

Supplementary Materials: Surfactants for Electrophoretic Deposition of Polyvinylidene Fluoride–Silica Composites

Zhengzheng Wang and Igor Zhitomirsky *

Department of Materials Science and Engineering, McMaster University, Hamilton, ON L8S4L7, Canada; wangz338@mcmaster.ca (Z.W.); zhitom@mcmaster.ca (I.Z.)

* Correspondence: zhitom@mcmaster.ca (I.Z.)

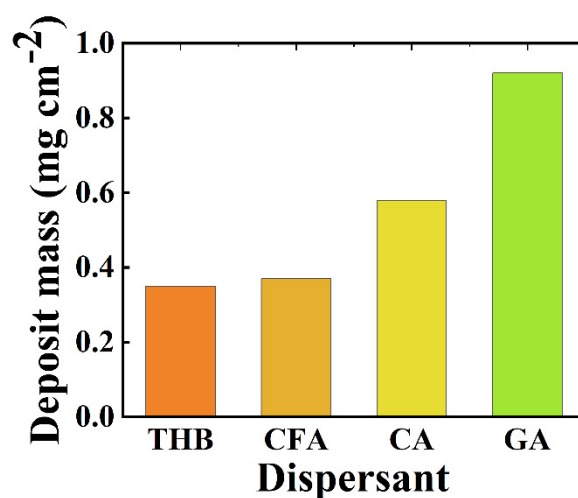


Figure S1. Deposit mass achieved for different dispersants using 5 g L⁻¹ PVDF suspensions containing 1 g L⁻¹ dispersants for deposition time of 5 min and deposition voltage of 50V.

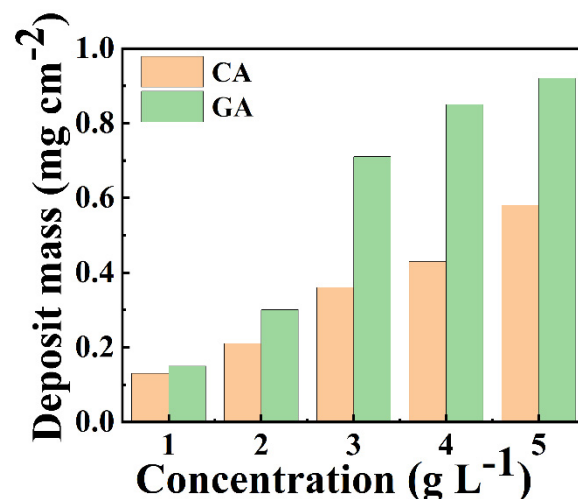


Figure S2. Deposit mass versus PVDF concentration in suspensions, containing 1 g L⁻¹ dispersants for deposition time of 5 min and deposition voltage of 50V.

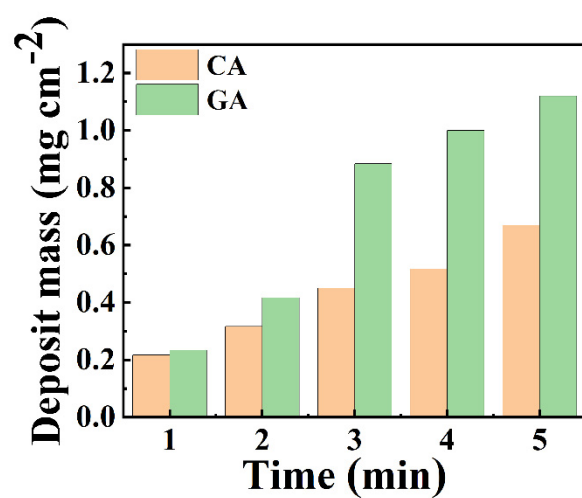


Figure S3. Deposit mass versus deposition time at a deposition voltage of 100 V for 5 g L⁻¹ PVDF suspensions.

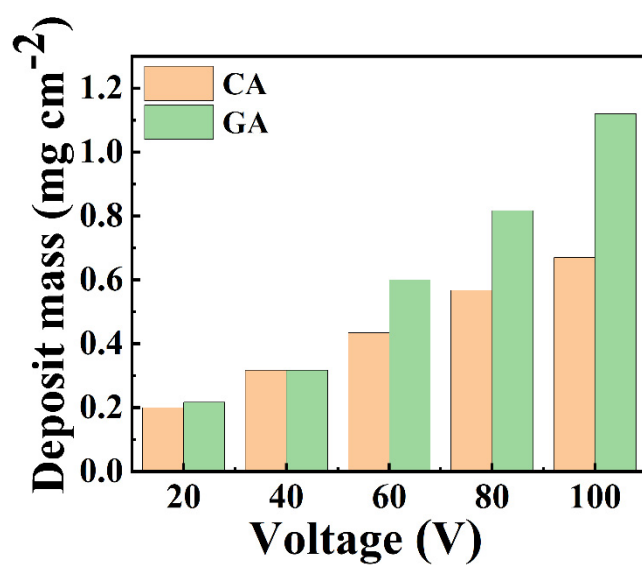


Figure S4. Deposit mass versus deposition voltage for 5 g L⁻¹ PVDF suspensions at deposition time of 5 min.

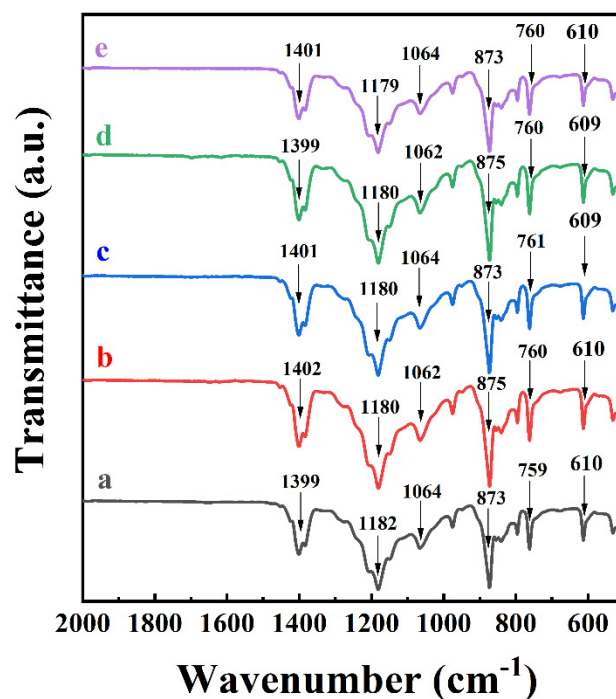


Figure S5. FTIR spectra of deposits, prepared from using 5 g L⁻¹ PVDF suspensions containing 1 g L⁻¹ dispersants: (a) THB, (b) CFA, (c) CA, (d) GA for deposition time of 5 min and deposition voltage of 100V and (e) as-received PVDF.

The spectra of deposited materials contains peaks similar to the peaks of as-received PVDF. The strong broad absorption at 1401 cm⁻¹ in the spectrum of as-received PVDF resulted from wagging of CH₂ and antisymmetric stretching of C-C bonds[1]. Other strong bands at 1179 and 873cm⁻¹ were associated with the stretching and rocking of CF₂ bonds[1, 2].

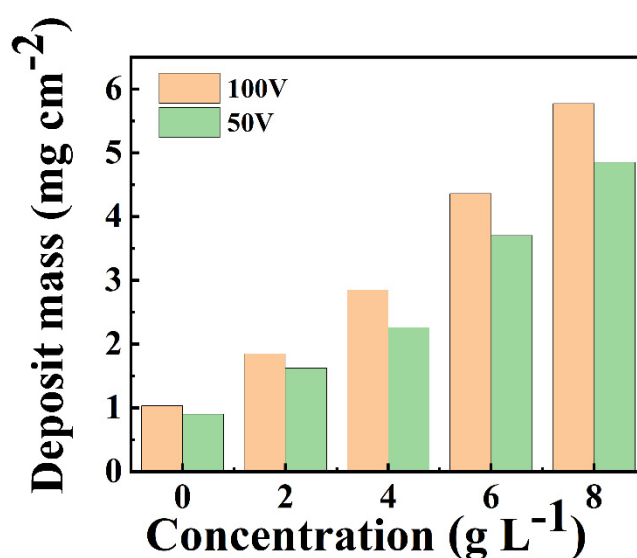


Figure S6. Deposit mass versus nanosilica concentration in 5 g L⁻¹ PVDF suspension containing 2 g L⁻¹ GA at deposition time of 5 min at voltages of 50 V and 100 V.

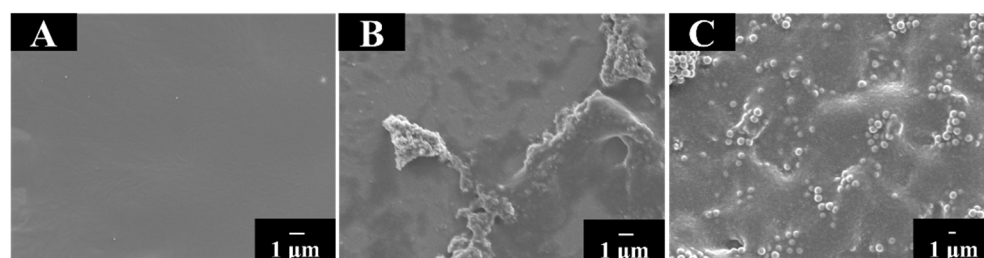


Figure S7. SEM images of coatings, prepared from 5 g L⁻¹ PVDF solution, containing 1 g L⁻¹ GA (A) without silica, (B) with 1 g L⁻¹ nanosilica and (C) with 1 g L⁻¹ micron size silica deposited at a deposition voltage of 100V and deposition time of 5 min and annealed at 200°C for 1 h.

1. Kobayashi, M.; Tashiro, K.; Tadokoro, H. Molecular vibrations of three crystal forms of poly (vinylidene fluoride). *Macromolecules* **1975**, *8*, 158-171.
2. Zeng, Z.; Yu, D.; He, Z.; Liu, J.; Xiao, F.-X.; Zhang, Y.; Wang, R.; Bhattacharyya, D.; Tan, T. T. Y. Graphene oxide quantum dots covalently functionalized PVDF membrane with significantly-enhanced bactericidal and antibiofouling performances. *Scientific reports* **2016**, *6*, 1-11.