

Supplementary Information

Facile Synthesis of Fmoc-N-Me-AA-OH using CTC resin as Temporary Protecting Group

Tanya Román 1,2,3, Gerardo Acosta 3,4, Constanza Cárdenas 1, Beatriz G. de la Torre 5, Fanny Guzmán 1,* and Fernando Albericio 3,4,6,*

1. Núcleo Biotecnología Curauma, Pontificia Universidad Católica de Valparaíso, Valparaíso 2373223, Chile; tanya.roman.21@gmail.com (T.R.); constanza.cardenas@pucv.cl (C.C.)
2. Doctorado en Biotecnología, Pontificia Universidad Católica de Valparaíso, Universidad Técnica Federico Santa María, Valparaíso 2373223, Chile
3. Department of Organic Chemistry and CIBER-BBN, Networking Centre on Bioengineering, Biomaterials and Nanomedicine, University of Barcelona, 08028 Barcelona, Spain; gerardoacosta@ub.edu
4. Institute for Advanced Chemistry of Catalonia (IQAC-CSIC), Jordi Girona 18-26, 08034 Barcelona, Spain
5. KwaZulu-Natal Research Innovation and Sequencing Platform (KRISP), School of Laboratory Medicine and Medical Sciences, College of Health Sciences, University of KwaZulu-Natal, Durban 4041, South Africa; garciadelatorreb@ukzn.ac.za; (B.G.d.l.T)
6. School of Chemistry and Physics, University of KwaZulu-Natal, Durban 4001, South Africa

*Correspondence: fanny.guzman@pucv.cl (F.G.); albericio@ub.edu (F.A.)

Table S1: ESI-MS analysis of Fmoc-N-Me- β Ala-OH and Fmoc-N-Me-Thr(tBu)-OH with first activation of CTC resin using dimethyl sulfate and iodomethane, both as N-methylation reagents. A) Analysis of mini-cleavage after *o*-NBS deprotection. B) Analysis of mini-cleavage after Fmoc protection.

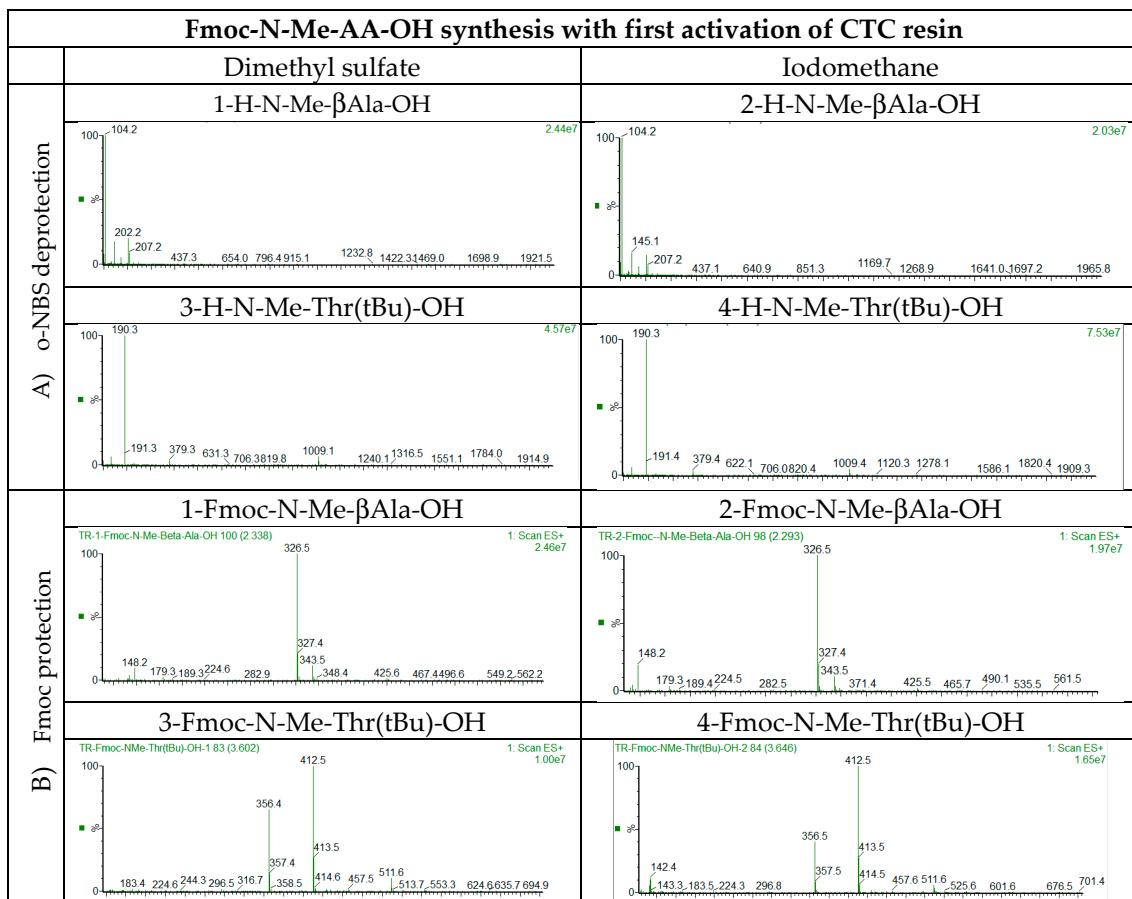
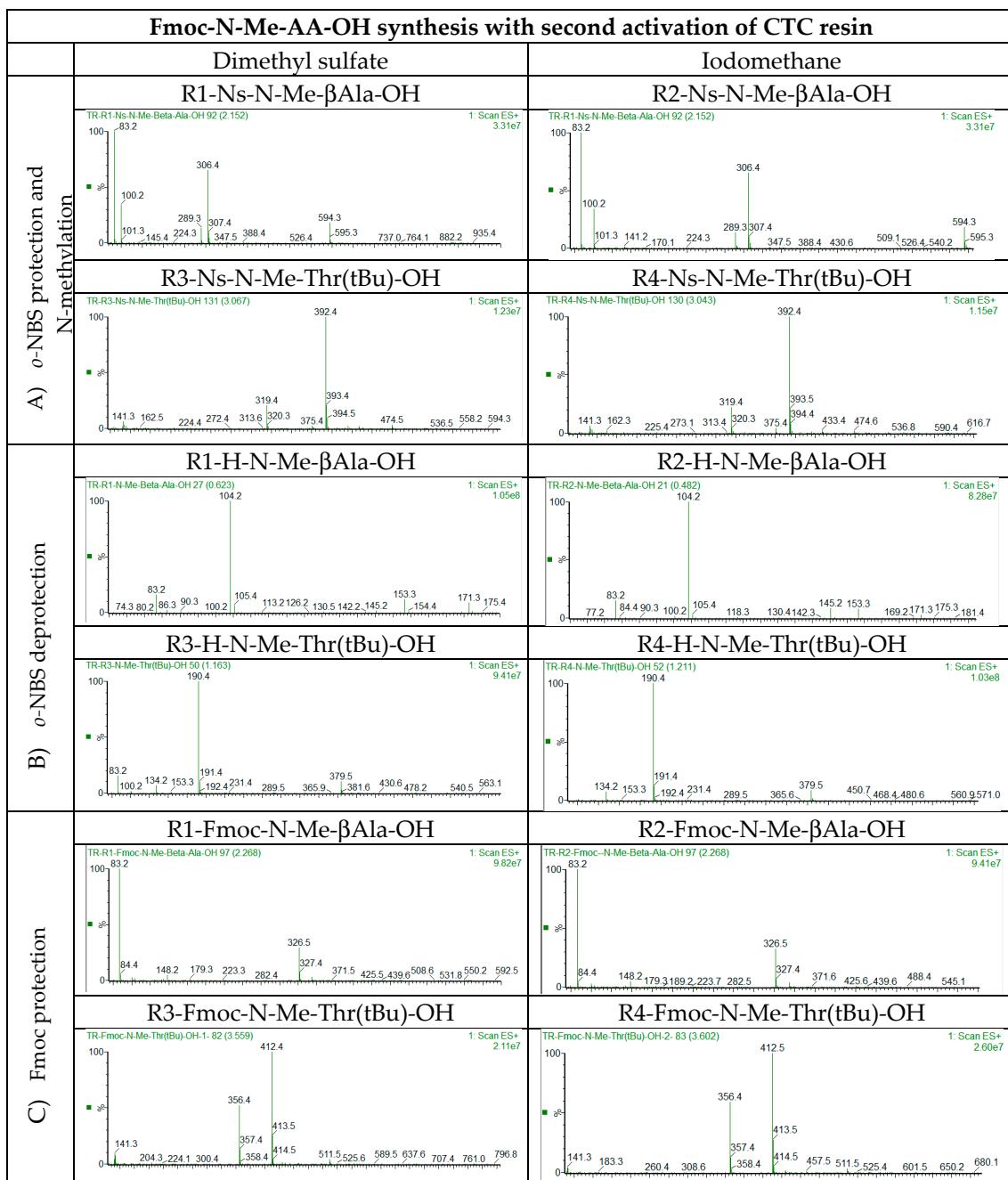


Table S2: ESI-MS analysis of Fmoc-N-Me- β Ala-OH and Fmoc-N-Me-Thr(tBu)-OH with second activation of CTC resin using dimethyl sulfate and iodomethane, both as N-methylation reagents. A) Analysis of mini-cleavage after *o*-NBS protection and N-methylation. B) Analysis of mini-cleavage after *o*-NBS deprotection. C) Analysis of mini-cleavage after Fmoc protection.



R1-Fmoc-N-Me- β Ala-OH

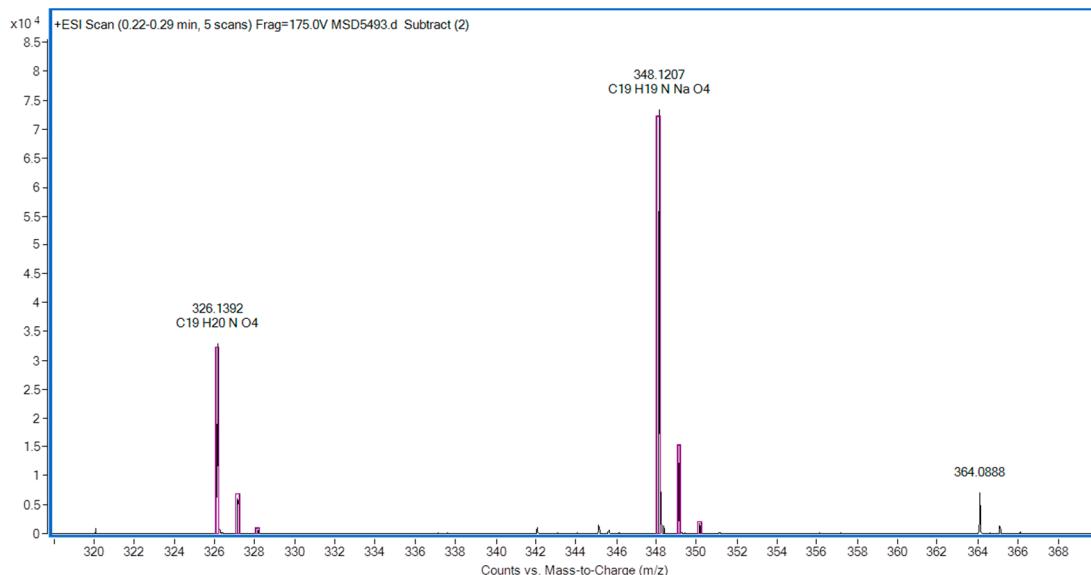


Figure S1: ESI-TOF analysis of Fmoc-N-Me- β Ala-OH using dimethyl sulfate as N-methylation reagent.

R2-Fmoc-N-Me- β Ala-OH

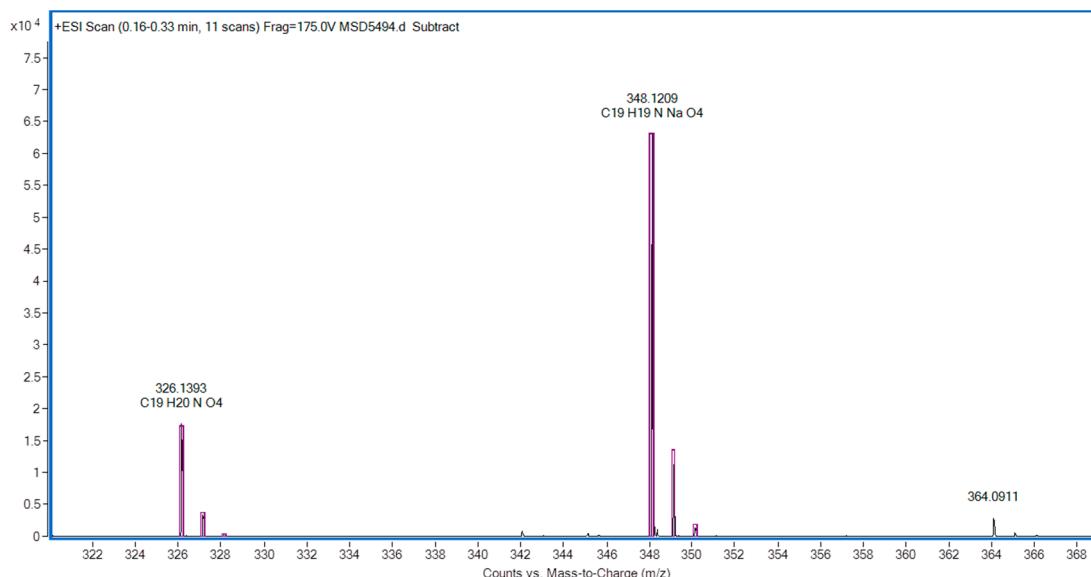


Figure S2: ESI-TOF analysis of Fmoc-N-Me- β Ala-OH using iodomethane as N-methylation reagent.

R3-Fmoc-N-Me-Thr(tBu)-OH

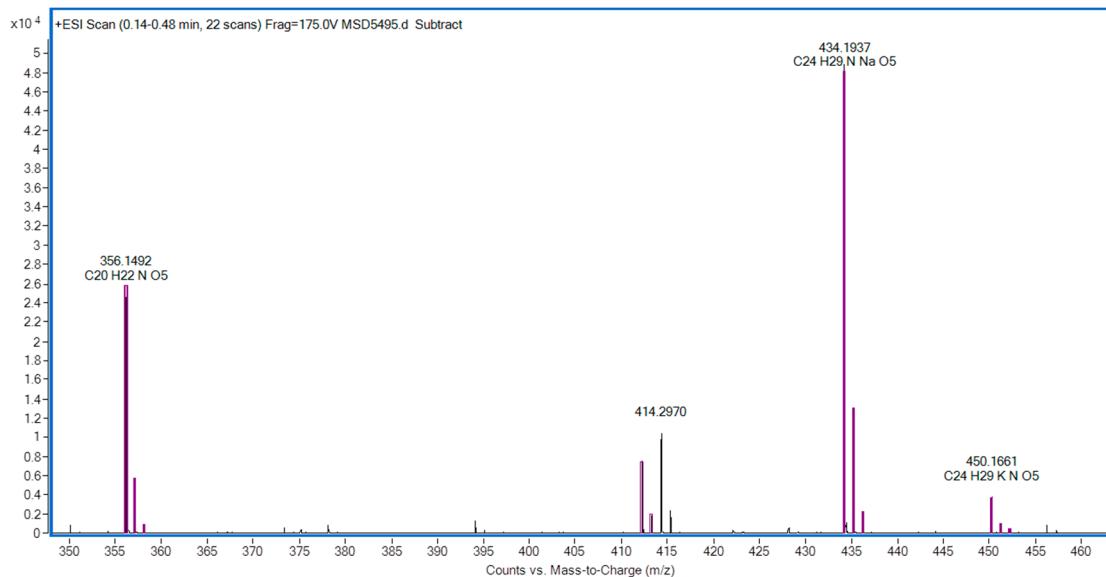


Figure S3: ESI-TOF analysis of Fmoc-N-Me-Thr(tBu)-OH using dimethyl sulfate as N-methylation reagent.

R4-Fmoc-N-Me-Thr(tBu)-OH

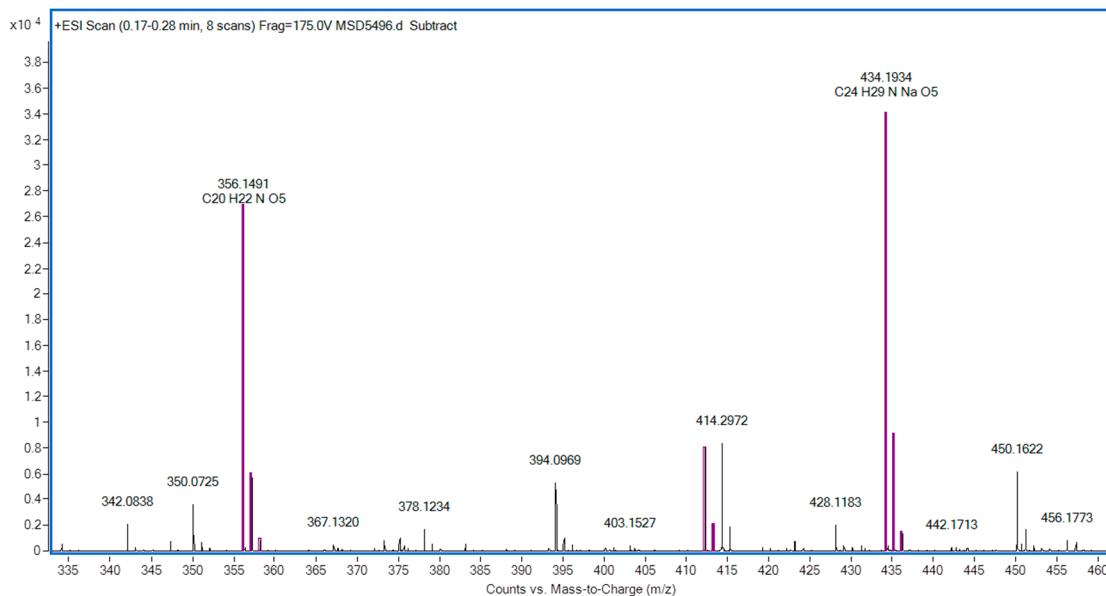


Figure S4: ESI-TOF analysis of Fmoc-N-Me-Thr(tBu)-OH using iodomethane as N-methylation reagent.

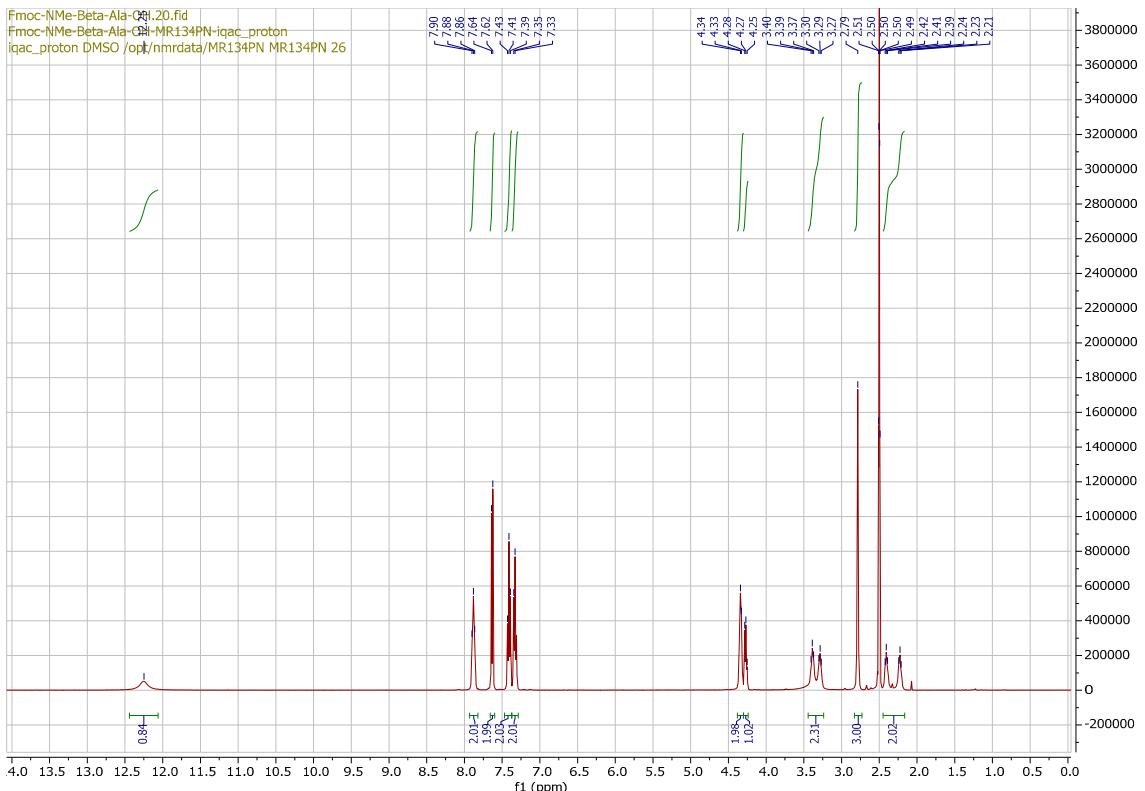


Figure S5: NMR ^1H spectrum of Fmoc-N-Me- β Ala-OH at 400 MHz in DMSO(d6).

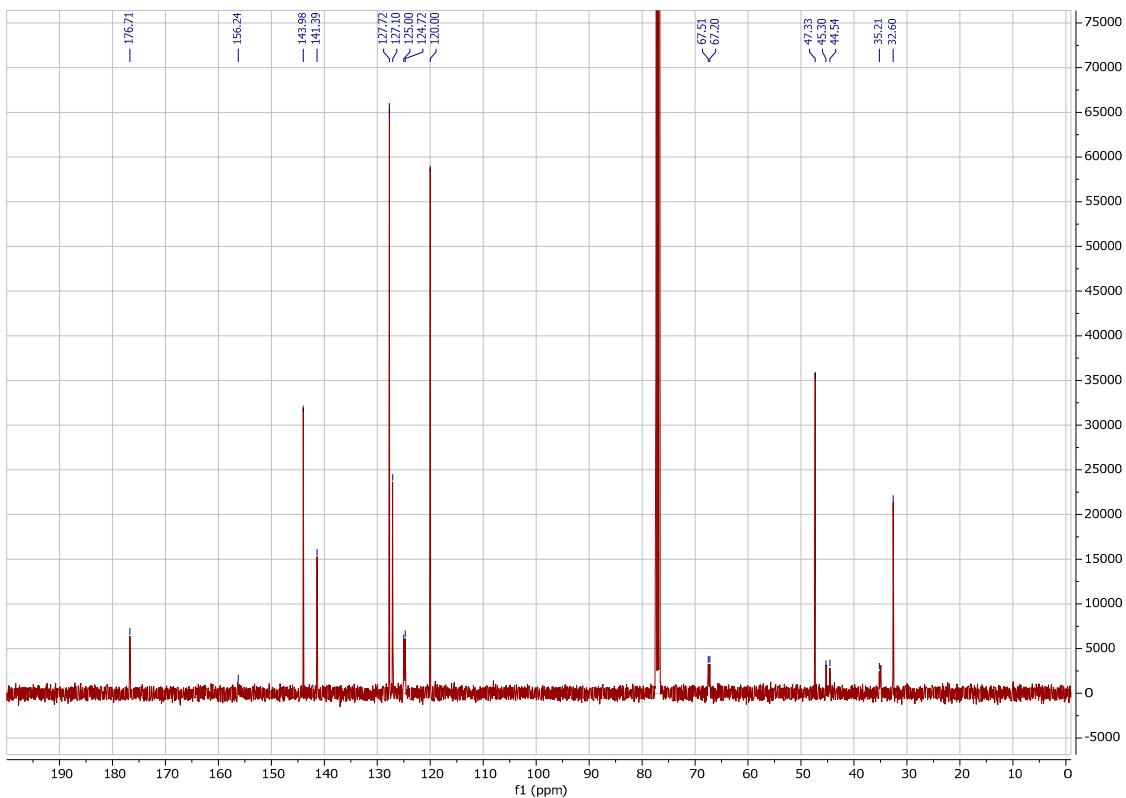


Figure S6: NMR ^1H spectrum of Fmoc-N-Me- β Ala-OH at 400 MHz in CDCl_3 .

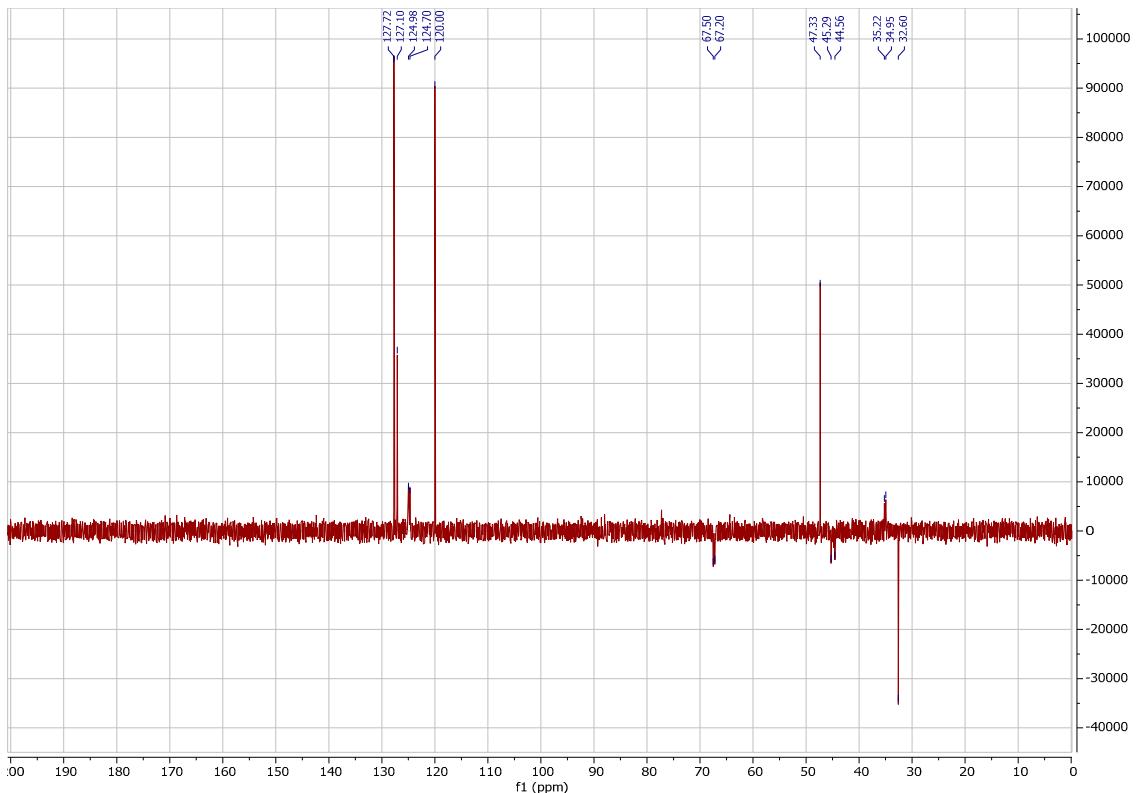


Figure S7: NMR DEP 135 spectrum of Fmoc-N-Me- β Ala-OH at 400 MHz in CDCl_3 .

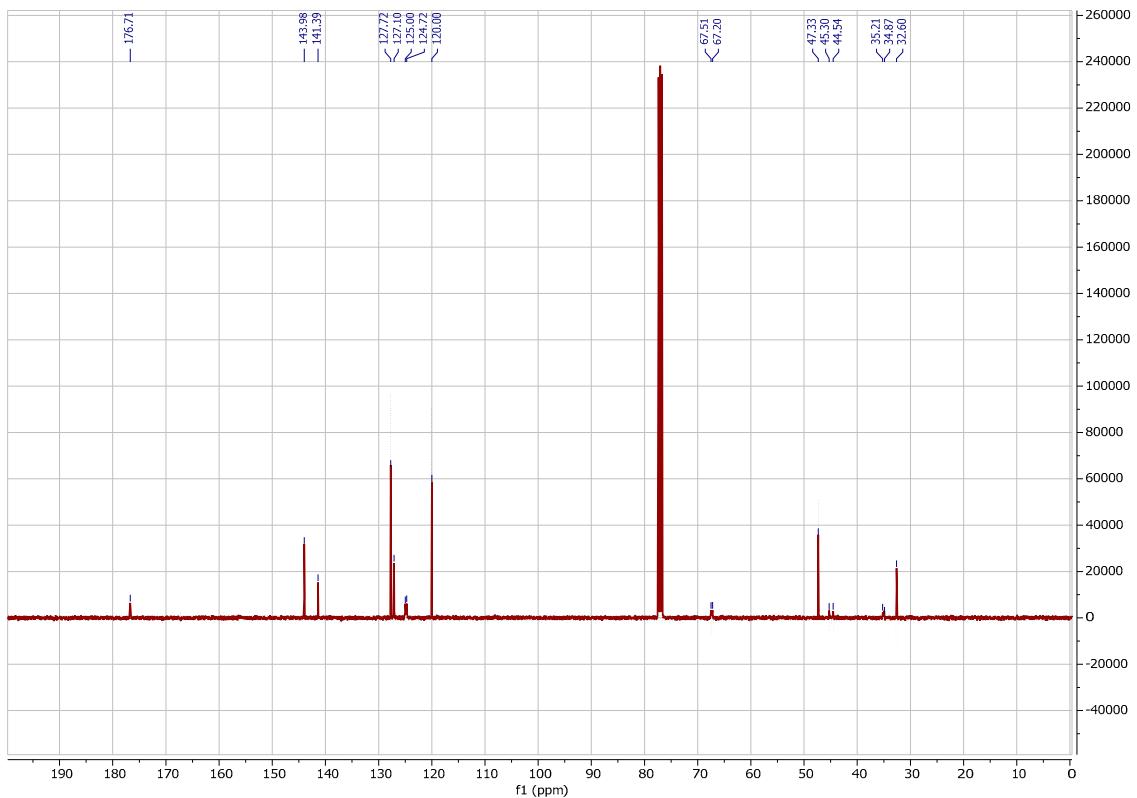


Figure S8: Overlaid ${}^{13}\text{C}$ and DEP 135 spectra of Fmoc-N-Me- β Ala-OH at 400 MHz in CDCl_3 .

Figure S9: NMR ^1H spectrum of Fmoc-N-Me-Thr(tBu)-OH at 400 MHz in DMSO(d₆).

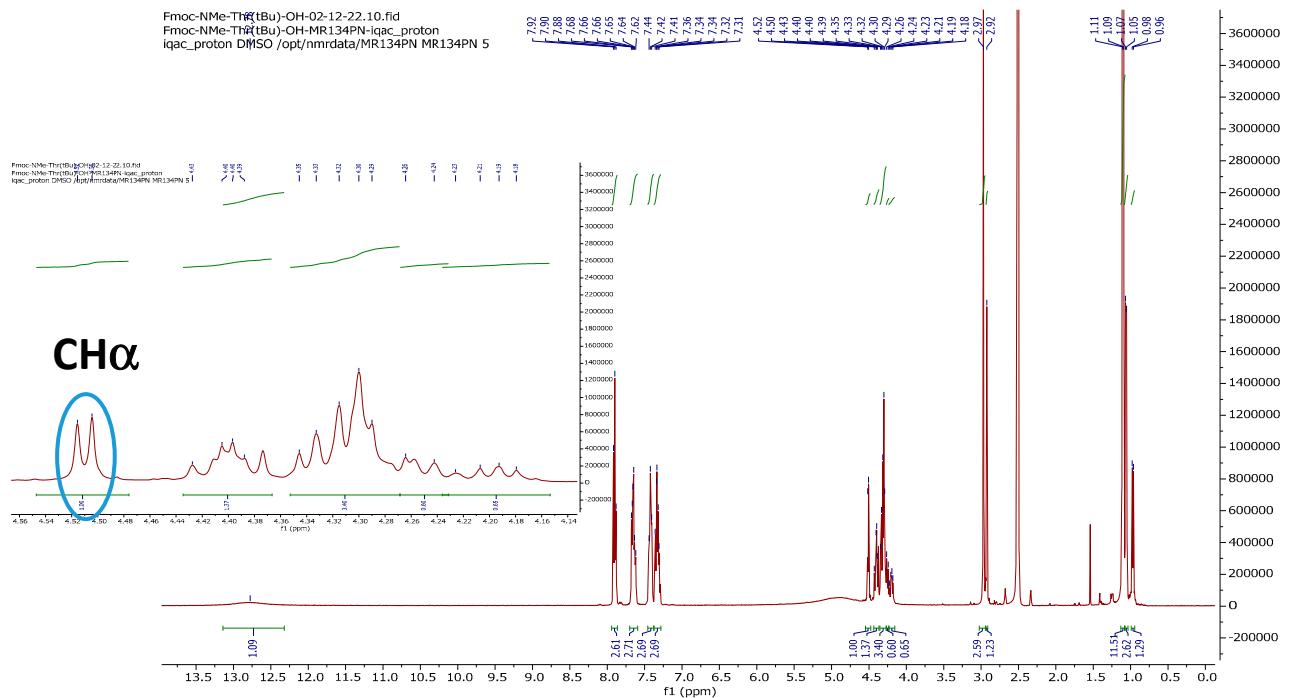


Figure S10: NMR ^1H spectrum of Fmoc-NMe-Thr(tBu)-OH enlarged aliphatic zone.

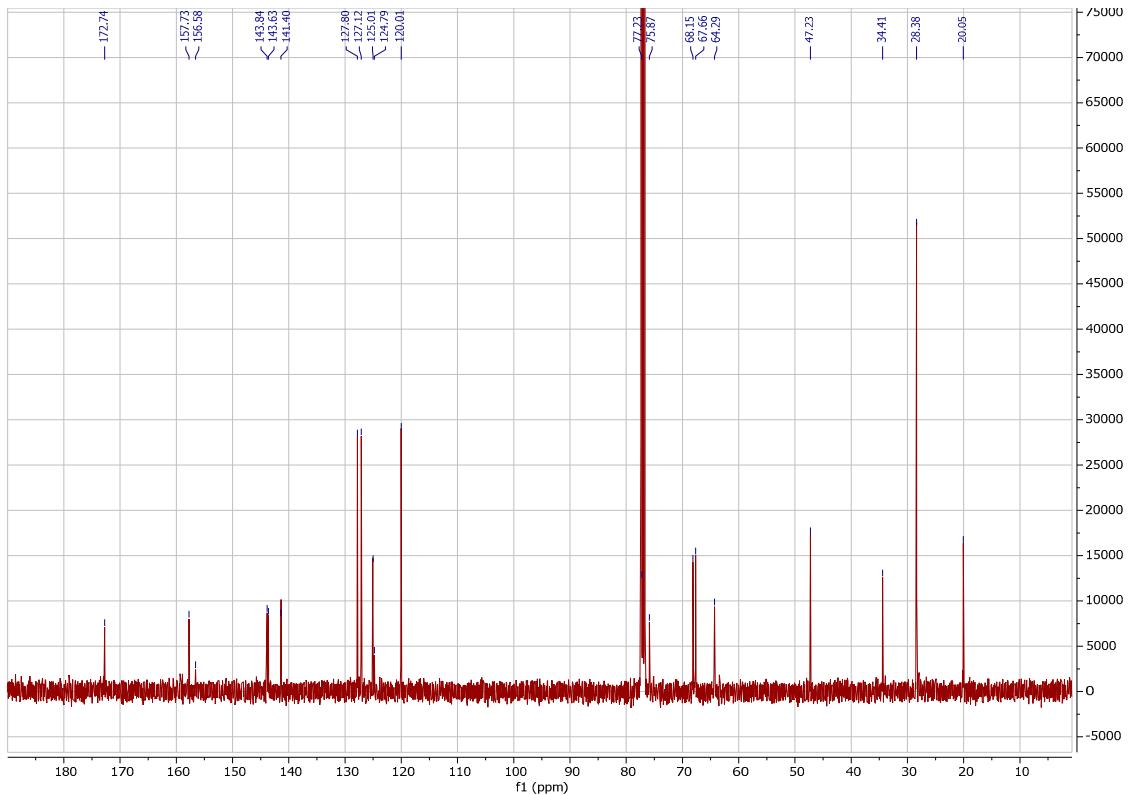


Figure S11: NMR ${}^{13}\text{C}$ spectrum of Fmoc-N-Me-Thr(tBu)-OH at 400 MHz in CDCl_3 .

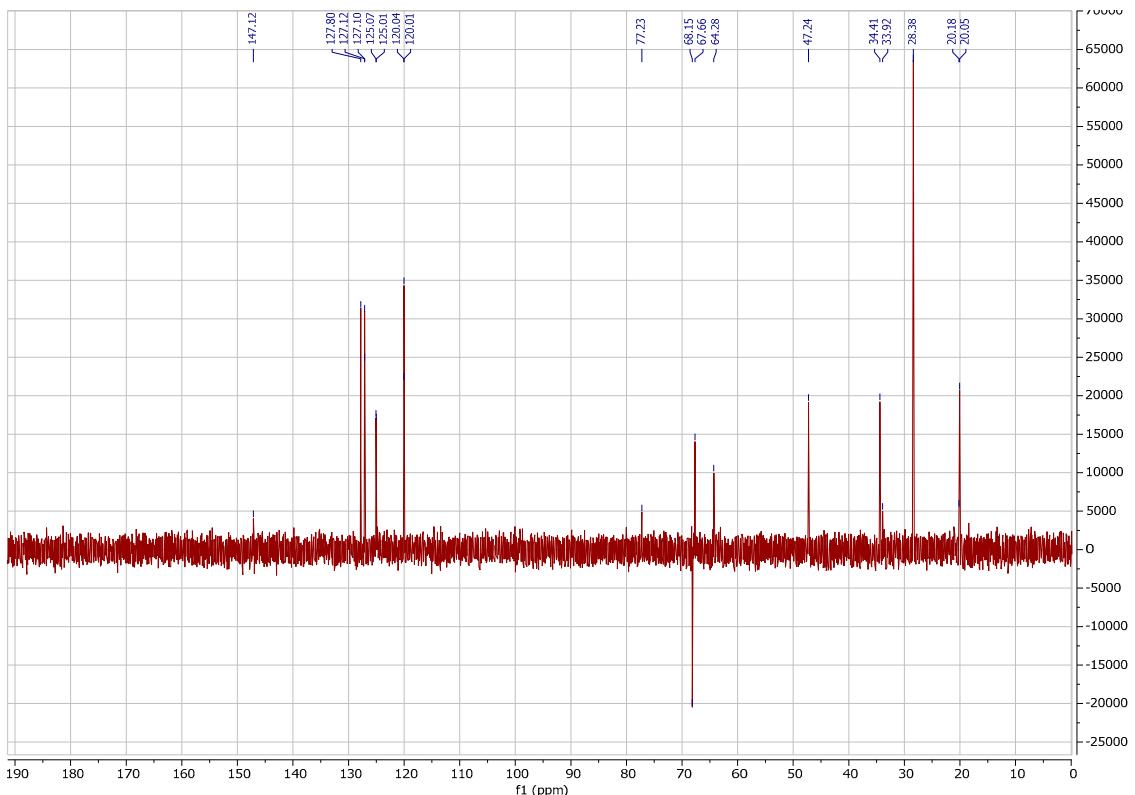


Figure S12: NMR DEP 135 spectrum of Fmoc-N-Me-Thr(tBu)-OH at 400 MHz in CDCl_3 .

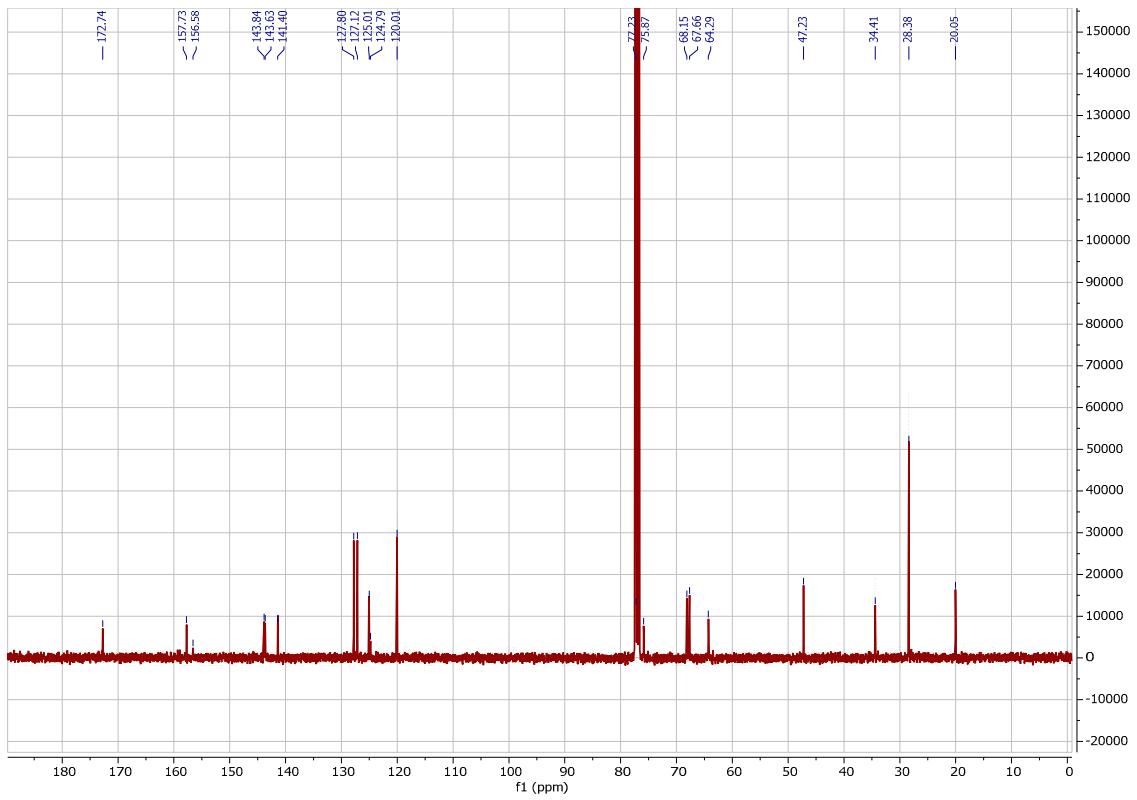


Figure S13: Overlaid ¹³C and DEP 135 spectra of Fmoc-N-Me-Thr(tBu)-OH at 400 MHz in CDCl₃.