

## Article

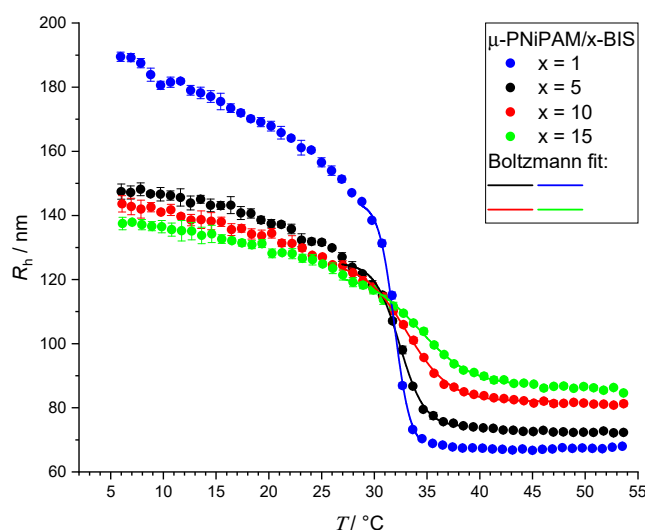
# Interplay of the Influence of Crosslinker Content and Model Drugs on the Phase Transition of Thermoresponsive PNiPAM-BIS Microgels

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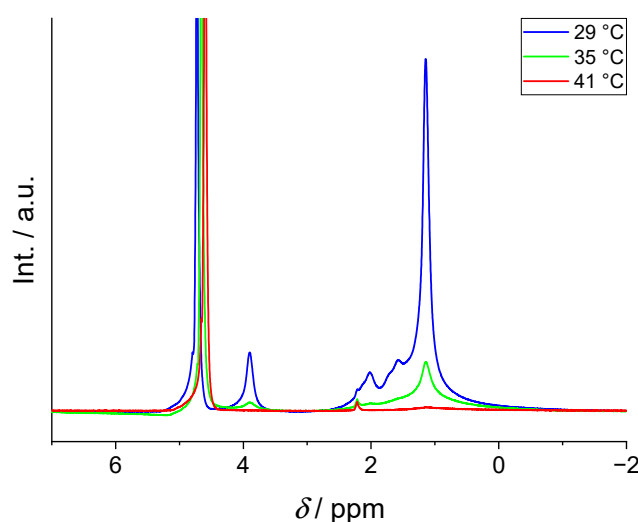
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## Supplementary Materials



**Figure S1.** Temperature dependent hydrodynamic radii  $R_h$  of microgel structures containing different amounts of crosslinker. Results shown from a heating cycle starting at low temperatures. Lines are fits according to equation (4).



**Figure S2.**  $^1\text{H}$ -NMR temperature series of a microgel acquired with equal spectral parameters.

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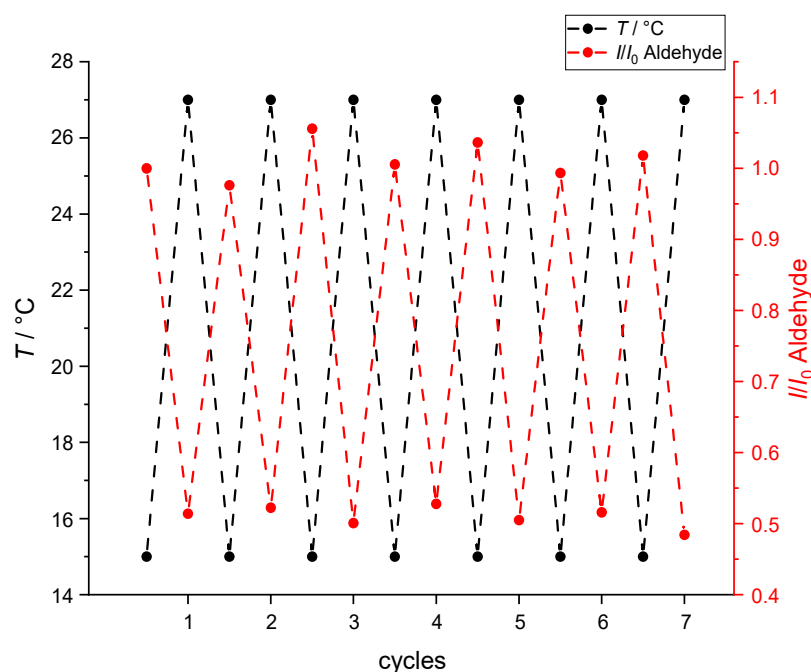
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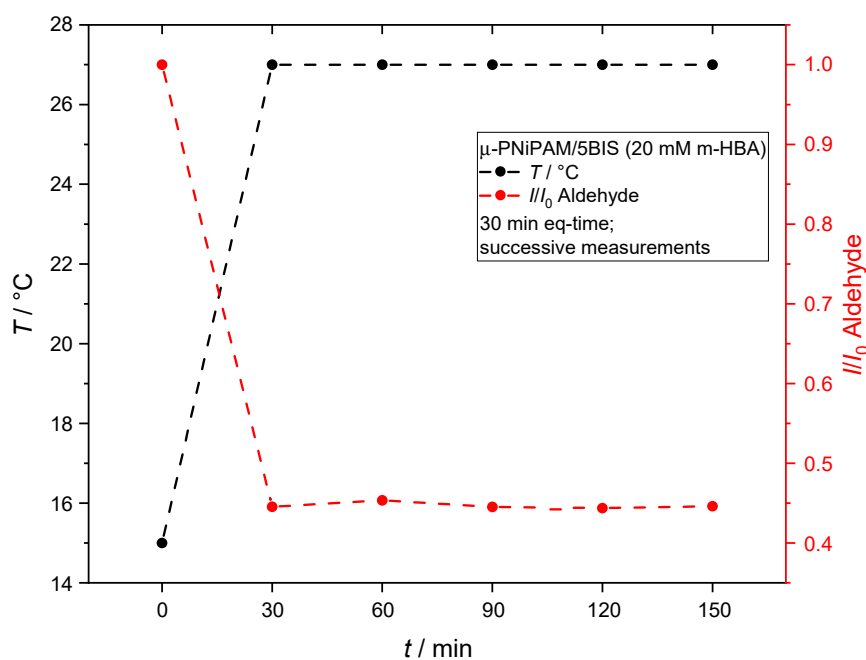
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**Figure S3.** Integrated aldehyde  $^1\text{H}$  signal for  $\mu\text{-PNiPAM}/1\text{-BIS}$  with 20 mM *m*-HBA for a series of temperature jumps. Integrals are normalized to the value measured at 15 °C.

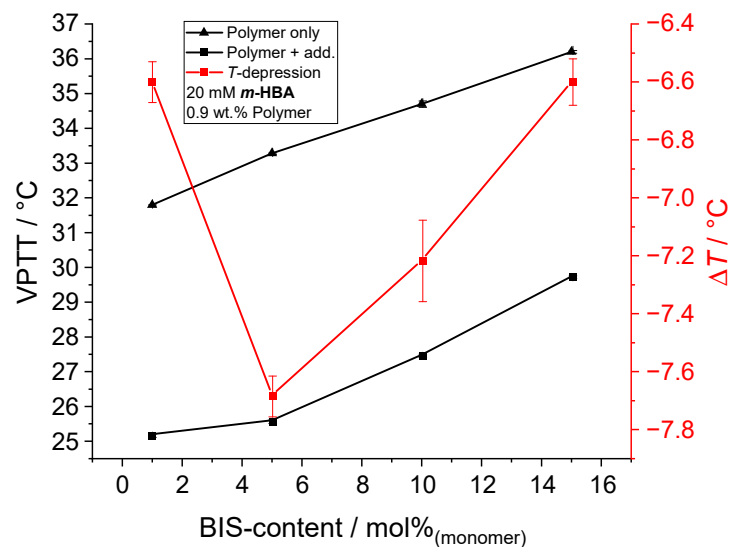
It could be shown, that the incorporation of additive is fully reversible and can be accomplished in the investigated time intervals.

NMR also revealed that incorporation of additives is *not* time dependent in the investigated period of time (2.5 h). Integrals are normalized to 15 °C. Equilibration time was 30 min at least. Data show no significant deviation from reached value at 27 °C and is forming a plateau, indicating no further reduction of  $P_{\text{inc.}}$  with time:

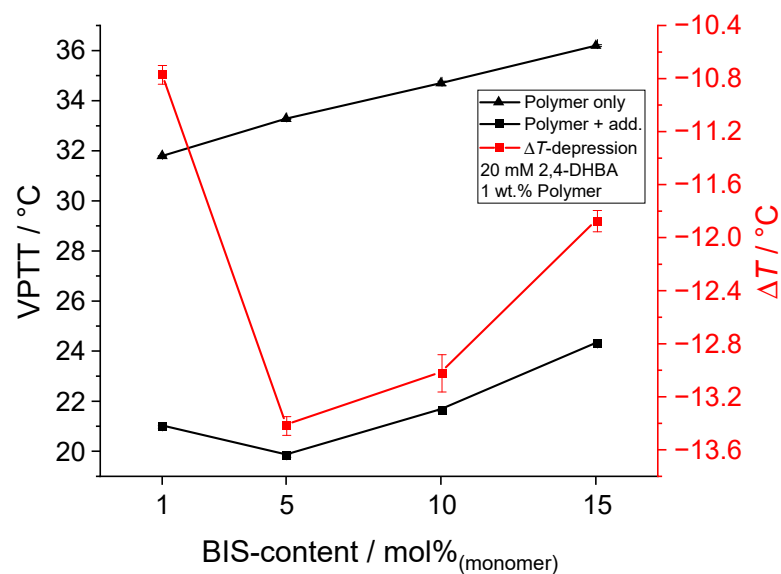


**Figure S4.** Integrated aldehyde signal for  $\mu\text{-PNiPAM}/1\text{-BIS}$  with 20 mM *m*-HBA in dependency of time, following a temperature jump.

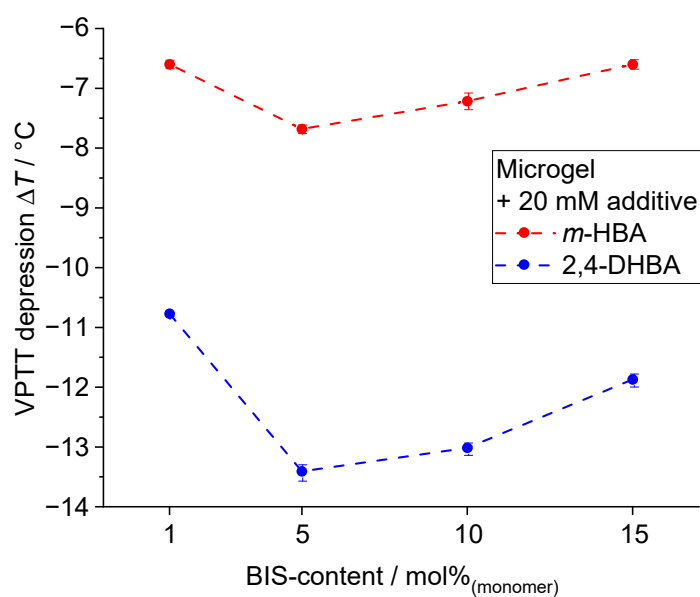
## DSC



**Figure S5.** Plot of the VPTT (triangles: polymer only, black squares: depressed temperature) and calculated  $\Delta T$  (red squares) against the crosslinker content for *m*-HBA.



**Figure S6.** Plot of the VPTT (triangles: polymer only, black squares: depressed Temperature) and calculated  $\Delta T$  (red squares) against the crosslinker content for 2,4-DHBA.



**Figure S7.** Calculated microgel VPTT depression  $\Delta T$  from DSC data in presence of two different aromatic additive structures. Concentrations were 1 wt.% for the microgels and 20 mM for each additive. Error bars do reflect standard deviation. Colorcode corresponds to the microgels VPTT influenced by additives relative to each other (low  $T$  = blue; high  $T$  = red).

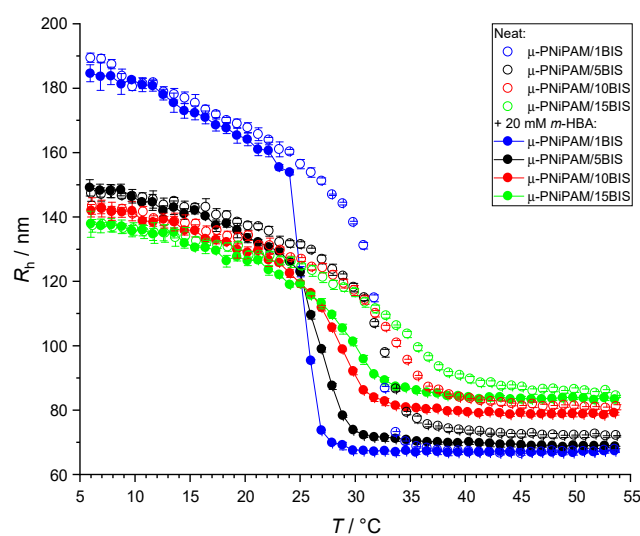
## DLS

**Table S1.** Summary of DLS data for neat  $\mu$ -PNiPAM/ $x$ -BIS microgels in  $H_2O$  and comparison with reported data from Friesen et al.

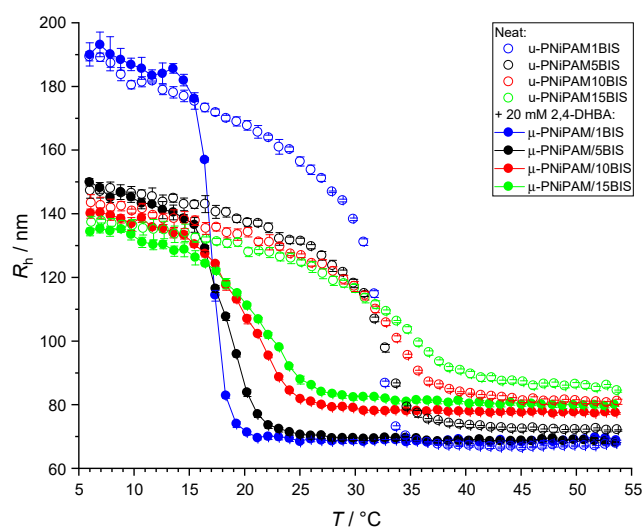
	$R_{h,15\text{ }^\circ\text{C}} / \text{nm}$	$R_{h,50\text{ }^\circ\text{C}} / \text{nm}$	$R_{sw,15\text{ }^\circ\text{C}/50\text{ }^\circ\text{C}}$	VPTT/ $^\circ\text{C}$	VPTT/ $^\circ\text{C}$	$\text{PDI}_{25\text{ }^\circ\text{C}} \pm 0.02$
$\mu$ -PNiPAM/1BIS	176	67	2.6	31.9	-	0.06
$\mu$ -PNiPAM/2.5BIS	-	-	-	-	34.1*	-
$\mu$ -PNiPAM/5BIS	143	72	2.0	32.4	35.7*	0.02
$\mu$ -PNiPAM/10BIS	138	81	1.7	33.0	37.1*	0.04
$\mu$ -PNiPAM/15BIS	132	87	1.5	34.3	38.6*	0.03

\*Friesen, S., et al., Accounting for Cooperativity in the Thermotropic Volume Phase Transition of Smart Microgels. Gels, 2021. 7(2): p. 42.

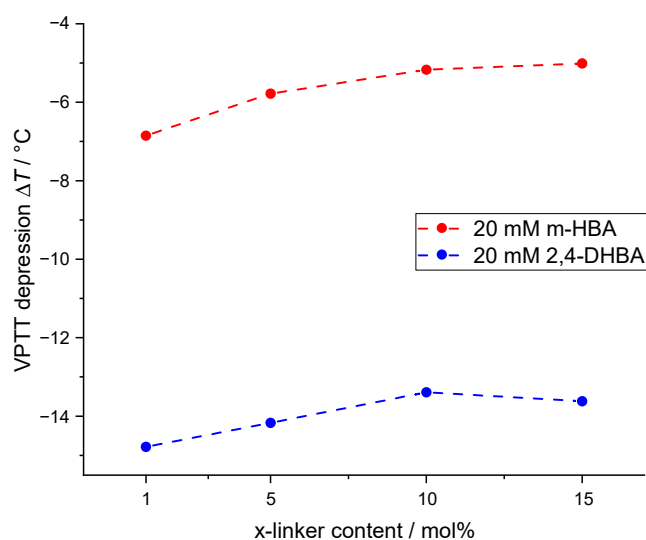
We note that absolute phase transition temperatures may differ, caused by different synthesis protocols. Nonetheless, temperature shifting trend remains the same.



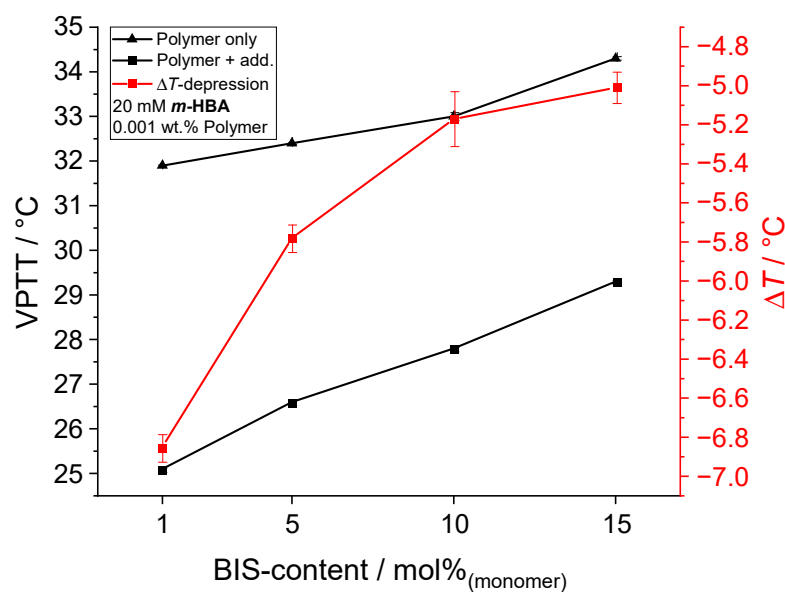
**Figure S8.** Temperature dependent hydrodynamic radii  $R_h$  of microgel particles containing different amounts of crosslinker (hollow circles: neat Polymer; solid circles: 20 mM *m*-HBA). Results shown from a heating cycle starting at low temperatures. Lines are guide to the eye.



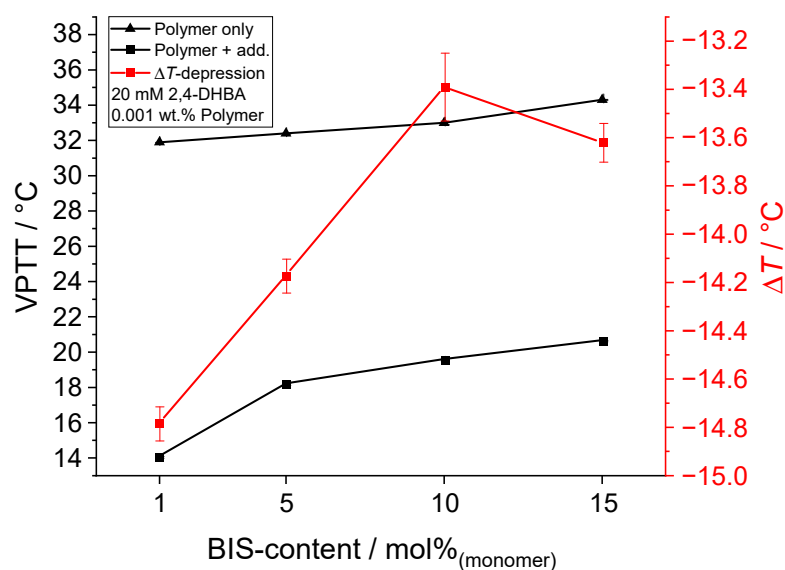
**Figure S9.** Temperature dependent hydrodynamic radii  $R_h$  of microgel particles containing different amounts of crosslinker (hollow circles: neat Polymer; solid circles: 20 mM 2,4-DHBA). Results shown from a heating cycle starting at low temperatures. Lines are guide to the eye.



**Figure S10.** Calculated microgel VPTT depression  $\Delta T$  from DLS data in presence of two different aromatic additive structures. Concentrations were 1 wt.% for the microgels and 20 mM for each additive. Error bars do reflect standard deviation. Colorcode corresponds to the microgels VPTT influenced by additives relative to each other (low  $T$  = blue; high  $T$  = red).

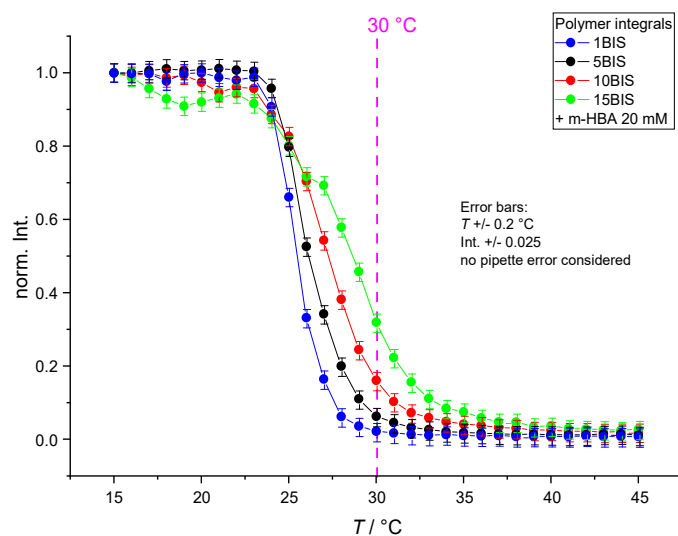


**Figure S11.** Plot of the VPTT (triangles: polymer only, black squares: depressed Temperature) and calculated  $\Delta T$  (red squares) against the crosslinker content for *m*-HBA.

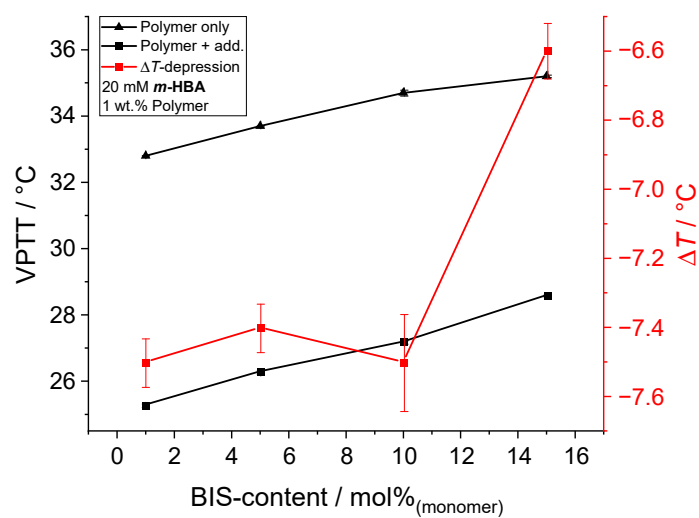


**Figure S12.** Plot of the VPTT (triangles: polymer only, black squares: depressed Temperature) and calculated  $\Delta T$  (red squares) against crosslinker content for 2,4-DHBA.

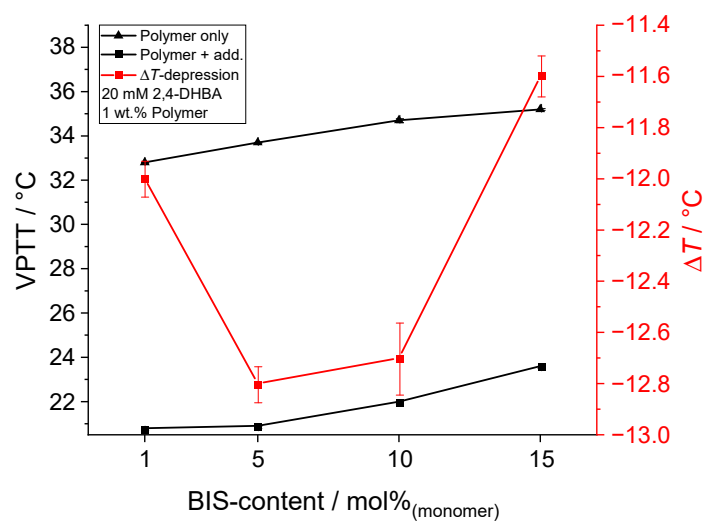
## NMR



**Figure S13.** Integrated normalized polymer signals as acquired in  $^1\text{H}$ -NMR temperature series for neat microgels containing different crosslinker content and 20 mM of *m*-HBA. Lines are guide to the eye.

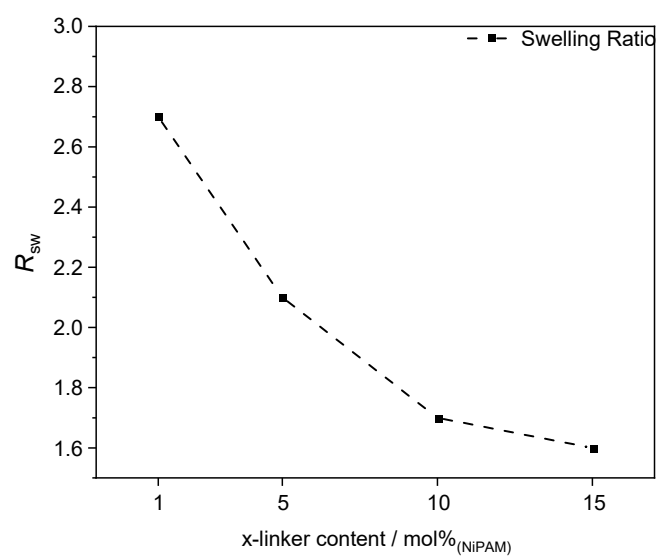


**Figure S14.** VPTT (triangles: polymer only, black squares: depressed Temperature) and calculated  $\Delta T$  (red squares) against the crosslinker content for *m*-HBA.



**Figure S15.** Plot of the VPTT (triangles: polymer only, black squares: depressed Temperature) and calculated  $\Delta T$  (red squares) against crosslinker content for 2,4-DHBA.





**Figure S16.** Swelling ratios  $R_{sw}$  for differently crosslinked microgels.