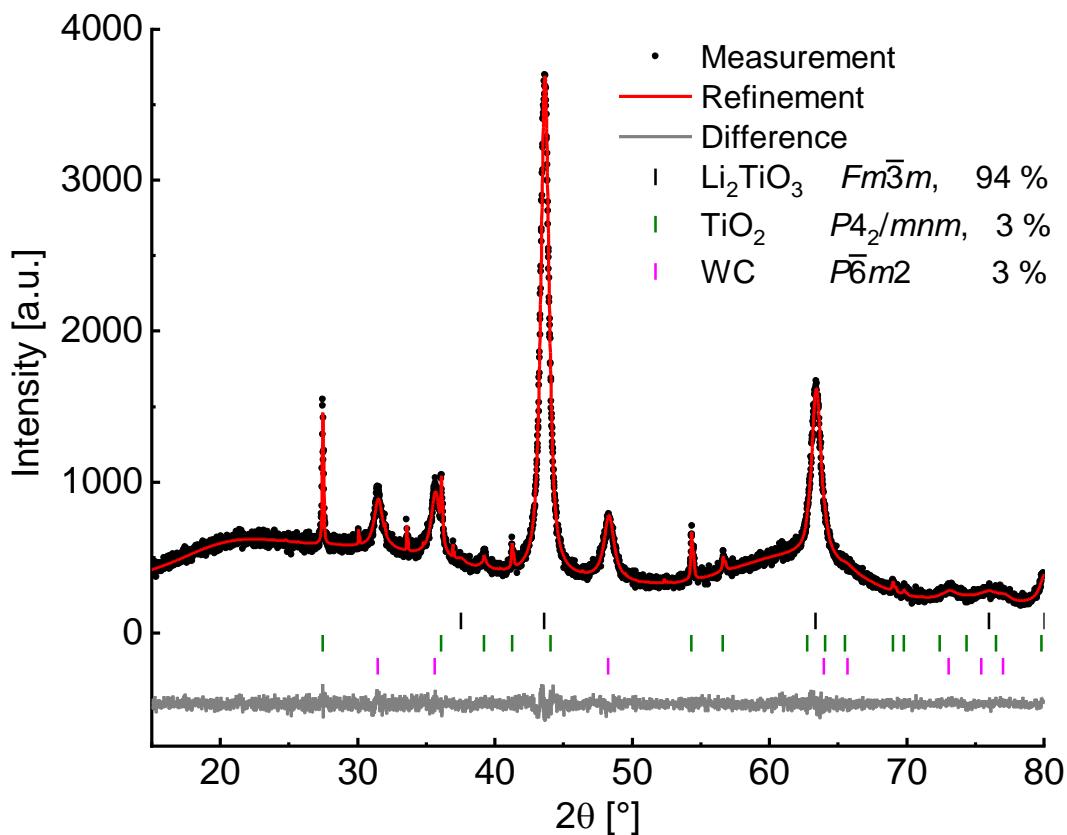


Figure S2: Rietveld refinement of the milling product from LiOH and rutile (WC tools, 600 rpm, 6h) after handling in the glovebox. Measurement was conducted with an air-tight dome sample holder (responsible for the background signal). The small percentage of crystalline TiO_2 is due to a small unreacted residue between milling vial and lid.

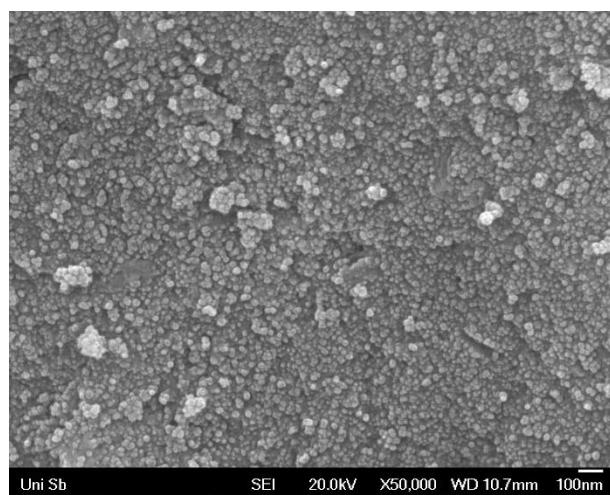


Figure S3: SEM photograph of the milling product from LiOH and rutile in WC tools at 600 rpm and 6h.

Table S1: Refined structure parameters of α -Li₂TiO₃ produced by milling of LiOH and rutile in WC tools at 600 rpm for 6 h. B_{iso} was constrained at the same value for all atoms in the refinement. Global refinement parameters: number of independent parameters = 40, $R_{wp} = 3.44\%$, $R_{exp} = 2.60\%$, $GOF = 1.32$. WP = Wyckoff position, sof = site occupancy factor.

SG	<i>Fm</i> $\bar{3}m$						
R _{Bragg}	1.17%						
No. of reflections	8						
Wt% Rietveld	68.3(4)						
Cell Mass	150.6(5)						
Cell Volume [Å ³]	71.93(1)						
Lattice Parameter [Å]	4.1588(2)						
Crystallite Size L _{vol} -IB [nm]	15.7(4)						
Strain e_0	0.18742						
Lin. Abs. Coeff. [cm ⁻¹]	358(4)						
Crystal Density [g/cm ³]	3.48(1)						
Site	WP	x	y	z	Atom	sof	B_{iso}
Li1	4a	0	0	0	Li ⁺	0.46(2)	1.34(2)
	4a				Ti ⁴⁺	0.39(6)	1.34(2)
O1	4b	1/2	1/2	1/2	O ²⁻	1	1.34(2)

Thermogravimetric analysis

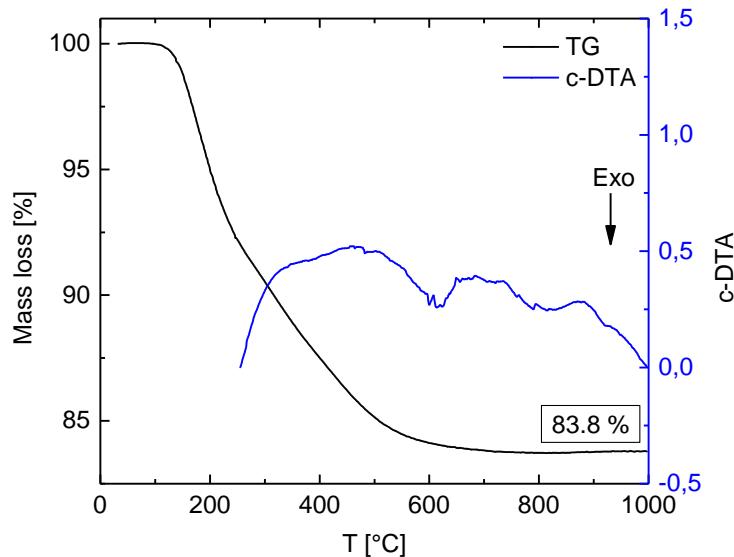


Figure S4: TGA and calculated DTA curve of Li_2TiO_3 from milling of LiOH and rutile in WC tools at 600 rpm for 6 h. Intermediate drying in N_2 atmosphere at 100 °C led to mass loss of 3%.

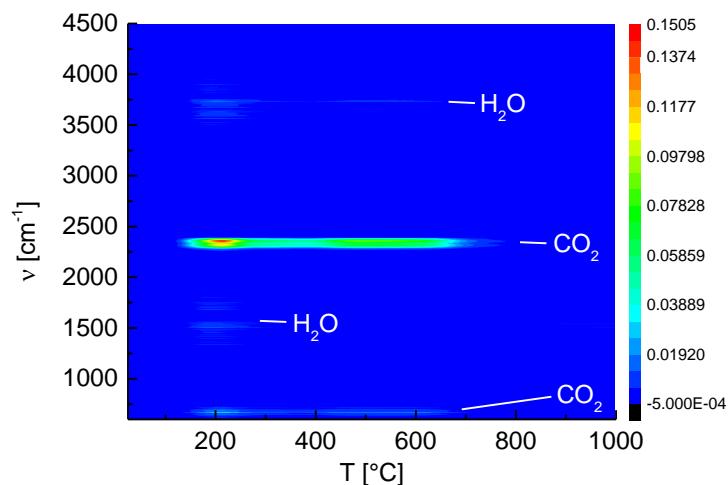


Figure S5: Temperature dependent IR signals from coupling of TGA gas flow to IR detector. Colors indicate intensity of the signals.

Thermal transformation of α -Li₂TiO₃ to β -Li₂TiO₃

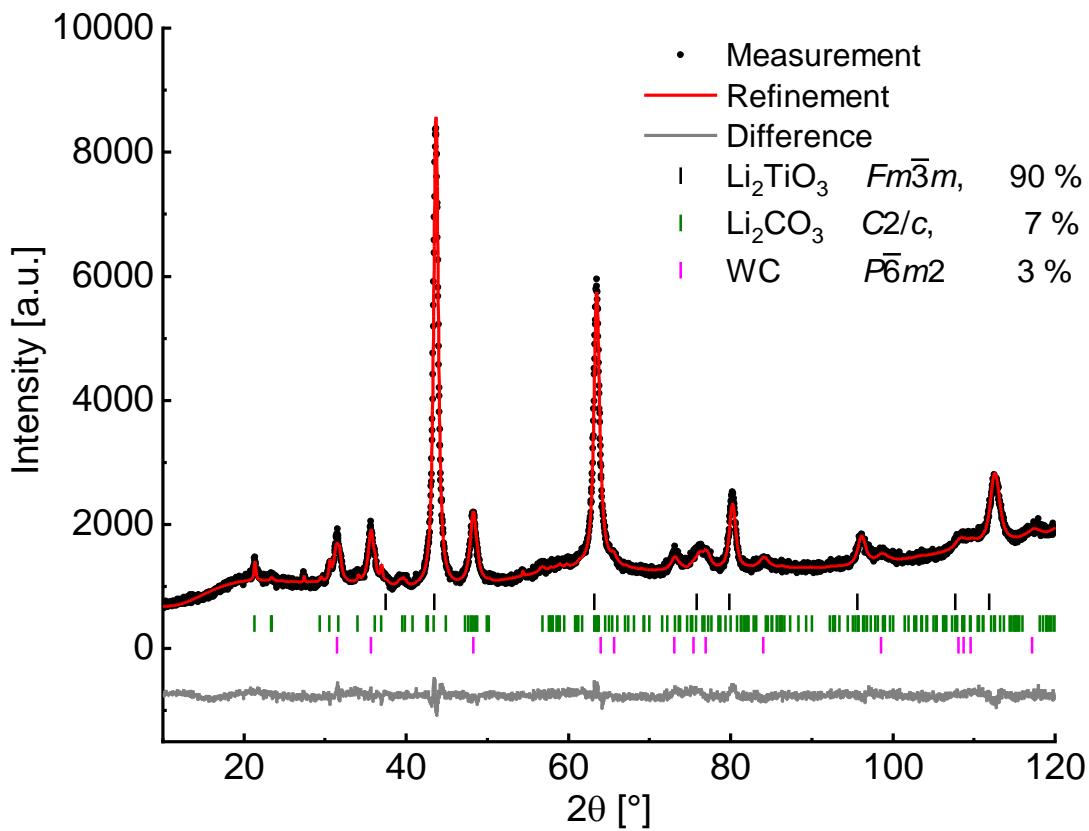


Figure S6: Rietveld refinement of the milling product from LiOH and rutile (WC tools, 600 rpm, 6h) after heating to 200 °C for 1 h. Li₂TiO₃ in SG $Fm\bar{3}m$ with $a = 4.1435(1)$ Å, $V = 71.14(1)$ Å³, number of reflections = 8 and $R_{Bragg} = 1.50\%$. Refinement parameters: number of independent parameters = 48, $R_{wp} = 3.08\%$, $R_{exp} = 2.63\%$, $GOF = 1.17$.

Transformation of β -Li₂TiO₃ and spinel Li₄Ti₅O₁₂ to α -Li₂TiO₃

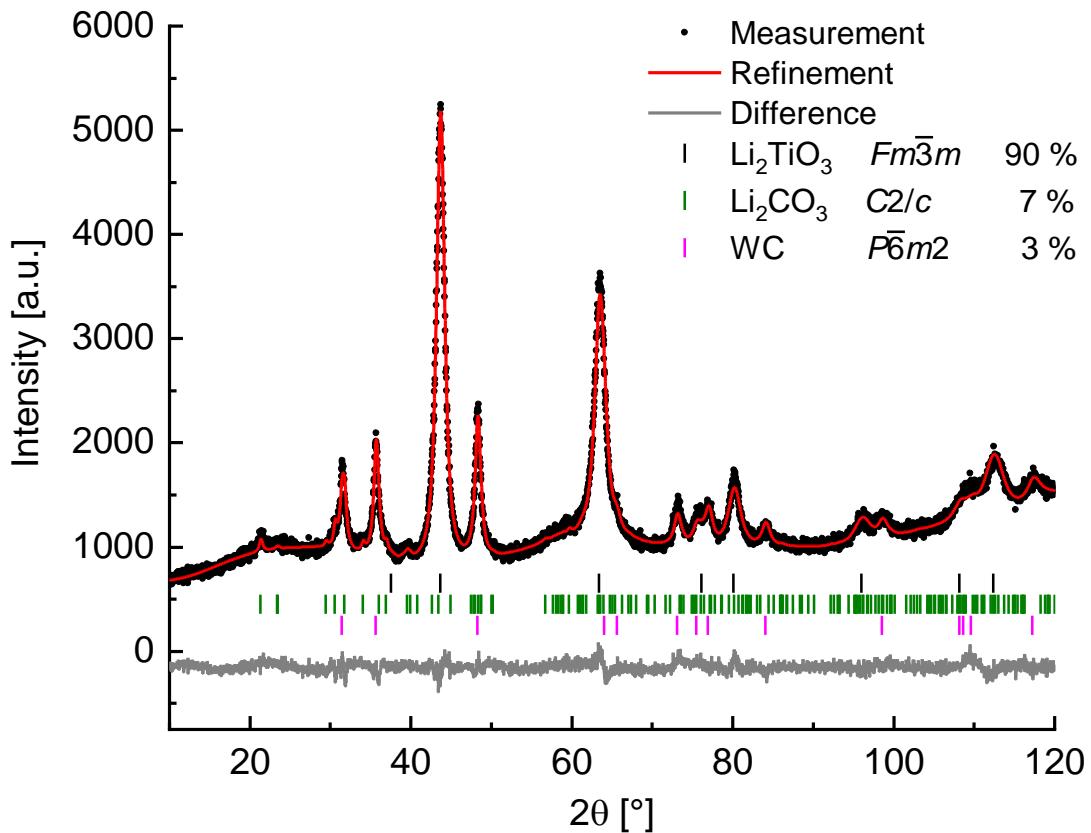


Figure S7: Rietveld refinement of β -Li₂TiO₃ after ball milling with WC tools for 6 h at 600 rpm. Li₂TiO₃ in SG $Fm\bar{3}m$ with $a = 4.1463(2)$ Å, $V = 71.28(1)$ Å³, number of reflections = 8 and $R_{Bragg} = 0.60\%$. Refinement parameters: number of independent parameters = 31, $R_{wp} = 3.60\%$, $R_{exp} = 2.85\%$, $GOF = 1.26$.

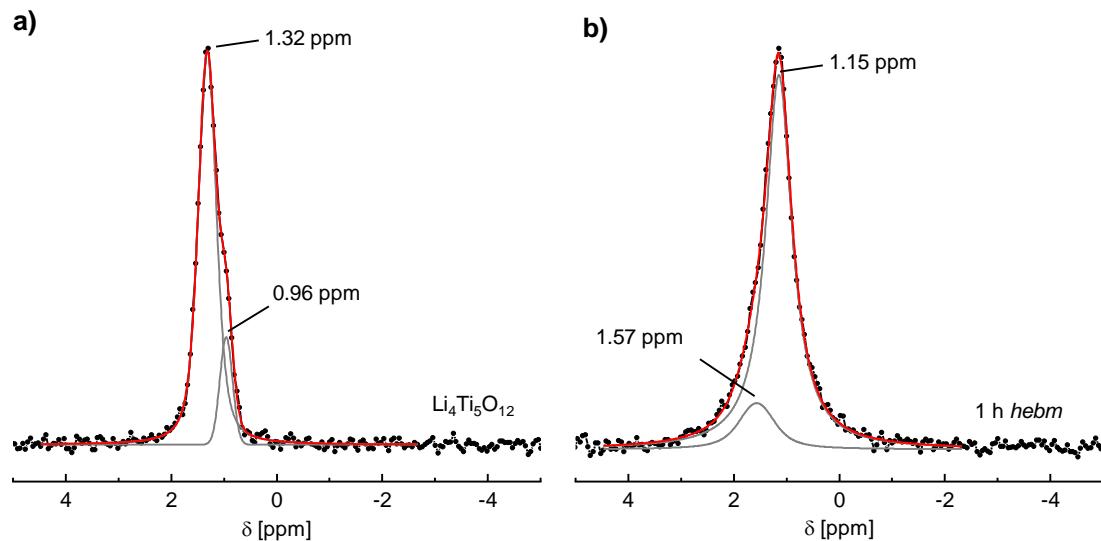
⁶Li SPE MAS NMR of Li₄Ti₅O₁₂

Figure S8: ⁶Li SPE MAS NMR spectra of a) spinel Li₄Ti₅O₁₂, b) after *hebm* for 1 h. The spectra were referenced against solid LiCl and fitted with two independent Voigt functions with R² > 0.99.