



The Landauer Principle and Its Implementations in Physics, Chemistry and Biology: Current Status, Critics and Controversies

Guest Editor:

**Prof. Dr. Edward
Bormashenko**

Chemical Engineering
Department, Engineering
Sciences Faculty, Ariel University,
Ariel 407000, Israel

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Message from the Guest Editor

The Landauer principle, establishing the energy equivalent of information, has remained as a focus of investigations in the last decade. Although non-equilibrium and quantum extensions of the Landauer principle have been reported, the exact meaning and formulation of the principle remain debatable, and both aspects have been the subject of intense discussion. In its strictest, tightest, and simplest meaning, the Landauer principle states that the erasure of one bit of information requires a minimum energy cost equal to $kT \ln 2$, where T is the temperature of a thermal reservoir used in the process and k is Boltzmann's constant. The Landauer principle was also extended to the transmission of information.

This Special Issue aims to present different approaches to the implementation of the Landauer principle in physics, chemistry and biology. Submissions addressing engineering applications of the Landauer principle are especially welcome. Review papers are encouraged.

- Landauer principle
- entropy
- information
- the second law of thermodynamics





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Editor-in-Chief

Prof. Dr. Kevin H. Knuth

Department of Physics, University
at Albany, 1400 Washington
Avenue, Albany, NY 12222, USA

Message from the Editor-in-Chief

The concept of entropy is traditionally a quantity in physics that has to do with temperature. However, it is now clear that entropy is deeply related to information theory and the process of inference. As such, entropic techniques have found broad application in the sciences.

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Entropy Editorial Office
MDPI, Grosspeteranlage 5
4052 Basel, Switzerland

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