



Diamond Formation and Decarbonation under Lithospheric Mantle Pressures and Temperatures

Guest Editor:

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Message from the Guest Editor

Dear Colleagues,

Studies on the stability of natural carbonates and the features of CO₂ fluid generation during mantle-crust interaction are critical for the reconstruction of the processes of the global carbon cycle, including mantle metasomatism, natural diamond formation, as well as formation evolution of carbonated eclogites and peridotites. The key factors that determine the stability of carbonates in the mantle are pressure, temperature, oxygen fugacity, and environmental composition. Their variations can lead to phase transitions and changes in the structure of carbonates, initiate processes of partial melting, decomposition or various reactions involving carbonates. The latter include diamond-forming redox reactions between carbonates and reduced phases (metallic iron, carbides, sulfides, reduced fluids and melts) and decarbonation reactions that occur when carbonates interact with silicates and/or oxides and lead to the formation of CO₂ fluid and the crystallization of newly formed silicates. Decarbonation is one of the most common fluid-generating processes occurring during the interaction of the subducting slab with mantle rocks.





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Message from the Editor-in-Chief

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