

Supplementary Information

Effect of Bacterial Amyloid Protein Phenol–Soluble Modulin Alpha 3 on the Aggregation of Amyloid Beta Protein Associated with Alzheimer’s Disease

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Table S1. Maximum ThT fluorescence intensity (y_{\max}) and lag phase time (T_{lag}) of aggregation kinetics of A β_{40} (25 μM) with different concentrations of PSM $\alpha 3$ monomers. The standard errors of the mean were calculated from three repeats.

A β_{40} : PSM $\alpha 3$	y_{\max} (a.u.)	T_{lag} (h)
1 : 0	13162.2 \pm 470.6	52.8 \pm 1.0
1 : 0.05	9863.1 \pm 182.5	56.6 \pm 0.9
1 : 0.1	8071.9 \pm 516.8	62.3 \pm 2.2
1 : 0.2	4171.4 \pm 87.1	64.7 \pm 1.3
1 : 0.5	1499.3 \pm 210.9	77.1 \pm 6.3
1 : 1	1027.2 \pm 44.2	92.1 \pm 1.2

Table S2. Maximum ThT fluorescence intensity (y_{\max}) and lag phase time (T_{lag}) of aggregation kinetics of A β_{40} (25 μM) incubated without or with PSM $\alpha 3$ seeds. The standard errors of the mean were calculated from three repeats.

Sample	y_{\max} (a.u.)	T_{lag} (h)
A β_{40}	13162.2 ± 470.6	52.8 ± 1.0
with PSM $\alpha 3$ seeds at 1 h	12700.0 ± 585.0	39.1 ± 2.3
with PSM $\alpha 3$ seeds at 2.5 h	14410.4 ± 644.1	33.8 ± 2.6
with PSM $\alpha 3$ seeds at 5 h	14311.0 ± 513.0	44.4 ± 1.8
with PSM $\alpha 3$ seeds at 24 h	13958.4 ± 459.0	47.6 ± 1.4

Table S3. Top 10 binding energies of the PSM α 3m–A β ₄₀ complex calculated by molecular docking using Autodock Vina [1].

Mode	Binding energy (kcal/mol)
1	–6.2
2	–5.9
3	–5.3
4	–5.2
5	–5.1
6	–5.1
7	–5.0
8	–4.9
9	–4.6
10	–4.5

Table S4. Top 10 binding energies of PSM α 3o and A β ₄₀ calculated by molecular docking using Autodock Vina [1].

Mode	Binding energy (kcal/mol)
1	−13.0
2	−12.7
3	−12.7
4	−12.1
5	−11.6
6	−11.5
7	−10.9
8	−10.7
9	−10.7
10	−10.5

Table S5. Structural parameters of the A β ₄₀ monomer in different systems in the equilibrium state. The standard errors of the mean were calculated from three repeats.

System	R_g^a (nm)	SASA^b (nm²)	H-bond^c
A β ₄₀ only	1.1 \pm 0.03	38.5 \pm 1.1	22
A β ₄₀ –PSM α 3m	1.3 \pm 0.03	40.3 \pm 1.7	20
A β ₄₀ –PSM α 3o	1.3 \pm 0.04	42.1 \pm 1.0	19

^a Radius of gyration.

^b Solvent accessible surface area.

^c Number of intermolecular hydrogen bonds.

Table S6. Secondary structure composition of the A β ₄₀ monomer in different systems in the equilibrium state.

System (%)	Coil	β-bridge	Bend	Turn	α-helix
A β ₄₀ only	43.9	4.5	13.2	24.6	13.8
A β ₄₀ -PSM α 3m	29.1	0.0	10.5	24.1	36.3
A β ₄₀ -PSM α 3o	32.5	0.5	11.2	23.3	32.5

Table S7. Binding free energy components of the A β ₄₀–PSM α 3m and A β ₄₀–PSM α 3o systems calculated using the MM–PBSA method [2].

System (kJ/mol)	$\Delta G_{\text{vdw}}^{\text{a}}$	$\Delta G_{\text{elec}}^{\text{b}}$	$\Delta G_{\text{ps}}^{\text{c}}$	$\Delta G_{\text{nps}}^{\text{d}}$	$\Delta G_{\text{bind}}^{\text{e}}$
A β ₄₀ – PSM α 3m	–373.5	–963.9	1106.5	–39.8	–270.7
A β ₄₀ –PSM α 3o	–251.0	–1274.7	1418.9	–28.6	–135.5

^a ΔG_{vdw} represents the van der Waals hydrophobic energy.

^b ΔG_{elec} represents electrostatic energy under vacuum.

^c ΔG_{ps} represents polar solvation energy.

^d ΔG_{nps} represents nonpolar solvation energy.

^e ΔG_{bind} represents total binding free energy.

Table S8. Energy components of key residues of A β ₄₀–PSM α 3m system.

System (kJ/mol)	Residues	$\Delta G_{\text{MM}}^{\text{a}}$	$\Delta G_{\text{ps}}^{\text{b}}$	$\Delta G_{\text{nps}}^{\text{c}}$	$\Delta G_{\text{TOT}}^{\text{d}}$
A β ₄₀	ASP-1	−136.7	118.9	−1.9	−19.7
	LEU-17	−10.8	−3.1	−2.2	−16.1
PSM α 3m	MET-1	−269.0	241.3	−2.2	−29.9
	GLU-2	107.7	−128.9	−0.5	−21.7
	PHE-3	−35.8	14.6	−2.3	−23.5
	LEU-7	−24.9	5.1	−2.6	−22.5
	LEU-14	−30.5	9.5	−2.5	−23.5
	PHE-18	−23.6	7.3	−1.7	−18.0

^a $\Delta G_{\text{MM}} = \Delta G_{\text{elec}} + \Delta G_{\text{vdw}}$, vacuum energy.

^b ΔG_{ps} represents polar solvation energy.

^c ΔG_{nps} represents nonpolar solvation energy.

^d ΔG_{TOT} represents total contribution of residues to binding free energy.

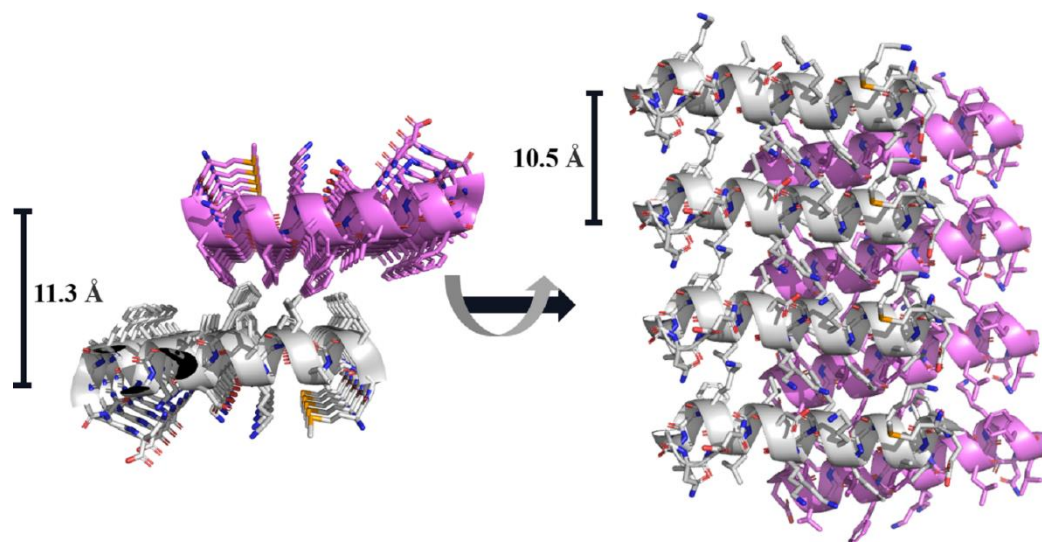


Figure S1. Crystal structure of PSMα3 (PBD ID: 5I55) [3].

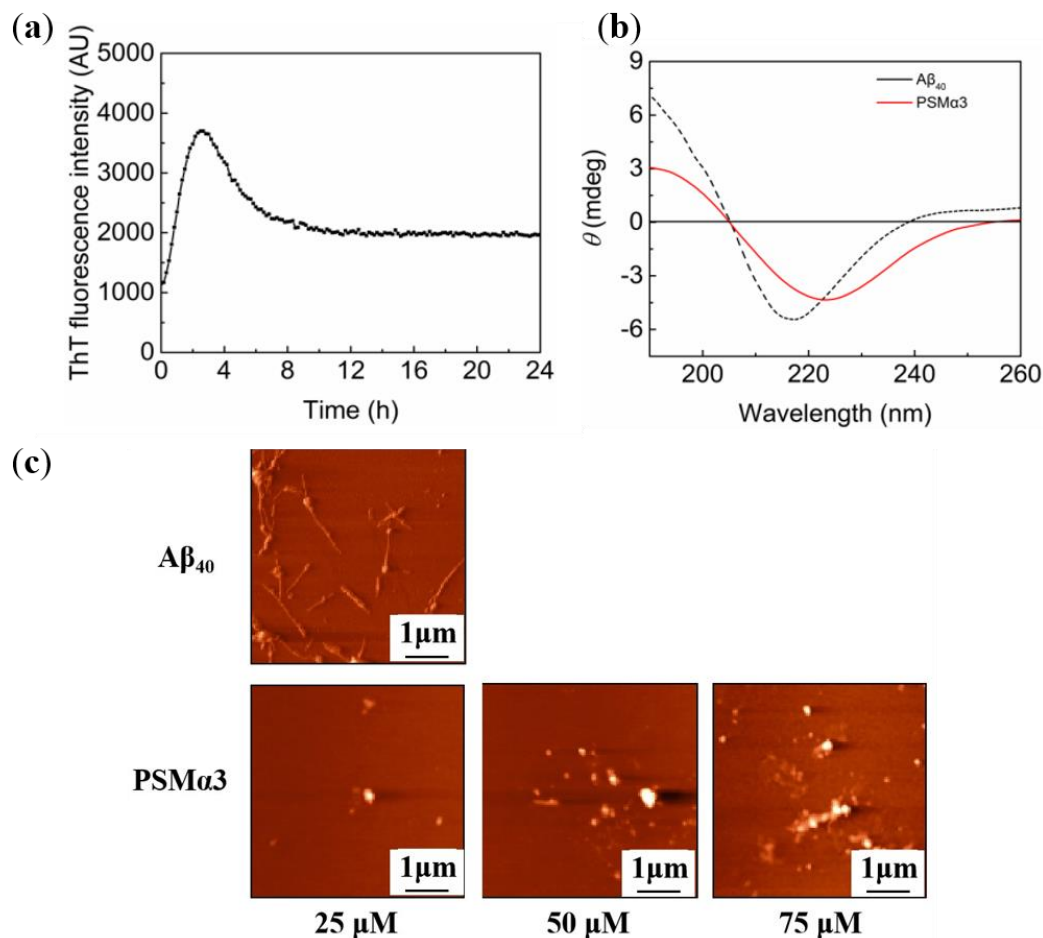


Figure S2. Aggregation kinetics of PSM $\alpha 3$: (a) Aggregation kinetics of PSM $\alpha 3$ (25 μ M) by ThT fluorescence assays; (b) Far-UV circular dichroism spectra of $A\beta_{40}$ (25 μ M) and PSM $\alpha 3$ (25 μ M) aggregates at 72 h; (c) AFM images of $A\beta_{40}$ (25 μ M) and PSM $\alpha 3$ (25 μ M, 50 μ M, 75 μ M) aggregates obtained by incubation at 37 $^{\circ}$ C for 72 h.

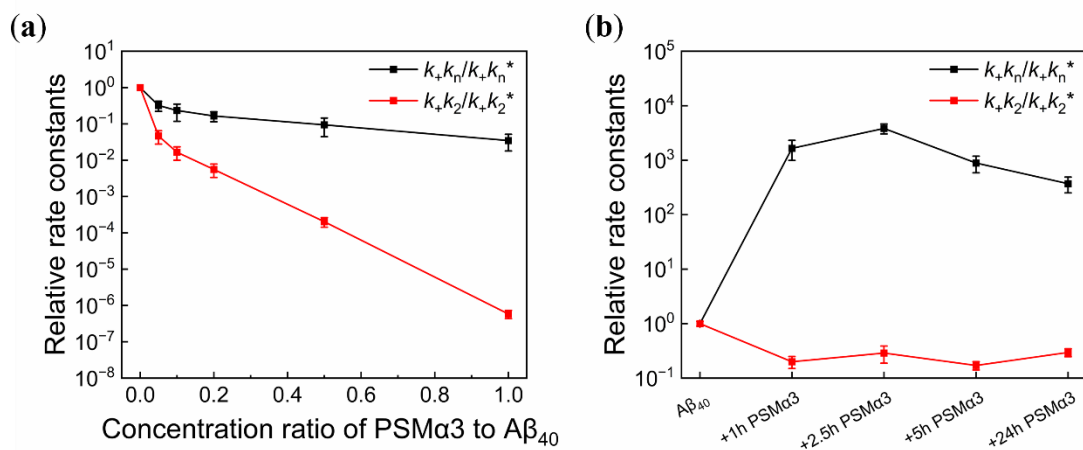


Figure S3. Relative kinetic rate constants of Aβ₄₀ aggregation in the presence of (a) PSMα3 monomers or (b) PSMα3 seeds. The kinetic curve was fitted using the multistep secondary nucleation dominated, unseeded model with both k_nk_+ and k_2k_+ as free-fitting parameters [4]. An asterisk (*) denotes rate constants from Aβ₄₀ aggregation in the absence of PSMα3. Error bars are standard deviations (s.d.) calculated from three repeats.

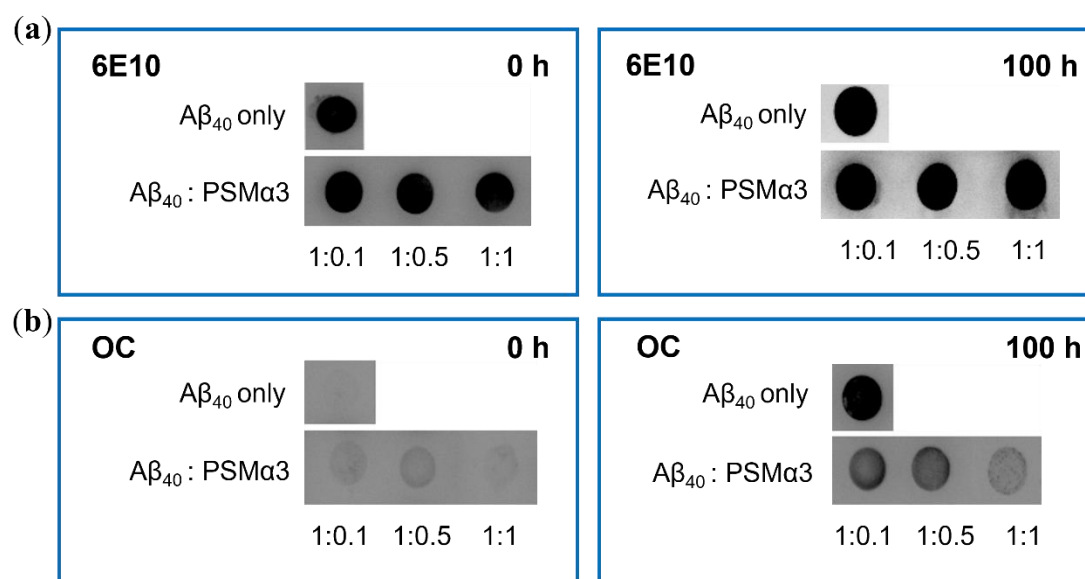


Figure S4. Effects of $PSM\alpha 3$ monomers at different concentrations on the formation of $A\beta_{40}$ (25 μM) fibrils at 0 h and 100 h. (a) $A\beta_{1-16}$ sequence-specific 6E10 antibody assay for all forms of $A\beta$ species; (b) $A\beta$ fibril-specific OC antibody assay.

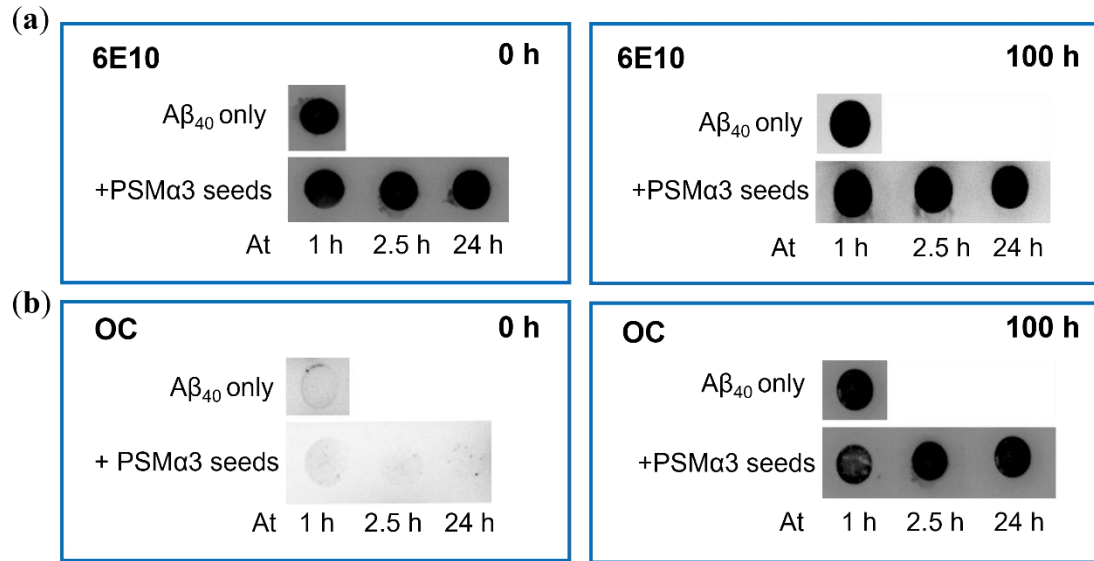


Figure S5. Effects of PSM α 3 seeds (5 μ M) at 1 h, 2.5 h, and 24 h on the formation of A β_{40} (25 μ M) fibrils at 0 h and 100 h. (a) A β_{1-16} sequence-specific 6E10 antibody assay for all forms of A β species; (b) A β fibril-specific OC antibody assay.

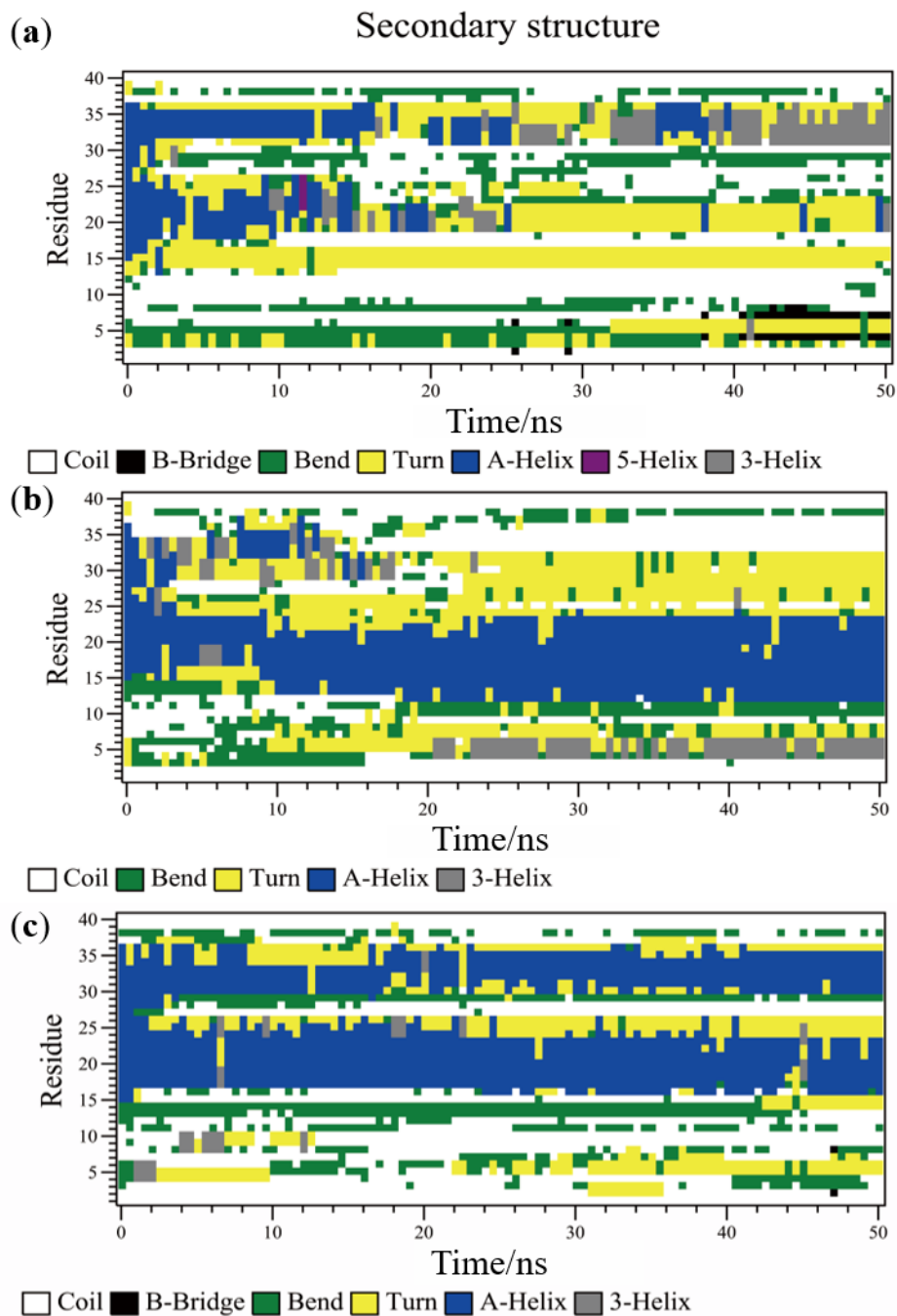


Figure S6. The secondary structure components of A β ₄₀ during the 50 ns MD simulations in (a) A β ₄₀ only, (b) A β ₄₀-PSM α 3m, and (c) A β ₄₀-PSM α 3o systems.

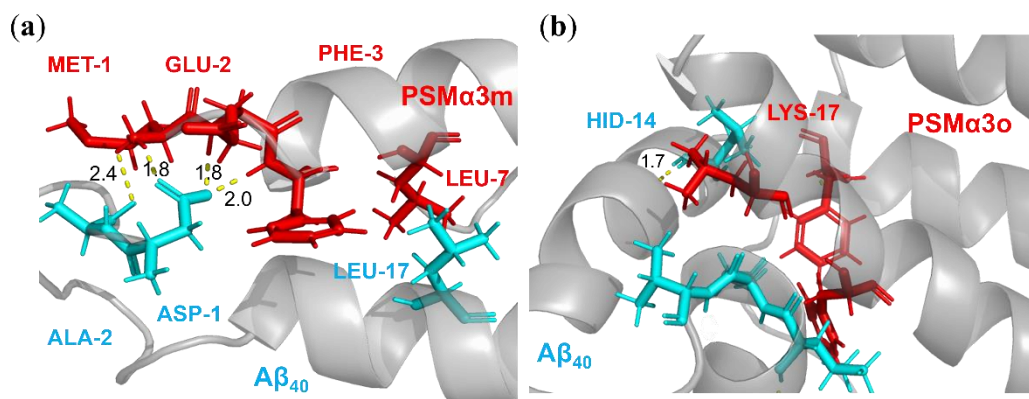


Figure S7. H-bonds between the A β ₄₀ monomer and (a) PSM α 3m or (b) PSM α 3o. Blue represents residues of A β ₄₀, red represents residues of PSM α 3m and PSM α 3o.

References

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