

Abstract

# Two-Dimensional Layered Amorphous Metal Oxide Gas Sensors (LAMOS) Perspectives and Gas Sensing Properties<sup>†</sup>

Valentina Paolucci<sup>1,\*</sup> , Jessica De Santis<sup>1</sup> , Vittorio Ricci<sup>1</sup> , Giacomo Giorgi<sup>2,3</sup>  and Carlo Cantalini<sup>1,\*</sup>

<sup>1</sup> Department of Industrial and Information Engineering and Economics, University of L'Aquila, UdR INSTM of L'Aquila, 67100 L'Aquila, Italy

<sup>2</sup> Department of Civil & Environmental Engineering, Università Degli Studi di Perugia, 06125 Perugia, Italy

<sup>3</sup> CNR-SCITEC, 06123 Perugia, Italy

\* Correspondence: valentina.paolucci2@univaq.it (V.P.); carlo.cantalini@univaq.it (C.C.)

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**Abstract:** Two-dimensional Layered Amorphous Metal Oxide Sensors (LAMOS) represent a new class of 2D amorphous oxide (*a*-MOx) interfaces with unveiled properties in gas sensing applications. Herein, we report the humidity and gas sensing response of *p*- and *n*-type chemoresistive few-layered (2D) amorphous *a*-SnO<sub>2</sub>, *a*-In<sub>2</sub>O<sub>3</sub>, and *a*-Cr<sub>2</sub>O<sub>3</sub>, discussing their reaction mechanisms using DFT modelling and electrical tests. LAMOS interfaces can be easily prepared by controlled oxidation in air of a large class of exfoliated 2D TMDs, MCs, and TMTH (Transition Metal Dichalcogenides, Chalcogenides, and Trihalides) like WS<sub>2</sub>, MoS<sub>2</sub>, SnSe<sub>2</sub>, In<sub>2</sub>Se<sub>3</sub>, NiCl<sub>2</sub>, and CrCl<sub>3</sub>, yielding 2D amorphous *a*-MOx interfaces. LAMOS platforms preserving all the surface-to-volume advantages of their 2D precursors show excellent gas sensing properties representing a new class of material for gas sensing applications.

**Keywords:** 2D; amorphous metal oxides; oxidation; TMDs; MCs; TMTH



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## 1. Introduction

The intrinsic thermodynamic instability ( $\Delta G < 0$ ) of 2D exfoliated TMDs/MCs/TMTHs (Transition Metal Dichalcogenides/Metal Chalcogenides/Transition Metal Trihalides), demonstrated by their spontaneous oxidation in dry/wet air laboratory conditions, represents a great opportunity to develop, via suitable thermal treatment, template-self-assembled, amorphous-metal-oxide (*a*-MOx) skin layers over crystalline 2D exfoliated TMDs/MCs/TMTH.

Departing from liquid-phase exfoliated TMDs/MCs/TMTHs, annealing in air at temperatures below the crystallization temperature of the native oxide, either amorphous/crystalline 2D heterostructures of *a*-MO/TMDs [1,2], or fully oxidized amorphous 2D *a*-MOx interfaces can be prepared [3] with unexploited surface properties.

Herein, we demonstrate that the oxidation/amorphization process can be extended to a large variety of exfoliated TMDs (WS<sub>2</sub>), MCs (SnSe<sub>2</sub>), and TMTH (CrCl<sub>3</sub>) where sulfur, selenium, or chlorine atoms can be easily displaced by O<sub>2</sub> atoms under controlled oxidation conditions, producing 2D layered *n*-type *a*-WO<sub>3</sub>, *a*-SnO<sub>2</sub>, and *p*-type *a*-Cr<sub>2</sub>O<sub>3</sub> 2D flakes spin-coated as thin films, with excellent sensing properties to H<sub>2</sub>, NH<sub>3</sub>, H<sub>2</sub>S, and NO<sub>2</sub>, and long-term stability properties. This research opens new perspectives for a novel generation of layered interfaces (LAMOS), exploiting new interaction mechanisms of these van der Waals amorphous semiconductor interfaces with the environment.

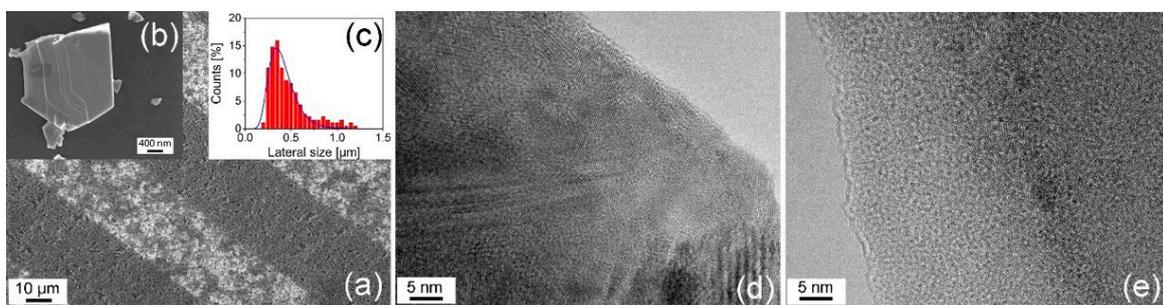
## 2. Materials and Methods

Liquid-phase exfoliated commercial SnSe<sub>2</sub>, WS<sub>2</sub>, and CrCl<sub>3</sub> powders were annealed in air at different temperatures (180 °C–300 °C) and times (24–70 h), and spin-coated over interdigital electrodes provided with platinum electrodes and a back side heater. Platforms

have been tested to sub ppm H<sub>2</sub>, NH<sub>3</sub>, H<sub>2</sub>S, NO<sub>2</sub> gases and humidity at a 100 °C operating temperature.

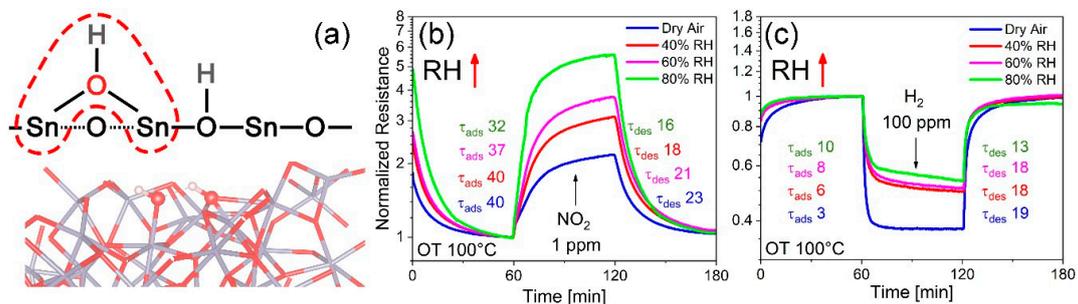
### 3. Discussion

Figure 1a shows the SEM picture of a spin-coated thin film over interdigitated electrodes (light regions) of 2D *a*-SnO<sub>2</sub> flakes of around 300 nm lateral size (Figure 1b,c), with a vertical height of approximately 20 nm, forming localized inter-sheet junctions between the flakes. Figure 1d shows the HRTEM of exfoliated flakes of 2D SnSe<sub>2</sub> exhibiting a fully crystalline and ordered 2D texture extending up to the edge of the flake (see electronic magnification, Figure 1d). After annealing, the 2D SnSe<sub>2</sub> is transformed in *a*-SnO<sub>2</sub>, whose amorphous structure is shown in Figure 1e. Grazing incidence XRD and XPS analysis of the annealed SnSe<sub>2</sub> flakes confirms the formation of the fully amorphous *a*-SnO<sub>2</sub> layer with a chemical composition matching that of SnO<sub>2</sub>.



**Figure 1.** (a) SEM picture of *a*-SnO<sub>2</sub> spin-coated flakes over Si<sub>3</sub>N<sub>4</sub> substrates with Pt-finger-type electrodes; (b) high-magnification SEM of terraced as-exfoliated SnSe<sub>2</sub> flake; (c) lateral size distribution of exfoliated SnSe<sub>2</sub>; (d) HRTEM of the as exfoliated crystalline 2D-SnSe<sub>2</sub>; (e) HRTEM of the *a*-SnO<sub>2</sub> flake after oxidation of the 2D-SnSe<sub>2</sub>.

Considering humid air as a natural background in practical gas sensing applications, we preliminary applied combined density function theory and ab initio molecular dynamics, demonstrating that a dissociative water mechanism occurs over *a*-MO<sub>x</sub> surfaces, leading to the formation of chemisorbed hydroxyls, as shown in Figure 2a. Experiments that aimed to investigate the humidity cross-response on NO<sub>2</sub> and H<sub>2</sub> sensing highlighted that increasing the relative humidity increases the degree of hydroxylation, resulting in an increase/decrease in the sensor signal response (i.e., R<sub>g</sub>/R<sub>a</sub> or R<sub>a</sub>/R<sub>g</sub>) to 1 ppm NO<sub>2</sub> and 100 ppm H<sub>2</sub>, as shown in Figure 2b,c, respectively.



**Figure 2.** (a) Schematization of H<sub>2</sub>O dissociative chemisorption mechanism over *a*-SnO<sub>2</sub> at a 100 °C operating temperature; (b,c) adsorption/desorption responses to 1 ppm NO<sub>2</sub> and 100 ppm H<sub>2</sub> with increasing RH.

Adsorption/desorption mechanisms of water and gases over amorphous interfaces (*a*-MO<sub>x</sub>), investigated via theory and experiments, resulted in being congruent with those of crystalline metal oxides. Long-term stability properties of the electrical response to

humidity and different gases, over a period of one year, exhibit no remarkable fluctuations in the base line resistance (BLR) or the sensor's signal response (i.e., RRs), demonstrating that the amorphization/oxidation strategy effectively passivates the material from further degradation, while preserving an excellent gas sensing response.

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