

Supplementary Materials

Multicomponent Reactions Promoted by Ecocatalyst from Metal Hyperaccumulating Plant *Pluchea sagittalis*

Leonardo H. R. Alporti ¹, Monize Picinini ², Ernesto A. Urquieta-Gonzalez ^{2,*}, Caroline S. da Silva ³, Simone Y. S. Silva ⁴, Sebastião C. Silva ⁴, Marilene N. de Oliveira ⁴, Juliana Viera ^{3,5}, Maria Fatima das G. F. da Silva ¹ and Arlene G. Corrêa ^{1,*}

¹ Centre of Excellence for Research in Sustainable Chemistry, Department of Chemistry, Federal University of São Carlos, São Carlos 13565-905, SP, Brazil

² Department of Chemical Engineering, Federal University of São Carlos, São Carlos 13565-905, SP, Brazil

³ Group for Applied Instrumental Analysis, Department of Chemistry, Federal University of São Carlos, São Carlos 13565-905, SP, Brazil

⁴ Department of Chemistry, Federal University of South and Southeast of Pará, Marabá 68507-590, PA, Brazil

⁵ Faculty of Chemistry, Universidad de la República, Montevideo 11800, Uruguay

* Correspondence: urquieta@ufscar.br (E.A.U.-G.); agcorrea@ufscar.br (A.G.C.)



(a)



(b)

Figure S1. *Pluchea sagittalis* leaves (a) dried and ground. (b) ashes after calcination.

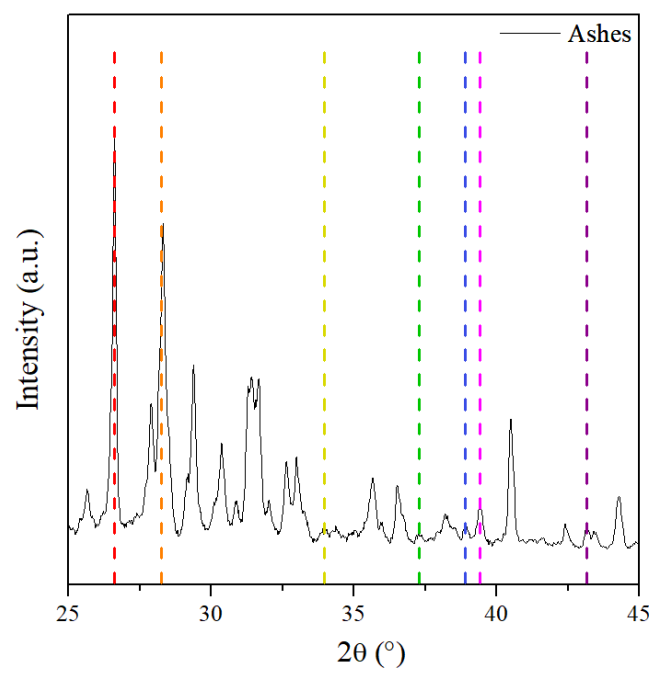


Figure S2. Enlarged ashes diffractogram between 25 to 45° 2θ

Table S1. Metal concentrations (ICP-OES) in plants collected at a mining area on northern Brazil.

Sample	Ba (mg kg⁻¹)	Ca (%)	Cd (mg kg⁻¹)	Cr (mg kg⁻¹)	Cu (mg kg⁻¹)	Fe (%)	K (%)	Mg (%)	Na (mg kg⁻¹)	Ni (mg kg⁻¹)	Pb (mg kg⁻¹)	S (mg kg⁻¹)
<i>C. pachystachya</i> leaves	21.93	1.12	<0.015	<2.54	83.0	0.35	2.60	0.28	0.52	3.6	<0.49	0.47
<i>C. pachystachya</i> branches	5.19	0.72	<0.015	<2.54	374	0.39	1.69	0.39	0.01	52.5	<0.49	0.56
<i>P. sagittalis</i> leaves	1.39	1.53	<0.015	<2.54	567	1.0	2.84	0.50	0.42	31.4	<0.49	1.46
<i>P. sagittalis</i> flowers	4.64	1.00	<0.015	<2.54	174.1	0.55	0.48	0.51	0.05	49.8	<0.49	1.61
<i>T. domingensis</i> leaves	4.25	0.89	<0.015	<2.54	56.8	0.55	3.85	0.47	0.80	23.7	<0.49	0.25

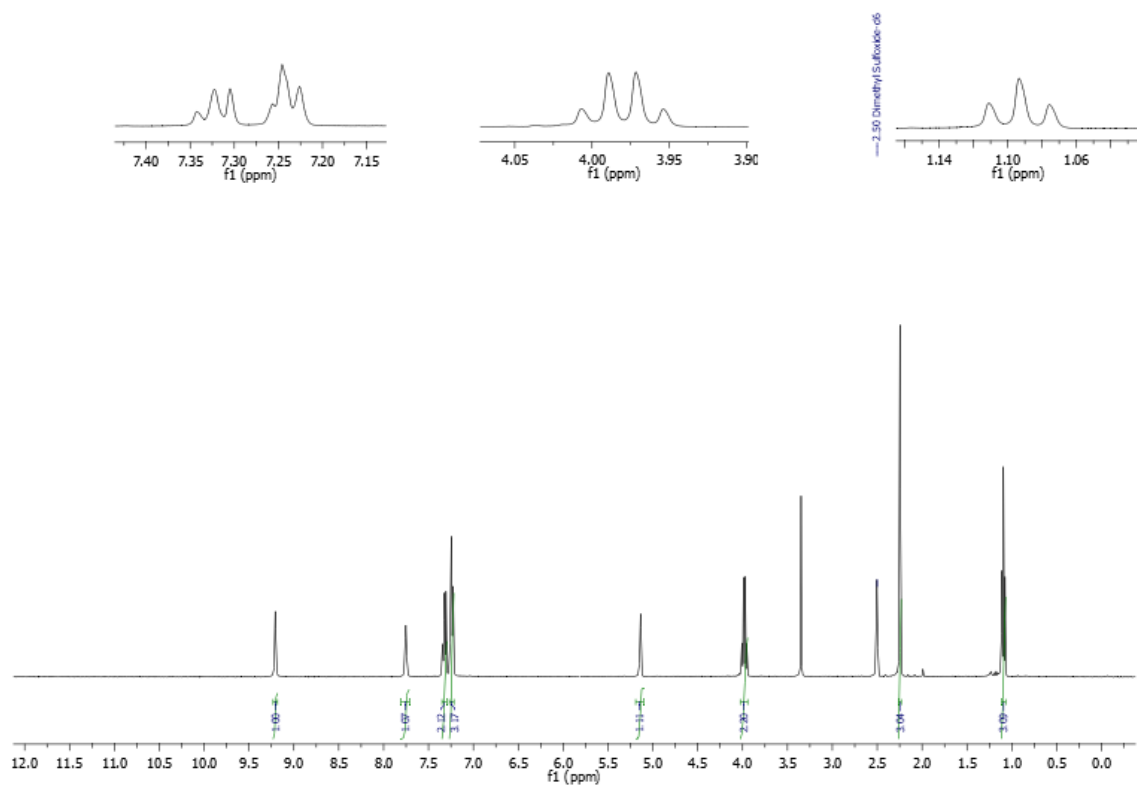


Figure S3. ^1H NMR ($\text{DMSO}-d_6$) of compound **3a**.

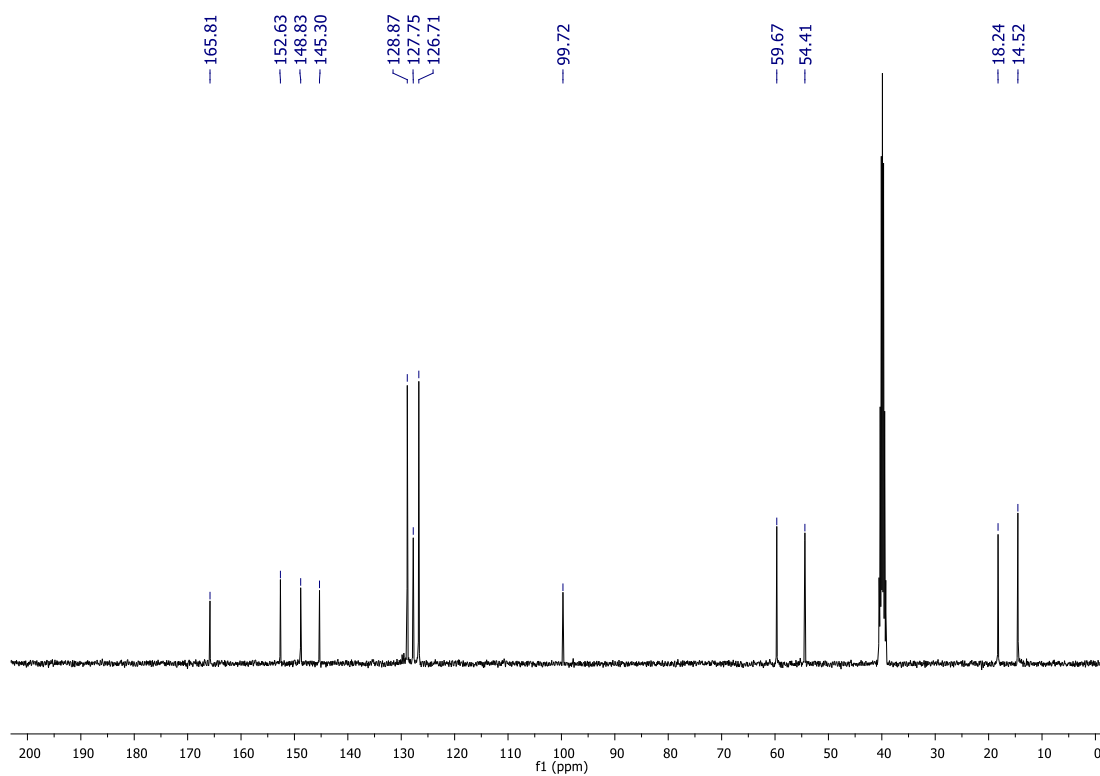


Figure S4. ^{13}C NMR ($\text{DMSO}-d_6$) of compound **3a**.

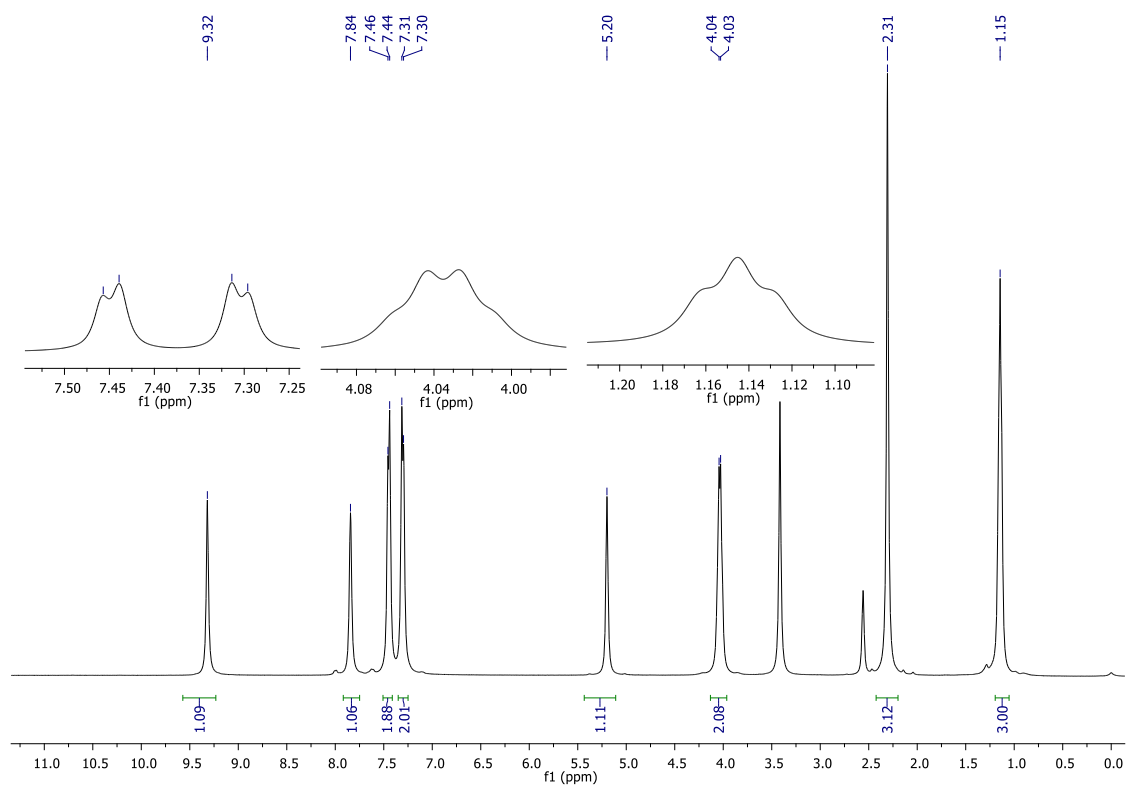


Figure S5. ¹H NMR (DMSO-*d*₆) of compound **3b**.

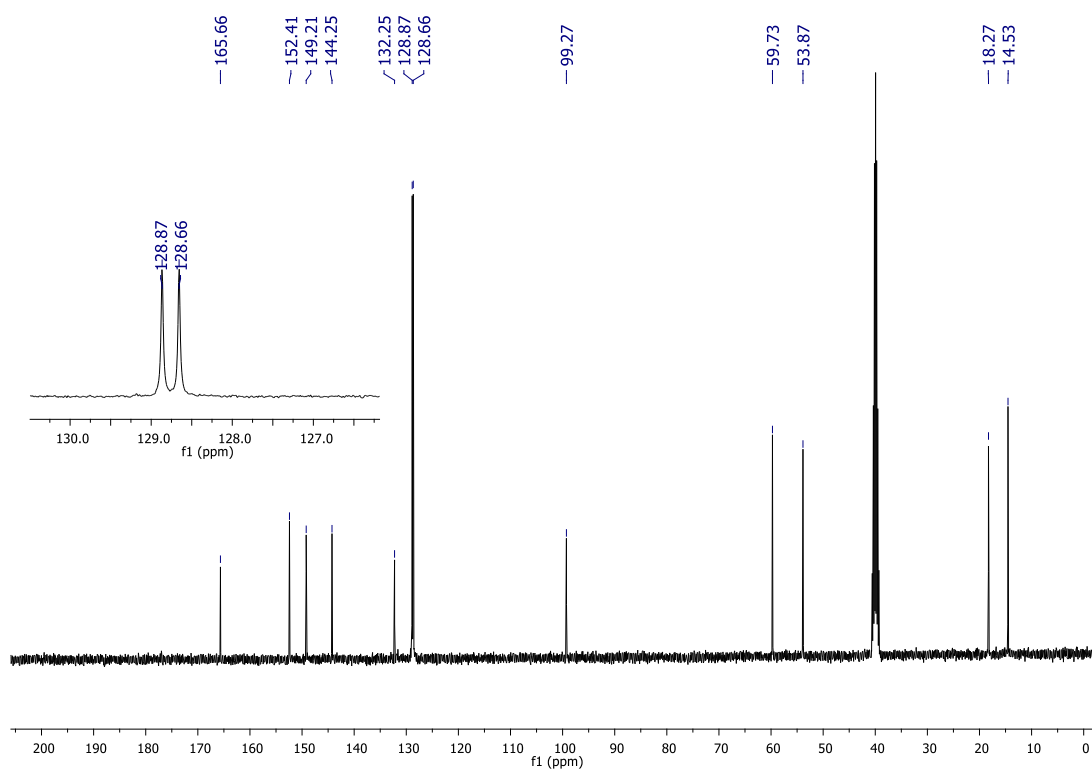


Figure S6. ¹³C NMR (DMSO-*d*₆) of compound **3b**.

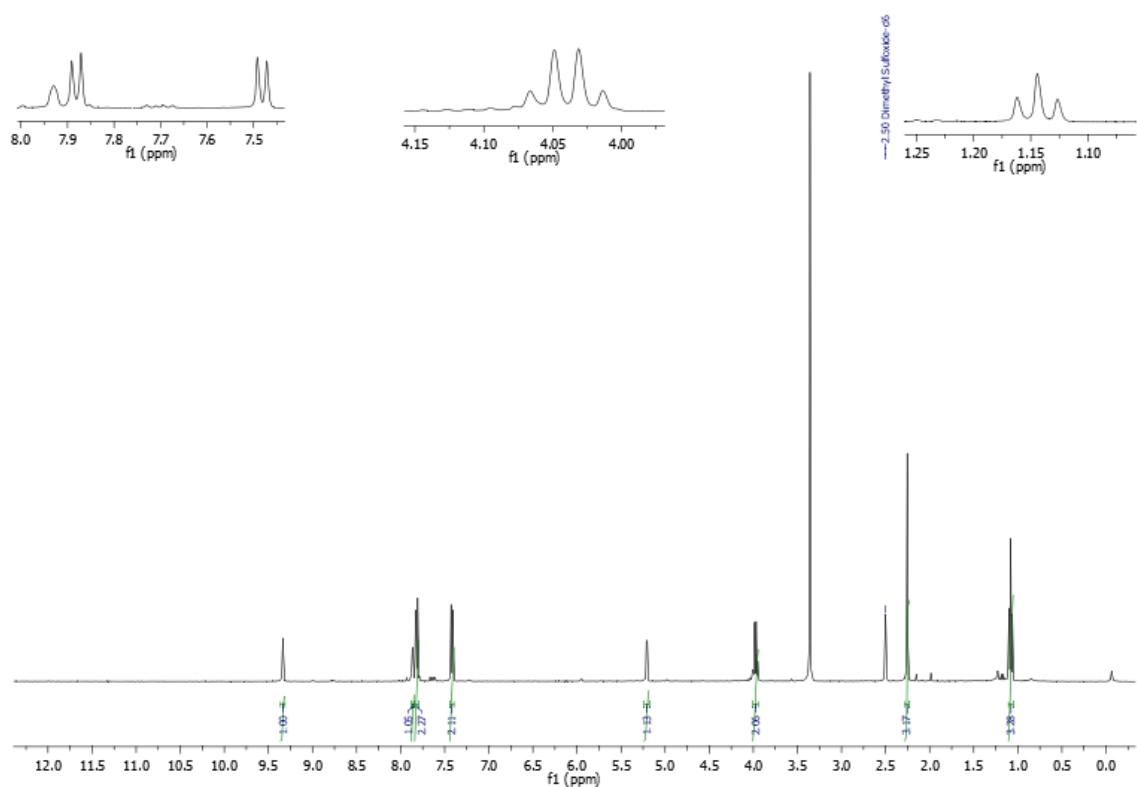


Figure S7. ^1H NMR (DMSO- d_6) of compound **3c**.

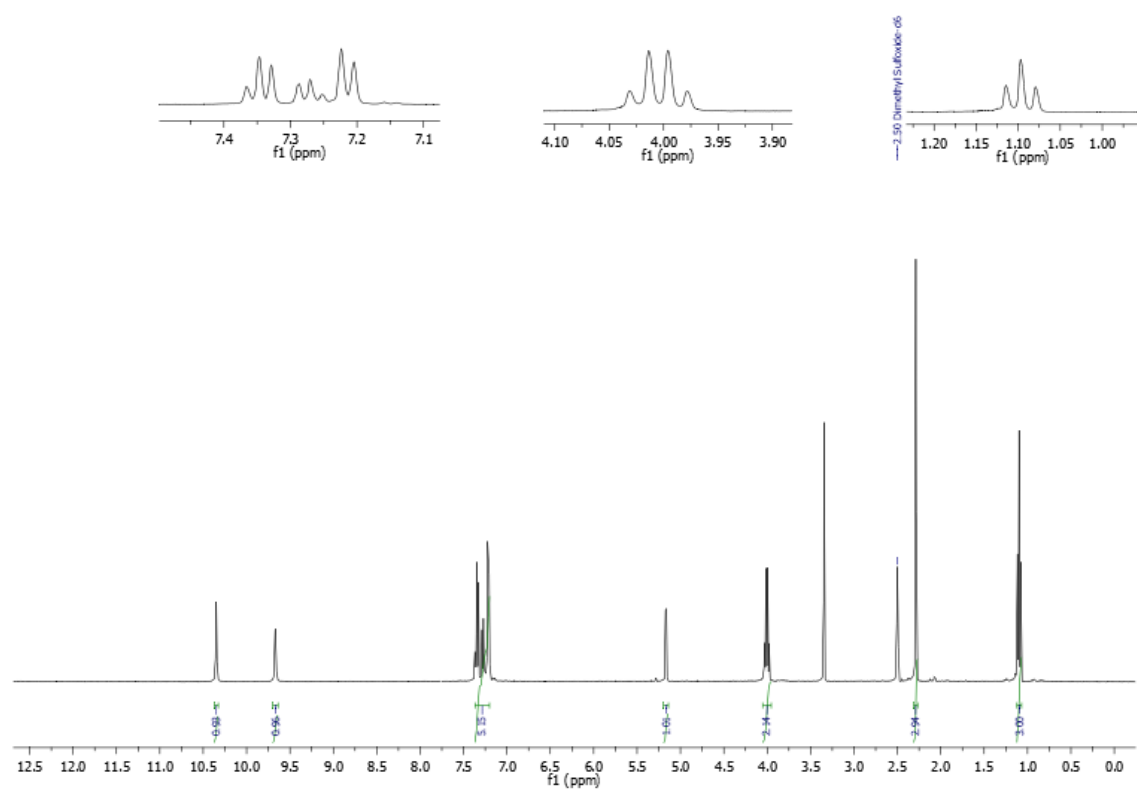


Figure S8. ^1H NMR (DMSO- d_6) of compound **3d**.

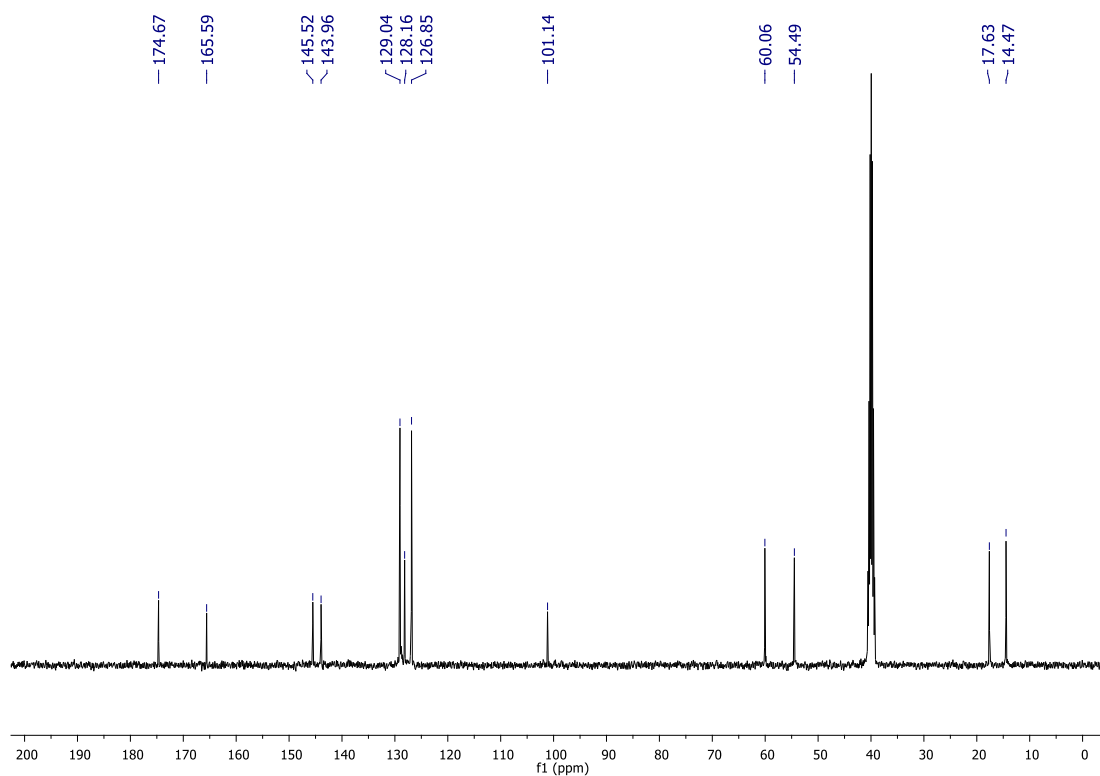


Figure S9. ^{13}C NMR (DMSO- d_6) of compound **3d**.

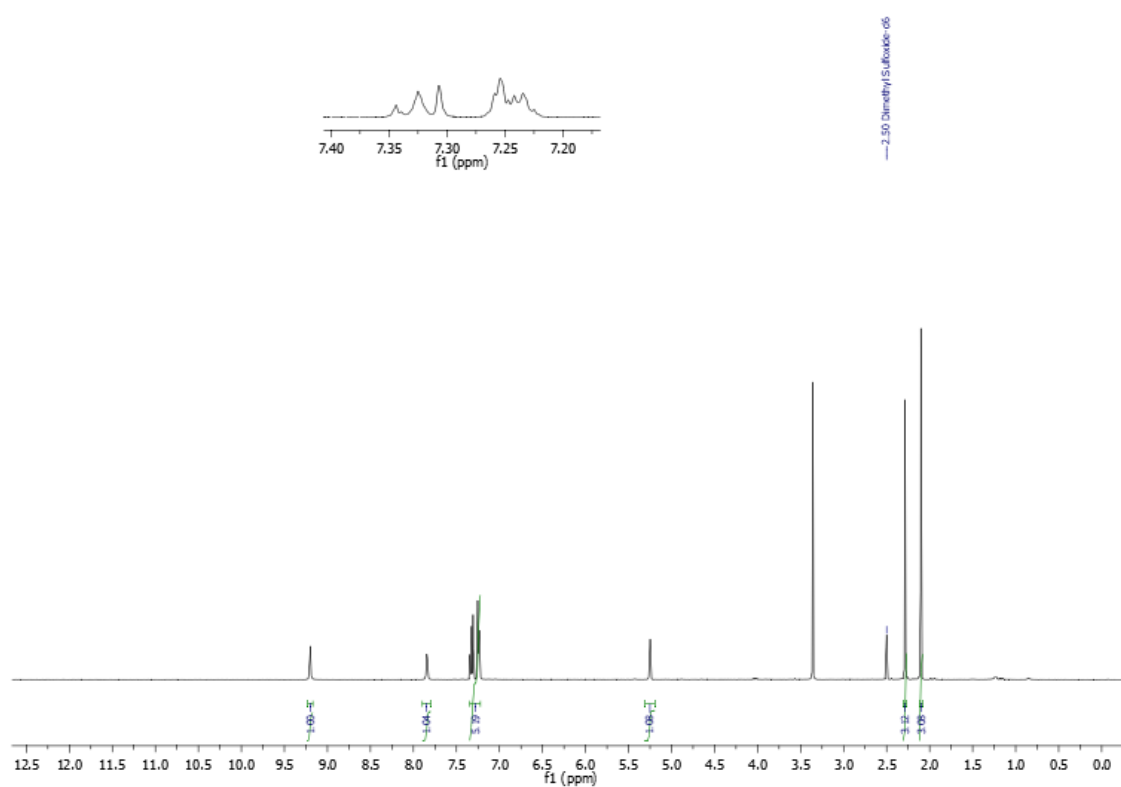


Figure. S10. ^1H NMR (DMSO- d_6) of compound **3e**.

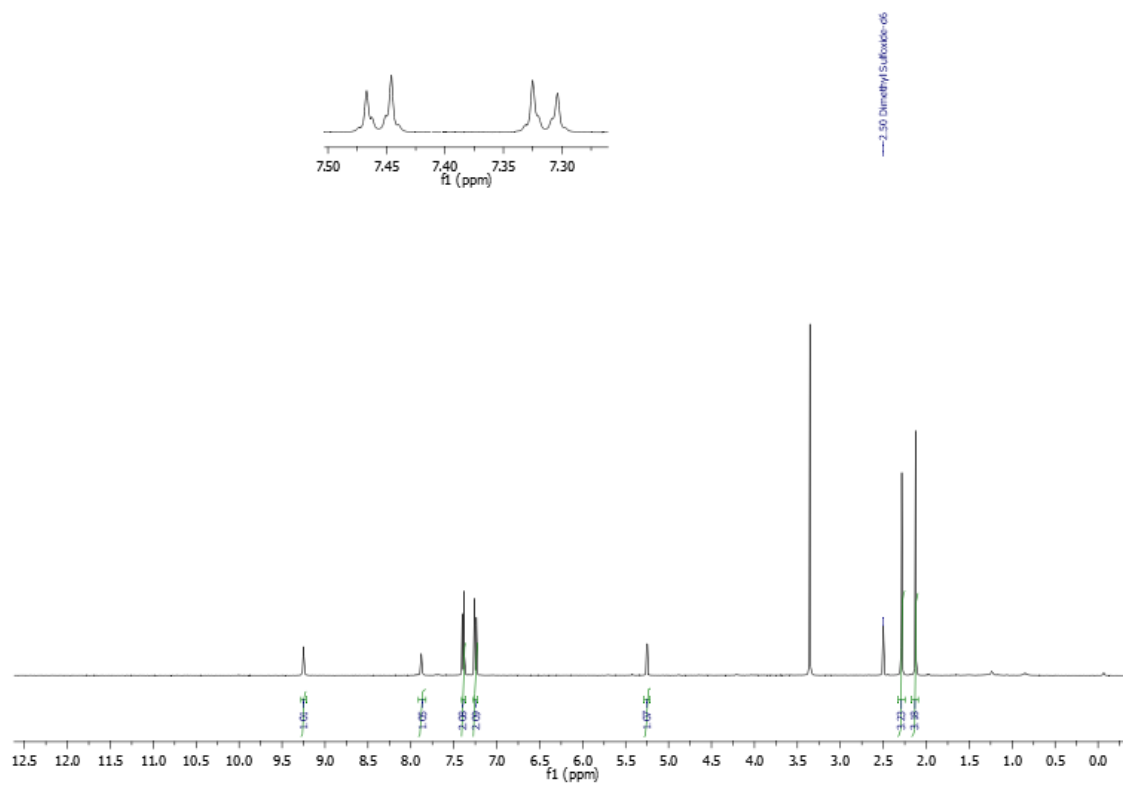


Figure S11. ^1H NMR (DMSO- d_6) of compound **3f**.

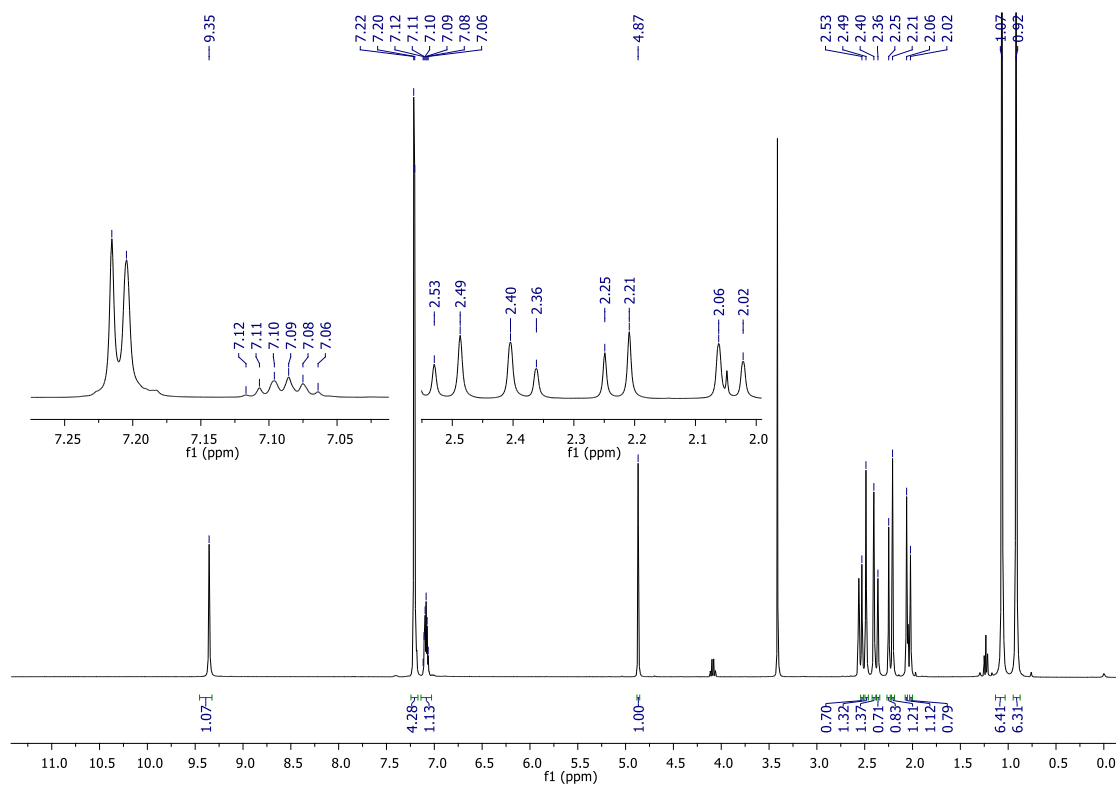


Figure S12. ^1H NMR (DMSO- d_6) of compound **5a**.

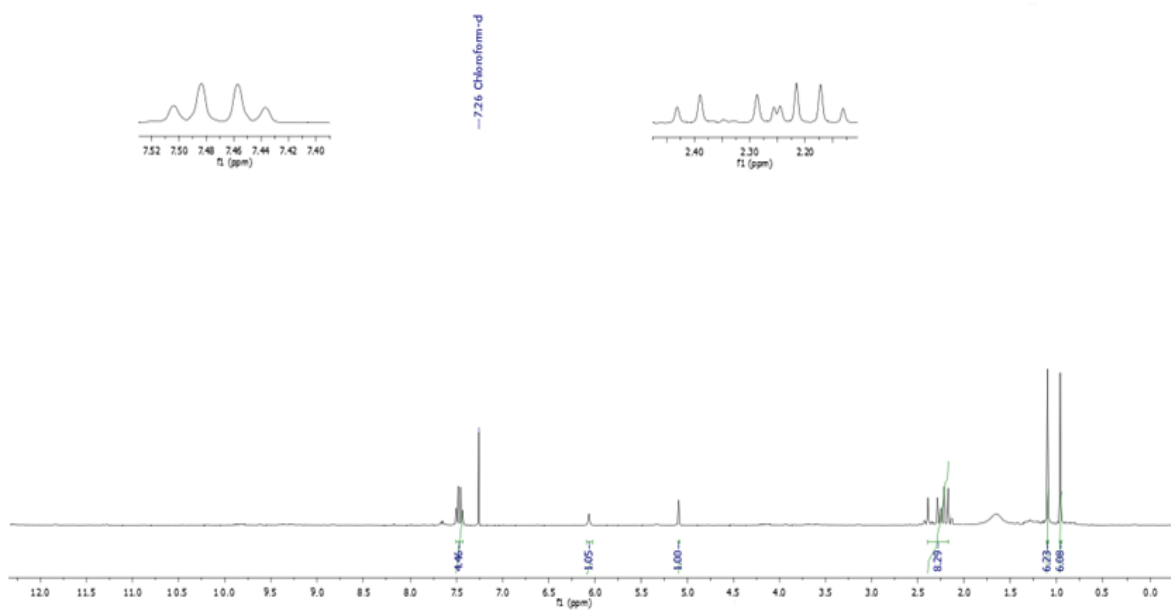


Figure S15. ¹H NMR (CDCl₃) of compound **5c**.

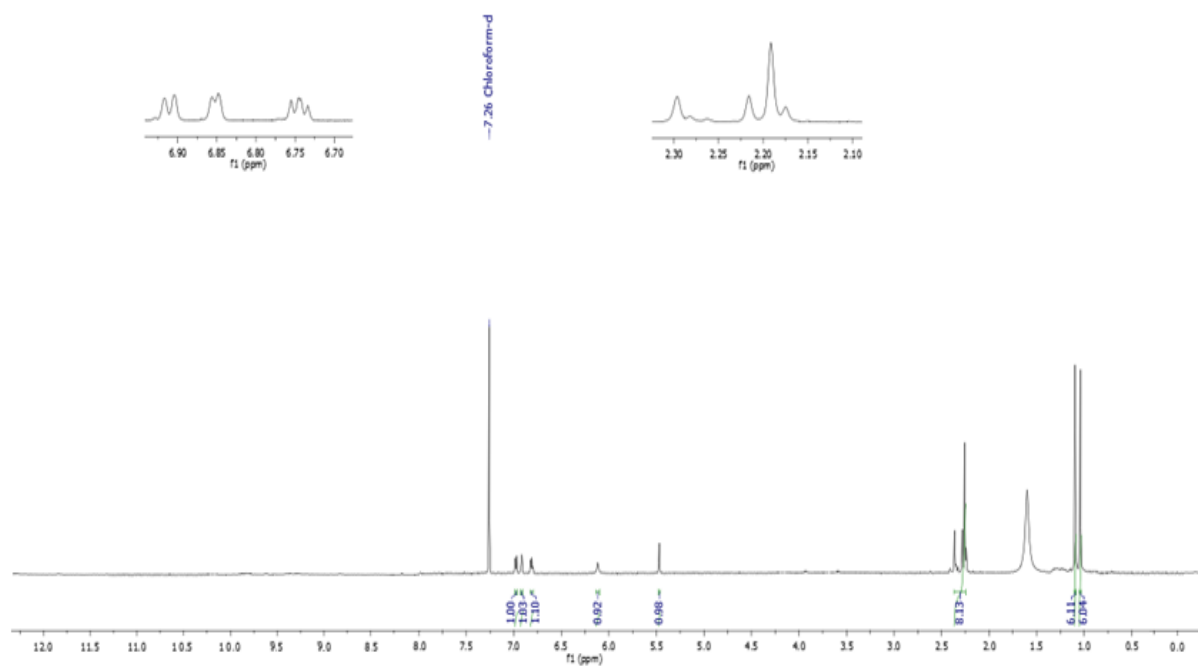


Figure S16. ¹H NMR (CDCl₃) of compound **5d**.

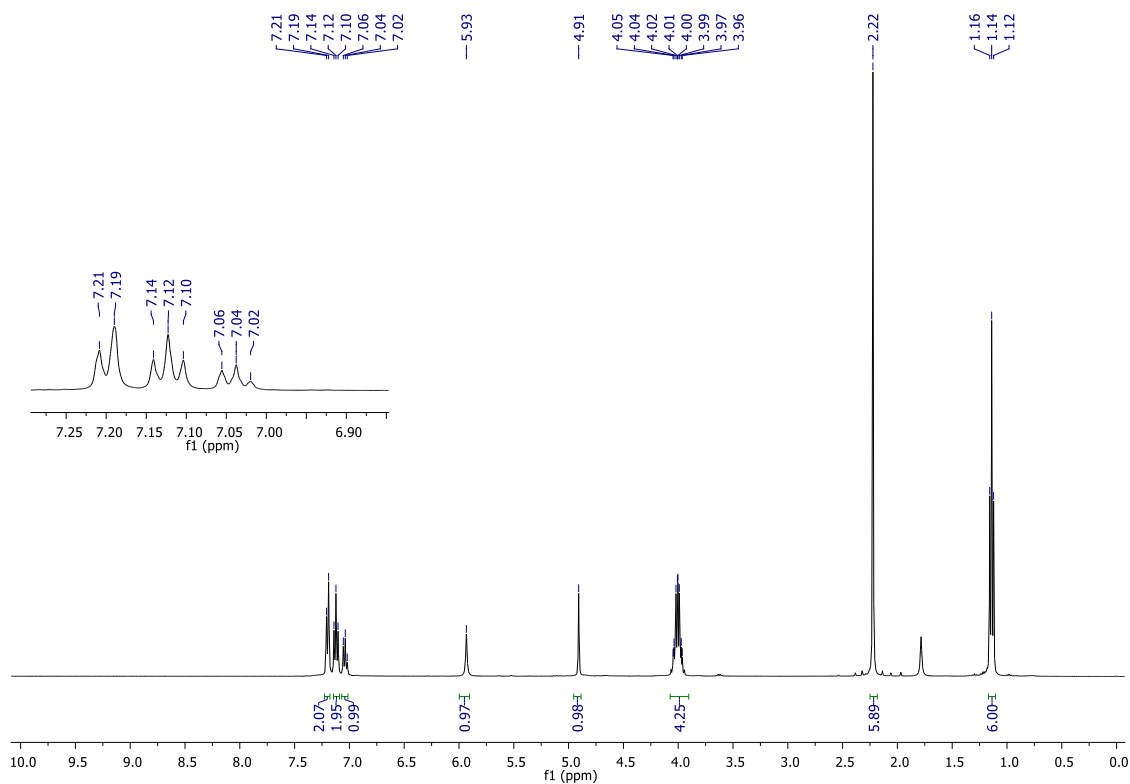


Figure S17. ¹H NMR (CDCl₃) of compound **5e**.

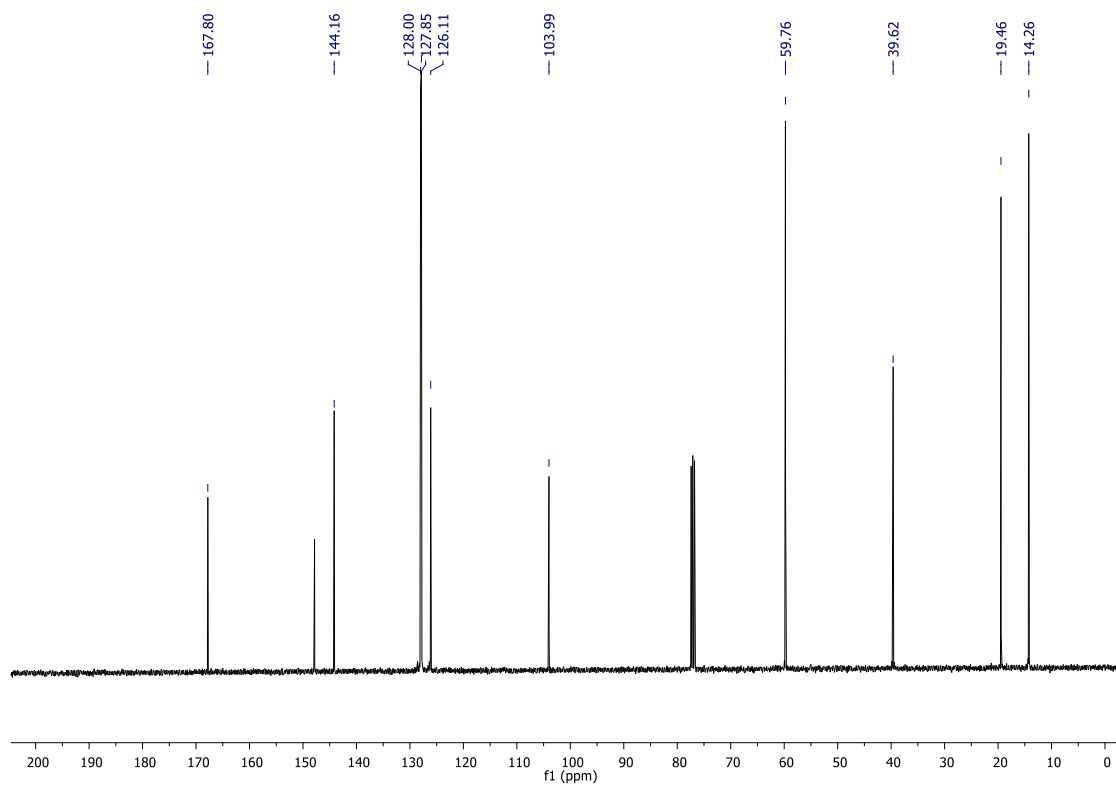


Figure S18. ¹³C NMR (CDCl₃) of compound **5e**.

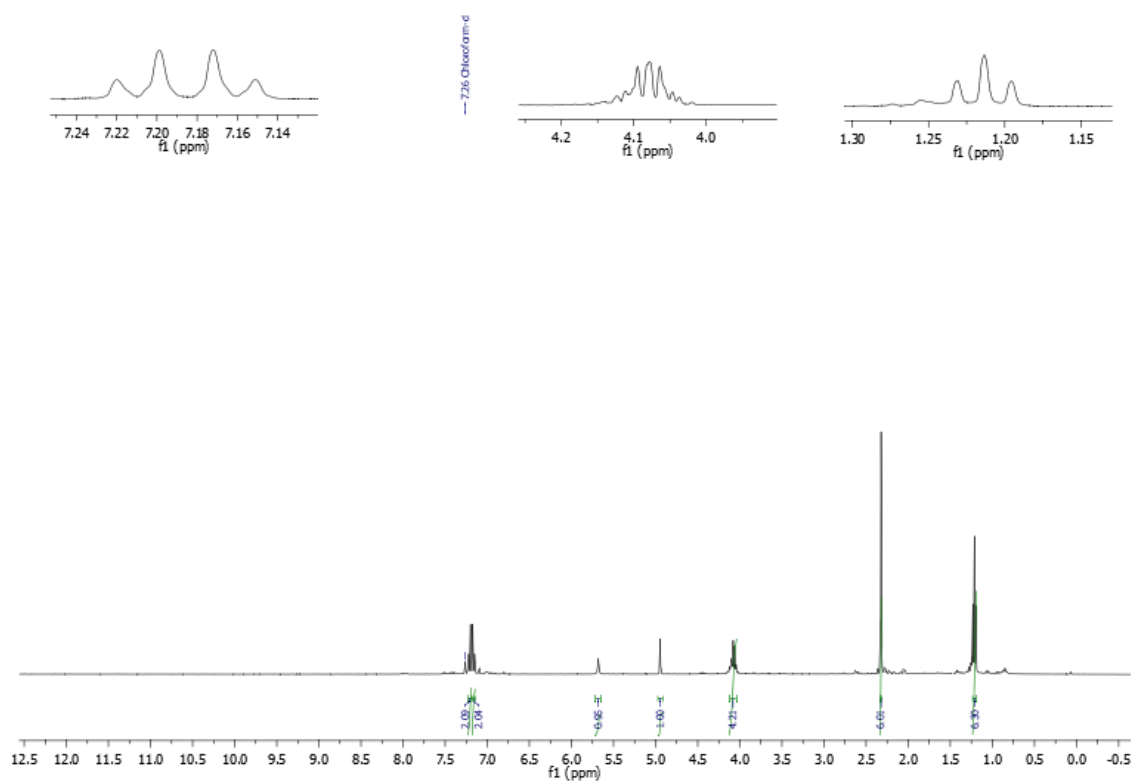


Figure S19. ^1H NMR (CDCl_3) of compound **5f**.