

*Supplementary Materials*

# **Ionogels Derived from Fluorinated Ionic Liquids to Enhance Aqueous Drug Solubility for Local Drug Administration**

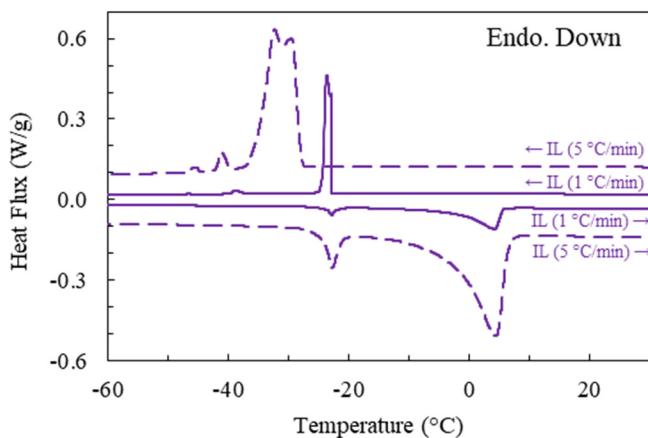
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**Table S1.** pHs of the IL and IL/H<sub>2</sub>O emulsions.

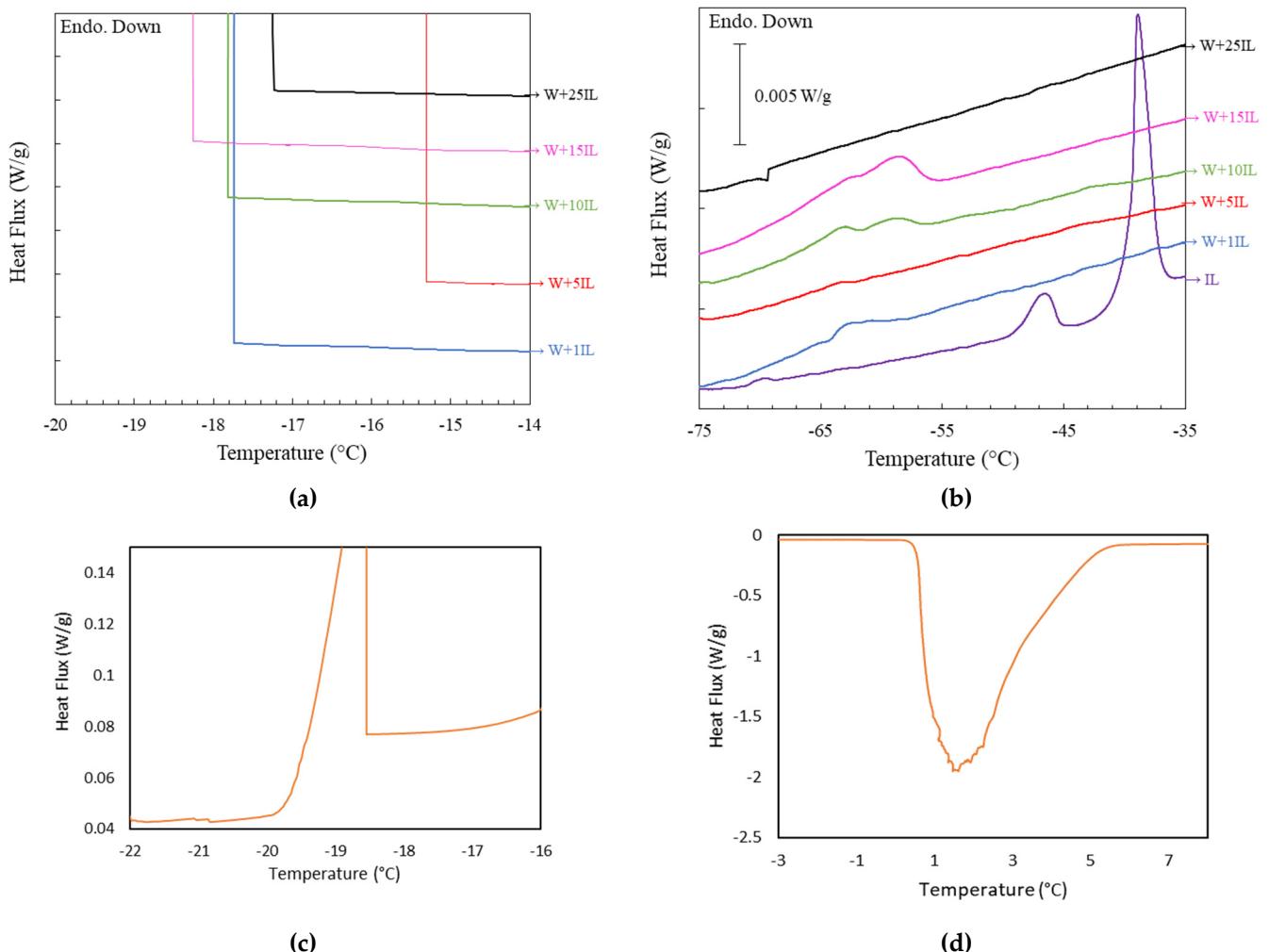
Samples	pH	T (°C)
H <sub>2</sub> O	6.23	27.4°C
1IL/H <sub>2</sub> O	7.44	28.5°C
5IL/H <sub>2</sub> O	7.45	28.6°C
10IL/H <sub>2</sub> O	7.49	29.1°C
15IL/H <sub>2</sub> O	7.51	29.1°C
25IL/H <sub>2</sub> O	7.51	29.1°C
IL	6.67	27.9°C

**Table S2.** Summary of the fitting of the IL/Water emulsions SAXS profiles to adjusted to the weakly correlated nanoscale mass fractal aggregates model with a radius of gyration  $R_g$  associated with the fractal size and the fractal dimension D that define their compactness and Ionogels and  $\xi$  being the correlation length. The 1IL/Water was not fitted due to the lack of structure and therof the fitting parameters are filled in the table as not applicable (NA).

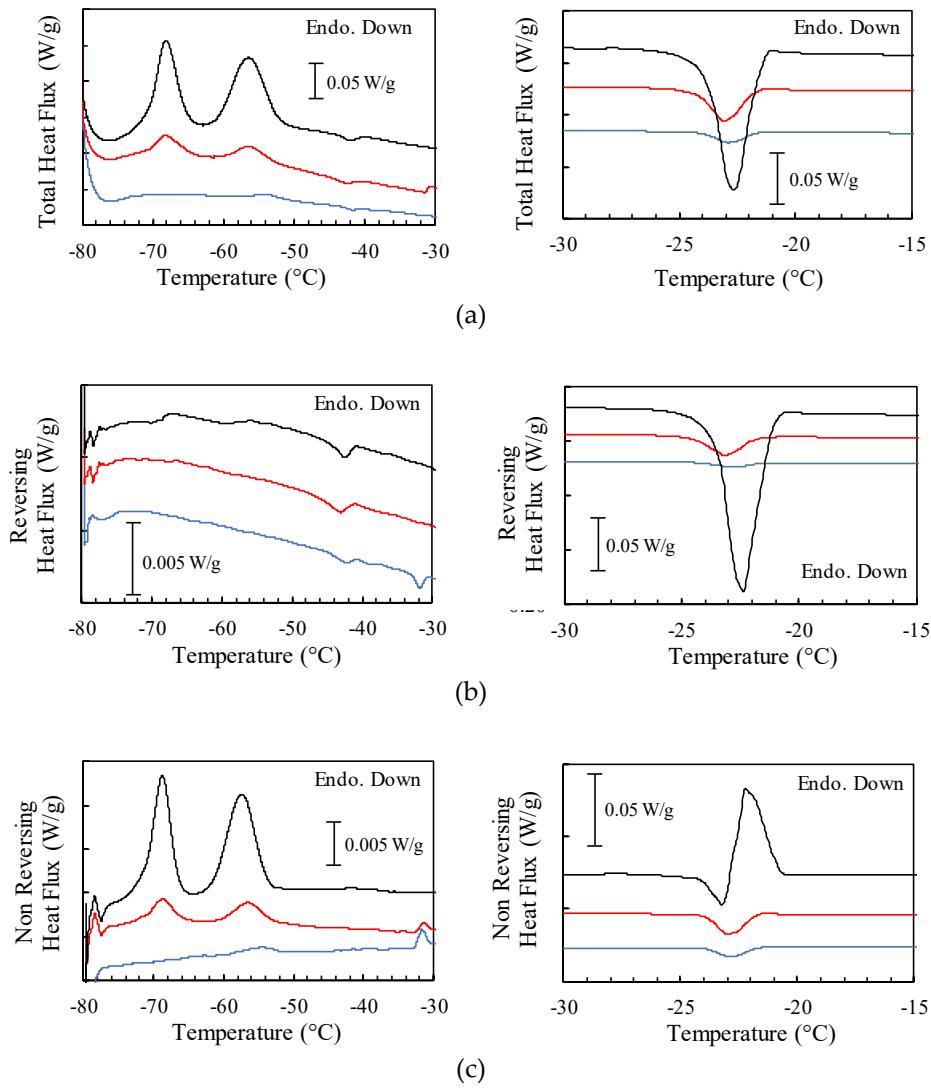
	<b>1IL/Water</b>	<b>5IL/Water</b>	<b>10IL/Water</b>	<b>15IL/Water</b>	<b>25IL/Water</b>
$R_g$	NA	1.73	1.72	1.68	1.55
$\xi$	NA	2.99	2.82	2.5	2.34
D <sub>f</sub>	NA	3.28	3.55	4	4



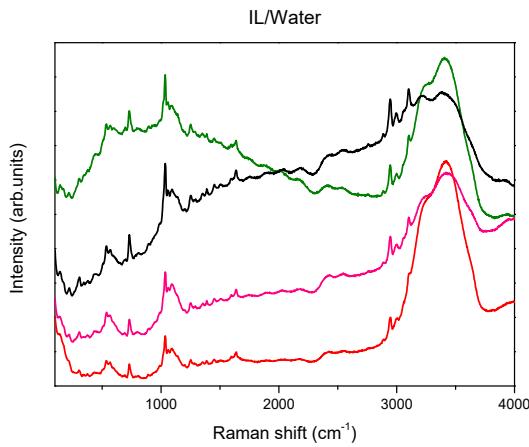
**Figure S1.** DSC thermogram of the cooling and heating cycles obtained for  $[\text{C}_2\text{C}_1\text{py}][\text{C}_4\text{F}_9\text{SO}_3]$  ionic liquid at 1 °C/min and 5 °C/min.



**Figure S2.** DSC thermogram (a, b) Cooling ramp of (—) Water, (—) 1IL/Water, (—) 5IL/Water, (—) 10IL/Water, (—) 15IL/Water, (—) 25IL/Water and (—) IL at 1 °C/min; (c) Cooling ramp of (—) Water at 1 °C/min; (d) Heating ramp of (—) Water at 1 °C/min.



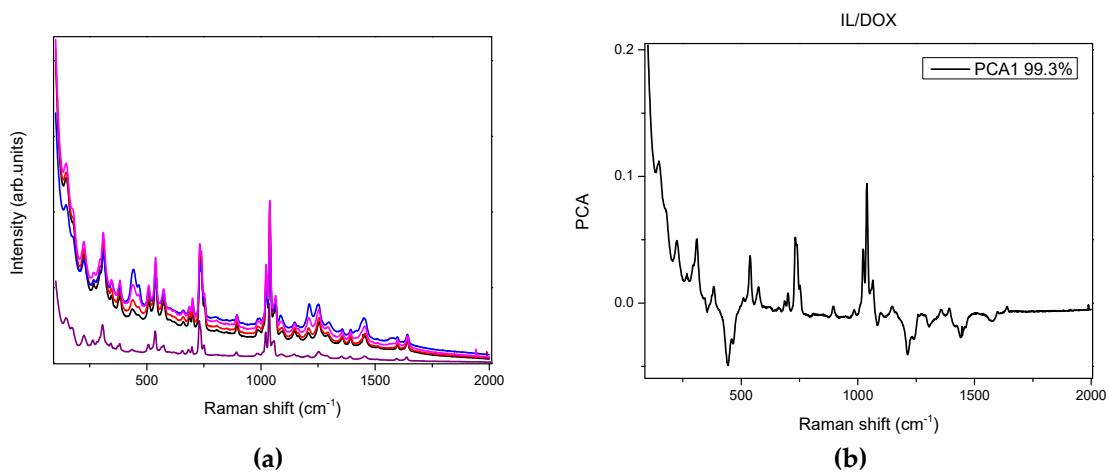
**Figure S3.** (a) Total; (b) reversing and; (c) non-reversing heat fluxes from heating MDSC runs at 1 °C/min for some representative IL/water emulsions: (—) 1IL/Water, (—) 5IL/Water, and (—) 25IL/Water.



**Figure S4.** Raman spectra of (—) 5IL/Water, (—) 10IL/Water, (—) 15IL/Water, (—) 25IL/Water.

**Table S3.** Summary of the fitting of the 25IL/Water emulsion loaded with DOX SAXS profiles to adjusted to the weakly correlated nanoscale mass fractal aggregates model with a radius of gyration  $R_g$  associated with the fractal size and the fractal dimension D that define their compactness and Ionogels and  $\xi$  being the correlation length.

	25IL/Water/ DOX_1mg/ml	25IL/Water/ DOX_2mg/ml	25IL/Water/ DOX_5mg/ml	25IL/Water/ DOX_10mg/ml
$R_g$	1.64	1.66	1.78	1.07
$\xi$	2.31	2.3	2.36	1.63
$D_f$	4.00	3.96	3.75	3.8

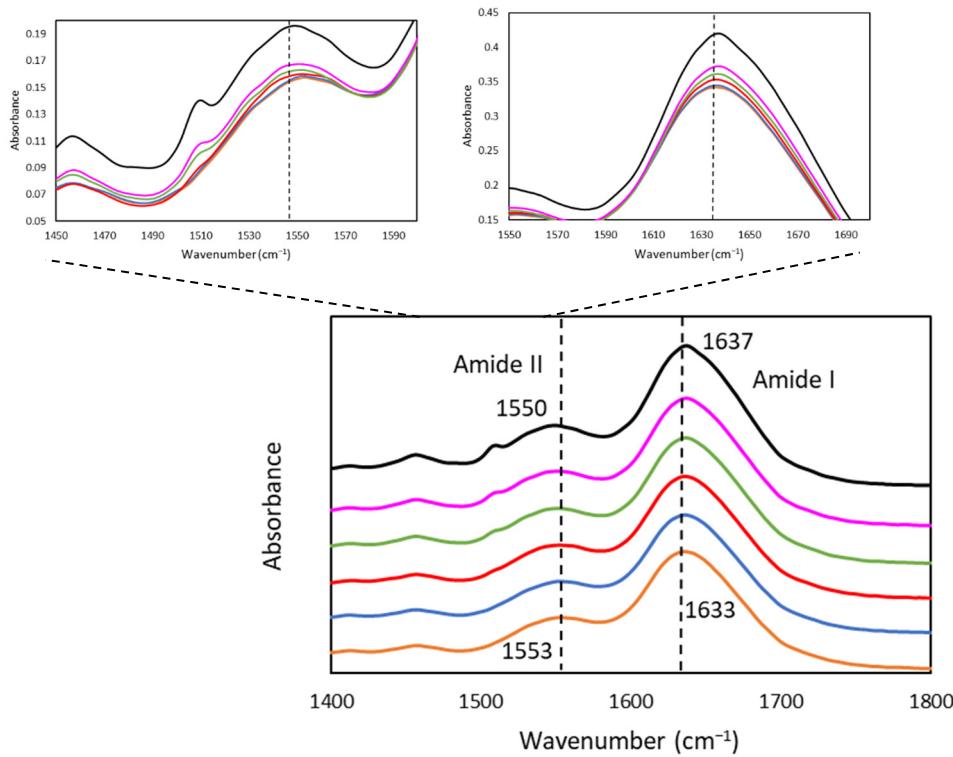


**Figure S5.** Raman spectra of: (a) 25IL/Water loaded with DOX; (b) and the corresponding PCA.

**Table S4.** Main FTIR vibration bands associated with IL and gelatin nanostructure.

Functional group	GE	GE/1IL	GE/5IL	GE/10IL	GE/15IL	GE/25I L	IL
	Wavenumber ( $\text{cm}^{-1}$ )						
N-H , O-H	3321	3334	3272	3353	3353, 3300	3353, 3300	---
Aliphatic tension CH	2850- 3000	2850-3000	2850-3000	2850- 3000	2850- 3000	2850- 3000	2951, 3000
Aromatic CH tension	---	---	---	---	---	---	3067
Aliphatic CH flexion	1337, 1458	1388, 1458	1384, 1457	1381, 1457	1338, 1457	1338, 1457	1390, 1459
Amide I,II,III	1633, 1553, 1250	1634, 1553, 1250	1635, 1552, 1250	1637, 1552, 1251	1637, 1551, 1251	1637, 1550, 1251	---
Tension C-O,C-O-C carbohydrates	1163, 1082	1163, 1082, 1064	1137,1082 , 1062	---	---	---	---
Water	600	600	600	600	600	600	---
C-F (CF <sub>2</sub> group)	---	---	1130-1290	---	1137- 1290	1137- 1290	1207-1253, 1131

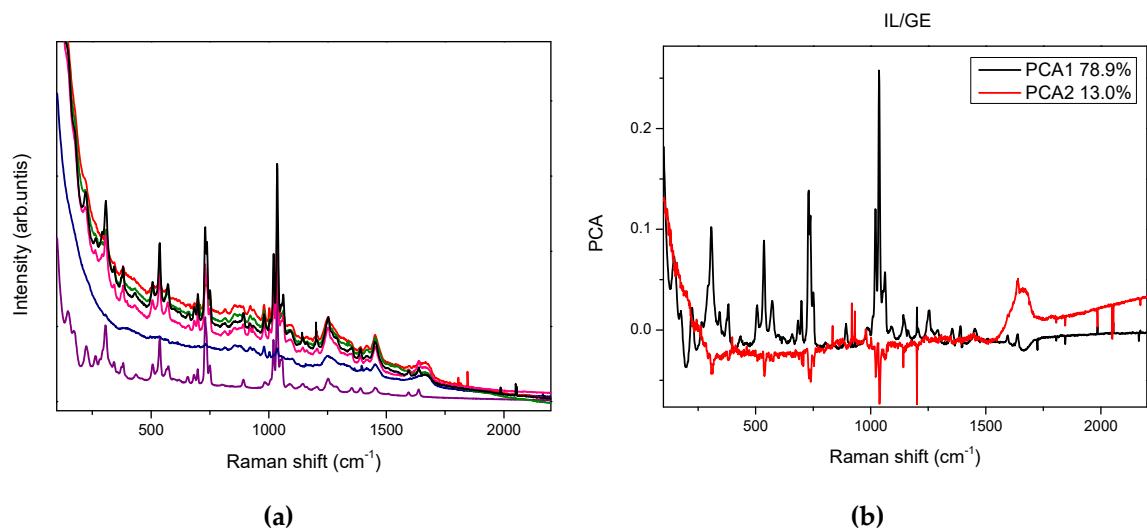
<b>C=C (pyridinium ring)</b>	---	---	---	---	---	---	1638-1600
<b>C=N (pyridinium ring)</b>	---	---	---	---	---	---	1508, 1459
<b>Pyridinium ring strain</b>	---	---	1000	1000	1007	1000	1006
<b>S-O stress mode (SO<sub>3</sub> group)</b>	---	---	1020, 1060	1000, 1061	1000, 1060	1000, 1060	1055, 1048, 1020
<b>S-O bending mode (SO<sub>3</sub> group)</b>	---	---	600-700	600-700	600-700	600-700	600-700
<b>S=O bond (SO<sub>3</sub> group)</b>	---	---	1350	1352	1352	1352	1352
<b>Flexion O=S=O</b>	---	---	---	---	525	525	522-530



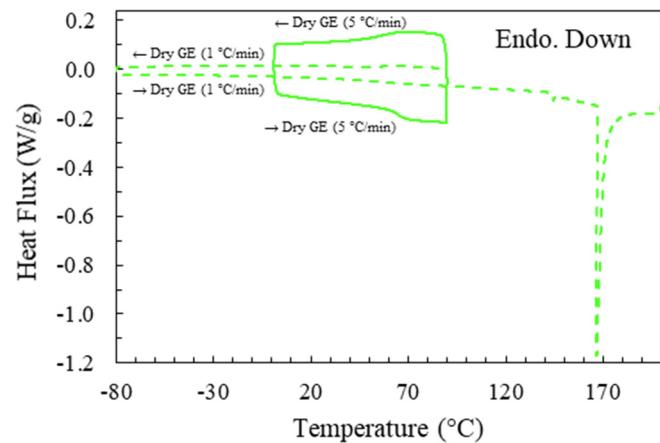
**Figure S6.** FTIR Spectra of Amide I and Amide II for: (—) GE, (—) GE/1IL, (—) GE/5IL, (—) GE/10IL, (—) GE/15IL and (—) GE/25IL.

**Table S5.** FTIR bands of Amide I and Amide II for: GE, GE/1IL, GE/5IL, GE/10IL, GE/15IL and GE/25IL.

Samples	Amide I (cm <sup>-1</sup> )	Amide II (cm <sup>-1</sup> )
GE	1633	1553
GE/1IL	1634	1553
GE/5IL	1635	1552
GE/10IL	1637	1552
GE/15IL	1637	1551
GE/25IL	1637	1550



**Figure S7.** (a) Raman spectra of the Ionogels and; (b) their corresponding PC1 and PC2.



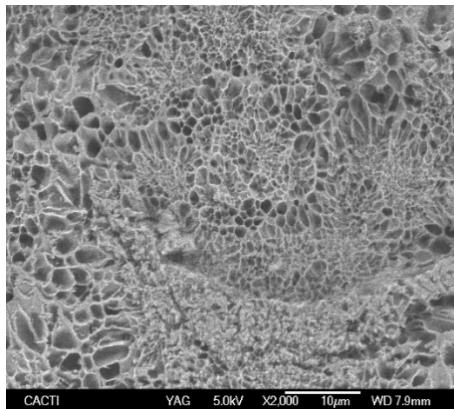
**Figure S8.** DSC cooling and heating thermograms obtained for dry shark gelatin.

**Table S6.** Enthalpies and Temperature of crystallization of water in the different hydrogel and Ionogels.

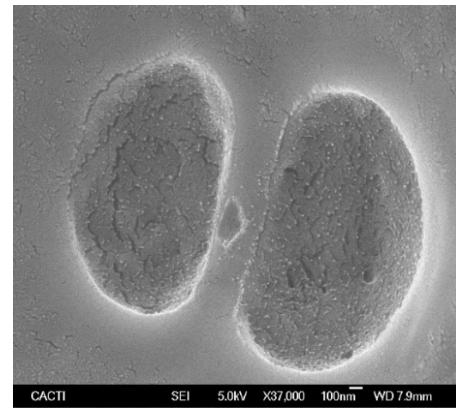
	<b>GE</b>	<b>IL</b>	<b>W</b>	<b>T (°C)</b>	<b>Δh (J/g)</b>
GE	25	0	75	-0.29	178.3
GE+1IL	25	1	74	-1.25	152.6
GE+5IL	25	5	70	-1.09	149.3
GE+10IL	25	10	65	-1.10	164.4
GE+15IL	25	15	60	-1.43	135.6
GE+25IL	25	25	50	-2.04	110.2
Water	0	0	100	1.47	308.7

**Table S7.** Summary of the fitting of the SAXS profiles to a heterogeneous sphere with fractal structure with a radius of gyration  $R_g$  associated with the fractal size and the fractal dimension  $D$  that define their compactness and Ionogels and  $\xi$  being the correlation length.

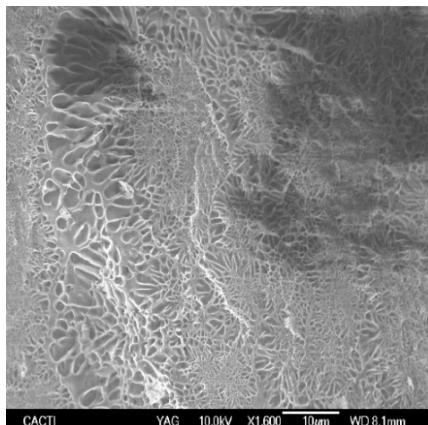
	<b>Ge/1IL</b>	<b>Ge/1IL</b>	<b>Ge/5IL</b>	<b>Ge/5IL</b>	<b>Ge/10IL</b>	<b>Ge/10IL</b>	<b>Ge/15IL</b>	<b>Ge/15IL</b>	<b>Ge/25IL</b>	<b>Ge/25IL</b>
	<b>10°C</b>	<b>25°C</b>	<b>10°C</b>	<b>25°C</b>	<b>10°C</b>	<b>25°C</b>	<b>10°C</b>	<b>25°C</b>	<b>0°C</b>	<b>25°C</b>
$R_g$	0.89	0.92	1.00	1.06	1.00	1.08	1.00	1.06	1.01	0.98
$\xi$	1.22	2.13	1.68	2.46	1.99	1.92	2.13	1.62	1.28	1.14
$D_f$	3.27	2.63	3.0	2.6	2.92	2.93	2.53	2.9	3.84	3.22



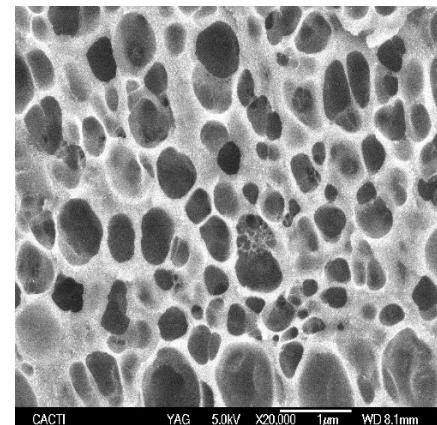
(a)



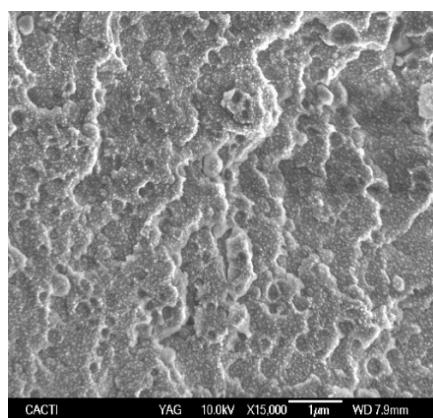
(b)



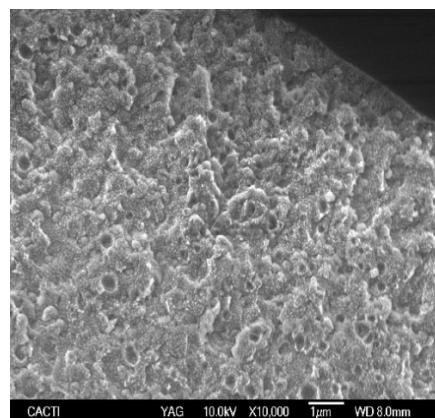
(c)



(d)

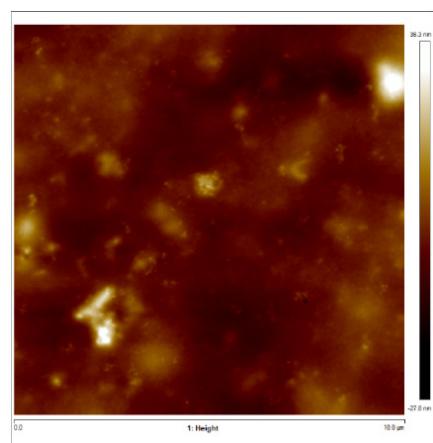


(e)

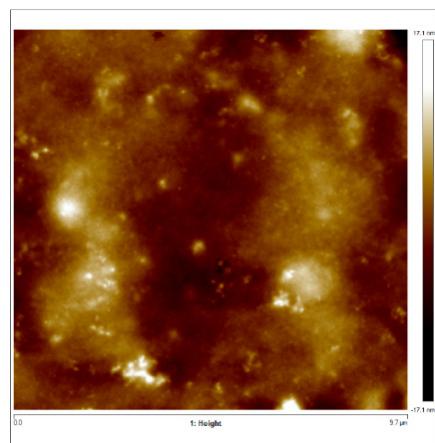


(f)

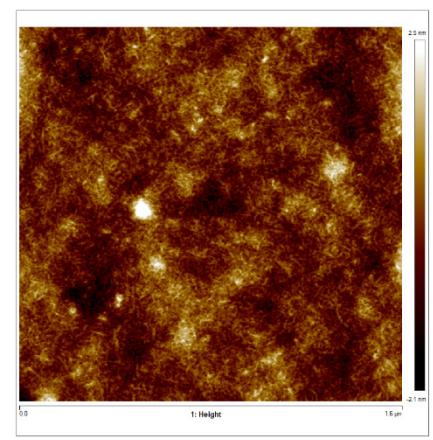
**Figure S9.** Cryo-scanning electron microscopy of: (a, b) Shark gelatin hydrogel, (a) X2000; (b) X37000; (c) GE/5IL at X1600; (d) GE/15IL at X20000; (e) and (f) GE/25IL at X15000.



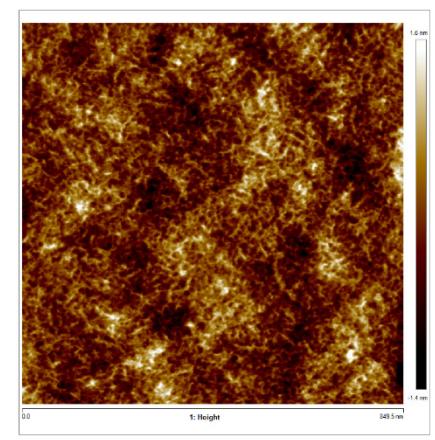
(a)



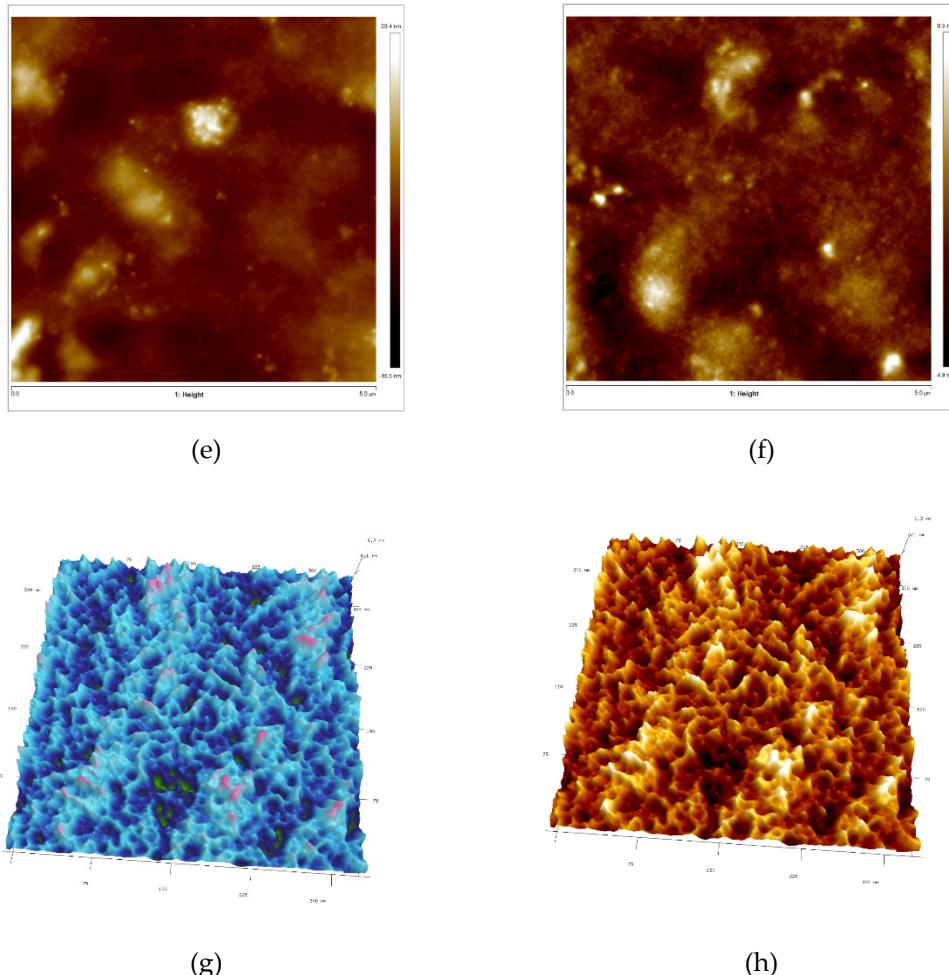
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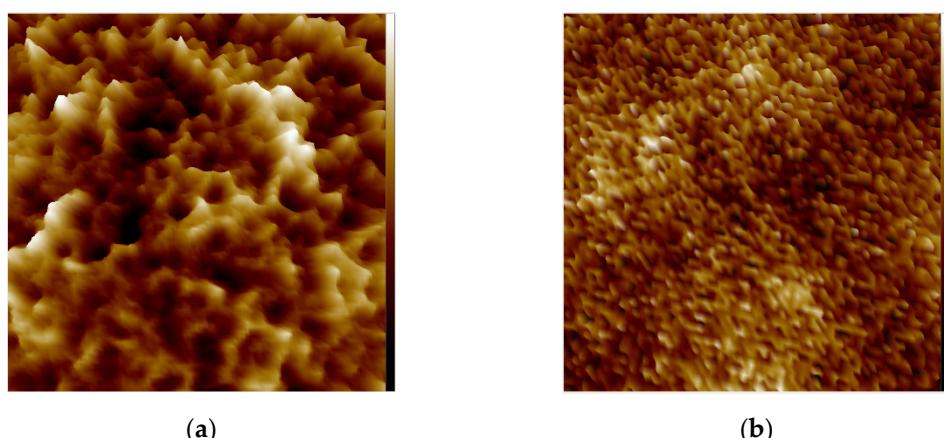
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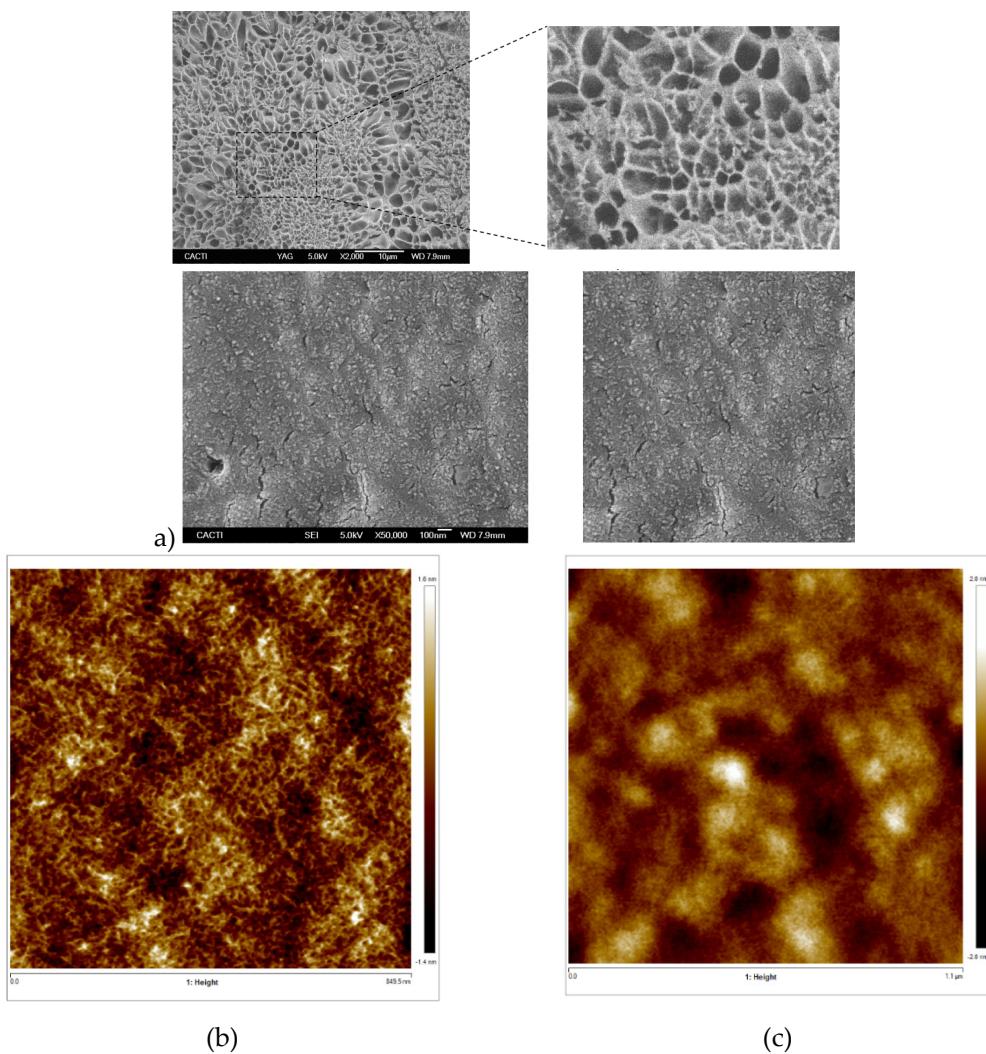
(d)



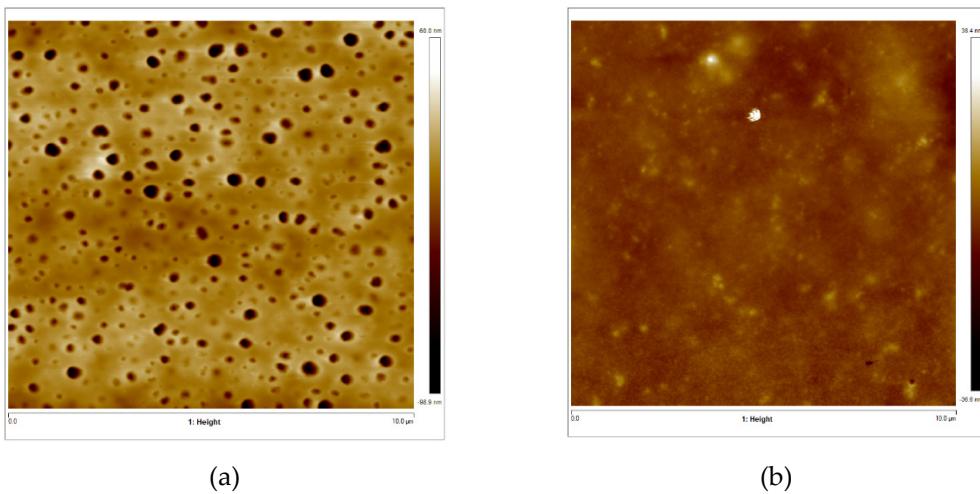
**Figure S10.** FM images of shark gelatin hydrogel. (a) Zone 1 at a magnification of  $10 \times 10 \mu\text{m}^2$ , (b) Zone 2 at a magnification of  $10 \times 10 \mu\text{m}^2$ . (c) Image 2D with a field of view  $1.6 \times 1.6 \mu\text{m}^2$ , (d) Image 2D with a field of view  $849.5 \times 849.5 \mu\text{m}^2$ , (e) Image 2D in Zone 1 at a magnification of  $5 \times 5 \mu\text{m}^2$ , (f) Zone 2 at a magnification of  $5 \times 5 \mu\text{m}^2$ ; (g), (h) Zone 1, Image 3D.

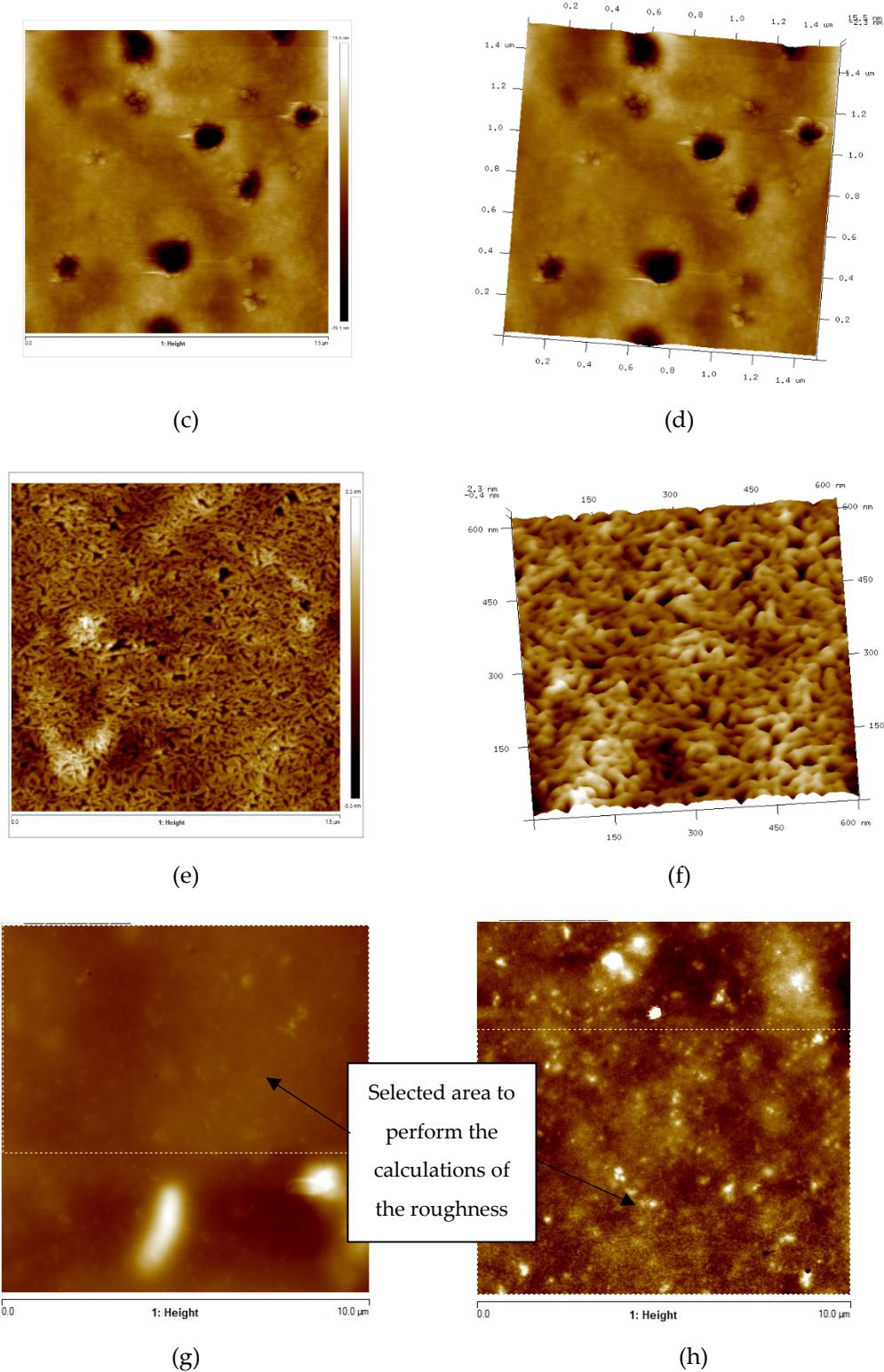


**Figure S11.** Zoom images 3D with magnifications  $120 \times 120 \text{ nm}^2$  of shark gelatin hydrogel. (a) Zone 1; (b) Zone 2.

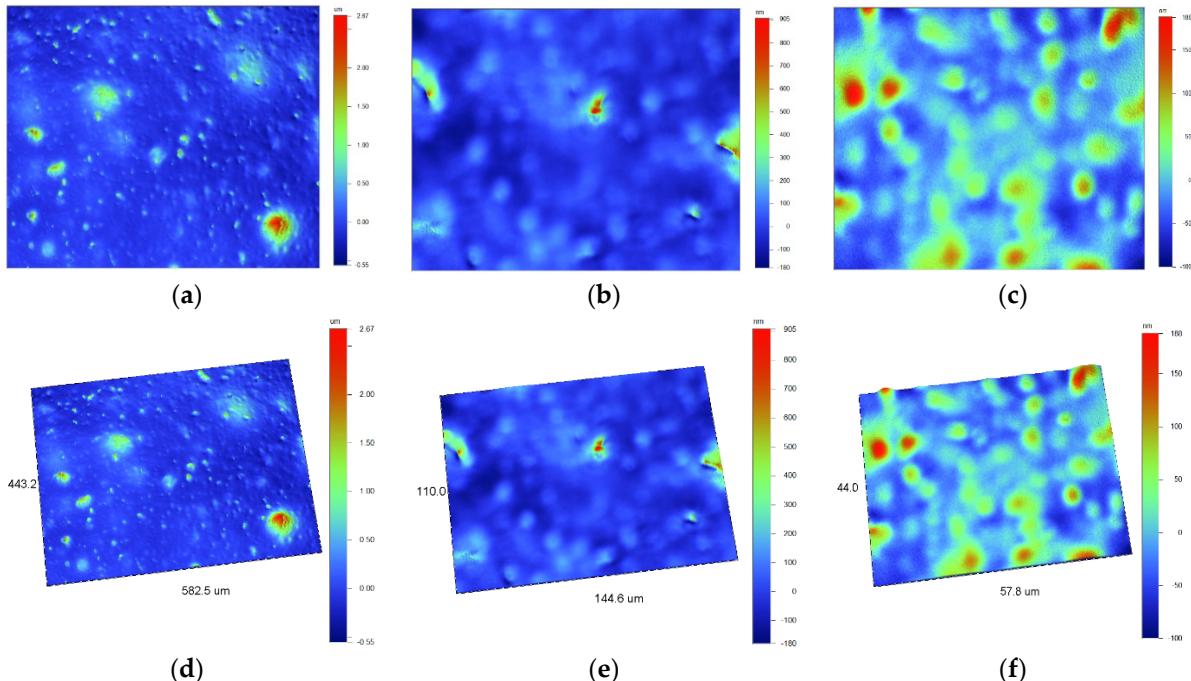


**Figure S12.** (a) Cryo-SEM images; (b) Image of Zone 1 at a magnification of  $0.84 \times 0.84 \mu\text{m}^2$ ; (c) Image of Zone 2 at a magnification of  $1.1 \times 1.1 \mu\text{m}^2$  of hydrogel gelatin.





**Figure S13.** (a) Zone 2 of GE/25IL; (b) Zone 2 of GE/5IL Topographic 2D images Zone 2,  $10 \times 10 \mu\text{m}^2$ ; (c) Zone 1, Image 2D Field of View  $1.5 \times 1.5 \mu\text{m}^2$  of GE/25IL; (d) Zone 1, Image 3D Field of View  $1.5 \times 1.5 \mu\text{m}^2$  of GE/25IL; (e) Zone 1, Image 2D, Field of View  $1.5 \times 1.5 \mu\text{m}^2$  of GE/5IL; (f) Zone 1, Image 3D Field of View  $1.5 \times 1.5 \mu\text{m}^2$  of GE/5IL; (g) GE/5IL of Zone 1; (h) GE/5IL of Zone 2.



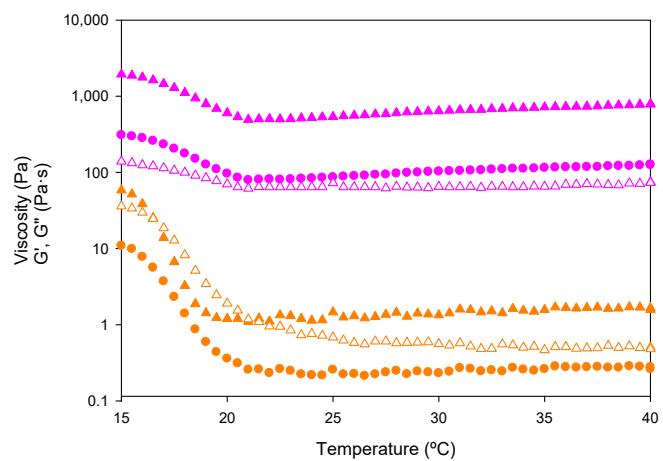
**Figure S14.** Imaging of WLOP method. (a) 2D image at 5x\_2x (443.2X582.5  $\mu\text{m}^2$ ); (b) 20x\_2x (110X144  $\mu\text{m}^2$ ); (c) 250x\_2x (44X58  $\mu\text{m}^2$ ); (d) 3D magnification 5X\_FOV 2X (443.2X582.5  $\mu\text{m}^2$ ); (e) 3D magnification 20X\_2X (110X144 $\mu\text{m}^2$ ), (f) 3D magnification 50X\_2X (44X58 $\mu\text{m}^2$ ).

The development of the calculations of the three-dimensional roughness parameters ( $S_a$ ,  $S_q$ ,  $S_z$ ,  $S_{sk}$  and  $S_{ku}$ ), have been developed in previous studies [1].

**Table S8.** Roughness parameter values obtained by WLOP of shark gelatin hydrogel.

INTERFEROMETRIC OPTICAL PROFILOMETRY / 3D Amplitude Roughness Parameters				
	$S_a$ (nm)	$S_q$ (nm)	$S_z$ (nm)	$S_{sk}$
<b>5X_2X</b>	148.07	241.84	2160.00	3.53
<b>20X_2X</b>	54.31	80.55	780.03	2.53
<b>50X_2X</b>	27.73	37.31	251.27	1.14

The gelation point of the Ionogels was determined by monitoring the loss and storage moduli, as well as the complex viscosity (Figure S15). A transition in the complex viscosity occurs in the thermal range between 15 °C and 22 °C as well as in the storage and loss moduli during the heating ramp, which is indicative of the interruption of the elastic network of the hydrogel and of in agreement with the sol/gel transition observed for the gelatin hydrogel and which is slightly lower for the Ionogels.



**Figure S15.** Complex viscosity variation (filled  $\circ$ ); and storage modulus ( $G'$ , filled  $\Delta$ ) and loss modulus ( $G''$ , open  $\Delta$ ) versus temperature of gelatin hydrogel (orange color) and Ionogel (fuchsia color).